

FUNDAMENTAL PHYSICAL CONSTANTS

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These tables give the 2002 self-consistent set of values of the basic constants and conversion factors of physics and chemistry recommended by the Committee on Data for Science and Technology (CODATA) for international use. The 2002 set replaces the previously recommended 1998 CODATA set. The 2002 adjustment takes into account the data considered in the 1998 adjustment as well as the data that became available between 31 December 1998, the closing date of that adjustment, and 31 December 2002, the closing date of the new adjustment.

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Reference

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TABLE I. An Abbreviated List of the CODATA Recommended Values of the Fundamental Constants of Physics and Chemistry Based on the 2002 Adjustment

Quantity	Symbol	Numerical value	Unit	Relative std. uncert. u_r
speed of light in vacuum	c, c_0	299 792 458	m s^{-1}	(exact)
magnetic constant	μ_0	$4\pi \times 10^{-7}$ $= 12.566 370 614... \times 10^{-7}$	N A^{-2}	(exact)
electric constant $1/\mu_0 c^2$	ϵ_0	$8.854 187 817... \times 10^{-12}$	F m^{-1}	(exact)
Newtonian constant of gravitation	G	$6.6742(10) \times 10^{-11}$	$\text{m}^3 \text{kg}^{-1} \text{s}^{-2}$	1.5×10^{-4}
Planck constant	h	$6.626 0693(11) \times 10^{-34}$	J s	1.7×10^{-7}
$h/2\pi$	\hbar	$1.054 571 68(18) \times 10^{-34}$	J s	1.7×10^{-7}
elementary charge	e	$1.602 176 53(14) \times 10^{-19}$	C	8.5×10^{-8}
magnetic flux quantum $h/2e$	Φ_0	$2.067 833 72(18) \times 10^{-15}$	Wb	8.5×10^{-8}
conductance quantum $2e^2/h$	G_0	$7.748 091 733(26) \times 10^{-5}$	S	3.3×10^{-9}
electron mass	m_e	$9.109 3826(16) \times 10^{-31}$	kg	1.7×10^{-7}
proton mass	m_p	$1.672 621 71(29) \times 10^{-27}$	kg	1.7×10^{-7}
proton-electron mass ratio	m_p/m_e	1836.152 672 61(85)		4.6×10^{-10}
fine-structure constant $e^2/4\pi\epsilon_0\hbar c$	α	$7.297 352 568(24) \times 10^{-3}$		3.3×10^{-9}
inverse fine-structure constant	α^{-1}	137.035 999 11(46)		3.3×10^{-9}
Rydberg constant $\alpha^2 m_e c/2h$	R_∞	10 973 731.568 525(73)	m^{-1}	6.6×10^{-12}
Avogadro constant	N_A, L	$6.022 1415(10) \times 10^{23}$	mol^{-1}	1.7×10^{-7}
Faraday constant $N_A e$	F	96 485.3383(83)	C mol^{-1}	8.6×10^{-8}
molar gas constant	R	8.314 472(15)	$\text{J mol}^{-1} \text{K}^{-1}$	1.7×10^{-6}
Boltzmann constant R/N_A	k	$1.380 6505(24) \times 10^{-23}$	J K^{-1}	1.8×10^{-6}
Stefan-Boltzmann constant $(\pi^2/60)k^4/\hbar^3 c^2$	σ	$5.670 400(40) \times 10^{-8}$	$\text{W m}^{-2} \text{K}^{-4}$	7.0×10^{-6}
<i>Non-SI units accepted for use with the SI</i>				
electron volt: (e/C) J	eV	$1.602 176 53(14) \times 10^{-19}$	J	8.5×10^{-8}
(unified) atomic mass unit	u	$1.660 538 86(28) \times 10^{-27}$	kg	1.7×10^{-7}
$1 \text{ u} = m_u = (1/12)m(^{12}\text{C}) = 10^{-3} \text{ kg mol}^{-1}/N_A$				

TABLE II. The CODATA Recommended Values of the Fundamental Constants of Physics and Chemistry Based on the 2002 Adjustment

Quantity	Symbol	Numerical value	Unit	Relative std. uncert. u_r
Universal				
speed of light in vacuum	c, c_0	299 792 458	m s^{-1}	(exact)
magnetic constant	μ_0	$4\pi \times 10^{-7}$ $= 12.566 370 614\dots \times 10^{-7}$	N A^{-2}	(exact)
electric constant $1/\mu_0 c^2$	ϵ_0	$8.854 187 817\dots \times 10^{-12}$	F m^{-1}	(exact)
characteristic impedance of vacuum $\sqrt{\mu_0/\epsilon_0} = \mu_0 c$	Z_0	376.730 313 461...	Ω	(exact)
Newtonian constant of gravitation	G	$6.6742(10) \times 10^{-11}$	$\text{m}^3 \text{kg}^{-1} \text{s}^{-2}$	1.5×10^{-4}
Planck constant	G/hc	$6.7087(10) \times 10^{-39}$	$(\text{GeV}/c^2)^{-2}$	1.5×10^{-4}
in eV s	h	$6.626 0693(11) \times 10^{-34}$	J s	1.7×10^{-7}
$h/2\pi$	\hbar	$4.135 667 43(35) \times 10^{-15}$	eV s	8.5×10^{-8}
in eV s	\hbar	$1.054 571 68(18) \times 10^{-34}$	J s	1.7×10^{-7}
$\hbar c$ in MeV fm		$6.582 119 15(56) \times 10^{-16}$	eV s	8.5×10^{-8}
Planck mass $(\hbar c/G)^{1/2}$		197.326 968(17)	MeV fm	8.5×10^{-8}
Planck temperature $(\hbar c^5/G)^{1/2}/k$	m_p	$2.176 45(16) \times 10^{-8}$	kg	7.5×10^{-5}
Planck length $\hbar/m_p c = (\hbar G/c^3)^{1/2}$	T_p	$1.416 79(11) \times 10^{32}$	K	7.5×10^{-5}
Planck time $\hbar/c = (\hbar G/c^5)^{1/2}$	l_p	$1.616 24(12) \times 10^{-35}$	m	7.5×10^{-5}
	t_p	$5.391 21(40) \times 10^{-44}$	s	7.5×10^{-5}
Electromagnetic				
elementary charge	e	$1.602 176 53(14) \times 10^{-19}$	C	8.5×10^{-8}
magnetic flux quantum $h/2e$	e/h	$2.417 989 40(21) \times 10^{14}$	A J^{-1}	8.5×10^{-8}
conductance quantum $2e^2/h$	Φ_0	$2.067 833 72(18) \times 10^{-15}$	Wb	8.5×10^{-8}
inverse of conductance quantum	G_0	$7.748 091 733(26) \times 10^{-5}$	S	3.3×10^{-9}
Josephson constant ¹ $2e/h$	G_0^{-1}	12 906.403 725(43)	Ω	3.3×10^{-9}
von Klitzing constant ² $h/e^2 = \mu_0 c/2\alpha$	K_J	$483 597.879(41) \times 10^9$	Hz V^{-1}	8.5×10^{-8}
Bohr magneton $eh/2m_e$	R_K	25 812.807 449(86)	Ω	3.3×10^{-9}
in eV T ⁻¹	μ_B	$927.400 949(80) \times 10^{-26}$	J T^{-1}	8.6×10^{-8}
		$5.788 381 804(39) \times 10^{-5}$	eV T ⁻¹	6.7×10^{-9}
nuclear magneton $eh/2m_p$	μ_B/h	$13.996 2458(12) \times 10^9$	Hz T^{-1}	8.6×10^{-8}
in eV T ⁻¹	μ_B/hc	46.686 4507(40)	$\text{m}^{-1} \text{T}^{-1}$	8.6×10^{-8}
	μ_B/k	0.671 7131(12)	K T^{-1}	1.8×10^{-6}
	μ_N	$5.050 783 43(43) \times 10^{-27}$	J T^{-1}	8.6×10^{-8}
		$3.152 451 259(21) \times 10^{-8}$	eV T ⁻¹	6.7×10^{-9}
	μ_N/h	7.622 593 71(65)	MHz T^{-1}	8.6×10^{-8}
	μ_N/hc	$2.542 623 58(22) \times 10^{-2}$	$\text{m}^{-1} \text{T}^{-1}$	8.6×10^{-8}
	μ_N/k	$3.658 2637(64) \times 10^{-4}$	K T^{-1}	1.8×10^{-6}
Atomic and Nuclear				
General				
Fine-structure constant $e^2/4\pi\epsilon_0\hbar c$	α	$7.297 352 568(24) \times 10^{-3}$		3.3×10^{-9}
inverse fine-structure constant	α^{-1}	137.035 999 11(46)		3.3×10^{-9}
Rydberg constant $\alpha^2 m_e c/2\hbar$	R_∞	10 973 731.568 525(73)	m^{-1}	6.6×10^{-12}
	$R_\infty c$	$3.289 841 960 360(22) \times 10^{15}$	Hz	6.6×10^{-12}
	$R_\infty \hbar c$	$2.179 872 09(37) \times 10^{-18}$	J	1.7×10^{-7}
$R_\infty \hbar c$ in eV		13.605 6923(12)	eV	8.5×10^{-8}
Bohr radius $\alpha/4\pi R_\infty = 4\pi\epsilon_0\hbar^2/m_e e^2$	a_0	$0.529 177 2108(18) \times 10^{-10}$	m	3.3×10^{-9}
Hartree energy $e^2/4\pi\epsilon_0 a_0 = 2R_\infty \hbar c = \alpha^2 m_e c^2$	E_h	$4.359 744 17(75) \times 10^{-18}$	J	1.7×10^{-7}
in eV		27.211 3845(23)	eV	8.5×10^{-8}
quantum of circulation	$h/2m_e$	$3.636 947 550(24) \times 10^{-4}$	$\text{m}^2 \text{s}^{-1}$	6.7×10^{-9}
	h/m_e	$7.273 895 101(48) \times 10^{-4}$	$\text{m}^2 \text{s}^{-1}$	6.7×10^{-9}
Electroweak				
Fermi coupling constant ³	$G_F/(\hbar c)^3$	$1.166 39(1) \times 10^{-5}$	GeV^{-2}	8.6×10^{-6}
weak mixing angle ⁴ θ_w (on-shell scheme) $\sin^2 \theta_w = s_w^2 \equiv 1 - (m_w/m_z)^2$	$\sin^2 \theta_w$	0.222 15(76)		3.4×10^{-3}
Electron, e^-				
electron mass	m_e	$9.109 3826(16) \times 10^{-31}$	kg	1.7×10^{-7}
in u, $m_e = A_1(e)$ u (electron relative atomic mass times u)		$5.485 799 0945(24) \times 10^{-4}$	u	4.4×10^{-10}

Quantity	Symbol	Numerical value	Unit	Relative std. uncert. u_r
energy equivalent in MeV	$m_e c^2$	8.187 1047(14)×10 ⁻¹⁴	J	1.7×10 ⁻⁷
electron-muon mass ratio	m_e/m_μ	0.510 998 918(44)	MeV	8.6×10 ⁻⁸
electron-tau mass ratio	m_e/m_τ	4.836 331 67(13)×10 ⁻³		2.6×10 ⁻⁸
electron-proton mass ratio	m_e/m_p	2.875 64(47)×10 ⁻⁴		1.6×10 ⁻⁴
electron-neutron mass ratio	m_e/m_n	5.446 170 2173(25)×10 ⁻⁴		4.6×10 ⁻¹⁰
electron-deuteron mass ratio	m_e/m_d	5.438 673 4481(38)×10 ⁻⁴		7.0×10 ⁻¹⁰
electron to alpha particle mass ratio	m_e/m_α	2.724 437 1095(13)×10 ⁻⁴		4.8×10 ⁻¹⁰
electron charge to mass quotient	$-e/m_e$	1.370 933 555 75(61)×10 ⁻⁴		4.4×10 ⁻¹⁰
electron molar mass $N_A m_e$	$M(e), M_e$	-1.758 820 12(15)×10 ¹¹	C kg ⁻¹	8.6×10 ⁻⁸
Compton wavelength $h/m_e c$	λ_C	5.485 799 0945(24)×10 ⁻⁷	kg mol ⁻¹	4.4×10 ⁻¹⁰
$\lambda_C/2\pi = \alpha a_0 = \alpha^2/4\pi R_\infty$	$\tilde{\lambda}_C$	2.426 310 238(16)×10 ⁻¹²	m	6.7×10 ⁻⁹
classical electron radius $\alpha^2 a_0$	r_e	386.159 2678(26)×10 ⁻¹⁵	m	6.7×10 ⁻⁹
Thomson cross section $(8\pi/3)r_e^2$	σ_e	2.817 940 325(28)×10 ⁻¹⁵	m	1.0×10 ⁻⁸
electron magnetic moment to Bohr magneton ratio	μ_e	0.665 245 873(13)×10 ⁻²⁸	m ²	2.0×10 ⁻⁸
to nuclear magneton ratio	μ_e/μ_B	-928.476 412(80)×10 ⁻²⁶	J T ⁻¹	8.6×10 ⁻⁸
electron magnetic moment anomaly $ \mu_e /\mu_B - 1$	μ_e/μ_N	-1.001 159 652 1859(38)		3.8×10 ⁻¹²
electron g -factor $-2(1 + a_e)$	a_e	-1838.281 971 07(85)		4.6×10 ⁻¹⁰
electron-muon magnetic moment ratio	g_e	1.159 652 1859(38)×10 ⁻³		3.2×10 ⁻⁹
electron-proton magnetic moment ratio	μ_e/μ_μ	-2.002 319 304 3718(75)		3.8×10 ⁻¹²
electron to shielded proton magnetic moment ratio (H ₂ O, sphere, 25°C)	μ_e/μ'_p	206.766 9894(54)		2.6×10 ⁻⁸
electron-neutron magnetic moment ratio	μ_e/μ_p	-658.210 6862(66)		1.0×10 ⁻⁸
electron-deuteron magnetic moment ratio	μ_e/μ'_n	-658.227 5956(71)		1.1×10 ⁻⁸
electron to shielded helion ⁵ magnetic moment ratio (gas, sphere, 25°C)	μ_e/μ'_h	960.920 50(23)		2.4×10 ⁻⁷
electron gyromagnetic ratio $2 \mu_e /\hbar$	γ_e	-2 143.923 493(23)		1.1×10 ⁻⁸
	$\gamma_e/2\pi$	μ_e/μ'_h	864.058 255(10)	1.2×10 ⁻⁸
		γ_e	1.760 859 74(15)×10 ¹¹	s ⁻¹ T ⁻¹
		$\gamma_e/2\pi$	28 024.9532(24)	MHz T ⁻¹
Muon, μ^-				
muon mass	m_μ	1.883 531 40(33)×10 ⁻²⁸	kg	1.7×10 ⁻⁷
in u, $m_\mu = A_r(\mu)$ u (muon relative atomic mass times u)		0.113 428 9264(30)	u	2.6×10 ⁻⁸
energy equivalent	$m_\mu c^2$	1.692 833 60(29)×10 ⁻¹¹	J	1.7×10 ⁻⁷
in MeV		105.658 3692(94)	MeV	8.9×10 ⁻⁸
muon-electron mass ratio	m_μ/m_e	206.768 2838(54)		2.6×10 ⁻⁸
muon-tau mass ratio	m_μ/m_τ	5.945 92(97)×10 ⁻²		1.6×10 ⁻⁴
muon-proton mass ratio	m_μ/m_p	0.112 609 5269(29)		2.6×10 ⁻⁸
muon-neutron mass ratio	m_μ/m_n	0.112 454 5175(29)		2.6×10 ⁻⁸
muon molar mass $N_A m_\mu$	$M(\mu), M_\mu$	0.113 428 9264(30)×10 ⁻³	kg mol ⁻¹	2.6×10 ⁻⁸
muon Compton wavelength $h/m_\mu c$	$\lambda_{C,\mu}$	11.734 441 05(30)×10 ⁻¹⁵	m	2.5×10 ⁻⁸
$\lambda_{C,\mu}/2\pi$	$\tilde{\lambda}_{C,\mu}$	1.867 594 298(47)×10 ⁻¹⁵	m	2.5×10 ⁻⁸
muon magnetic moment	μ_μ	-4.490 447 99(40)×10 ⁻²⁶	J T ⁻¹	8.9×10 ⁻⁸
to Bohr magneton ratio	μ_μ/μ_B	-4.841 970 45(13)×10 ⁻³		2.6×10 ⁻⁸
to nuclear magneton ratio	μ_μ/μ_N	-8.890 596 98(23)		2.6×10 ⁻⁸
muon magnetic moment anomaly $ \mu_\mu /(e\hbar/2m_\mu) - 1$	α_μ	1.165 919 81(62)×10 ⁻³		5.3×10 ⁻⁷
muon g -factor $-2(1 + \alpha_\mu)$	g_μ	-2.002 331 8396(12)		6.2×10 ⁻¹⁰
muon-proton magnetic moment ratio	μ_μ/μ_p	-3.183 345 118(89)		2.8×10 ⁻⁸
Tau, τ				
tau mass ⁶	m_τ	3.167 77(52)×10 ⁻²⁷	kg	1.6×10 ⁻⁴
in u, $m_\tau = A_r(\tau)$ u (tau relative atomic mass times u)		1.907 68(31)	u	1.6×10 ⁻⁴
energy equivalent	$m_\tau c^2$	2.847 05(46)×10 ⁻¹⁰	J	1.6×10 ⁻⁴
in MeV		1776.99(29)	MeV	1.6×10 ⁻⁴
tau-electron mass ratio	m_τ/m_e	3477.48(57)		1.6×10 ⁻⁴
tau-muon mass ratio	m_τ/m_μ	16.8183(27)		1.6×10 ⁻⁴
tau-proton mass ratio	m_τ/m_p	1.893 90(31)		1.6×10 ⁻⁴
tau-neutron mass ratio	m_τ/m_n	1.891 29(31)		1.6×10 ⁻⁴
tau molar mass $N_A m_\tau$	$M(\tau), M_\tau$	1.907 68(31)×10 ⁻³	kg mol ⁻¹	1.6×10 ⁻⁴
tau Compton wavelength $h/m_\tau c$	$\lambda_{C,\tau}$	0.697 72(11)×10 ⁻¹⁵	m	1.6×10 ⁻⁴
$\lambda_{C,\tau}/2\pi$	$\tilde{\lambda}_{C,\tau}$	0.111 046(18)×10 ⁻¹⁵	m	1.6×10 ⁻⁴

Quantity	Symbol	Numerical value	Unit	Relative std. uncert. u_r
Proton, p				
proton mass	m_p	$1.672\,621\,71(29)\times 10^{-27}$	kg	1.7×10^{-7}
in u, $m_p = A_r(p)$ u (proton relative atomic mass times u)		1.007 276 466 88(13)	u	1.3×10^{-10}
energy equivalent	$m_p c^2$	$1.503\,277\,43(26)\times 10^{-10}$	J	1.7×10^{-7}
in MeV		938.272 029(80)	MeV	8.6×10^{-8}
proton-electron mass ratio	m_p/m_e	1836.152 672 61(85)		4.6×10^{-10}
proton-muon mass ratio	m_p/m_μ	8.880 243 33(23)		2.6×10^{-8}
proton-tau mass ratio	m_p/m_τ	0.528 012(86)		1.6×10^{-4}
proton-neutron mass ratio	m_p/m_n	0.998 623 478 72(58)		5.8×10^{-10}
proton charge to mass quotient	e/m_p	$9.578\,833\,76(82)\times 10^7$	C kg ⁻¹	8.6×10^{-8}
proton molar mass $N_A m_p$	$M(p), M_p$	$1.007\,276\,466\,88(13)\times 10^{-3}$	kg mol ⁻¹	1.3×10^{-10}
proton Compton wavelength $h/m_p c$	$\lambda_{C,p}$	$1.321\,409\,8555(88)\times 10^{-15}$	m	6.7×10^{-9}
$\lambda_{C,p}/2\pi$	$\tilde{\lambda}_{C,p}$	$0.210\,308\,9104(14)\times 10^{-15}$	m	6.7×10^{-9}
proton rms charge radius	R_p	$0.8750(68)\times 10^{-15}$	m	7.8×10^{-3}
proton magnetic moment	μ_p	$1.410\,606\,71(12)\times 10^{-26}$	J T ⁻¹	8.7×10^{-8}
to Bohr magneton ratio	μ_p/μ_B	$1.521\,032\,206(15)\times 10^{-3}$		1.0×10^{-8}
to nuclear magneton ratio	μ_p/μ_N	2.792 847 351(28)		1.0×10^{-8}
proton g -factor $2\mu_p/\mu_N$	g_p	5.585 694 701(56)		1.0×10^{-8}
proton-neutron magnetic moment ratio	μ_p/μ_n	-1.459 898 05(34)		2.4×10^{-7}
shielded proton magnetic moment (H ₂ O, sphere, 25°C)	μ'_p	$1.410\,570\,47(12)\times 10^{-26}$	J T ⁻¹	8.7×10^{-8}
to Bohr magneton ratio	μ'_p/μ_B	$1.520\,993\,132(16)\times 10^{-3}$		1.1×10^{-8}
to nuclear magneton ratio	μ'_p/μ_N	2.792 775 604(30)		1.1×10^{-8}
proton magnetic shielding correction $1 - \mu'_p/\mu_p$ (H ₂ O, sphere, 25°C)	σ'_p	$25.689(15)\times 10^{-6}$		5.7×10^{-4}
proton gyromagnetic ratio $2\mu_p/\hbar$	γ_p	$2.675\,222\,05(23)\times 10^8$	s ⁻¹ T ⁻¹	8.6×10^{-8}
	$\gamma_p/2\pi$	42.577 4813(37)	MHz T ⁻¹	8.6×10^{-8}
shielded proton gyromagnetic ratio $2\mu'_p/\hbar$ (H ₂ O, sphere, 25°C)	γ'_p	$2.675\,153\,33(23)\times 10^8$	s ⁻¹ T ⁻¹	8.6×10^{-8}
	$\gamma'_p/2\pi$	42.576 3875(37)	MHz T ⁻¹	8.6×10^{-8}
Neutron, n				
neutron mass	m_n	$1.674\,927\,28(29)\times 10^{-27}$	kg	1.7×10^{-7}
in u, $m_n = A_r(n)$ u (neutron relative atomic mass times u)		1.008 664 915 60(55)	u	5.5×10^{-10}
energy equivalent	$m_n c^2$	$1.505\,349\,57(26)\times 10^{-10}$	J	1.7×10^{-7}
in MeV		939.565 360(81)	MeV	8.6×10^{-8}
neutron-electron mass ratio	m_n/m_e	1838.683 6598(13)		7.0×10^{-10}
neutron-muon mass ratio	m_n/m_μ	8.892 484 02(23)		2.6×10^{-8}
neutron-tau mass ratio	m_n/m_τ	0.528 740(86)		1.6×10^{-4}
neutron-proton mass ratio	m_n/m_p	1.001 378 418 70(58)		5.8×10^{-10}
neutron molar mass $N_A m_n$	$M(n), M_n$	$1.008\,664\,915\,60(55)\times 10^{-3}$	kg mol ⁻¹	5.5×10^{-10}
neutron Compton wavelength $h/m_n c$	$\lambda_{C,n}$	$1.319\,590\,9067(88)\times 10^{-15}$	m	6.7×10^{-9}
$\lambda_{C,n}/2\pi$	$\tilde{\lambda}_{C,n}$	$0.210\,019\,4157(14)\times 10^{-15}$	m	6.7×10^{-9}
neutron magnetic moment	μ_n	$-0.966\,236\,45(24)\times 10^{-26}$	J T ⁻¹	2.5×10^{-7}
to Bohr magneton ratio	μ_n/μ_B	$-1.041\,875\,63(25)\times 10^{-3}$		2.4×10^{-7}
to nuclear magneton ratio	μ_n/μ_N	-1.913 042 73(45)		2.4×10^{-7}
neutron g -factor $2\mu_n/\mu_N$	g_n	-3.826 085 46(90)		2.4×10^{-7}
neutron-electron magnetic moment ratio	μ_n/μ_e	$1.040\,668\,82(25)\times 10^{-3}$		2.4×10^{-7}
neutron-proton magnetic moment ratio	μ_n/μ_p	-0.684 979 34(16)		2.4×10^{-7}
neutron to shielded proton magnetic moment ratio (H ₂ O, sphere, 25°C)	μ_n/μ'_p	-0.684 996 94(16)		2.4×10^{-7}
neutron gyromagnetic ratio $2 \mu_n /\hbar$	γ_n	$1.832\,471\,83(46)\times 10^8$	s ⁻¹ T ⁻¹	2.5×10^{-7}
	$\gamma_n/2\pi$	29.164 6950(73)	MHz T ⁻¹	2.5×10^{-7}
Deuteron, d				
deuteron mass	m_d	$3.343\,583\,35(57)\times 10^{-27}$	kg	1.7×10^{-7}
in u, $m_d = A_r(d)$ u (deuteron relative atomic mass times u)		2.013 553 212 70(35)	u	1.7×10^{-10}
energy equivalent	$m_d c^2$	$3.005\,062\,85(51)\times 10^{-10}$	J	1.7×10^{-7}
in MeV		1875.612 82(16)	MeV	8.6×10^{-8}
deuteron-electron mass ratio	m_d/m_e	3670.482 9652(18)		4.8×10^{-10}
deuteron-proton mass ratio	m_d/m_p	1.999 007 500 82(41)		2.0×10^{-10}
deuteron molar mass $N_A m_d$	$M(d), M_d$	$2.013\,553\,212\,70(35)\times 10^{-3}$	kg mol ⁻¹	1.7×10^{-10}
deuteron rms charge radius	R_d	$2.1394(28)\times 10^{-15}$	m	1.3×10^{-3}

Quantity	Symbol	Numerical value	Unit	Relative std. uncert. u_r
deuteron magnetic moment	μ_d	$0.433\,073\,482(38)\times 10^{-26}$	J T^{-1}	8.7×10^{-8}
to Bohr magneton ratio	μ_d/μ_B	$0.466\,975\,4567(50)\times 10^{-3}$		1.1×10^{-8}
to nuclear magneton ratio	μ_d/μ_N	$0.857\,438\,2329(92)$		1.1×10^{-8}
deuteron-electron magnetic moment ratio	μ_d/μ_e	$-4.664\,345\,548(50)\times 10^{-4}$		1.1×10^{-8}
deuteron-proton magnetic moment ratio	μ_d/μ_p	$0.307\,012\,2084(45)$		1.5×10^{-8}
deuteron-neutron magnetic moment ratio	μ_d/μ_n	$-0.448\,206\,52(11)$		2.4×10^{-7}
Helion, h				
helion mass ⁵	m_h	$5.006\,412\,14(86)\times 10^{-27}$	kg	1.7×10^{-7}
in u, $m_h = A_r(\text{h})$ u (helion relative atomic mass times u)		$3.014\,932\,2434(58)$	u	1.9×10^{-9}
energy equivalent	$m_h c^2$	$4.499\,538\,84(77)\times 10^{-10}$	J	1.7×10^{-7}
in MeV		$2808.391\,42(24)$	MeV	8.6×10^{-8}
helion-electron mass ratio	m_h/m_e	$5495.885\,269(11)$		2.0×10^{-9}
helion-proton mass ratio	m_h/m_p	$2.993\,152\,6671(58)$		1.9×10^{-9}
helion molar mass $N_A m_h$	$M(\text{h}), M_h$	$3.014\,932\,2434(58)\times 10^{-3}$	kg mol ⁻¹	1.9×10^{-9}
shielded helion magnetic moment (gas, sphere, 25°C)	μ'_h	$-1.074\,553\,024(93)\times 10^{-26}$	J T^{-1}	8.7×10^{-8}
to Bohr magneton ratio	μ'_h/μ_B	$-1.158\,671\,474(14)\times 10^{-3}$		1.2×10^{-8}
to nuclear magneton ratio	μ'_h/μ_N	$-2.127\,497\,723(25)$		1.2×10^{-8}
shielded helion to proton magnetic moment ratio (gas, sphere, 25°C)	μ'_h/μ_p	$-0.761\,766\,562(12)$		1.5×10^{-8}
shielded helion to shielded proton magnetic moment ratio (gas/H ₂ O, spheres, 25°C)	μ'_h/μ'_p	$-0.761\,786\,1313(33)$		4.3×10^{-9}
shielded helion gyromagnetic ratio $2 \mu'_h /h$ (gas, sphere, 25°C)	γ'_h	$2.037\,894\,70(18)\times 10^8$	s ⁻¹ T ⁻¹	8.7×10^{-8}
	$\gamma'_h/2\pi$	$32.434\,1015(28)$	MHz T ⁻¹	8.7×10^{-8}
Alpha particle, α				
alpha particle mass	m_α	$6.644\,6565(11)\times 10^{-27}$	kg	1.7×10^{-7}
in u, $m_\alpha = A_r(\alpha)$ u (alpha particle relative atomic mass times u)		$4.001\,506\,179\,149(56)$	u	1.4×10^{-11}
energy equivalent	$m_\alpha c^2$	$5.971\,9194(10)\times 10^{-10}$	J	1.7×10^{-7}
in MeV		$3727.379\,17(32)$	MeV	8.6×10^{-8}
alpha particle to electron mass ratio	m_α/m_e	$7294.299\,5363(32)$		4.4×10^{-10}
alpha particle to proton mass ratio	m_α/m_p	$3.972\,599\,689\,07(52)$		1.3×10^{-10}
alpha particle molar mass $N_A m_\alpha$	$M(\alpha), M_\alpha$	$4.001\,506\,179\,149(56)\times 10^{-3}$	kg mol ⁻¹	1.4×10^{-11}
Physico-Chemical				
Avogadro constant	N_A, L	$6.022\,1415(10)\times 10^{23}$	mol ⁻¹	1.7×10^{-7}
atomic mass constant $m_u = (1/12)m(^{12}\text{C}) = 1\text{ u} = 10^{-3}\text{ kg mol}^{-1}/N_A$	m_u	$1.660\,538\,86(28)\times 10^{-27}$	kg	1.7×10^{-7}
energy equivalent	$m_u c^2$	$1.492\,417\,90(26)\times 10^{-10}$	J	1.7×10^{-7}
in MeV		$931.494\,043(80)$	MeV	8.6×10^{-8}
Faraday constant ⁷ $N_A e$	F	$96\,485.3383(83)$	C mol ⁻¹	8.6×10^{-8}
molar Planck constant	$N_A h$	$3.990\,312\,716(27)\times 10^{-10}$	J s mol ⁻¹	6.7×10^{-9}
	$N_A hc$	$0.119\,626\,565\,72(80)$	J m mol ⁻¹	6.7×10^{-9}
molar gas constant	R	$8.314\,472(15)$	J mol ⁻¹ K ⁻¹	1.7×10^{-6}
Boltzmann constant R/N_A	k	$1.380\,6505(24)\times 10^{-23}$	J K ⁻¹	1.8×10^{-6}
in eV K ⁻¹		$8.617\,343(15)\times 10^{-5}$	eV K ⁻¹	1.8×10^{-6}
	k/h	$2.083\,6644(36)\times 10^{10}$	Hz K ⁻¹	1.7×10^{-6}
	k/hc	$69.503\,56(12)$	m ⁻¹ K ⁻¹	1.7×10^{-6}
molar volume of ideal gas RT/p	V_m	$22.413\,996(39)\times 10^{-3}$	m ³ mol ⁻¹	1.7×10^{-6}
$T = 273.15\text{ K}, p = 101.325\text{ kPa}$	n_0	$2.686\,7773(47)\times 10^{25}$	m ⁻³	1.8×10^{-6}
Loschmidt constant N_A/V_m	V_m	$22.710\,981(40)\times 10^{-3}$	m ³ mol ⁻¹	1.7×10^{-6}
$T = 273.15\text{ K}, p = 100\text{ kPa}$				
Sackur-Tetrode constant (absolute entropy constant) ⁸	S_0/R	$-1.151\,7047(44)$		3.8×10^{-6}
$5/2 + \ln[(2\pi m_u k T_1/h^2)^{3/2} k T_1/p_0]$		$-1.164\,8677(44)$		3.8×10^{-6}
$T_1 = 1\text{ K}, p_0 = 100\text{ kPa}$				
$T_1 = 1\text{ K}, p_0 = 101.325\text{ kPa}$				
Stefan-Boltzmann constant $(\pi^2/60)k^4/h^3c^2$	σ	$5.670\,400(40)\times 10^{-8}$	W m ⁻² K ⁻⁴	7.0×10^{-6}
first radiation constant $2\pi\hbar c^2$	c_1	$3.741\,771\,38(64)\times 10^{-16}$	W m ²	1.7×10^{-7}
first radiation constant for spectral radiance $2hc^2$	c_{1L}	$1.191\,042\,82(20)\times 10^{-16}$	W m ² sr ⁻¹	1.7×10^{-7}
second radiation constant $\hbar c/k$	c_2	$1.438\,7752(25)\times 10^{-2}$	m K	1.7×10^{-6}
Wien displacement law constant $b = \lambda_{\text{max}} T = c_2/4.965\,114\,231\dots$	b	$2.897\,7685(51)\times 10^{-3}$	m K	1.7×10^{-6}

- ¹ See the "Adopted values" table for the conventional value adopted internationally for realizing representations of the volt using the Josephson effect.
- ² See the "Adopted values" table for the conventional value adopted internationally for realizing representations of the ohm using the quantum Hall effect.
- ³ Value recommended by the Particle Data Group (Hagiwara *et al.*, 2002).
- ⁴ Based on the ratio of the masses of the W and Z bosons m_W/m_Z recommended by the Particle Data Group (Hagiwara *et al.*, 2002). The value for $\sin^2\theta_W$ they recommend, which is based on a particular variant of the modified minimal subtraction (\overline{MS}) scheme, is $\sin^2\hat{\theta}_W(M_Z) = 0.231\,24(24)$.
- ⁵ The helion, symbol h, is the nucleus of the ^3He atom.
- ⁶ This and all other values involving m_e are based on the value of $m_e c^2$ in MeV recommended by the Particle Data Group (Hagiwara *et al.*, 2002), but with a standard uncertainty of 0.29 MeV rather than the quoted uncertainty of -0.26 MeV, $+0.29$ MeV.
- ⁷ The numerical value of F to be used in coulometric chemical measurements is $96\,485.336(16)$ [1.7×10^{-7}] when the relevant current is measured in terms of representations of the volt and ohm based on the Josephson and quantum Hall effects and the internationally adopted conventional values of the Josephson and von Klitzing constants K_{J-90} and R_{K-90} given in the "Adopted values" table.
- ⁸ The entropy of an ideal monoatomic gas of relative atomic mass A_r is given by $S = S_0 + (3/2)R \ln A_r - R \ln(p/p_0) + (5/2)R \ln(T/K)$.

TABLE III. Internationally Adopted Values of Various Quantities

Quantity	Symbol	Numerical value	Unit	Relative std. uncert. u_r
molar mass of ^{12}C	$M(^{12}\text{C})$	12×10^{-3}	kg mol $^{-1}$	(exact)
molar mass constant $M(^{12}\text{C})/12$	M_u	1×10^{-3}	kg mol $^{-1}$	(exact)
conventional value of Josephson constant	K_{J-90}	483 597.9	GHz V $^{-1}$	(exact)
conventional value of von Klitzing constant	R_{K-90}	25 812.807	Ω	(exact)
standard atmosphere		101 325	Pa	(exact)
standard acceleration of gravity	g_n	9.806 65	m s $^{-2}$	(exact)

TABLE IV. The Values of Some Energy Equivalents Derived From the Relations $E = mc^2 = hc/\lambda = hv = kT$, and Based on the 2002 CODATA Adjustment of the Values of the Constants

1 eV = (e/C) J, 1 u = $m_u = (1/12)m(^{12}\text{C}) = 10^{-3}$ kg mol $^{-1}/N_A$, and $E_h = 2R_\infty hc = \alpha^2 m_e c^2$ is the Hartree Energy (hartree)

		Relevant unit			
		J	kg	m $^{-1}$	Hz
1 J	(1 J) = 1 J		(1 J)/ $c^2 = 1.112\,650\,056 \times 10^{-17}$ kg	(1 J)/ $hc = 5.034\,117\,20(86) \times 10^{24}$ m $^{-1}$	(1 J)/ $h = 1.509\,190\,37(26) \times 10^{33}$ Hz
1 kg	(1 kg) $c^2 = 8.987\,551\,787 \times 10^{16}$ J		(1 kg) = 1kg	(1 kg) $c/h = 4.524\,438\,91(77) \times 10^{41}$ m $^{-1}$	(1 kg) $c^2/h = 1.356\,392\,66(23) \times 10^{30}$ Hz
1 m $^{-1}$	(1 m $^{-1}$) $hc = 1.986\,445\,61(34) \times 10^{-25}$ J		(1 m $^{-1}$) $h/c = 2.210\,218\,81(38) \times 10^{-42}$ kg	(1 m $^{-1}$) = 1 m $^{-1}$	(1 m $^{-1}$) $c = 299\,792\,458$ Hz
1 Hz	(1 Hz) $h = 6.626\,0693(11) \times 10^{-34}$ J		(1 Hz) $h/c^2 = 7.372\,4964(13) \times 10^{-51}$ kg	(1 Hz)/ $c = 3.335\,640\,952 \times 10^{-9}$ m $^{-1}$	(1 Hz) = 1 Hz
1 K	(1 K) $k = 1.380\,6505(24) \times 10^{-23}$ J		(1 K) $k/c^2 = 1.536\,1808(27) \times 10^{-40}$ kg	(1 K) $k/hc = 69.503\,56(12)$ m $^{-1}$	(1 K) $k/h = 2.083\,6644(36) \times 10^{10}$ Hz
1 eV	(1 eV) = $1.602\,176\,53(14) \times 10^{-19}$ J		(1 eV)/ $c^2 = 1.782\,661\,81(15) \times 10^{-36}$ kg	(1 eV)/ $hc = 8.065\,544\,45(69) \times 10^5$ m $^{-1}$	(1 eV)/ $h = 2.417\,989\,40(21) \times 10^{14}$ Hz
1 u	(1 u) $c^2 = 1.492\,417\,90(26) \times 10^{-10}$ J		(1 u) = $1.660\,538\,86(28) \times 10^{-27}$ kg	(1 u) $c/h = 7.513\,006\,608(50) \times 10^{14}$ m $^{-1}$	(1 u) $c^2/h = 2.252\,342\,718(15) \times 10^{23}$ Hz
1 E_h	(1 E_h) = $4.359\,744\,17(75) \times 10^{-18}$ J		(1 E_h)/ $c^2 = 4.850\,869\,60(83) \times 10^{-35}$ kg	(1 E_h)/ $hc = 2.194\,746\,313\,705(15) \times 10^7$ m $^{-1}$	(1 E_h)/ $h = 6.579\,683\,920\,721(44) \times 10^{15}$ Hz

TABLE V. The Values of Some Energy Equivalents Derived From the Relations $E = mc^2 = hc/\lambda = hv = kT$, and Based on the 2002 CODATA Adjustment of the Values of the Constants

1 eV = (e/C) J, 1 u = $m_u = (1/12)m(^{12}\text{C}) = 10^{-3}$ kg mol $^{-1}/N_A$, and $E_h = 2R_\infty hc = \alpha^2 m_e c^2$ is the Hartree Energy (hartree)

		Relevant unit			
		K	eV	u	E_h
1 J	(1 J)/ $k = 7.242\,963(13) \times 10^{22}$ K		(1 J) = $6.241\,509\,47(53) \times 10^{18}$ eV	(1 J)/ $c^2 = 6.700\,5361(11) \times 10^9$ u	(1 J) = $2.293\,712\,57(39) \times 10^{17}$ E_h
1 kg	(1 kg) $c^2/k = 6.509\,650(11) \times 10^{39}$ K		(1 kg) $c^2 = 5.609\,588\,96(48) \times 10^{35}$ eV	(1 kg) = $6.022\,1415(10) \times 10^{26}$ u	(1 kg) $c^2 = 2.061\,486\,05(35) \times 10^{34}$ E_h
1 m $^{-1}$	(1 m $^{-1}$) $hc/k = 1.438\,7752(25) \times 10^{-2}$ K		(1 m $^{-1}$) $hc = 1.239\,841\,91(11) \times 10^{-6}$ eV	(1 m $^{-1}$) $h/c = 1.331\,025\,0506(89) \times 10^{-15}$ u	(1 m $^{-1}$) $hc = 4.556\,335\,252\,760(30) \times 10^{-8}$ E_h
1 Hz	(1 Hz) $h/k = 4.799\,2374(84) \times 10^{-11}$ K		(1 Hz) $h = 4.135\,667\,43(35) \times 10^{-15}$ eV	(1 Hz) $h/c^2 = 4.439\,821\,667(30) \times 10^{-24}$ u	(1 Hz) $h = 1.519\,829\,846\,006(10) \times 10^{-16}$ E_h
1 K	(1 K) = 1 K		(1 K) $k = 8.617\,343(15) \times 10^{-5}$ eV	(1 K) $k/c^2 = 9.251\,098(16) \times 10^{-14}$ u	(1 K) $k = 3.166\,8153(55) \times 10^{-6}$ E_h
1 eV	(1 eV)/ $k = 1.160\,4505(20) \times 10^4$ K		(1 eV) = 1 eV	(1 eV)/ $c^2 = 1.073\,544\,171(92) \times 10^{-9}$ u	(1 eV) = $3.674\,932\,45(31) \times 10^{-2}$ E_h
1 u	(1 u) $c^2/k = 1.080\,9527(19) \times 10^{13}$ K		(1 u) $c^2 = 931.494\,043(80) \times 10^6$ eV	(1 u) = 1 u	(1 u) $c^2 = 3.423\,177\,686(23) \times 10^7$ E_h
1 E_h	(1 E_h)/ $k = 3.157\,7465(55) \times 10^5$ K		(1 E_h) = $27.211\,3845(23)$ eV	(1 E_h)/ $c^2 = 2.921\,262\,323(19) \times 10^{-8}$ u	(1 E_h) = 1 E_h

FUNDAMENTAL PHYSICAL CONSTANTS — FREQUENTLY USED CONSTANTS

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
speed of light in vacuum	c, c_0	299 792 458	m s^{-1}	(exact)
magnetic constant	μ_0	$4\pi \times 10^{-7}$	N A^{-2}	
		$= 12.566 370 614\dots \times 10^{-7}$	N A^{-2}	(exact)
electric constant $1/\mu_0 c^2$	ϵ_0	$8.854 187 817\dots \times 10^{-12}$	F m^{-1}	(exact)
Newtonian constant of gravitation	G	$6.6742(10) \times 10^{-11}$	$\text{m}^3 \text{kg}^{-1} \text{s}^{-2}$	1.5×10^{-4}
Planck constant	h	$6.626 0693(11) \times 10^{-34}$	J s	1.7×10^{-7}
		\hbar	$1.054 571 68(18) \times 10^{-34}$	J s
elementary charge	e	$1.602 176 53(14) \times 10^{-19}$	C	8.5×10^{-8}
magnetic flux quantum $h/2e$	Φ_0	$2.067 833 72(18) \times 10^{-15}$	Wb	8.5×10^{-8}
conductance quantum $2e^2/h$	G_0	$7.748 091 733(26) \times 10^{-5}$	S	3.3×10^{-9}
electron mass	m_e	$9.109 3826(16) \times 10^{-31}$	kg	1.7×10^{-7}
proton mass	m_p	$1.672 621 71(29) \times 10^{-27}$	kg	1.7×10^{-7}
proton-electron mass ratio	m_p/m_e	1836.152 672 61(85)		4.6×10^{-10}
fine-structure constant $e^2/4\pi\epsilon_0\hbar c$	α	$7.297 352 568(24) \times 10^{-3}$		3.3×10^{-9}
inverse fine-structure constant	α^{-1}	137.035 999 11(46)		3.3×10^{-9}
Rydberg constant $\alpha^2 m_e c/2\hbar$	R_∞	10 973 731.568 525(73)	m^{-1}	6.6×10^{-12}
Avogadro constant	N_A, L	$6.022 1415(10) \times 10^{23}$	mol^{-1}	1.7×10^{-7}
Faraday constant $N_A e$	F	96 485.3383(83)	C mol^{-1}	8.6×10^{-8}
molar gas constant	R	8.314 472(15)	$\text{J mol}^{-1} \text{K}^{-1}$	1.7×10^{-6}
Boltzmann constant R/N_A	k	$1.380 6505(24) \times 10^{-23}$	J K^{-1}	1.8×10^{-6}
Stefan-Boltzmann constant $(\pi^2/60)k^4/\hbar^3 c^2$	σ	$5.670 400(40) \times 10^{-8}$	$\text{W m}^{-2} \text{K}^{-4}$	7.0×10^{-6}
Non-SI units accepted for use with the SI				
electron volt: $(e/C) \text{ J}$	eV	$1.602 176 53(14) \times 10^{-19}$	J	8.5×10^{-8}
(unified) atomic mass unit $1 \text{ u} = m_u = \frac{1}{12} m(^{12}\text{C}) = 10^{-3} \text{ kg mol}^{-1}/N_A$	u	$1.660 538 86(28) \times 10^{-27}$	kg	1.7×10^{-7}