PHYSICAL CONSTANTS OF INORGANIC COMPOUNDS

The compounds in this table were selected on the basis of their laboratory and industrial importance, as well as their value in illustrating trends in the variation of physical properties with position in the periodic table. An effort has been made to include the most frequently encountered inorganic substances; a limited number of organometallics are also covered. Many, if not most, of the compounds that are solids at ambient temperature can exist in more than one crystalline modification. In the absence of other information, the data given here can be assumed to apply to the most stable or common crystalline form. In many cases however, two or more forms are of practical importance, and separate entries will be found in the table.

Compounds are arranged primarily in alphabetical order by the most commonly used name. However, adjustments are made in many instances so as to bring closely related compounds together. For example, hydrides of elements such as boron, silicon, and germanium are grouped together immediately following the entry for the parent element, since they would otherwise be scattered throughout the table. Likewise, the oxoacids of an element are given in one group whenever a strict alphabetical order would separate them (e.g., sulfuric acid and fluorosulfuric acid). The Formula Index following the table provides another means of locating a compound. There is also an index to CAS Registry Numbers.

The following data fields appear in the table:

- **Name**: Systematic name for the substance. The valence state of a metallic element is indicated by a Roman numeral, e.g., copper in the +1 state is written as copper(I) rather than cuprous, iron in the +3 state is iron(III) rather than ferric.
- **Formula**: The simplest descriptive formula is given, but this does not necessarily specify the actual structure of the compound. For example, aluminum chloride is designated as $AlCl_3$, even though a more accurate representation of the structure in the solid phase (and, under some conditions, in the gas phase) is Al_2Cl_6 A few exceptions are made, such as the use of Hg_2^{+2} for the mercury(I) ion.
- **CAS Registry Number**: Chemical Abstracts Service Registry Number. An asterisk* following the CAS RN for a hydrate indicates that the number refers to the anhydrous compound. In most cases the generic CAS RN for the compound is given rather than the number for a specific crystalline form or mineral.
- **Mol. Weight**: Molecular weight (relative molar mass) as calculated with the 2005 IUPAC Recommended Atomic Weights. The number of decimal places corresponds to the number of places in the atomic weight of the least accurately known element (e.g., one place for lead compounds, two places for compounds of selenium, germanium, etc.); a maximum of three places is given. For compounds of radioactive elements for which IUPAC makes no recommendation, the mass number of the isotope with longest half-life is used.
- **Physical Form**: The crystal system is given, when available, for compounds that are solid at room temperature, together with color and other descriptive features. Abbreviations are listed below.
- **mp**: Normal melting point in °C. The notation tp indicates the temperature where solid, liquid, and gas are in equilibrium at a pressure greater than one atmosphere (i.e., the normal melting point does not exist). When available, the triple point pressure is listed.
- **bp**: Normal boiling point in °C (referred to 101.325 kPa or 760 mmHg pressure). The notation sp following the number indicates the temperature where the pressure of the vapor in equilibrium with the solid reaches 101.325 kPa. See Reference 8, p. 23, for further discussion of sublimation points and triple points. A notation "sublimes" without a temperature being given indicates that there is a perceptible sublimation pressure above the solid at ambient temperatures.
- **Density**: Density values for solids and liquids are always in units of grams per cubic centimeter and can be assumed to refer to temperatures near room temperature unless otherwise stated. Values for gases are the calculated ideal gas densities in grams per liter at 25°C and 101.325 kPa; the unit is always specified for a gas value.
- **Aqueous Solubility:** Solubility is expressed as the number of grams of the compound (excluding any water of hydration) that will dissolve in 100 grams of water. The temperature in °C is given as a superscript. Solubility at other temperatures can be found for many compounds in the table "Aqueous Solubility of Inorganic Compounds at Various Temperatures" in Section 8.
- **Qualitative Solubility**: Qualitative information on the solubility in other solvents (and in water, if quantitative data are unavailable) is given here. The abbreviations are:
  - i insoluble
  - sl slightly soluble
  - s soluble
  - vs very soluble
  - reac reacts with the solvent

Data were taken from a wide variety of reliable sources, including monographs, treatises, review articles, evaluated compilations and databases, and in many cases the primary literature. Some of the most useful references for the properties covered here are listed below.

List of Abbreviations

Ac - acetyl
ace - acetone
acid - acid solutions
alk - alkaline solutions
amorp - amorphous
anh - anhydrous
aq - aqueous
blk - black
brn - brown
bz - benzene
chl - chloroform
col - colorless
conc - concentrated
cry - crystals, crystalline
cub - cubic
cyhex - cyclohexane
dec - decomposes
dil - dilute
diox - dioxane
eth - ethyl ether
EtOH - ethanol
exp - explodes, explosive
extrap - extrapolated
flam - flammable
gl - glass, glassy
grn - green
hc - hydrocarbon solvents
hex - hexagonal, hexane

hp - heptane
HT - high temperature
hyd - hydrate
hyg - hygroscopic
i - insoluble in
liq - liquid
LT - low temperature
MeOH - methanol
monocl - monoclinic
octahed - octahedral
oran - orange
orth - orthorhombic
os - organic solvents
peth - petroleum ether
pow - powder
prec - precipitate
pur - purple
py - pyridine
reac - reacts with
refrac - refractory
rhom - rhombohedral
r.t. - room temperature
s - soluble in
silv - silvery
sl - slightly soluble in
soln - solution
sp - sublimation point
stab - stable
subl - sublimes
temp - temperature
tetr - tetragonal
thf - tetrahydrofuran
tol - toluene
tp - triple point
trans - transition, transformation
tricl - triclinic
trig - trigonal
unstab - unstable
viol - violet
visc - viscous
vs - very soluble in
wh - white
xyl - xylene
yel - yellow

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No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1	Actinium	Ac	7440-34-8	227	silv metal; cub	1050	3198	10		
2	Actinium bromide	$AcBr_3$	33689-81-5	467	wh hex cry		800 subl	5.85		s H_2O
3	Actinium chloride	$AcCl_3$	22986-54-5	333	wh hex cry		960 subl	4.81		
4	Actinium fluoride	AcF_3	33689-80-4	284	wh hex cry			7.88		i H_2O
5	Actinium iodide	AcI_3	33689-82-6	608	wh cry					s H_2O
6	Actinium oxide	Ac_2O_3	12002-61-8	502	wh hex cry	1977		9.19		i H_2O
7	Aluminum	Al	7429-90-5	26.982	silv-wh metal; cub cry	660.32	2519	2.70		i H_2O; s acid, alk
8	Aluminum acetate	$Al(C_2H_3O_2)_3$	139-12-8	204.113	wh hyg solid	dec				s H_2O; sl ace
9	Aluminum diacetate	$Al(OH)(C_2H_3O_2)_2$	142-03-0	162.078	wh amorp powder					i H_2O
10	Aluminum ammonium sulfate	$AlNH_4(SO_4)_2$	7784-25-0	237.146	wh powder					sl H_2O; i EtOH
11	Aluminum ammonium sulfate dodecahydrate	$AlNH_4(SO_4)_2 \cdot 12H_2O$	7784-26-1	453.329	col cry or powder	94.5	>280 dec	1.65		s H_2O; i EtOH
12	Aluminum antimonide	AlSb	25152-52-7	148.742	brn cub cry	1065		4.26		
13	Aluminum arsenide	AlAs	22831-42-1	101.903	oran cub cry; hyg	1740		3.76		
14	Aluminum borate	$2Al_2O_3 \cdot B_2O_3$	11121-16-7	273.543	needles	≈1050				i H_2O
15	Aluminum borohydride	$Al(BH_4)_3$	16962-07-5	71.510	flam liq	-64.5	44.5			reac H_2O
16	Aluminum bromate nonahydrate	$Al(BrO_3)_3 \cdot 9H_2O$	11126-81-1*	572.826	wh hyg cry	62	>100 dec			s H_2O
17	Aluminum bromide	$AlBr_3$	7727-15-3	266.694	wh-yel monocl cry; hyg	97.5	255	3.2		reac H_2O; s bz, tol

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
18	Aluminum bromide hexahydrate	$AlBr_3 \cdot 6H_2O$	7784-11-4	374.785	col-yel hyg cry	93		2.54		s H_2O, EtOH, CS_2
19	Aluminum carbide	Al_4C_3	1299-86-1	143.958	yel hex cry	2100	>2200 dec	2.36		reac H_2O
20	Aluminum chlorate nonahydrate	$Al(ClO_3)_3 \cdot 9H_2O$	15477-33-5	439.473	hyg cry					vs H_2O; s EtOH
21	Aluminum chloride	$AlCl_3$	7446-70-0	133.341	wh hex cry or powder; hyg	192.6	180 sp	2.48	45.1^{25}	s bz, ctc, chl
22	Aluminum chloride hexahydrate	$AlCl_3 \cdot 6H_2O$	7784-13-6	241.432	col hyg cry	100 dec		2.398	45.1^{25}	s EtOH, eth
23	Dichloromethylaluminum	$AlCl_2CH_3$	917-65-7	112.923	cry	72.7	95^{10}0			s bz, eth, hc
24	Chlorodiethylaluminum	$AlCl(C_2H_5)_2$	96-10-6	120.557	col liq	-74		0.96		reac H_2O
25	Chlorodiisobutylaluminum	$AlCl(C_4H_9)_2$	1779-25-5	176.664	hyg col liq	-40		0.95		s eth, hx
26	Aluminum diboride	AlB_2	12041-50-8	48.604	powder	>920 dec		3.19		s dil HCl
27	Aluminum dodecaboride	AlB_{12}	12041-54-2	156.714	yel-brn prisms	2070		2.55		s hot HNO_3; i acid, alk
28	Aluminum ethanolate	$Al(C_2H_5O)_3$	555-75-9	162.163	liq, condenses to wh solid	140				reac H_2O; sl xyl
29	Aluminum fluoride	AlF_3	7784-18-1	83.977	wh hex cry	2250 tp (220 MPa)	1276 sp	3.10	0.50^{25}	
30	Aluminum fluoride monohydrate	$AlF_3 \cdot H_2O$	32287-65-3	101.992	orth cry			2.17	0.50^{25}	
31	Aluminum fluoride trihydrate	$AlF_3 \cdot 3H_2O$	15098-87-0	138.023	wh hyg cry			1.914	0.50^{25}	
32	Aluminum hexafluorosilicate nonahydrate	$Al_2(SiF_6)_3 \cdot 9H_2O$	17099-70-6	642.329	hex prisms	>500 dec				s H_2O
33	Aluminum hydride	AlH_3	7784-21-6	30.006	col hex cry	>150 dec				reac H_2O
34	Aluminum hydroxide	$Al(OH)_3$	21645-51-2	78.004	wh amorp powder			2.42		i H_2O; s alk, acid
35	Aluminum hydroxychloride	$Al_2(OH)_5Cl \cdot 2H_2O$	1327-41-9	210.483	gl solid					s H_2O
36	Aluminum iodide	AlI_3	7784-23-8	407.695	wh leaflets	188.28	382	3.98		reac H_2O
37	Aluminum iodide hexahydrate	$AlI_3 \cdot 6H_2O$	10090-53-6	515.786	yel hyg cry powder					vs H_2O; s EtOH, eth
38	Aluminum lactate	$Al(C_3H_5O_3)_3$	18917-91-4	294.192	powder					vs H_2O
39	Aluminum molybdate	$Al_2(MoO_4)_3$	15123-80-5	533.78	wh pow	≈950				
40	Aluminum nitrate	$Al(NO_3)_3$	13473-90-0	212.997	wh hyg solid	dec			68.9^{25}	vs EtOH; sl ace
41	Aluminum nitrate nonahydrate	$Al(NO_3)_3 \cdot 9H_2O$	7784-27-2	375.134	wh hyg monocl cry	73	135 dec	1.72	68.9^{25}	vs EtOH; i pyr
42	Aluminum nitride	AlN	24304-00-5	40.989	blue-wh hex cry	3000		3.255		reac H_2O
43	Aluminum oleate	$Al(C_{18}H_{33}O_2)_3$	688-37-9	871.342	yel solid					i H_2O; s EtOH, bz
44	Aluminum oxalate monohydrate	$Al_2(C_2O_4)_3 \cdot H_2O$	814-87-9	336.035	wh pow					i H_2O, EtOH; s acid
45	Aluminum oxide (α)	Al_2O_3	1344-28-1	101.961	wh powder; hex	2054	2977	3.99		i H_2O, os; sl alk
46	Aluminum oxide (γ)	Al_2O_3	1344-28-1	101.961	soft wh pow	trans to corundum 1200		3.97		i H_2O; s acid; sl alk
47	Aluminum oxyhydroxide (boehmite)	AlO(OH)	1318-23-6	59.989	wh orth cry	trans to diasphore 227		3.07		i H_2O; s hot acid, alk
48	Aluminum oxyhydroxide (diaspore)	AlO(OH)	14457-84-2	59.989	orth cry	dec 450		3.38		i H_2O; s acid, alk
49	Aluminum palmitate	$Al(C_{15}H_{31}COO)_3$	555-35-1	793.230	wh-yel powder					i H_2O, EtOH; s peth
50	Aluminum 2,4-pentanedioate	$Al(CH_3COCHCOCH_3)_3$	13963-57-0	324.306	pale yel prisms	194.6	315	1.27		i H20; s bz, EtOH; sl hex
51	Aluminum perchlorate	$Al(ClO_4)_3$	14452-39-2	325.334	wh hyg cry				55^{0}	s H_2O, eth; i ctc
52	Aluminum perchlorate nonahydrate	$Al(ClO_4)_3 \cdot 9H_2O$	14452-39-2	487.471	wh hyg cry	82 dec		2.0	182.4^{0}	
53	Aluminum phosphate	$AlPO_4$	7784-30-7	121.953	wh rhomb plates	>1460		2.56		i H_2O; sl acid
54	Aluminum phosphate dihydrate	$AlPO_4 \cdot 2H_2O$	13477-75-3	157.984	wh rhom cry	dec 1500		2.54		i H_2O
55	Aluminum phosphate trihydroxide	$Al_2(OH)_3PO_4$	12004-29-4	199.957	wh or yel monocl cry			2.7		
56	Aluminum metaphosphate	$Al(PO_3)_3$	32823-06-6	263.898	col powder; tetr	≈1525		2.78		i H_2O
57	Aluminum hypophosphite	$Al(H_2PO_2)_3$	7784-22-7	221.948	cry powder	220 dec				i H_2O; s alk, acid
58	Aluminum phosphide	AlP	20859-73-8	57.956	grn or yel cub cry	2550		2.40		reac H_2O
59	Aluminum selenide	Al_2Se_3	1302-82-5	290.84	yel-brown powder	960		3.437		reac H_2O
60	Aluminum silicate (andalusite)	Al_2SiO_5	12183-80-1	162.046	gray-grn cry			3.145		
61	Aluminum silicate (kyanite)	Al_2SiO_5	1302-76-7	162.046	blue or gray tricl cry	dec 1000		3.68		
62	Aluminum silicate (mullite)	$3Al_2O_3 \cdot 2SiO_2$	1302-93-8	426.052	col orth cry	1750		3.17		i H_2O, acid, HF
63	Aluminum silicate (sillimanite)	Al_2SiO_5	12141-45-6	162.046	wh orth cry	1816		3.25		
64	Aluminum silicate dihydrate	$Al_2O_3 \cdot 2SiO_2 \cdot 2H_2O$	1332-58-7	258.161	wh-yel powder; tricl			2.59		i H_2O, acid, alk
65	Aluminum stearate	$Al(C_{18}H_{35}O_2)_3$	637-12-7	877.390	wh powder	115		1.070		i H_2O, EtOH, eth; s alk
66	Aluminum monostearate	$Al(OH)_2(C_{18}H_{35}O_2)$	7047-84-9	344.467	yel-wh pow	155		1.02		i H_2O
67	Aluminum distearate	$Al(OH)(C_{18}H_{35}O_2)_2$	300-92-5	610.928	wh pow	145				i H_2O
68	Aluminum sulfate	$Al_2(SO_4)_3$	10043-01-3	342.151	wh cry	1040 dec			38.5^{25}	i EtOH
69	Aluminum sulfate octadecahydrate	$Al_2(SO_4)_3 \cdot 18H_2O$	7784-31-8	666.426	col monocl cry	86 dec		1.69	38.5^{25}	
70	Aluminum sulfide	Al_2S_3	1302-81-4	150.158	yel-gray powder	1100		2.02		
71	Aluminum telluride	Al_2Te_3	12043-29-7	436.76	gray-blk hex cry	≈895		4.5		

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
72	Aluminum thiocyanate	$Al(SCN)_3$	538-17-0	201.229	yel powder					s H_2O; i EtOH, eth
73	Aluminum titanate	Al_2TiO_5	12004-39-6	181.827	refrac solid	1860				
74	Aluminum zirconium	Al_2Zr	12004-50-1	145.187	metallic solid	1645				
75	Americium	Am	7440-35-9	243	silv metal; hex or cub	1176	2011	12		s acid
76	Americium(III) oxide	Am_2O_3	12254-64-7	534	tan hex cry			11.77		s acid
77	Americium(III) bromide	$AmBr_3$	14933-38-1	483	wh orth cry			6.85		s H_2O
78	Americium(III) chloride	$AmCl_3$	13464-46-5	349	pink hex cry	500		5.87		
79	Americium(III) fluoride	AmF_3	13708-80-0	300	pink hex cry	1393		9.53		
80	Americium(III) iodide	AmI_3	13813-47-3	624	yel ortho cry	≈950		6.9		
81	Americium(IV) fluoride	AmF_4	15947-41-8	319	tan monocl cry			7.23		
82	Americium(IV) oxide	AmO_2	12005-67-3	275	blk cub cry	>1000 dec		11.68		s acid
83	Ammonia	NH_3	7664-41-7	17.031	col gas	-77.73	-33.33	0.696 g/L		vs H_2O; s EtOH, eth
84	Ammonium acetate	$NH_4C_2H_3O_2$	631-61-8	77.083	wh hyg cry	114		1.073	148[4]	s EtOH; sl ace
85	Ammonium azide	NH_4N_3	12164-94-2	60.059	orth cry; flam	160	exp	1.346	20.2[30]	
86	Ammonium benzoate	$NH_4C_7H_5O_2$	1863-63-4	139.152	wh cry or powder	198		1.26		s H_2O; sl EtOH
87	Ammonium bromate	NH_4BrO_3	13843-59-9	145.941	col hex cry	exp				vs H_2O
88	Ammonium bromide	NH_4Br	12124-97-9	97.943	wh hyg tetr cry	542 dec	396 sp	2.429	78.3[25]	s EtOH, ace; sl eth
89	Ammonium caprylate	$NH_4C_8H_{15}O_2$	5972-76-9	161.243	hyg monocl cry	≈75				reac H_2O; s EtOH; i chl, bz
90	Ammonium carbamate	NH_2COONH_4	1111-78-0	78.071	cry powder					vs H_2O; s EtOH
91	Ammonium carbonate	$(NH_4)_2CO_3$	506-87-6	96.086	col cry powder	58 dec			100[15]	
92	Ammonium chlorate	NH_4ClO_3	10192-29-7	101.490	wh cry	102 exp		1.80	28.7[0]	
93	Ammonium chloride	NH_4Cl	12125-02-9	53.492	col cub cry	520.1 tp (dec)	338 sp	1.519	39.5[25]	
94	Ammonium chromate	$(NH_4)_2CrO_4$	7788-98-9	152.071	yel cry	185 dec		1.90	37[25]	sl ace, MeOH; i EtOH
95	Ammonium chromic sulfate dodecahydrate	$NH_4Cr(SO_4)_2 \cdot 12H_2O$	10022-47-6	478.343	blue-viol cry	94 dec		1.72		s H_2O; sl EtOH
96	Ammonium cobalt(II) phosphate	$CoNH_4PO_4$	14590-13-7	171.943	red-viol powder (hyd)					i H_2O; s acid
97	Ammonium cobalt(II) phosphate monohydrate	$CoNH_4PO_4 \cdot H_2O$	16827-96-6	189.959	red-purp orth plates	dec 450				s acid
98	Ammonium cobalt(II) sulfate hexahydrate	$(NH_4)_2Co(SO_4)_2 \cdot 6H_2O$	13586-38-4	395.227	red monocl prisms			1.90		s H_2O; i EtOH
99	Ammonium copper(II) chloride	$CuCl_2 \cdot 2NH_4Cl$	10060-13-6*	241.435	yel hyg orth cry					s H_2O
100	Ammonium copper(II) chloride dihydrate	$CuCl_2 \cdot 2NH_4Cl \cdot 2H_2O$	10060-13-6	277.465	blue-grn tetr cry	110 dec		1.993		s H_2O, EtOH
101	Ammonium cyanide	NH_4CN	12211-52-8	44.056	col tetr cry	dec		1.10		vs H_2O
102	Ammonium dichromate	$(NH_4)_2Cr_2O_7$	7789-09-5	252.065	oran-red monocl cry; hyg	180 dec		2.155	35.6[20]	
103	Ammonium dihydrogen arsenate	$NH_4H_2AsO_4$	13462-93-6	158.975	tetr cry	300 dec		2.311	52.7[25]	
104	Ammonium dihydrogen phosphate	$NH_4H_2PO_4$	7722-76-1	115.026	wh tetr cry	190		1.80	40.4[25]	sl EtOH; i ace
105	Ammonium *O,O*-diethyldithiophosphate	$(C_2H_5O)_2P(S)SNH_4$	1068-22-0	203.264	cry	165				
106	Ammonium dithiocarbamate	NH_4NH_2CSS	513-74-6	110.202	yel ortho cry	99 dec		1.45		s H_2O
107	Ammonium ferricyanide trihydrate	$(NH_4)_3Fe(CN)_6 \cdot 3H_2O$	14221-48-8*	320.110	red cry					s H_2O; i EtOH
108	Ammonium ferrocyanide trihydrate	$(NH_4)_4Fe(CN)_6 \cdot 3H_2O$	14481-29-9*	338.149	yel cry	dec				s H_2O; i EtOH
109	Ammonium fluoride	NH_4F	12125-01-8	37.037	wh hex cry; hyg	238		1.015	83.5[25]	sl EtOH
110	Ammonium fluorosulfonate	NH_4SO_3F	13446-08-7	117.100	col needles	245				s H_2O, EtOH, MeOH
111	Ammonium formate	NH_4CHO_2	540-69-2	63.057	hyg cry	116		1.27	143[20]	s EtOH
112	Ammonium heptafluorotantalate	$(NH_4)_2TaF_7$	12022-02-5	350.014	hyg cry					
113	Ammonium hexabromoosmate(IV)	$(NH_4)_2OsBr_6$	24598-62-7	705.73	small blk cubes					sl H_2O; s glycerol; i EtOH
114	Ammonium hexabromoplatinate(IV)	$(NH_4)_2PtBr_6$	17363-02-9	710.585	powder	145 dec			0.59[20]	
115	Ammonium hexachloroiridate(III)	$(NH_4)_3IrCl_6$	15752-05-3	459.050	grn pow					
116	Ammonium hexachloroiridate(IV)	$(NH_4)_2IrCl_6$	16940-92-4	441.012	blk cry powder	dec		2.856	1.09[25]	
117	Ammonium hexachloroosmate(IV)	$(NH_4)_2OsCl_6$	12125-08-5	439.03	red cry or powder		subl	2.93		s H_2O, EtOH
118	Ammonium hexachloropalladate(IV)	$(NH_4)_2PdCl_6$	19168-23-1	355.22	red-brn hyg cry	dec		2.418		
119	Ammonium hexachloroplatinate(IV)	$(NH_4)_2PtCl_6$	16919-58-7	443.879	red-oran cub cry	380 dec		3.065	0.5[20]	i EtOH
120	Ammonium hexachlororuthenate(IV)	$(NH_4)_2RuCl_6$	18746-63-9	349.87	red cry					
121	Ammonium hexafluoroaluminate	$(NH_4)_3AlF_6$	7784-19-2	195.087	cub cry			1.78		s H_2O
122	Ammonium hexafluorogallate	$(NH_4)_3GaF_6$	14639-94-2	237.828	col cub cry	>200 dec		2.10		
123	Ammonium hexafluorogermanate	$(NH_4)_2GeF_6$	16962-47-3	222.71	wh cry	380	subl	2.564		s H_2O; i EtOH
124	Ammonium hexafluorophosphate	NH_4PF_6	16941-11-0	163.003	wh cub cry	58 dec		2.180		vs H_2O; s ace, EtOH, MeOH
125	Ammonium hexafluorosilicate	$(NH_4)_2SiF_6$	16919-19-0	178.153	wh cub or trig cry	dec		2.011	22.7[25]	i EtOH, ace
126	Ammonium hexafluorotitanate	$(NH_4)_2TiF$	16962-40-6	197.934	wh solid					s H_2O

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
127	Ammonium hexafluorozirconate(IV)	$(NH_4)_2ZrF_6$	16919-31-6	241.291	wh hex cry			1.154		s H_2O
128	Ammonium hydrogen arsenate	$(NH_4)_2HAsO_4$	7784-44-3	176.004	wh powder			1.99		s H_2O
129	Ammonium hydrogen carbonate	NH_4HCO_3	1066-33-7	79.056	col or wh prisms	107 dec		1.586	24.8^{25}	i EtOH, bz
130	Ammonium hydrogen citrate	$(NH_4)_2HC_6H_5O_7$	3012-65-5	226.184	col cry			1.48		vs H_2O; sl EtOH
131	Ammonium hydrogen fluoride	NH_4HF_2	1341-49-7	57.044	wh orth cry	125	240 dec	1.50	60.2^{20}	
132	Ammonium hydrogen malate	$NH_4C_4H_5O_5$	5972-71-4	151.118	orth cry	160		1.15		s H_2O; sl EtOH
133	Ammonium hydrogen oxalate monohydrate	$NH_4HC_2O_4 \cdot H_2O$	5972-72-5*	125.081	col rhomb cry	dec		1.56		sl H_2O, EtOH
134	Ammonium hydrogen phosphate	$(NH_4)_2HPO_4$	7783-28-0	132.055	wh cry	155 dec		1.619	69.5^{25}	i EtOH, ace
135	Ammonium hydrogen phosphite monohydrate	$(NH_4)_2HPO_3 \cdot H_2O$	51503-61-8	134.071	hyg cry					s H_2O
136	Ammonium hydrogen selenate	NH_4HSeO_4	10294-60-7	162.01	rhom cry	dec		2.162		
137	Ammonium hydrogen sulfate	NH_4HSO_4	7803-63-6	115.110	wh hyg cry	147		1.78	100^{20}	i EtOH, ace, py
138	Ammonium hydrogen sulfide	NH_4HS	12124-99-1	51.112	wh tetr or orth cry	dec		1.17	128^{0}	sl ace; i bz, eth
139	Ammonium hydrogen sulfite	NH_4HSO_3	10192-30-0	99.110	col cry	dec		2.03	71.8^{0}	
140	Ammonium hydrogen tartrate	$NH_4HC_4H_4O_6$	3095-65-6	167.117	wh cry			1.68		sl H_2O; s alk; i EtOH
141	Ammonium hydroxide	NH_4OH	1336-21-6	35.046	exists only in soln					
142	Ammonium hypophosphite	$NH_4H_2PO_2$	7803-65-8	83.028	wh hyg cry	dec				vs H_2O; sl EtOH; i ace
143	Ammonium iodate	NH_4IO_3	13446-09-8	192.941	wh powder	150		3.3	3.84^{25}	
144	Ammonium iodide	NH_4I	12027-06-4	144.943	wh tetr cry; hyg	551 dec	405 sp	2.514	178^{25}	sl EtOH, MeOH
145	Ammonium iron(II) sulfate hexahydrate	$(NH_4)_2Fe(SO_4)_2 \cdot 6H_2O$	7783-85-9	392.139	blue-grn monocl cry	≈100 dec		1.86		s H_2O; i EtOH
146	Ammonium iron(III) chromate	$NH_4Fe(CrO_4)_2$	7789-08-4	305.871	red powder					i H_2O
147	Ammonium iron(III) oxalate trihydrate	$(NH_4)_3Fe(C_2O_4)_3 \cdot 3H_2O$	13268-42-3	428.063	grn monocl cry; hyg	≈160 dec		1.780		vs H_2O; i EtOH
148	Ammonium iron(III) sulfate dodecahydrate	$NH_4Fe(SO_4)_2 \cdot 12H_2O$	7783-83-7	482.192	col to viol cry	≈37		1.71		vs H_2O; i EtOH
149	Ammonium lactate	$NH_4C_3H_5O_3$	52003-58-4	107.108	col cry	92				s H_2O, EtOH; sl MeOH; i ace, eth
150	Ammonium magnesium chloride hexahydrate	$NH_4MgCl_3 \cdot 6H_2O$	39733-35-2	256.794	hyg cry	dec 100		1.46	17^{20}	s H_2O
151	Ammonium mercuric chloride dihydrate	$(NH_4)_2HgCl_4 \cdot 2H_2O$	33445-15-7*	414.51	powder					s H_2O; sl EtOH
152	Ammonium metatungstate hexahydrate	$(NH_4)_6W_7O_{24} \cdot 6H_2O$	12028-48-7	1887.19	wh cry					s H_2O; i EtOH
153	Ammonium metavanadate	NH_4VO_3	7803-55-6	116.979	wh-yel cry	200 dec		2.326	4.8^{20}	
154	Ammonium molybdate(VI) tetrahydrate	$(NH_4)_6Mo_7O_{24} \cdot 4H_2O$	12054-85-2	1235.86	col or grn-yel cry	90 dec		2.498	43	i EtOH
155	Ammonium dimolybdate	$(NH_4)_2Mo_2O_7$	27546-07-2	339.95	cry					s H_2O
156	Ammonium molybdophosphate	$(NH_4)_3PO_4 \cdot 12MoO_3$	12026-66-3	1876.35	grn or yel cry	dec			0.02^{20}	sl H_2O; s alk
157	Ammonium nitrate	NH_4NO_3	6484-52-2	80.043	wh hyg cry; orth	169.7	dec 200-260	1.72	213^{25}	sl MeOH
158	Ammonium nitrite	NH_4NO_2	13446-48-5	64.044	wh-yel cry	60 exp		1.69	221^{25}	i eth
159	Ammonium nitroferricyanide	$(NH_4)_2Fe(CN)_5NO$	14402-70-1	252.016	red-brn cry					s H_2O, EtOH
160	Ammonium oleate	$NH_4C_{18}H_{33}O_2$	544-60-5	299.493	yel-brn paste	21				s H_2O; sl ace
161	Ammonium oxalate	$(NH_4)_2C_2O_4$	1113-38-8	124.096	col sol			1.5	5.20^{25}	
162	Ammonium oxalate monohydrate	$(NH_4)_2C_2O_4 \cdot H_2O$	6009-70-7	142.110	wh orth cry	dec		1.50	5.20^{25}	sl EtOH
163	Ammonium palmitate	$NH_4C_{15}H_{31}CO_2$	593-26-0	273.455	yel-wh powder	22				s H_2O; sl bz, xyl; i ace, EtOH, ctc
164	Ammonium pentaborate tetrahydrate	$NH_4B_5O_8 \cdot 4H_2O$	12007-89-5	272.150	wh cry				7.03^{18}	
165	Ammonium pentachlororhodate(III) monohydrate	$(NH_4)_2RhCl_5 \cdot H_2O$	63771-33-5	334.262	red cry	dec 210				
166	Ammonium pentachlorozincate	$(NH_4)_3ZnCl_5$	14639-98-6	296.789	hyg orth cry			1.81		vs H_2O
167	Ammonium perchlorate	NH_4ClO_4	7790-98-9	117.490	wh orth cry	dec, exp		1.95	24.5^{25}	s MeOH; sl EtOH, ace; i eth
168	Ammonium permanganate	NH_4MnO_4	13446-10-1	136.975	purp rhomb cry	70 dec		2.22	7.9^{15}	
169	Ammonium peroxydisulfate	$(NH_4)_2S_2O_8$	7727-54-0	228.202	monocl cry or wh powder	dec		1.982	83.5^{25}	
170	Ammonium perrhenate	NH_4ReO_4	13598-65-7	268.244	col powder			3.97	6.23^{20}	
171	Ammonium phosphate trihydrate	$(NH_4)_3PO_4 \cdot 3H_2O$	10361-65-6*	203.133	wh prisms				25.0^{25}	i ace
172	Ammonium phosphomolybdate monohydrate	$(NH_4)_3PO_4 \cdot 12MoO_3 \cdot H_2O$	54723-94-3	1894.36	yel cry or powder	dec			0.02	
173	Ammonium phosphotungstate dihydrate	$(NH_4)_3PO_4 \cdot 12WO_3 \cdot 2H_2O$	1311-90-6	2967.18	cry powder					sl H_2O
174	Ammonium picrate	$NH_4C_6H_2N_3O_7$	131-74-8	246.135	yel orth cry	exp		1.72		sl H_2O
175	Ammonium polysulfide	$(NH_4)_2S_x$	9080-17-5		yel unstab soln					reac acids
176	Ammonium salicylate	$NH_4C_7H_5O_3$	528-94-9	155.151	wh cry powder					vs H_2O; s EtOH

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
177	Ammonium selenate	$(NH_4)_2SeO_4$	7783-21-3	179.04	wh monocl cry	dec		2.194	117^{25}	i EtOH, ace
178	Ammonium selenite	$(NH_4)_2SeO_3$	7783-19-9	163.04	wh or red hyg cry	dec			121^{25}	
179	Ammonium stearate	$NH_4C_{18}H_{35}O_2$	1002-89-7	301.509	yel-wh powder	22		0.89		sl H_2O, bz; s EtOH, MeOH; i ace
180	Ammonium sulfamate	$NH_4NH_2SO_3$	7773-06-0	114.124	wh hyg cry	131	160 dec			vs H_2O; sl EtOH
181	Ammonium sulfate	$(NH_4)_2SO_4$	7783-20-2	132.140	wh or brn orth cry	280 dec		1.77	76.4^{25}	i EtOH, ace
182	Ammonium sulfide	$(NH_4)_2S$	12135-76-1	68.142	yel-oran cry	≈0 dec				s H_2O, EtOH, alk
183	Ammonium sulfite	$(NH_4)_2SO_3$	17026-44-7	116.140	wh hyg cry				64.2^{25}	
184	Ammonium sulfite monohydrate	$(NH_4)_2SO_3 \cdot H_2O$	7783-11-1	134.155	col cry	dec		1.41	64.2^{25}	i EtOH, ace
185	Ammonium tartrate	$(NH_4)_2C_4H_4O_6$	3164-29-2	184.147	wh cry	dec		1.601		s H_2O
186	Ammonium tellurate	$(NH_4)_2TeO_4$	13453-06-0	227.68	wh powder	dec		3.024		
187	Ammonium tetraborate tetrahydrate	$(NH_4)_2B_4O_7 \cdot 4H_2O$	12228-87-4	263.377	wh tetr cry	dec 87				vs H_2O; s HNO_3
188	Ammonium tetrachloroaluminate	NH_4AlCl_4	7784-14-7	186.833	wh hyg solid	304				s H_2O, eth
189	Ammonium tetrachloropalladate(II)	$(NH_4)_2PdCl_4$	13820-40-1	284.31	grn cry or red-brn pow					s H_2O
190	Ammonium tetrachloroplatinate(II)	$(NH_4)_2PtCl_4$	13820-41-2	372.973	red cry	dec		2.936		s H_2O; i EtOH
191	Ammonium tetrachlorozincate	$(NH_4)_2ZnCl_4$	14639-97-5	243.298	wh orth plates; hyg	150 dec		1.879		vs H_2O
192	Ammonium tetrafluoroantimonate	NH_4SbF_4	14972-90-8	215.793	col cry					s H_2O
193	Ammonium tetrafluoroborate	NH_4BF_4	13826-83-0	104.844	wh powder; orth	487 dec		1.871	25^{20}	
194	Ammonium tetrathiocyanodiammonochromate(III) monohydrate	$NH_4[Cr(NH_3)_2(SCN)_4] \cdot H_2O$	13573-16-5	354.440	red cry	270 dec				s H_2O, EtOH, ace; i bz
195	Ammonium tetrathiomolybdate	$(NH_4)_2MoS_4$	15060-55-6	260.28	red cry	100 dec				vs H_2O
196	Ammonium tetrathiotungstate	$(NH_4)_2WS_4$	13862-78-7	348.18	oran cry	dec		2.71		s H_2O
197	Ammonium tetrathiovandate	$(NH_4)_3VS_4$	14693-56-2	233.317	dark viol cry					
198	Ammonium thiocyanate	NH_4SCN	1762-95-4	76.121	col hyg cry	≈149	dec	1.30	181^{25}	vs EtOH; s ace; i chl
199	Ammonium thiosulfate	$(NH_4)_2S_2O_3$	7783-18-8	148.205	wh cry	150 dec		1.678		vs H_2O; i EtOH, eth
200	Ammonium titanium oxalate monohydrate	$(NH_4)_2TiO(C_2O_4)_2 \cdot H_2O$	10580-03-7	293.996	hyg cry					vs H_2O
201	Ammonium tungstate(VI)	$(NH_4)_{10}W_{12}O_{41}$	11120-25-5	3042.44	cry powder			2.3		s H_2O; i EtOH
202	Ammonium tungstate(VI) pentahydrate	$(NH_4)_{10}W_{12}O_{41} \cdot 5H_2O$	1311-93-9	3132.52	cry pow or plates			2.3		vs H_2O; i EtOH
203	Ammonium uranate(VI)	$(NH_4)_2U_2O_7$	7783-22-4	624.131	red-yel amorp powder					i H_2O, alk; s acid
204	Ammonium uranium fluoride	$UO_2(NH_4)_3F_5$	18433-40-4	419.135	grn-yel monocl cry					s H_2O; i EtOH
205	Ammonium valerate	$NH_4C_4H_9CO_2$	42739-38-8	119.163	hyg cry	108				vs H_2O, EtOH; s eth
206	Antimony (gray)	Sb	7440-36-0	121.760	silv metal; hex	630.628	1587	6.68		i dil acid
207	Antimony (black)	Sb	7440-36-0	121.760	blk amorp solid	trans gray 0				
208	Stibine	SbH_3	7803-52-3	124.784	col gas; flam	-88	-17	5.100 g/L		sl H_2O; s EtOH
209	Trimethylstibine	$Sb(CH_3)_3$	594-10-5	166.863	col flam liq	-62	81	1.52		
210	Pentamethylstibine	$Sb(CH_3)_5$	15120-50-0	196.933	col hyg liq	-19	127			reac H_2O
211	Tetramethyldistibine	$[Sb(CH_3)_2]_2$	41422-43-9	303.658	yel flam liq or red solid	17				
212	Antimony arsenide	SbAs	12322-34-8	196.682	hex cry	≈680		6.0		
213	Antimony potassium tartrate trihydrate	$K_2(SbC_4H_4O_6)_2 \cdot 3H_2O$	28300-74-5	669.889	col cry			2.6		sl H_2O
214	Antimony(III) acetate	$Sb(C_2H_3O_2)_3$	3643-76-3	298.891	wh pow					
215	Antimony(III) bromide	$SbBr_3$	7789-61-9	361.472	yel orth cry; hyg	97	288	4.35		reac H_2O; s ace, bz, chl
216	Antimony(III) chloride	$SbCl_3$	10025-91-9	228.119	col orth cry; hyg	73.4	220.3	3.14	987^{25}	s acid, EtOH, bz, ace
217	Antimony(III) fluoride	SbF_3	7783-56-4	178.755	wh orth cry; hyg	287	376	4.38	492^{25}	
218	Antimony(III) iodide	SbI_3	7790-44-5	502.473	red rhomb cry	171	400	4.92		reac H_2O; s EtOH, ace; i ctc
219	Antimony(III) iodide sulfide	SbIS	13816-38-1	280.729	dark red prisms or needles	400				
220	Antimony(III) oxide (senarmontite)	Sb_2O_3	1309-64-4	291.518	col cub cry	570 trans	1425	5.58		sl H_2O; i os
221	Antimony(III) oxide (valentinite)	Sb_2O_3	1309-64-4	291.518	wh orth cry	655	1425	5.7		sl H_2O; i os
222	Antimony(III) oxychloride	SbOCl	7791-08-4	173.212	wh momo cry	170 dec				reac H_2O; i EtOH, eth
223	Antimony(III) phosphate	$SbPO_4$	12036-46-3	216.731	cry pow					reac H_2O
224	Antimony(III) potassium oxalate trihydrate	$K_3Sb(C_2O_4)_3 \cdot 3H_2O$	5965-33-3*	557.158	cry pow					s H_2O
225	Antimony(III) selenide	Sb_2Se_3	1315-05-5	480.40	grn orth cry	611		5.81		sl H_2O
226	Antimony(III) sulfate	$Sb_2(SO_4)_3$	7446-32-4	531.708	wh cry powder; hyg	dec		3.62		sl H_2O
227	Antimony(III) sulfide	Sb_2S_3	1345-04-6	339.715	gray-blk orth cry	550		4.562		i H_2O; s conc HCl
228	Antimony(III) telluride	Sb_2Te_3	1327-50-0	626.32	gray cry	620		6.5		

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
229	Antimony(III,V) oxide	Sb_2O_4	1332-81-6	307.518	yel orth cry			6.64		
230	Antimony(V) chloride	$SbCl_5$	7647-18-9	299.025	col or yel liq	4	140 dec	2.34		reac H_2O; s chl, ctc
231	Antimony(V) fluoride	SbF_5	7783-70-2	216.752	hyg visc liq	8.3	141	3.10		reac H_2O
232	Antimony(V) dichlorotrifluoride	$SbCl_2F_3$	7791-16-4	249.661	visc liq					reac H_2O
233	Antimony(V) oxide	Sb_2O_5	1314-60-9	323.517	yel powder; cub	dec		3.78	0.3^{20}	
234	Antimony(V) sulfide	Sb_2S_5	1315-04-4	403.845	oran-yel powder	75 dec		4.120		i H_2O; s acid, alk
235	Argon	Ar	7440-37-1	39.948	col gas	-189.36 tp (69 kPa)	-185.847	1.633 g/L		sl H_2O
236	Arsenic (gray)	As	7440-38-2	74.922	gray metal; rhomb	817 tp (3.70 MPa)	616 sp	5.75		i H_2O
237	Arsenic (black)	As	7440-38-2	74.922	blk amorp solid	trans gray As 270		4.9		
238	Arsenic (yellow)	As	7440-38-2	74.922	soft yel cub cry	trans gray As 358		1.97		s CS_2
239	Arsine	AsH_3	7784-42-1	77.946	col gas	-116	-62.5	3.186 g/L		sl H_2O
240	Diarsine	As_2H_4	15942-63-9	153.875	unstab liq		≈100			
241	Arsenic acid	H_3AsO_4	7778-39-4	141.944	exists only in soln					
242	Arsenic acid hemihydrate	$H_3AsO_4 \cdot 0.5H_2O$	7778-39-4*	150.951	wh hyg cry	36.1		2.5		vs H_2O, EtOH
243	Arsenious acid	H_3AsO_3	13464-58-9	125.944	exists only in soln					
244	Arsenic diiodide	As_2I_4	13770-56-4	657.461	red cry	137				reac H_2O; s os
245	Arsenic hemiselenide	As_2Se	1303-35-1	228.80	blk cry					i H_2O, os; dec acid, alk
246	Arsenic sulfide	As_4S_4	12279-90-2	427.946	red monocl cry	320	565	3.5		i H_2O; sl bz; s alk
247	Arsenic(III) bromide	$AsBr_3$	7784-33-0	314.634	col or yel orth cry; hyg	31.1	221	3.40		reac H_2O; s hc, ctc; vs eth, bz
248	Arsenic(III) chloride	$AsCl_3$	7784-34-1	181.281	col liq	-16	130	2.150		reac H_2O; vs chl, ctc, eth
249	Arsenic(III) ethoxide	$As(C_2H_5O)_3$	3141-12-6	210.103	liq		166	1.21		
250	Arsenic(III) fluoride	AsF_3	7784-35-2	131.917	col liq	-5.9	57.13	2.7		reac H_2O; s EtOH, eth, bz
251	Arsenic(III) iodide	AsI_3	7784-45-4	455.635	red hex cry	141	424	4.73		sl H_2O, EtOH, eth; s bz. tol
252	Arsenic(III) oxide (arsenolite)	As_2O_3	1327-53-3	197.841	wh cub cry	274	460	3.86	2.05^{25}	
253	Arsenic(III) oxide (claudetite)	As_2O_3	1327-53-3	197.841	wh monocl cry	314	460	3.74	2.05^{25}	s dil acid, alk; i EtOH
254	Arsenic(III) selenide	As_2Se_3	1303-36-2	386.72	brn-blk solid	260		4.75		i H_2O; s alk
255	Arsenic(III) sulfide	As_2S_3	1303-33-9	246.038	yel-oran monocl cry	312	707	3.46		i H_2O; s alk
256	Arsenic(III) telluride	As_2Te_3	12044-54-1	532.64	blk monocl cry	621		6.50		
257	Arsenic(V) chloride	$AsCl_5$	22441-45-8	252.187	stab at low temp	≈-50 dec				
258	Arsenic(V) fluoride	AsF_5	7784-36-3	169.914	col gas	-79.8	-52.8	6.945 g/L		reac H_2O; s EtOH, bz, eth
259	Arsenic(V) oxide	As_2O_5	1303-28-2	229.840	wh amorp powder	315		4.32	65.8^{20}	vs EtOH
260	Arsenic(V) selenide	As_2Se_5	1303-37-3	544.64	blk solid	dec				i H_2O, EtOH, eth; s alk
261	Arsenic(V) sulfide	As_2S_5	1303-34-0	310.168	brn-yel amorp solid	dec				i H_2O; s alk
262	Astatine	At	7440-68-8	210	cry	302				s HNO_3, os
263	Barium	Ba	7440-39-3	137.327	silv-yel metal; cub	727	1897	3.62		reac H_2O; sl EtOH
264	Barium acetate	$Ba(C_2H_3O_2)_2$	543-80-6	255.416	wh powder			2.47	79.2^{25}	
265	Barium acetate monohydrate	$Ba(C_2H_3O_2)_2 \cdot H_2O$	5908-64-5	273.431	wh cry	110 dec		2.19	79.2^{25}	sl EtOH
266	Barium aluminate	$BaAl_2O_4$	12004-04-5	255.288	hex cry	1827				
267	Barium aluminide	$BaAl_4$	12672-79-6	245.253	metallic solid	1097				
268	Barium azide	$Ba(N_3)_2$	18810-58-7	221.367	monocl cry; exp	≈120 dec		2.936	17.3^{20}	sl EtOH; i eth
269	Barium bismuthate	$BaBiO_3$	12785-50-1	394.305	bronze cry	1040 dec				
270	Barium bromate	$Ba(BrO_3)_2$	13967-90-3	393.131	col monocl cry				0.79^{25}	s ace
271	Barium bromate monohydrate	$Ba(BrO_3)_2 \cdot H_2O$	10326-26-8	411.147	wh monocl cry	260 dec		3.99	0.831^{25}	i EtOH
272	Barium bromide	$BaBr_2$	10553-31-8	297.135	wh orth cry	857	1835	4.781	100^{25}	
273	Barium bromide dihydrate	$BaBr_2 \cdot 2H_2O$	7791-28-8	333.166	wh cry	75 dec		3.7	100^{25}	s MeOH; i EtOH, ace, diox
274	Barium calcium tungstate	Ba_2CaWO_6	15552-14-4	594.57	cub cry	1420				
275	Barium carbide	BaC_2	50813-65-5	161.348	gray tetr cry	dec		3.74		reac H_2O
276	Barium carbonate	$BaCO_3$	513-77-9	197.336	wh orth cry	1380 dec; 1555 (high pres.)		4.308	0.0014^{20}	s acid
277	Barium chlorate	$Ba(ClO_3)_2$	13477-00-4	304.229	wh cry	414			37.9^{25}	sl EtOH, ace
278	Barium chlorate monohydrate	$Ba(ClO_3)_2 \cdot H_2O$	10294-38-9	322.245	wh monocl cry	120 dec		3.179	37.9^{25}	s acid; sl EtOH, ace
279	Barium chloride	$BaCl_2$	10361-37-2	208.233	wh orth cry; hyg	961	1560	3.9	37.0^{25}	
280	Barium chloride dihydrate	$BaCl_2 \cdot 2H_2O$	10326-27-9	244.264	wh monocl cry	≈120 dec		3.097	37.0^{25}	i EtOH
281	Barium chloride fluoride	BaClF	13718-55-3	191.778	wh cry					
282	Barium chromate(V)	$Ba_3(CrO_4)_2$	12345-14-1	643.968	grn-blk hex cry			5.25		s H_2O

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
283	Barium chromate(VI)	$BaCrO_4$	10294-40-3	253.321	yel orth cry	1380		4.50	0.00026^{20}	reac acid
284	Barium citrate monohydrate	$Ba_3(C_6H_5O_7)_2 \cdot H_2O$	512-25-4*	808.195	gray-wh cry					s H_2O, acid
285	Barium copper yttrium oxide	$BaCuY_2O_5$	82642-06-6	458.682	grn cry; not superconductor					
286	Barium copper yttrium oxide	$Ba_2Cu_3YO_7$	109064-29-1	666.194	blk solid; HT superconductor					
287	Barium copper yttrium oxide	$Ba_2Cu_4YO_8$	114104-80-2	745.739	HT superconductor					
288	Barium copper yttrium oxide	$Ba_4Cu_7Y_2O_{15}$	124365-83-9	1411.933	HT superconductor					
289	Barium cyanide	$Ba(CN)_2$	542-62-1	189.361	wh cry powder					vs H_2O; s EtOH
290	Barium dichromate dihydrate	$BaCr_2O_7 \cdot 2H_2O$	10031-16-0	389.346	brn-red needles	dec				reac H_2O
291	Barium disilicate	$BaSi_2O_5$	12650-28-1	273.495	wh orth cry	1420		3.70		
292	Barium dithionate dihydrate	$BaS_2O_6 \cdot 2H_2O$	13845-17-5	333.484	wh cry	140 dec		4.54	22.1^{20}	sl EtOH
293	Barium ferrite	$BaFe_{12}O_{19}$	11138-11-7	1111.456	magnetic solid					
294	Barium ferrocyanide hexahydrate	$Ba_2Fe(CN)_6 \cdot 6H_2O$	13821-06-2*	594.694	yel monocl cry	80 dec				i H_2O, EtOH
295	Barium fluoride	BaF_2	7787-32-8	175.324	wh cub cry	1368	2260	4.893	0.161^{25}	
296	Barium formate	$Ba(CHO_2)_2$	541-43-5	227.362	cry			3.21		s H_2O; i EtOH
297	Barium hexaboride	BaB_6	12046-08-1	202.193	blk cub cry	2070		4.36		i H_2O; s acid; i EtOH
298	Barium hexafluorogermanate	$BaGeF_6$		323.96	wh cry	≈665		4.56		
299	Barium hexafluorosilicate	$BaSiF_6$	17125-80-3	279.403	wh orth needles	300 dec		4.29		i H_2O, EtOH; sl acid
300	Barium hydride	BaH_2	13477-09-3	139.343	gray orth cry	1200		4.16		reac H_2O
301	Barium hydrogen phosphate	$BaHPO_4$	10048-98-3	233.306	wh cry powder	400 dec		4.16	0.015^{20}	s dil acid
302	Barium hydrosulfide	$Ba(HS)_2$	25417-81-6	203.473	yel hyg cry					s H_2O
303	Barium hydrosulfide tetrahydrate	$Ba(HS)_2 \cdot 4H_2O$	12230-74-9	275.534	yel rhomb cry	50 dec				s H_2O
304	Barium hydroxide	$Ba(OH)_2$	17194-00-2	171.342	wh powder	408			4.91^{25}	
305	Barium hydroxide monohydrate	$Ba(OH)_2 \cdot H_2O$	22326-55-2	189.357	wh powder			3.743	4.91^{25}	s acid
306	Barium hydroxide octahydrate	$Ba(OH)_2 \cdot 8H_2O$	12230-71-6	315.464	wh monocl cry	78 dec		2.18	4.91^{25}	
307	Barium hypophosphite monohydrate	$Ba(H_2PO_2)_2 \cdot H_2O$	14871-79-5*	285.320	monocl plates			2.90		s H_2O; i EtOH
308	Barium iodate	$Ba(IO_3)_2$	10567-69-8	487.132	wh cry powder	476 dec		5.23	0.0396^{25}	
309	Barium iodate monohydrate	$Ba(IO_3)_2 \cdot H_2O$	7787-34-0	505.148	cry	130 dec		5.00	0.0396^{25}	s acid; i EtOH
310	Barium iodide	BaI_2	13718-50-8	391.136	wh orth cry	711		5.15	221^{25}	
311	Barium iodide dihydrate	$BaI_2 \cdot 2H_2O$	7787-33-9	427.167	col cry	740 dec		5.0	221^{25}	s EtOH, ace
312	Barium manganate(VI)	$BaMnO_4$	7787-35-1	256.263	grn-gray hyg cry			4.85	0.00041^{20}	
313	Barium metaborate monohydrate	$Ba(BO_2)_2 \cdot H_2O$	26124-86-7	240.962	wh powder	>900		3.3		sl H_2O
314	Barium metaborate dihydrate	$Ba(BO_2)_2 \cdot 2H_2O$	23436-05-7	258.977	wh prec	dec			1.3^{25}	sl H_2O
315	Barium metaphosphate	$Ba(PO_3)_2$	13466-20-1	295.271	wh powder	1560				i H_2O; sl acid
316	Barium metasilicate	$BaSiO_3$	13255-26-0	213.411	col rhomb powder	1605		4.40		i H_2O; s acid
317	Barium molybdate	$BaMoO_4$	7787-37-3	297.27	wh powder	1450		4.975	0.0021^{20}	
318	Barium niobate	$Ba(NbO_3)_2$	12009-14-2	419.136	yel orth cry	1455		5.44		i H_2O
319	Barium nitrate	$Ba(NO_3)_2$	10022-31-8	261.336	wh cub cry	590		3.24	10.3^{25}	sl EtOH, ace
320	Barium nitride	Ba_3N_2	12047-79-9	439.994	yel-brn cry	>500 dec		4.78		reac H_2O
321	Barium nitrite	$Ba(NO_2)_2$	13465-94-6	229.338	col hex cry	267		3.234	79.5^{25}	
322	Barium nitrite monohydrate	$Ba(NO_2)_2 \cdot H_2O$	7787-38-4	247.353	yel-wh hex cry	217 dec		3.18	79.5^{25}	i EtOH
323	Barium orthovanadate	$Ba_3(VO_4)_2$	39416-30-3	641.859	hex cry	707		5.14		
324	Barium oxalate	BaC_2O_4	516-02-9	225.346	wh powder	400 dec		2.658	0.0075	
325	Barium oxalate monohydrate	$BaC_2O_4 \cdot H_2O$	13463-22-4	243.361	wh cry powder			2.66	0.0075^{20}	s acid
326	Barium oxide	BaO	1304-28-5	153.326	wh-yel powder; cub and hex	1973		5.72(cub)	1.5^{20}	s dil acid, EtOH; i ace
327	Barium 2,4-pentanedioate octahydrate	$Ba(CH_3COCHCOCH_3)_2 \cdot 8H_2O$	12084-29-6*	479.665	col hyg cry	320 (anh)				
328	Barium perchlorate	$Ba(ClO_4)_2$	13465-95-7	336.228	col hex cry	505		3.20	312^{25}	vs EtOH
329	Barium perchlorate trihydrate	$Ba(ClO_4)_2 \cdot 3H_2O$	10294-39-0	390.274	col cry			2.74	312^{25}	s MeOH; sl EtOH, ace; i eth
330	Barium permanganate	$Ba(MnO_4)_2$	7787-36-2	375.198	brn-viol cry	200 dec		3.77	62.5^{20}	reac EtOH
331	Barium peroxide	BaO_2	1304-29-6	169.326	gray-wh tetr cry	450 dec		4.96	0.091^{20}	reac dil acid
332	Barium plumbate	$BaPbO_3$	12047-25-5	392.5	orth cry					
333	Barium potassium chromate	$BaK_2(CrO_4)_2$	27133-66-0	447.511	yel hex cry			3.63		vs H_2O
334	Barium pyrophosphate	$Ba_2P_2O_7$	13466-21-2	448.597	wh powder	1430		3.9	0.0088^{20}	s acid
335	Barium selenate	$BaSeO_4$	7787-41-9	280.29	wh rhomb cry	dec		4.75	0.015^{20}	
336	Barium selenide	BaSe	1304-39-8	216.29	cub cry powder	1780		5.02		reac H_2O
337	Barium selenite	$BaSeO_3$	13718-59-7	264.29	solid					i H_2O
338	Barium silicide	$BaSi_2$	1304-40-1	193.498	gray lumps	1180				reac H_2O
339	Barium sodium niobate	$Ba_2Na(NbO_3)_5$	12323-03-4	1002.167	wh orth cry	1437		5.40		i H_2O
340	Barium stannate	$BaSnO_3$	12009-18-6	304.035	cub cry			7.24		sl H_2O
341	Barium stannate trihydrate	$BaSnO_3 \cdot 3H_2O$	12009-18-6*	358.081	wh cry powder					sl H_2O; s acid

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
342	Barium stearate	$Ba(C_{18}H_{35}O_2)_2$	6865-35-6	704.266	wh powder	160		1.145		i H_2O, EtOH
343	Barium strontium niobate	$BaSr(NbO_3)_4$	37185-09-4	788.57	pale yel solid					
344	Barium strontium tungstate	Ba_2SrWO_6	14871-56-8	642.11	hyg pow	1400				
345	Barium sulfate	$BaSO_4$	7727-43-7	233.390	wh orth cry	1580		4.49	0.00031^{20}	i EtOH
346	Barium sulfide	BaS	21109-95-5	169.392	col cub cry or gray powder	2227		4.3	8.94^{25}	
347	Barium sulfite	$BaSO_3$	7787-39-5	217.390	wh monocl cry	dec		4.44	0.0011^{25}	i EtOH
348	Barium tartrate	$BaC_4H_4O_6$	5908-81-6	285.398	wh cry			2.98		s H_2O; i EtOH
349	Barium tetracyanoplatinate(II) tetrahydrate	$BaPt(CN)_4 \cdot 4H_2O$	13755-32-3	508.543	yel powder or cry			2.076		sl H_2O; i EtOH
350	Barium tetraiodomercurate(II)	$BaHgI_4$	10048-99-4	845.54	yel-red hyg cry					vs H_2O, EtOH
351	Barium thiocyanate	$Ba(SCN)_2$	2092-17-3	253.491	hyg cry				167^{25}	s ace, MeOH, EtOH
352	Barium thiocyanate dihydrate	$Ba(SCN)_2 \cdot 2H_2O$	2092-17-3*	289.522	hyg wh cry				167^{25}	s EtOH
353	Barium thiocyanate trihydrate	$Ba(SCN)_2 \cdot 3H_2O$	68016-36-4	307.537	wh needles; hyg			2.286	167^{25}	s EtOH
354	Barium thiosulfate	BaS_2O_3	35112-53-9	249.455	wh cry powder	220 dec			0.2^{20}	i EtOH
355	Barium thiosulfate monohydrate	$BaS_2O_3 \cdot H_2O$	7787-40-8	267.471	wh cry powder	dec		3.5	0.2	i EtOH
356	Barium titanate ($BaTiO_3$)	$BaTiO_3$	12047-27-7	233.192	wh tetr cry	1625		6.02		i H_2O
357	Barium titanate ($BaTi_2O_5$)	$BaTi_2O_5$	12009-27-7	313.058	wh solid					
358	Barium titanate ($BaTi_4O_9$)	$BaTi_4O_9$	12009-31-3	472.790	wh solid					
359	Barium titanium silicate	$BaTi(SiO_3)_3$	15491-35-7	413.446	rhom blue-pur cry					
360	Barium tungstate	$BaWO_4$	7787-42-0	385.17	wh tetr cry	1475	1730	5.04	0.0016^{20}	
361	Barium uranium oxide	BaU_2O_7	10380-31-1	725.381	oran-yel powder					i H_2O; s acid
362	Barium yttrium tungsten oxide	$Ba_3Y_3WO_9$	37265-86-4	1006.53	cub cry	1470				
363	Barium zirconate	$BaZrO_3$	12009-21-1	276.549	gray-wh cub cry	2500		5.52		i H_2O, alk; sl acid
364	Barium zirconium silicate	$BaO \cdot ZrO_2 \cdot SiO_2$		336.634	wh pow					
365	Berkelium (α form)	Bk	7440-40-6	247	hex cry	trans to · 930		14.78		
366	Berkelium (β form)	Bk	7440-40-6	247	cub cry	986		13.25		
367	Beryllium	Be	7440-41-7	9.012	hex cry	1287	2471	1.85		s acid, alk
368	Beryllium acetate	$Be(C_2H_3O_2)_2$	543-81-7	127.101	wh cry	60 dec				i H_2O, EtOH
369	Beryllium basic acetate	$Be_4O(C_2H_3O_2)_6$	1332-52-1	406.312	wh cry	285	330	1.25		i H_2O; s eth, os
370	Beryllium aluminate	$BeAl_2O_4$	12004-06-7	126.973	orth cry			3.65		
371	Beryllium aluminum metasilicate	$Be_3Al_2(SiO_3)_6$	1302-52-9	537.502	col or grn-yel cry; hex			2.64		
372	Beryllium boride (BeB_2)	BeB_2	12228-40-9	30.634	refrac solid	>1970				
373	Beryllium boride (BeB_6)	BeB_6	12429-94-6	73.878	red solid	2070				
374	Beryllium boride (Be_2B)	Be_2B	12536-51-5	28.835	pink cry	1520				
375	Beryllium boride (Be_4B)	Be_4B	12536-52-6	46.860	refrac solid	1160				
376	Beryllium borohydride	$Be(BH_4)_2$	17440-85-6	36.682	solid	125 dec	subl			reac H_2O
377	Beryllium bromide	$BeBr_2$	7787-46-4	168.820	orth cry; hyg	508	473 sp	3.465		vs H_2O; s EtOH, pyr
378	Beryllium carbide	Be_2C	506-66-1	30.035	red cub cry	2127		1.90		reac H_2O
379	Beryllium carbonate tetrahydrate	$BeCO_3 \cdot 4H_2O$	60883-64-9	93.085	wh solid	100 dec			0.36^0	
380	Beryllium basic carbonate	$Be_3(OH)_2(CO_3)_2$	66104-24-3	181.069	wh powder					i H_2O; s acid, alk
381	Beryllium chloride	$BeCl_2$	7787-47-5	79.918	wh-yel orth cry; hyg	415	482	1.90	71.5^{25}	s EtOH, eth, py; i bz, tol
382	Beryllium fluoride	BeF_2	7787-49-7	47.009	tetr cry or gl; hyg	552	1283	2.1		vs H_2O; sl EtOH
383	Beryllium formate	$Be(CHO_2)_2$	1111-71-3	99.047	powder	>250 dec				reac H_2O; i os
384	Beryllium hydride	BeH_2	7787-52-2	11.028	wh amorp solid	250 dec		0.65		reac H_2O; i eth, tol
385	Beryllium hydrogen phosphate	$BeHPO_4$	13598-15-7	104.991	cry					i H_2O
386	Beryllium hydroxide (α)	$Be(OH)_2$	13327-32-7	43.027	wh powder or cry	≈200 dec		1.92		sl H_2O, alk; s acid
387	Beryllium hydroxide (β)	$Be(OH)_2$	13327-32-7	43.027	col tetr cry	dec 138				i H_2O; s, acid, alk
388	Beryllium iodide	BeI_2	7787-53-3	262.821	hyg needles	480	590	4.32		reac H_2O; s EtOH
389	Beryllium nitrate trihydrate	$Be(NO_3)_2 \cdot 3H_2O$	13597-99-4	187.068	yel-wh hyg cry	≈30	dec		107^{20}	s EtOH
390	Beryllium nitride	Be_3N_2	1304-54-7	55.050	gray refrac cry; cub	2200		2.71		reac acid, alk
391	Beryllium oxalate trihydrate	$BeC_2O_4 \cdot 3H_2O$	15771-43-4	151.077	rhom cry	dec 320				vs H_2O
392	Beryllium oxide	BeO	1304-56-9	25.011	wh hex cry	2578		3.01		i H_2O; sl acid, alk
393	Beryllium 2,4-pentanedioate	$Be(CH_3COCHCOCH_3)_2$	10210-64-7	207.228	monocl cry powder	108	270	1.168		i H_2O; vs EtOH, eth
394	Beryllium perchlorate tetrahydrate	$Be(ClO_4)_2 \cdot 4H_2O$	7787-48-6	279.975	hyg cry	250 dec			198^{25}	
395	Beryllium selenate tetrahydrate	$BeSeO_4 \cdot 4H_2O$	10039-31-3	224.03	orth cry	100 dec		2.03		vs H_2O
396	Beryllium sulfate	$BeSO_4$	13510-49-1	105.075	col tetr cry; hyg	1127		2.5	41.3^{25}	
397	Beryllium sulfate dihydrate	$BeSO_4 \cdot 2H_2O$	14215-00-0	141.105	col cry	dec 92				
398	Beryllium sulfate tetrahydrate	$BeSO_4 \cdot 4H_2O$	7787-56-6	177.136	col tetr cry	≈100 dec		1.71	41.3^{25}	i EtOH
399	Beryllium sulfide	BeS	13598-22-6	41.077	col cub cry	dec		2.36		reac hot H_2O
400	Bismuth	Bi	7440-69-9	208.980	gray-wh soft metal	271.406	1564	9.79		s acid
401	Bismuth acetate	$Bi(C_2H_3O_2)_3$	22306-37-2	386.111	col tablets	250				i H_2O
402	Bismuth subacetate	$BiOC_2H_3O_2$	5142-76-7	284.023	thin cry plates					i H_2O; s dil acid
403	Bismuth antimonide	BiSb	12323-19-2	330.740	cry	475				

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
404	Bismuth arsenate	$BiAsO_4$	13702-38-0	347.900	wh monocl cry			7.14		i H_2O; sl conc HNO_3
405	Bismuth basic carbonate	$(BiO)_2CO_3$	5892-10-4	509.969	wh powder			6.86		i H_2O; s acid
406	Bismuth basic dichromate	$Bi_2O_3 \cdot 2CrO_3$		665.948	red-oran amorp pow					i H_2O; s acid, alk
407	Bismuth citrate	$BiC_6H_5O_7$	813-93-4	398.080	wh powder			3.458		i H_2O; sl EtOH
408	Bismuth hydride	BiH_3	18288-22-7	212.004	col gas; unstab	-67	≈17	8.665 g/L		
409	Bismuth hydroxide	$Bi(OH)_3$	10361-43-0	260.002	wh-yel amorp powder			4.962		i H_2O; s acid
410	Bismuth germanium oxide	$2Bi_2O_3 \cdot 3GeO_2$	12233-56-6	1245.84	wh pow	1044				
411	Bismuth hexafluoro-2,4-pentanedioate	$Bi(CF_3COCHCOCF_3)_3$	142617-56-9	830.132	powder	96				
412	Bismuth molybdate	Bi_2MoO_6	13565-96-3	609.90	yel solid			9.32		
413	Bismuth molybdate	$Bi_2(MoO_4)_3$	51898-99-8	897.77	monocl cry			5.95		
414	Bismuth nitrate pentahydrate	$Bi(NO_3)_3 \cdot 5H_2O$	10035-06-0	485.071	col tricl cry; hyg	≈75 dec		2.83		reac H_2O; s ace; i EtOH
415	Bismuth subnitrate	$Bi_5O(OH)_9(NO_3)_4$	1304-85-4	1461.987	hyg cry powder	260 dec		4.928		i H_2O, EtOH; s dil acid
416	Bismuth oleate	$Bi(C_{18}H_{33}O_2)_3$	52951-38-9	1053.340	soft yel-brn solid					i H_2O; s eth; sl bz
417	Bismuth oxalate	$Bi_2(C_2O_4)_3$	6591-55-5	682.018	wh powder					i H_2O, EtOH; s dil acid
418	Bismuth oxide	Bi_2O_3	1304-76-3	465.959	yel monocl cry or powder	825	1890	8.9		i H_2O; s acid
419	Bismuth tetroxide	Bi_2O_4	12048-50-9	481.959	red-oran powder	305		5.6		reac H_2O
420	Bismuth oxybromide	BiOBr	7787-57-7	304.883	col tetr cry	560 dec		8.08		i H_2O, EtOH; s acid
421	Bismuth oxychloride	BiOCl	7787-59-9	260.432	wh tetr cry	575 dec		7.72		i H_2O
422	Bismuth oxyiodide	BiOI	7787-63-5	351.883	red tetr cry	300 dec		7.92		i H_2O, EtOH, chl; s HCl
423	Bismuth oxynitrate	$BiONO_3$	10361-46-3	286.985	wh powder	260 dec		4.93		i H_2O, EtOH; s acid
424	Bismuth phosphate	$BiPO_4$	10049-01-1	303.951	monocl cry			6.32		sl H_2O, dil acid; i EtOH
425	Bismuth potassium iodide	BiK_4I_7	41944-01-8	1253.704	red cry					reac H_2O; s alk iodide soln
426	Bismuth selenide	Bi_2Se_3	12068-69-8	654.84	blk hex cry	710 dec		7.5		i H_2O
427	Bismuth stannate pentahydrate	$Bi_2(SnO_3)_3 \cdot 5H_2O$	12777-45-6	1008.162	wh cry					i H_2O
428	Bismuth sulfate	$Bi_2(SO_4)_3$	7787-68-0	706.149	wh needles or powder	405 dec		5.08		reac H_2O, EtOH
429	Bismuth sulfide	Bi_2S_3	1345-07-9	514.156	blk-brn orth cry	850		6.78		i H_2O; s acid
430	Bismuth telluride	Bi_2Te_3	1304-82-1	800.76	gray hex plates	580		7.74		i H_2O; s EtOH
431	Bismuth tribromide	$BiBr_3$	7787-58-8	448.692	yel cub cry	219	462	5.72		reac H_2O; s dil acid, ace; i EtOH
432	Bismuth trichloride	$BiCl_3$	7787-60-2	315.339	col or yel cub cry; hyg	234	441	4.75		reac H_2O; s acid, EtOH, ace
433	Bismuth trifluoride	BiF_3	7787-61-3	265.975	wh-gray cub cry	727	900	8.3		i H_2O
434	Bismuth pentafluoride	BiF_5	7787-62-4	303.972	wh tetr needles; hyg	151.4	230	5.55		reac H_2O
435	Bismuth triiodide	BiI_3	7787-64-6	589.693	blk-brn hex cry	408.6	542	5.778	0.00078[20]	s EtOH
436	Bismuth trimethyl	$Bi(CH_3)_3$	593-91-9	254.083	col flam liq	-86	110	2.3		
437	Bismuth titanate	$Bi_4(TiO_4)_3$	12048-51-0	1171.516	wh orth cry			7.85		
438	Bismuth tungstate	$Bi_2(WO_4)_3$	13595-87-4	1161.47	wh powder					
439	Bismuth vanadate	$BiVO_4$	14059-33-7	323.920	orth cry	trans 500		6.25		i H_2O; s acid
440	Bismuth zirconate	$2Bi_2O_3 \cdot 3ZrO_2$	37306-42-6	1301.587	wh pow					
441	Boron	B	7440-42-8	10.811	blk rhomb cry	2075	4000	2.34		i H_2O
442	Diborane	B_2H_6	19287-45-7	27.670	col gas; flam	-164.85	-92.49	1.131 g/L		reac H_2O
443	Tetraborane(10)	B_4H_{10}	18283-93-7	53.323	unstab col gas	-120	18	2.180 g/L		reac H_2O
444	Pentaborane(9)	B_5H_9	19624-22-7	63.126	flam col liq	-46.74	60.10	0.60		reac hot H_2O
445	Pentaborane(11)	B_5H_{11}	18433-84-6	65.142	col liq; unstab	-122	65			reac H_2O
446	Hexaborane(10)	B_6H_{10}	23777-80-2	74.945	col liq	-62.3	108 dec	0.67		reac hot H_2O
447	Hexaborane(12)	B_6H_{12}	12008-19-4	76.961	col liq	-82.3	≈85			reac H_2O
448	Nonaborane(15)	B_9H_{15}	19465-30-6	112.418	col liq	2.7				
449	Decaborane(14)	$B_{10}H_{14}$	17702-41-9	122.221	wh orth cry	98.78	213	0.94		sl H_2O; s EtOH, bz, CS_2, ctc
450	Decaborane(16)	$B_{10}H_{16}$	71595-75-0	124.237	col cry	≈81	dec 170	subl		
451	Dodecaborane(16)	$B_{12}H_{16}$	89711-39-7	145.859	col cry	65				s bz, hx
452	Tridecaborane(19)	$B_{13}H_{19}$	43093-20-5	159.694	yel cry	44				s hx, CH2Cl2
453	Tetradecaborane(18)	$B_{14}H_{18}$	55606-55-8	169.497	visc yel oil		dec 100			s cyhex, CS_2
454	Hexadecaborane(20)	$B_{16}H_{20}$	28265-11-4	193.135	col cry	≈110				s ctc, cyhex, thf
455	Octadecaborane(22)	$B_{18}H_{22}$	11071-61-7	216.773	yel cry	180				s os
456	Tetrabromodiborane	B_2Br_4	14355-29-4	341.238	col liq	≈1	dec 20			
457	Tetrachlorodiborane	B_2Cl_4	13701-67-2	163.434	col liq; flam	-92.6	66.5			reac H_2O

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
458	Tetrafluorodiborane	B_2F_4	13965-73-6	97.616	col gas; flam	-56	-34.0	3.990 g/L		reac H_2O
459	Borane carbonyl	BH_3CO	13205-44-2	41.845	col gas	-137	-64	1.710 g/L		reac H_2O
460	Borazine	$B_3N_3H_6$	6569-51-3	80.501	col liq	-58	53	0.824		reac H_2O
461	Boric acid	H_3BO_3	10043-35-3	61.833	col tricl cry	170.9		1.5	5.80^{25}	sl EtOH
462	Metaboric acid (α form)	HBO_2	13460-50-9	43.818	col orth cry; hyg	176		1.784		s H_2O
463	Metaboric acid (β form)	HBO_2	13460-50-9	43.818	col monocl cry; hyg	201		2.045		s H_2O
464	Metaboric acid (γ form)	HBO_2	13460-50-9	43.818	col cub cry	236		2.487		s H_2O
465	Tetrafluoroboric acid	HBF_4	16872-11-0	87.813	col liq		130 dec	≈1.8		vs H_2O, EtOH
466	Boron arsenide	BAs	12005-69-5	85.733	brn cub cry	1100 dec		5.22		
467	Boron carbide	B_4C	12069-32-8	55.255	hard blk cry	2350	>3500	2.50		i H_2O, acid
468	Boron nitride	BN	10043-11-5	24.818	wh powder; hex or cub cry	2967		2.18		i H_2O, acid
469	Boron oxide	B_2O_3	1303-86-2	69.620	col gl or hex cry; hyg	450		2.55	2.2^{20}	s EtOH
470	Boron phosphide	BP	20205-91-8	41.785	red cub cry or powder	1125 dec				reac H_2O, acid
471	Boron silicide	B_6Si	12008-29-6	92.952	blk cry	1980				
472	Boron sulfide	B_2S_3	12007-33-9	117.817	yel amorp solid	563		≈1.7		
473	Boron tribromide	BBr_3	10294-33-4	250.523	col liq; hyg	-46	91.3	2.6		reac H_2O, EtOH
474	Boron trichloride	BCl_3	10294-34-5	117.170	col liq or gas	-107.3	12.5	4.789 g/L		reac H_2O, EtOH
475	Boron trifluoride	BF_3	7637-07-2	67.806	col gas	-126.8	-99.9	2.772 g/L		s H_2O
476	Boron trifluoride etherate	$BF_3(C_2H_5)_2O$	109-63-7	141.927	liq	-60.4	125.5	1.125^{25}		reac H_2O; vs eth, EtOH
477	Boron triiodide	BI_3	13517-10-7	391.524	wh needles	49.7	209.5	3.35		i H_2O
478	Bromine	Br_2	7726-95-6	159.808	red liq	-7.2	58.8	3.1028		sl H_2O
479	Bromic acid	$HBrO_3$	7789-31-3	128.910	stab only in aq soln					s H_2O
480	Hypobromous acid	HOBr	13517-11-8	96.911	exists aq soln					s H_2O
481	Bromine dioxide	BrO_2	21255-83-4	111.903	unstab yel cry	≈0 dec				
482	Bromine monoxide	Br_2O	21308-80-5	175.807	unstab brn solid	-17.5 dec				
483	Dibromine trioxide	Br_2O_3	53809-75-9	207.806	oran needles (LT)	dec -40				
484	Dibromine pentoxide	Br_2O_5	58572-43-3	239.805	col cry (low temp)	-20 dec				
485	Bromine azide	BrN_3	13973-87-0	121.924	red cry; exp	≈45	exp			
486	Bromine chloride	BrCl	13863-41-7	115.357	dark red liq (<5°C)	-66	5 dec			reac H_2O; s eth, CS_2
487	Bromine fluoride	BrF	13863-59-7	98.902	unstab red-brn gas	≈-33	≈20 dec	4.043 g/L		
488	Bromine trifluoride	BrF_3	7787-71-5	136.899	col hyg liq	8.77	125.8	2.803		reac H_2O
489	Bromine pentafluoride	BrF_5	7789-30-2	174.896	col liq	-60.5	41.3	2.460		reac H_2O (exp)
490	Bromosyl trifluoride	$BrOF_3$	61519-37-7	152.898	col liq	-5	dec >20			reac H_2O
491	Bromyl fluoride	BrO_2F	22585-64-4	130.901	col liq	-9	dec 55			reac H_2O
492	Perbromyl fluoride	BrO_3F	37265-91-1	146.900	col gas	-110	dec 20			reac H_2O
493	Cadmium	Cd	7440-43-9	112.411	silv-wh metal	321.069	767	8.69		i H_2O; reac acid
494	Cadmium acetate	$Cd(C_2H_3O_2)_2$	543-90-8	230.500	col cry	255		2.34		s H_2O, EtOH
495	Cadmium acetate dihydrate	$Cd(C_2H_3O_2)_2 \cdot 2H_2O$	5743-04-4	266.529	wh cry	130 dec		2.01		vs H_2O; s EtOH
496	Cadmium antimonide	CdSb	12014-29-8	234.171	orth cry	456		6.92		
497	Cadmium arsenide	Cd_3As_2	12006-15-4	487.076	gray tetr cry	721		6.25		
498	Cadmium azide	$Cd(N_3)_2$	14215-29-3	196.451	yel-wh orth cry; exp	exp		3.24		
499	Cadmium borotungstate octadecahydrate	$Cd_5(BW_{12}O_{40}) \cdot 18H_2O$	1306-26-9	3743.20	yel cry					vs H_2O
500	Cadmium bromide	$CdBr_2$	7789-42-6	272.219	wh-yel hex cry; hyg	568	863	5.19	115^{25}	sl ace, eth
501	Cadmium bromide tetrahydrate	$CdBr_2 \cdot 4H_2O$	13464-92-1	344.281	wh-yel cry				115^{25}	s ace, EtOH
502	Cadmium carbonate	$CdCO_3$	513-78-0	172.420	wh hex cry	500 dec		5.026		i H_2O; s acid
503	Cadmium chlorate dihydrate	$Cd(ClO_3)_2 \cdot 2H_2O$	22750-54-5*	315.344	col hyg cry	80 dec		2.28	2.64^0	
504	Cadmium chloride	$CdCl_2$	10108-64-2	183.317	rhom cry; hyg	568	964	4.08	120^{25}	s ace; sl EtOH; i eth
505	Cadmium chloride monohydrate	$CdCl_2 \cdot H_2O$	34330-64-8	201.332	wh cry				120^{25}	
506	Cadmium chloride hemipentahydrate	$CdCl_2 \cdot 2.5H_2O$	7790-78-5	228.354	wh rhomb leaflets			3.327	120^{25}	s ace
507	Cadmium chromate	$CdCrO_4$	14312-00-6	228.405	yel orth cry			4.5		i H_2O
508	Cadmium cyanide	$Cd(CN)_2$	542-83-6	164.445	wh cub cry			2.23	1.7^{15}	
509	Cadmium dichromate monohydrate	$CdCr_2O_7 \cdot H_2O$	69239-51-6	346.414	oran solid					s H_2O
510	Cadmium 2-ethylhexanoate	$Cd(C_8H_{15}O_2)_2$	2420-98-6	398.818	powder					
511	Cadmium fluoride	CdF_2	7790-79-6	150.408	cub cry	1075	1750	6.33	4.36^{25}	s acid; i EtOH
512	Cadmium hydroxide	$Cd(OH)_2$	21041-95-2	146.426	wh trig or hex cry	130 dec		4.79	0.00015^{20}	s dil acid
513	Cadmium iodate	$Cd(IO_3)_2$	7790-81-0	462.216	wh powder			6.48	0.091^{25}	s HNO_3

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
514	Cadmium iodide	CdI_2	7790-80-9	366.220	col hex flakes	388	744	5.64	86.2^{25}	vs H_2O; s EtOH, eth, ace
515	Cadmium metasilicate	$CdSiO_3$	13477-19-5	188.495	grn monocl cry	1252		5.10		
516	Cadmium molybdate	$CdMoO_4$	13972-68-4	272.35	col tetr cry	≈900 dec		5.4		i H_2O; s acid
517	Cadmium niobate	$Cd_2Nb_2O_7$	12187-14-3	522.631	cub cry	≈1410		6.28		i H_2O
518	Cadmium nitrate	$Cd(NO_3)_2$	10325-94-7	236.420	wh cub cry; hyg	360		3.6	156^{25}	s EtOH
519	Cadmium nitrate tetrahydrate	$Cd(NO_3)_2 \cdot 4H_2O$	10022-68-1	308.482	col orth cry; hyg	59.5		2.45	156^{25}	s EtOH, ace
520	Cadmium oxalate	CdC_2O_4	814-88-0	200.430	wh solid			3.32	0.0060^{25}	
521	Cadmium oxalate trihydrate	$CdC_2O_4 \cdot 3H_2O$	20712-42-9	254.476	wh amorp powder	340 dec			0.0060^{25}	i EtOH; s dil acid
522	Cadmium oxide	CdO	1306-19-0	128.410	brn cub cry		1559 sp	8.15		i H_2O; s dil acid
523	Cadmium 2,4-pentanedioate	$Cd(CH_3COCHCOCH_3)_2$	14689-45-3	310.627	wh solid or red cry	235				
524	Cadmium perchlorate hexahydrate	$Cd(ClO_4)_2 \cdot 6H_2O$	10326-28-0	419.404	wh hex cry			2.37	191.5^{25}	
525	Cadmium phosphate	$Cd_3(PO_4)_2$	13477-17-3	527.176	powder	≈1500				i H_2O
526	Cadmium phosphide	Cd_3P_2	12014-28-7	399.181	grn tetr needles	700		5.96		s dil HCl
527	Cadmium selenate dihydrate	$CdSeO_4 \cdot 2H_2O$	10060-09-0	291.40	orth cry	100 dec		3.62	70.5^{25}	
528	Cadmium selenide	CdSe	1306-24-7	191.37	wh cub cry	1240		5.81		i H_2O
529	Cadmium selenite	$CdSeO_3$	13814-59-0	239.37	col prisms					
530	Cadmium stearate	$Cd(C_{18}H_{35}O_2)_2$	2223-93-0	679.350	wh cry pow	134		1.21		
531	Cadmium succinate	$CdC_4H_4O_4$	141-00-4	228.484	wh pow or needles				0.37^{40}	sl H_2O; i EtOH
532	Cadmium sulfate	$CdSO_4$	10124-36-4	208.474	col orth cry	1000		4.69	76.7^{25}	i EtOH
533	Cadmium sulfate monohydrate	$CdSO_4 \cdot H_2O$	7790-84-3	226.489	monocl cry	105		3.79	76.7^{25}	
534	Cadmium sulfate octahydrate	$CdSO_4 \cdot 8H_2O$	15244-35-6	352.596	col monocl cry	40 dec		3.08	76.7^{25}	
535	Cadmium sulfide	CdS	1306-23-6	144.476	yel-oran hex cry	≈1480		4.826		i H_2O; s acid
536	Cadmium sulfite	$CdSO_3$	13477-23-1	192.474	col prisms	dec ≈400			0.05^{20}	sl H_2O
537	Cadmium telluride	CdTe	1306-25-8	240.01	brn-blk cub cry	1042		6.2		i H_2O, dil acid
538	Cadmium tellurite	$CdTeO_3$	15851-44-2	288.01	col monocl cry	695	dec 1050			
539	Cadmium tetrafluoroborate	$Cd(BF_4)_2$	14486-19-2	286.020	col hyg liq			1.6		vs H_2O, EtOH
540	Cadmium titanate	$CdTiO_3$	12014-14-1	208.276	orth cry			6.5		
541	Cadmium tungstate	$CdWO_4$	7790-85-4	360.25	wh monocl cry			8.0		i H_2O, acid; s NH_4OH
542	Calcium	Ca	7440-70-2	40.078	silv-wh metal	842	1484	1.54		reac H_2O; i bz
543	Calcium acetate	$Ca(C_2H_3O_2)_2$	62-54-4	158.167	wh hyg cry	160 dec		1.50		s H_2O; sl EtOH
544	Calcium acetate monohydrate	$Ca(C_2H_3O_2)_2 \cdot H_2O$	5743-26-0	176.182	wh needles or powder	≈150 dec				s H_2O; sl EtOH
545	Calcium acetate dihydrate	$Ca(C_2H_3O_2)_2 \cdot 2H_2O$	14977-17-4	194.196	long col needles					s H_2O
546	Calcium aluminate	$CaAl_2O_4$	12042-68-1	158.039	wh monocl cry	1605		2.98		reac H_2O
547	Calcium aluminate (β form)	$Ca_3Al_2O_6$	12042-78-3	270.193	wh cub cry; refr	1535		3.04		i H_2O
548	Calcium arsenate	$Ca_3(AsO_4)_2$	7778-44-1	398.072	wh powder	dec		3.6	0.0036^{20}	s dil acid
549	Calcium arsenite (1:1)	$CaAsO_3$	52740-16-6	162.998	wh powder					sl H_2O; s acid
550	Calcium borate hexahydrate	$CaB_4O_7 \cdot 6H_2O$	13701-64-9*	303.409	wh cry pow	1162 (anh)				
551	Calcium boride	CaB_6	12007-99-7	104.944	refrac solid	2235		2.49		
552	Calcium bromate	$Ca(BrO_3)_2$	10102-75-7	295.882	wh pow	180				
553	Calcium bromate monohydrate	$Ca(BrO_3)_2 \cdot H_2O$	10102-75-7*	313.898	wh monocl cry	dec 180		3.33		vs H_2O
554	Calcium bromide	$CaBr_2$	7789-41-5	199.886	rhom cry; hyg	742	1815	3.38	156^{25}	s EtOH, ace
555	Calcium bromide dihydrate	$CaBr_2 \cdot 2H_2O$	22208-73-7	235.917	wh cry pow					vs H_2O
556	Calcium bromide hexahydrate	$CaBr_2 \cdot 6H_2O$	13477-28-6	307.977	wh hyg powder	38 dec		2.29	156^{25}	
557	Calcium carbide	CaC_2	75-20-7	64.099	gray-blk orth cry	2300		2.22		reac H_2O
558	Calcium carbonate (aragonite)	$CaCO_3$	471-34-1	100.087	wh orth cry or powder	825 dec		2.930	0.00066^{20}	s dil acid
559	Calcium carbonate (calcite)	$CaCO_3$	471-34-1	100.087	wh hex cry or powder	1330		2.710	0.00066^{20}	s dil acid
560	Calcium carbonate (vaterite)	$CaCO_3$	471-34-1	100.087	col hex cry			2.653	0.0011^{25}	s dil acid
561	Calcium chlorate	$Ca(ClO_3)_2$	10137-74-3	206.980	wh cry	340			197^{25}	
562	Calcium chlorate dihydrate	$Ca(ClO_3)_2 \cdot 2H_2O$	10035-05-9	243.011	wh monocl cry; hyg	100 dec		2.711	197^{25}	s EtOH
563	Calcium chloride	$CaCl_2$	10043-52-4	110.984	wh cub cry or powder; hyg	775	1935	2.15	81.3^{25}	vs EtOH
564	Calcium chloride monohydrate	$CaCl_2 \cdot H_2O$	13477-29-7	128.999	wh hyg cry	260 dec		2.24	81.3^{25}	s EtOH
565	Calcium chloride dihydrate	$CaCl_2 \cdot 2H_2O$	10035-04-8	147.015	hyg flakes or powder	175 dec		1.85	81.3^{25}	vs EtOH
566	Calcium chloride tetrahydrate	$CaCl_2 \cdot 4H_2O$	25094-02-4	183.046	col tricl cry			1.83		
567	Calcium chloride hexahydrate	$CaCl_2 \cdot 6H_2O$	7774-34-7	219.075	wh hex cry; hyg	30 dec		1.71	81.3^{25}	
568	Calcium chlorite	$Ca(ClO_2)_2$	14674-72-7	174.982	wh cub cry			2.71		reac H_2O
569	Calcium chromate	$CaCrO_4$	13765-19-0	156.072	yel cry	1000 dec				sl H_2O; i EtOH, ace
570	Calcium chromate dihydrate	$CaCrO_4 \cdot 2H_2O$	10060-08-9	192.102	yel orth cry	dec 200		2.50	13.2^{20}	s dil acids
571	Calcium citrate tetrahydrate	$Ca_3(C_6H_5O_7)_2 \cdot 4H_2O$	5785-44-4	570.494	wh needles or pow	dec 100			0.096^{23}	i eth
572	Calcium cyanamide	$CaCN_2$	156-62-7	80.102	col hex cry	≈1340	subl	2.29		reac H_2O
573	Calcium cyanide	$Ca(CN)_2$	592-01-8	92.112	wh rhomb cry; hyg					s H_2O, EtOH

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
574	Calcium dichromate trihydrate	$CaCr_2O_7 \cdot 3H_2O$	14307-33-6*	310.112	red-oran cry	100 dec		2.37		vs H_2O; reac EtOH; i eth, ctc
575	Calcium dihydrogen phosphate monohydrate	$Ca(H_2PO_4)_2 \cdot H_2O$	10031-30-8	252.068	col tricl plates	100 dec		2.220		sl H_2O; s dil acid
576	Calcium 2-ethylhexanoate	$Ca(C_8H_{15}O_2)_2$	136-51-6	326.485	powder					
577	Calcium ferrocyanide dodecahydrate	$Ca_2Fe(CN)_6 \cdot 12H_2O$		508.289	yel tricl cry	dec		1.68	87^{25}	vs H_2O; i EtOH
578	Calcium fluoride	CaF_2	7789-75-5	78.075	wh cub cry or powder	1418	2500	3.18	0.0016^{25}	sl acid
579	Calcium fluorophosphate	$Ca_5(PO_4)_3F$	12015-73-5	504.302	col hex cry	1650		3.201		i H_2O
580	Calcium fluorophosphate dihydrate	$CaPO_3F \cdot 2H_2O$	37809-19-1	174.079	col monocl cry				0.42^{27}	i os
581	Calcium formate	$Ca(CHO_2)_2$	544-17-2	130.113	orth cry	300 dec		2.02	16.6^{20}	i EtOH
582	Calcium hexaborate pentahydrate	$2CaO \cdot 3B_2O_3 \cdot 5H_2O$	12291-65-5	411.091	col monocl cry	dec 375 (exp)		2.42	1^{25}	sl acid
583	Calcium hexafluoro-2,4-pentanedioate	$Ca(CF_3COCHCOCF_3)_2$	121012-90-6	454.180	powder	135				
584	Calcium hexafluorosilicate dihydrate	$CaSiF_6 \cdot 2H_2O$	16925-39-6	218.185	col tetr cry			2.25	0.52^{20}	i ace; reac hot H_2O
585	Calcium hydride	CaH_2	7789-78-8	42.094	gray orth cry or powder	1000		1.7		reac H_2O, EtOH
586	Calcium hydrogen phosphate	$CaHPO_4$	7757-93-9	136.057	wh tricl cry	dec		2.92	0.02^{25}	i EtOH
587	Calcium hydrogen phosphate dihydrate	$CaHPO_4 \cdot 2H_2O$	7789-77-7	172.088	monocl cry	≈100 dec		2.31	0.02^{25}	i EtOH; s dil acid
588	Calcium hydrogen sulfite	$CaH_2(SO_3)_2$	13780-03-5	202.220				1.06		s H_2O
589	Calcium hydrosulfide hexahydrate	$Ca(HS)_2 \cdot 6H_2O$		214.315	col cry	dec				s H_2O, EtOH
590	Calcium hydroxide	$Ca(OH)_2$	1305-62-0	74.093	soft hex cry			≈2.2	0.160^{20}	s acid
591	Calcium hydroxide phosphate	$Ca_5(OH)(PO_4)_3$	12167-74-7	502.311	col hex cry	dec >900		3.155		i H_2O
592	Calcium hypochlorite	$Ca(OCl)_2$	7778-54-3	142.983	powder	100		2.350		
593	Calcium hypophosphite	$Ca(H_2PO_2)_2$	7789-79-9	170.055	wh monocl cry	300 dec				s H_2O; i EtOH
594	Calcium iodate	$Ca(IO_3)_2$	7789-80-2	389.883	wh monocl cry			4.52	0.306^{25}	s HNO_3; i EtOH
595	Calcium iodide	CaI_2	10102-68-8	293.887	hyg hex cry	783	1100	3.96	215^{25}	s MeOH, EtOH, ace; i eth
596	Calcium iodide hexahydrate	$CaI_2 \cdot 6H_2O$	71626-98-7	401.978	wh hex needles or powder	42 dec		2.55	215^{25}	vs EtOH
597	Calcium metaborate	$Ca(BO_2)_2$	13701-64-9	125.698	powder				0.13^{20}	
598	Calcium metasilicate	$CaSiO_3$	1344-95-2	116.162	wh monocl cry	1540		2.92		i H_2O
599	Calcium molybdate	$CaMoO_4$	7789-82-4	200.02	wh tetr cry	1520		4.35	0.0011^{20}	i EtOH; s conc acid
600	Calcium nitrate	$Ca(NO_3)_2$	10124-37-5	164.087	wh cub cry; hyg	561		2.5	144^{25}	s EtOH, MeOH, ace
601	Calcium nitrate tetrahydrate	$Ca(NO_3)_2 \cdot 4H_2O$	13477-34-4	236.149	wh cry	≈40 dec		1.82	144^{25}	s EtOH, ace
602	Calcium nitride	Ca_3N_2	12013-82-0	148.247	red-brn cub cry	1195		2.67		s H_2O, acid; i EtOH
603	Calcium nitrite	$Ca(NO_2)_2$	13780-06-8	132.089	wh-yel hex cry; hyg	392		2.23	94.6^{25}	sl EtOH
604	Calcium nitrite monohydrate	$Ca(NO_2)_2 \cdot H_2O$	10031-34-2	150.104	col or yel cry	dec 100				vs H_2O; sl EtOH
605	Calcium oleate	$Ca(C_{18}H_{33}O_2)_2$	142-17-6	602.985	pale yel solid	dec 140			0.04^{25}	sl H_2O; s bz; i EtOH, ace, eth
606	Calcium oxalate	CaC_2O_4	563-72-4	128.097	wh cry powder			2.2	0.00061^{20}	
607	Calcium oxalate monohydrate	$CaC_2O_4 \cdot H_2O$	5794-28-5	146.112	cub cry	200 dec		2.2	0.00061^{20}	s dil acid
608	Calcium oxide	CaO	1305-78-8	56.077	gray-wh cub cry	2613		3.34		reac H_2O; s acid
609	Calcium oxide silicate	Ca_3OSiO_4	12168-85-3	228.317	refrac solid	2150				
610	Calcium palmitate	$Ca(C_{16}H_{31}O_2)_2$	542-42-7	550.910	wh-yel pow	dec 155				i H_2O, EtOH, eth, ace; sl bz
611	Calcium perborate heptahydrate	$Ca(BO_3)_2 \cdot 7H_2O$		283.803	gray-wh pow					s H_2O, acid
612	Calcium 2,4-pentanedioate	$Ca(CH_3COCHCOCH_3)_2$	19372-44-2	238.294	cry	dec 175				
613	Calcium perchlorate	$Ca(ClO_4)_2$	13477-36-6	238.979	wh cry	270 dec		2.65	188^{25}	s EtOH, MeOH
614	Calcium perchlorate tetrahydrate	$Ca(ClO_4)_2 \cdot 4H_2O$	15627-86-8	311.041	wh cry					vs H_2O
615	Calcium permanganate	$Ca(MnO_4)_2$	10118-76-0	277.949	purp hyg cry			2.4	331^{20}	reac EtOH
616	Calcium peroxide	CaO_2	1305-79-9	72.077	wh-yel tetr cry; hyg	≈200 dec		2.9		sl H_2O; s acid
617	Calcium phosphate	$Ca_3(PO_4)_2$	7758-87-4	310.177	wh amorp powder	1670		3.14	0.00012^{20}	i EtOH; s dil acid
618	Calcium phosphide	Ca_3P_2	1305-99-3	182.182	red-brn hyg cry	≈1600		2.51		reac H_2O; i EtOH, eth
619	Calcium phosphonate monohydrate	$CaHPO_3 \cdot H_2O$	25232-60-4	138.073	col monocl cry	dec 150				sl H_2O; i EtOH
620	Calcium plumbate	Ca_2PbO_4	12013-69-3	351.4	oran-brn orth cry	dec		5.71		i H_2O; s acid
621	Calcium propanoate	$Ca(C_3H_5O_2)_2$	4075-81-4	186.219	monocl cry, powder					s H_2O; sl MeOH, EtOH; i ace, bz
622	Calcium pyrophosphate	$Ca_2P_2O_7$	7790-76-3	254.099	wh powder	1353		3.09		i H_2O; s dil acid
623	Calcium selenate dihydrate	$CaSeO_4 \cdot 2H_2O$	7790-74-1	219.07	wh monocl cry			2.75	8.3^{18}	
624	Calcium selenide	$CaSe$	1305-84-6	119.04	wh-brn cub cry	1400 dec		3.8		reac H_2O
625	Calcium silicide (CaSi)	$CaSi$	12013-55-7	68.164	orth cry	1324		2.39		

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
626	Calcium silicide ($CaSi_2$)	$CaSi_2$	12013-56-8	96.249	gray hex cry	1040		2.50		i cold H_2O; reac hot H_2O; s acid
627	Calcium stannate trihydrate	$CaSnO_3 \cdot 3H_2O$	12013-46-6*	260.832	wh cry pow	dec ≈350				i H_2O
628	Calcium stearate	$Ca(C_{18}H_{35}O_2)_2$	1592-23-0	607.017	granular powder	180				i H_2O, EtOH
629	Calcium succinate trihydrate	$CaC_4H_4O_4 \cdot 3H_2O$	140-99-8	210.196	needles					sl H_2O; s dil acid; i EtOH
630	Calcium sulfate	$CaSO_4$	7778-18-9	136.141	orth cry	1460		2.96	0.205^{25}	
631	Calcium sulfate hemihydrate	$CaSO_4 \cdot 0.5H_2O$	10034-76-1	145.149	wh powder				0.205^{25}	
632	Calcium sulfate dihydrate	$CaSO_4 \cdot 2H_2O$	10101-41-4	172.171	monocl cry or powder	150 dec		2.32	0.205^{20}	i os
633	Calcium sulfide	CaS	20548-54-3	72.143	wh-yel cub cry; hyg	2524		2.59		sl H_2O; i EtOH
634	Calcium sulfite dihydrate	$CaSO_3 \cdot 2H_2O$	10257-55-3	156.172	wh powder				0.0070^{25}	sl EtOH; s acid
635	Calcium tartrate tetrahydrate	$CaC_4H_4O_6 \cdot 4H_2O$	3164-34-9*	260.210	wh pow				0.04^{10}	s dil acid; sl EtOH
636	Calcium telluride	CaTe	12013-57-9	167.68	wh cub cry	1600 dec		4.87		
637	Calcium tetrahydroaluminate	$Ca(AlH_4)_2$	16941-10-9	102.105	gray powder; flam					reac H_2O; s thf; i eth, bz
638	Calcium thiocyanate tetrahydrate	$Ca(SCN)_2 \cdot 4H_2O$	2092-16-2	228.304	hyg cry	160 dec				vs H_2O; s EtOH, ace
639	Calcium thiosulfate hexahydrate	$CaS_2O_3 \cdot 6H_2O$	10124-41-1	260.298	tricl cry	45 dec		1.87		s H_2O; i EtOH
640	Calcium titanate	$CaTiO_3$	12049-50-2	135.943	cub cry	1980		3.98		
641	Calcium tungstate	$CaWO_4$	7790-75-2	287.92	wh tetr cry	1620		6.06	0.2^{18}	s hot acid
642	Calcium zirconate	$CaZrO_3$	12013-47-7	179.300	powder	2550				
643	Californium	Cf	7440-71-3	251	hex or cub metal	900		15.1		
644	Carbon (diamond)	C	7782-40-3	12.011	col cub cry	4440 (12.4 GPa)		3.513		i H_2O
645	Carbon (graphite)	C	7782-42-5	12.011	soft blk hex cry	4489 tp (10.3 MPa)	3825 sp	2.2		i H_2O
646	Carbon black	C	1333-86-4	12.011	fine blk pow					i H_2O
647	Carbon (fullerene-C_{60})	C_{60}	99685-96-8	720.642	yel needles or plates	>280				s os
648	Carbon (fullerene-C_{70})	C_{70}	115383-22-7	840.749	red-brn solid	>280				s bz, tol
649	Fullerene fluoride	$C_{60}F_{60}$	134929-59-2	1860.546	col plates	287				vs ace; s thf; i chl
650	Carbon monoxide	CO	630-08-0	28.010	col gas	-205.02	-191.5	1.145 g/L		sl H_2O; s chl, EtOH
651	Carbon dioxide	CO_2	124-38-9	44.010	col gas	-56.558 tp	-78.464 sp	1.799 g/L		s H_2O
652	Carbon suboxide	C_3O_2	504-64-3	68.031	col gas	-112.5	6.8	2.781 g/L		reac H_2O
653	Carbon disulfide	CS_2	75-15-0	76.141	col or yel liq	-112.1	46	1.2632^{20}		i H_2O; vs EtOH, bz, os
654	Carbon subsulfide	C_3S_2	627-34-9	100.162	red liq	-1	90 dec	1.27		reac H_2O
655	Carbon diselenide	CSe_2	506-80-9	169.93	yel liq	-43.7	125.5	2.6823^{20}		i H_2O; vs ctc, tol
656	Carbon oxysulfide	COS	463-58-1	60.075	col gas	-138.8	-50	2.456 g/L		s H_2O, EtOH
657	Carbon oxyselenide	COSe	1603-84-5	106.97	col gas; unstab	-124.4	-21.7	4.372 g/L		reac H_2O
658	Carbon sulfide selenide	CSSe	5951-19-9	123.04	yel liq	-85	84.5	1.99		i H_2O
659	Carbon sulfide telluride	CSTe	10340-06-4	171.68	red-yel liq; unstab	-54	20 dec			reac H_2O
660	Carbonyl bromide	$COBr_2$	593-95-3	187.818	col liq		64.5	2.5		reac H_2O
661	Carbonyl chloride	$COCl_2$	75-44-5	98.916	col gas	-127.78	8	4.043 g/L		sl H_2O; s bz, tol
662	Carbonyl fluoride	COF_2	353-50-4	66.007	col gas	-111.26	-84.57	2.698 g/L		reac H_2O
663	Cyanogen	C_2N_2	460-19-5	52.034	col gas	-27.83	-21.1	2.127 g/L		sl H_2O, eth; s EtOH
664	Cyanogen azide	N_3CN	764-05-6	68.038	col oily liq	exp				
665	Cyanogen bromide	BrCN	506-68-3	105.922	wh hyg needles	52	61.5	2.015		s H_2O, EtOH, eth
666	Cyanogen chloride	ClCN	506-77-4	61.471	col vol liq or gas	-6.55	13	2.513 g/L		s H_2O, EtOH, eth
667	Cyanogen fluoride	FCN	1495-50-7	45.016	col gas	-82	-46	1.840 g/L		
668	Cyanogen iodide	ICN	506-78-5	152.922	col needles	146.7		2.84		s H_2O, EtOH, eth
669	Cerium	Ce	7440-45-1	140.116	silv metal; cub or hex	799	3443	6.770		s dil acid
670	Cerium boride	CeB_6	12008-02-5	204.982	blue refrac solid; hex	2550		4.87		i H_2O, HCl
671	Cerium carbide	CeC_2	12012-32-7	164.137	red hex cry	2250		5.47		reac H_2O
672	Cerium carbide	Ce_2C_3	12115-63-8	316.264	yel-brn cub cry	1505		6.9		
673	Cerium nitride	CeN	25764-08-3	154.123	refrac cub cry	2557		7.89		
674	Cerium silicide	$CeSi_2$	12014-85-6	196.287	tetr cry	1420		5.31		i H_2O
675	Cerium(II) hydride	CeH_2	13569-50-1	142.132	cub cry			5.45		reac H_2O
676	Cerium(II) iodide	CeI_2	19139-47-0	393.925	bronze cry	808				
677	Cerium(II) sulfide	CeS	12014-82-3	172.181	yel cub cry	2445		5.9		
678	Cerium(III) acetate sesqihydrate	$Ce(C_2H_3O_2)_3 \cdot 1.5H_2O$	17829-82-2		col cry	dec 115			26^{15}	s H_2O
679	Cerium(III) ammonium nitrate tetrahydrate	$(NH_4)_2Ce(NO_3)_5 \cdot 4H_2O$	13083-04-0	558.279	col monocl cry	74				vs H_2O
680	Cerium(III) ammonium sulfate tetrahydrate	$NH_4Ce(SO_4)_2 \cdot 4H_2O$	21995-38-0*	422.341	monocl cry					s H_2O
681	Cerium(III) bromide	$CeBr_3$	14457-87-5	379.828	wh hex cry; hyg	732	1457			s H_2O

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682	Cerium(III) bromide heptahydrate	$CeBr_3 \cdot 7H_2O$	7789-56-2	505.935	col hyg needles	732				s H_2O, EtOH
683	Cerium(III) carbonate	$Ce_2(CO_3)_3$	537-01-9	460.259	wh pow	dec 500				i H_2O; s acid
684	Cerium(III) carbonate pentahydrate	$Ce_2(CO_3)_3 \cdot 5H_2O$	72520-94-6	550.335	wh powder					i H_2O; s dil acid
685	Cerium(III) chloride	$CeCl_3$	7790-86-5	246.475	wh hex cry	807		3.97		s H_2O, EtOH
686	Cerium(III) chloride heptahydrate	$CeCl_3 \cdot 7H_2O$	18618-55-8	372.582	yel orth cry; hyg	90 dec				vs H_2O, EtOH
687	Cerium(III) fluoride	CeF_3	7758-88-5	197.111	wh hex cry; hyg	1430	2180	6.157		i H_2O
688	Cerium(III) hydride	CeH_3	13864-02-3	143.140	blk pow or blue-blk cry	dec (flam)				reac H_2O
689	Cerium(III) hydroxide	$Ce(OH)_3$	15785-09-8	191.138	wh solid					i H_2O; s acid
690	Cerium(III) iodide	CeI_3	7790-87-6	520.829	yel orth cry; hyg	760				s H_2O
691	Cerium(III) iodide nonahydrate	$CeI_3 \cdot 9H_2O$	7790-87-6*	682.967	wh-red cry					vs H_2O; s EtOH
692	Cerium(III) nitrate hexahydrate	$Ce(NO_3)_3 \cdot 6H_2O$	10108-73-3*	434.222	col-red cry	150 dec			176^{25}	s ace
693	Cerium(III) oxalate nonahydrate	$Ce(C_2O_4)_3 \cdot 9H_2O$	13266-83-6	706.426	wh pow	dec				i H_2O, EtOH; s acid
694	Cerium(III) oxide	Ce_2O_3	1345-13-7	328.230	yel-grn cub cry	2210	3730	6.2		i H_2O; s acid
695	Cerium(III) 2,4-pentanedioate trihydrate	$Ce(CH_3COCHCOCH_3)_3 \cdot 3H_2O$	15653-01-7	491.486	yel hyg cry	≈150				vs EtOH
696	Cerium(III) perchlorate hexahydrate	$Ce(ClO_4)_3 \cdot 6H_2O$	36907-38-7	546.559	hyg col cry	dec 200				s H_2O, EtOH
697	Cerium(III) selenate	$Ce_2(SeO_4)_3$		709.11	rhom cry			4.46		s H_2O
698	Cerium(III) sulfate	$Ce_2(SO_4)_3$	13454-94-9	568.420	col hyg cry	920 dec				s H_2O
699	Cerium(III) sulfate octahydrate	$Ce_2(SO_4)_3 \cdot 8H_2O$	13454-94-9	712.542	wh orth cry	≈250 dec		2.87		s H_2O
700	Cerium(III) sulfide	Ce_2S_3	12014-93-6	376.427	red cub cry	2450		5.02		i H_2O
701	Cerium(III) tungstate	$Ce_2(WO_4)_3$	13454-74-5	1023.75	yel tetr cry	1089		6.77		i H_2O
702	Cerium(IV) ammonium nitrate	$(NH_4)_2Ce(NO_3)_6$	16774-21-3	548.223	red-oran cry					vs H_2O
703	Cerium(IV) ammonium nitrate dihydrate	$(NH_4)_4Ce(SO_4)_4 \cdot 2H_2O$	10378-47-9	632.551	cry pow	dec 450				
704	Cerium(IV) fluoride	CeF_4	10060-10-3	216.110	wh hyg powder	≈600 dec		4.77		i H_2O
705	Cerium(IV) hydroxide	$Ce(OH)_4$	12014-56-1	208.146	yel-wh pow					i H_2O; s conc acid
706	Cerium(IV) oxide	CeO_2	1306-38-3	172.115	wh-yel powder; cub	2480		7.216		i H_2O, dil acid
707	Cerium(IV) sulfate tetrahydrate	$Ce(SO_4)_2 \cdot 4H_2O$	10294-42-5	404.303	yel-oran orth cry	180 dec		3.91	9.66^{20}	
708	Cesium	Cs	7440-46-2	132.905	silv-wh metal	28.5	671	1.93		reac H_2O
709	Cesium acetate	$CsC_2H_3O_2$	3396-11-0	191.949	hyg lumps	194			10^{11}	
710	Cesium aluminum sulfate dodecahydrate	$CsAl(SO_4)_2 \cdot 12H_2O$	7784-17-0	568.196	col cub cry	117 dec		1.97		s H_2O; i EtOH
711	Cesium amide	$CsNH_2$	22205-57-8	148.928	wh tetr cry			3.70		
712	Cesium azide	CsN_3	22750-57-8	174.925	hyg tetr cry; exp	326		≈3.5	22^{40}	
713	Cesium bromate	$CsBrO_3$	13454-75-6	260.807	col hex cry			4.11	3.83^{25}	
714	Cesium bromide	CsBr	7787-69-1	212.809	wh cub cry; hyg	636	≈1300	4.43	123^{25}	s EtOH; i ace
715	Cesium carbonate	Cs_2CO_3	534-17-8	325.820	wh monocl cry; hyg	793		4.24	261^{15}	s EtOH, eth
716	Cesium chlorate	$CsClO_3$	13763-67-2	216.356	col hex cry	342		3.57	7.78^{25}	sl H_2O
717	Cesium chloride	CsCl	7647-17-8	168.358	wh cub cry; hyg	646	1297	3.988	191^{25}	s EtOH
718	Cesium chromate(IV)	Cs_4CrO_4	56320-90-2	647.616	yel hex cry	982		4.24		vs H_2O
719	Cesium cyanide	CsCN	21159-32-0	158.923	wh cub cry; hyg	350		3.34		vs H_2O
720	Cesium dibromoiodate	$CsIBr_2$	18278-82-5	419.617	dark oran cry	dec				s H_2O
721	Cesium fluoride	CsF	13400-13-0	151.903	wh cub cry; hyg	703		4.64	573^{25}	s MeOH; i diox, py
722	Cesium fluoroborate	$CsBF_4$	18909-69-8	219.710	wh orth cry	555 dec		3.2	1.6^{17}	sl H_2O
723	Cesium formate	$CsCHO_2$	3495-36-1	177.923	wh cry			1.017		vs H_2O
724	Cesium hexafluorogermanate	Cs_2GeF_6		452.44	wh cry	≈675		4.10		sl cold H_2O; s hot H_2O
725	Cesium hydride	CsH	58724-12-2	133.913	wh cub cry; flam	528		3.42		reac H_2O
726	Cesium hydrogen carbonate	$CsHCO_3$	15519-28-5	193.922	rhom cry	175 dec			209^{15}	s EtOH
727	Cesium hydrogen fluoride	$CsHF_2$	12280-52-3	171.910	tetr cry	170		3.86		
728	Cesium hydrogen sulfate	$CsHSO_4$	7789-16-4	229.976	col rhom prisms	dec		3.352		s H_2O
729	Cesium hydroxide	CsOH	21351-79-1	149.912	wh-yel hyg cry	342.3		3.68	300^{30}	s EtOH
730	Cesium iodate	$CsIO_3$	13454-81-4	307.807	wh monocl cry			4.85	2.6^{25}	
731	Cesium iodide	CsI	7789-17-5	259.809	col cub cry; hyg	632	≈1280	4.51	84.8^{25}	s EtOH, MeOH, ace
732	Cesium metaborate	$CsBO_2$	92141-86-1	175.715	cub cry	732		≈3.7		
733	Cesium molybdate	Cs_2MoO_4	13597-64-3	425.75	wh cry	956.3			67^{18}	s H_2O
734	Cesium nitrate	$CsNO_3$	7789-18-6	194.910	wh hex or cub cry	409		3.66	27.9^{25}	s ace; sl EtOH
735	Cesium nitrite	$CsNO_2$	13454-83-6	178.911	yel cry	406				s H_2O
736	Cesium oxide	Cs_2O	20281-00-9	281.810	yel-oran hex cry	495		4.65		vs H_2O
737	Cesium superoxide	CsO_2	12018-61-0	164.904	yel tetr cry	432		3.77		reac H_2O
738	Cesium trioxide	Cs_2O_3	12134-22-4	313.809	brn cry	≈400		4.25		reac H_2O
739	Cesium perchlorate	$CsClO_4$	13454-84-7	232.356	wh orth cry; hyg	≈600 dec		3.327	2.00^{25}	
740	Cesium periodate	$CsIO_4$	13478-04-1	323.807	wh rhom prisms			4.26	2.2^{15}	
741	Cesium sulfate	Cs_2SO_4	10294-54-9	361.874	wh orth cry or hex prisms; hyg	1005		4.24	182^{25}	i EtOH, ace, py

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
742	Cesium sulfide	Cs_2S	12214-16-3	297.876	yel orth hyg cry	520				vs H_2O
743	Cesium trifluoroacetate	$Cs(C_2F_3O_2)$	21907-50-6	245.920	hyg solid	115				vs H_2O
744	Chlorine	Cl_2	7782-50-5	70.906	grn-yel gas	-101.5	-34.04	2.898 g/L		sl H_2O
745	Hypochlorous acid	HOCl	7790-92-3	52.460	grn-yel; stable only in aq soln					s H_2O
746	Chloric acid	$HClO_3$	7790-93-4	84.459	exists only in aq soln					vs H_2O
747	Perchloric acid	$HClO_4$	7601-90-3	100.459	col hyg liq	-112	≈90 dec	1.77		s H_2O
748	Chlorine monoxide	Cl_2O	7791-21-1	86.905	yel-brn gas	-120.6	2.2	3.552 g/L		vs H_2O
749	Chlorine dioxide	ClO_2	10049-04-4	67.452	oran-grn gas	-59	11	2.757 g/L		sl H_2O
750	Dichlorine trioxide	Cl_2O_3	17496-59-2	118.904	dark brn solid	exp <25				
751	Dichlorine hexoxide	Cl_2O_6	12442-63-6	166.902	red liq	3.5	≈200			reac H_2O
752	Dichlorine heptoxide	Cl_2O_7	10294-48-1	182.902	col oily liq; exp	-91.5	82	1.9		reac H_2O
753	Chlorine fluoride	ClF	7790-89-8	54.451	col gas	-155.6	-101.1	2.226 g/L		reac H_2O
754	Chlorine trifluoride	ClF_3	7790-91-2	92.448	gas	-76.34	11.75	3.779 g/L		reac H_2O
755	Chlorine pentafluoride	ClF_5	13637-63-3	130.445	col gas	-103	-13.1	5.332 g/L		
756	Chlorosyl trifluoride	$ClOF_3$	30708-80-6	108.447	col liq	-42	27			reac H_2O
757	Chloryl fluoride	ClO_2F	13637-83-7	86.450	col gas	-115	-6	3.534 g/L		reac H_2O
758	Chloryl trifluoride	ClO_2F_3	38680-84-1	124.447	col gas	-81.2	-21.6	5.087 g/L		reac H_2O
759	Perchloryl fluoride	ClO_3F	7616-94-6	102.449	col gas	-147	-46.75	4.187 g/L		
760	Chlorine perchlorate	$ClOClO_3$	27218-16-2	134.904	unstab yel liq	-117	≈45 dec	1.81^0		
761	Chromium	Cr	7440-47-3	51.996	blue-wh metal; cub	1907	2671	7.15		reac dil acid
762	Chromic acid	H_2CrO_4	7738-94-5	118.010	aq soln only					s H_2O
763	Chromium antimonide	CrSb	12053-12-2	173.756	hex cry	1110		7.11		
764	Chromium arsenide	Cr_2As	12254-85-2	178.914	tetr cry			7.04		
765	Chromium boride (CrB)	CrB	12006-79-0	62.807	refrac orth cry	2100		6.1		
766	Chromium boride (CrB_2)	CrB_2	12007-16-8	73.618	refrac solid; hex	2200		5.22		
767	Chromium boride (Cr_2B)	Cr_2B	12006-80-3	114.803	refrac solid	1875				
768	Chromium boride (Cr_5B_3)	Cr_5B_3	12007-38-4	292.414	tetr cry	1900		6.10		
769	Chromium carbide	Cr_3C_2	12012-35-0	180.009	gray orth cry	1895		6.68		
770	Chromium carbonyl	$Cr(CO)_6$	13007-92-6	220.056	col orth cry	130 dec	subl	1.77		i H_2O, EtOH; s eth, chl
771	Chromium nitride (CrN)	CrN	24094-93-7	66.003	gray cub cry	1080 dec		5.9		
772	Chromium nitride (Cr_2N)	Cr_2N	12053-27-9	117.999	hex cry	1650		6.8		
773	Chromium phosphide	CrP	26342-61-0	82.970	orth cry			5.25		
774	Chromium selenide	CrSe	12053-13-3	130.96	hex cry	≈1500		6.1		
775	Chromium silicide ($CrSi_2$)	$CrSi_2$	12018-09-6	108.167	gray hex cry	1490		4.91		
776	Chromium silicide (Cr_3Si)	Cr_3Si	12018-36-9	184.074	cub cry	1770		6.4		
777	Chromium(II) acetate monohydrate	$Cr(C_2H_3O_2)_2 \cdot H_2O$	628-52-4*	188.100	red monocl cry			1.79		sl H_2O
778	Chromium(II) bromide	$CrBr_2$	10049-25-9	211.804	wh monocl cry; aq soln blue	842		4.236		s H_2O, EtOH
779	Chromium(II) chloride	$CrCl_2$	10049-05-5	122.902	wh hyg needles; aq soln blue	824	1120	2.88		s H_2O
780	Chromium(II) chloride tetrahydrate	$Cr(H_2O)_4Cl_2 \cdot 4H_2O$	13931-94-7	267.024	blue hyg cry	51 dec				s H_2O
781	Chromium(II) fluoride	CrF_2	10049-10-2	89.993	blue-grn monocl cry	894		3.79		sl H_2O; i EtOH
782	Chromium(II) formate monohydrate	$Cr(CHOO)_2 \cdot H_2O$	4493-37-2	160.046	red needles					s H_2O
783	Chromium(II) iodide	CrI_2	13478-28-9	305.805	red-brn cry; hyg	867		5.1		s H_2O
784	Chromium(II) oxalate monohydrate	$CrC_2O_4 \cdot H_2O$	814-90-4*	158.030	yel-grn powder			2.468		sl H_2O
785	Chromium(II) sulfate pentahydrate	$CrSO_4 \cdot 5H_2O$	13825-86-0	238.135	blue cry				21^0	s dil acid; sl EtOH; i ace
786	Chromium(II,III) oxide	Cr_3O_4	12018-34-7	219.986	cub cry			6.1		
787	Chromium(III) acetate	$Cr(C_2H_3O_2)_3$	1066-30-4	229.127	blue-grn pwd					sl H_2O
788	Chromium(III) acetate monohydrate	$Cr(C_2H_3O_2)_3 \cdot H_2O$	25013-82-5	247.143	gray-grn pow					sl H_2O; i EtOH
789	Chromium(III) acetate hexahydrate	$Cr(C_2H_3O_2)_3 \cdot 6H_2O$	1066-30-4*	337.220	blue needles					s H_2O
790	Chromium(III) acetate hydroxide	$Cr(C_2H_3O_2)_2(OH)$	39430-51-8	187.092	viol cry pow					vs H_2O
791	Chromium(III) bromide	$CrBr_3$	10031-25-1	291.708	dark grn hex cry	812		4.68		s hot H_2O, bz
792	Chromium(III) bromide hexahydrate (α)	$CrBr_3(H_2O)_4 \cdot 2H_2O$	18721-05-6	399.799	grn hyg cry					s H_2O, EtOH
793	Chromium(III) bromide hexahydrate (β)	$Cr(H_2O)_6Br_3$	10031-25-1*	399.799	viol hyg cry					s H_2O; i EtOH, eth
794	Chromium(III) chloride	$CrCl_3$	10025-73-7	158.355	red-viol cry	1152	1300 dec	2.76		sl H_2O
795	Chromium(III) chloride hexahydrate	$[CrCl_2(H_2O)_4]Cl \cdot 2H_2O$	10060-12-5	266.446	grn monocl cry; hyg					s H_2O, EtOH; sl ace; i eth
796	Chromium(III) fluoride	CrF_3	7788-97-8	108.991	grn needles	1425		3.8		i H_2O, EtOH
797	Chromium(III) fluoride trihydrate	$CrF_3 \cdot 3H_2O$	16671-27-5	163.037	grn hex cry			2.2		sl H_2O
798	Chromium(III) hydroxide sulfate	$Cr(OH)SO_4$	12336-95-7	165.066	grn cry					
799	Chromium(III) fluoride nonahydrate	$Cr(H_2O)_6F_3 \cdot 3H_2O$	102430-09-1	271.129	rhom viol cry					sl H_2O

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
800	Chromium(III) hydroxide trihydrate	$Cr(OH)_3 \cdot 3H_2O$	1308-14-1	157.063	blue-grn powder					i H_2O; s acid
801	Chromium(III) iodide	CrI_3	13569-75-0	432.709	dark grn hex cry	500 dec		5.32		sl H_2O
802	Chromium(III) nitrate	$Cr(NO_3)_3$	13548-38-4	238.011	grn hyg powder	>60 dec				vs H_2O
803	Chromium(III) nitrate nonahydrate	$Cr(NO_3)_3 \cdot 9H_2O$	7789-02-8	400.148	grn-blk monocl cry	66.3	>100 dec	1.80		vs H_2O
804	Chromium(III) oxide	Cr_2O_3	1308-38-9	151.990	grn hex cry	2320	≈3000	5.22		i H_2O, EtOH; sl acid, alk
805	Chromium(III) 2,4-pentanedioate	$Cr(CH_3COCHCOCH_3)_3$	21679-31-2	349.320	red monocl cry	208	345	1.34		i H_2O; s bz
806	Chromium(III) perchlorate	$Cr(ClO_4)_3$	27535-70-2	350.348	grn-blue cry				58^{25}	vs H_2O
807	Chromium(III) phosphate	$CrPO_4$	7789-04-0	146.967	blue orth cry	>1800		4.6		i H_2O, acid, aqua regia
808	Chromium(III) phosphate hemiheptahydrate	$CrPO_4 \cdot 3.5H_2O$	84359-31-9	210.021	blue-grn powder			2.15		i H_2O; s acid
809	Chromium(III) phosphate hexahydrate	$CrPO_4 \cdot 6H_2O$	84359-31-9	255.059	viol cry	>500 dec		2.121		i H_2O; s acid, alk
810	Chromium(III) potassium oxalate trihydrate	$K_3Cr(C_2O_4)_3 \cdot 3H_2O$	15275-09-9	487.394	blue-grn monocl cry					s H_2O
811	Chromium(III) potassium sulfate dodecahydrate	$CrK(SO_4)_2 \cdot 12H_2O$	7788-99-0	499.403	viol-blk cub cry	89 dec		1.83		s H_2O; i EtOH
812	Chromium(III) sulfate	$Cr_2(SO_4)_3$	10101-53-8	392.180	red pow	dec >700		3.1	64^{25}	s H_2O; vs acid
813	Chromium(III) sulfate octadecahydrate	$Cr_2(SO_4)_3 \cdot 18H_2O$	10101-53-8*	716.455	viol cry	dec 115		1.7		reac H_2O
814	Chromium(III) sulfide	Cr_2S_3	12018-22-3	200.187	brn-blk hex cry			3.8		
815	Chromium(III) telluride	Cr_2Te_3	12053-39-3	486.79	hex cry	≈1300		7.0		
816	Chromium(IV) chloride	$CrCl_4$	15597-88-3	193.808	gas, stable at HT		>600 dec	7.922 g/L		
817	Chromium(IV) fluoride	CrF_4	10049-11-3	127.990	grn cry	277		2.89		reac H_2O
818	Chromium(IV) oxide	CrO_2	12018-01-8	83.995	brn-blk tetr powder	≈400 dec		4.89		i H_2O; s acid
819	Chromium(V) fluoride	CrF_5	14884-42-5	146.988	red orth cry	34	117			reac H_2O
820	Chromium(V) oxide	Cr_2O_5	12218-36-9	183.989	blk needles	dec 200				
821	Chromium(VI) fluoride	CrF_6	13843-28-2	165.986	yel solid; stable at low temp	-100 dec				
822	Chromium(VI) oxide	CrO_3	1333-82-0	99.994	red orth cry	197	≈250 dec	2.7	169^{25}	
823	Chromium(VI) tetrafluoride oxide	$CrOF_4$	23276-90-6	143.989	dark red solid	55				reac H_2O, ace, dmso
824	Chromium(VI) dichloride dioxide	CrO_2Cl_2	14977-61-8	154.901	red liq	-96.5	117	1.91		reac H_2O; s ctc, chl, bz
825	Chromium(VI) difluoride dioxide	CrO_2F_2	7788-96-7	121.992	red-viol cry	30	subl			reac H_2O
826	Cobalt	Co	7440-48-4	58.933	gray metal; hex or cub	1495	2927	8.86		s dil acid
827	Cobaltocene	$Co(C_5H_5)_2$	1277-43-6	189.119	blk-purp cry	173				
828	Cobalt antimonide	CoSb	12052-42-5	180.693	hex cry	1202		8.8		
829	Cobalt arsenic sulfide	CoAsS	12254-82-9	165.920	silv-wh solid			≈6.1		
830	Cobalt arsenide (CoAs)	CoAs	27016-73-5	133.855	orth cry	1180		8.22		
831	Cobalt arsenide ($CoAs_2$)	$CoAs_2$	12044-42-7	208.776	monocl cry			7.2		
832	Cobalt arsenide ($CoAs_3$)	$CoAs_3$	12256-04-1	283.698	cub cry	942		6.84		
833	Cobalt boride (CoB)	CoB	12006-77-8	69.744	refrac solid	1460		7.25		reac H_2O, HNO_3
834	Cobalt boride (Co_2B)	Co_2B	12045-01-1	128.677	refrac solid	1280		8.1		
835	Cobalt carbonyl	$Co_2(CO)_8$	10210-68-1	341.947	oran cry	51 dec		1.78		i H_2O; s EtOH, eth, CS_2
836	Cobalt disulfide	CoS_2	12013-10-4	123.063	cub cry			4.3		
837	Cobalt dodecacarbonyl	$Co_4(CO)_{12}$	17786-31-1	571.854	blk cry	60 dec		2.09		
838	Cobalt phosphide	Co_2P	12134-02-0	148.840	gray needles	1386		6.4		i H_2O; s HNO_3
839	Cobalt silicide	$CoSi_2$	12017-12-8	115.104	gray cub cry	1326		4.9		s hot HCl
840	Cobalt(II) acetate	$Co(C_2H_3O_2)_2$	71-48-7	177.022	pink cry					vs H_2O; s EtOH
841	Cobalt(II) acetate tetrahydrate	$Co(C_2H_3O_2)_2 \cdot 4H_2O$	6147-53-1	249.082	red monocl cry			1.705		s H_2O, EtOH, dil acid
842	Cobalt(II) aluminate	$CoAl_2O_4$	13820-62-7	176.894	blue cub cry			4.37		i H_2O
843	Cobalt(II) arsenate octahydrate	$Co_3(AsO_4)_2 \cdot 8H_2O$	24719-19-5	598.760	red monocl needles	400 dec	1000 dec	3.0		i H_2O; s dil acid
844	Cobalt(II) bromate hexahydrate	$Co(BrO_3)_2 \cdot 6H_2O$	13476-01-2	422.829	viol cry			≈2.5		vs H_2O
845	Cobalt(II) bromide	$CoBr_2$	7789-43-7	218.741	grn hex cry; hyg	678		4.91	113.2^{20}	s MeOH, EtOH, ace
846	Cobalt(II) bromide hexahydrate	$CoBr_2 \cdot 6H_2O$	13762-12-4	326.832	red hyg cry	47 dec	100 dec	2.46	113.2	
847	Cobalt(II) carbonate	$CoCO_3$	513-79-1	118.942	pink rhomb cry	dec 280		4.2	0.00014^{20}	i EtOH
848	Cobalt(II) basic carbonate	$2CoCO_3 \cdot 3Co(OH)_2 \cdot H_2O$	7542-09-8	534.743	red-viol cry	dec				i H_2O; s acid
849	Cobalt(II) chlorate hexahydrate	$Co(ClO_3)_2 \cdot 6H_2O$		333.927	dark red hyg cry	dec 61				s H_2O
850	Cobalt(II) chloride	$CoCl_2$	7646-79-9	129.839	blue hyg leaflets	737	1049	3.36	56.2^{25}	s EtOH, eth, ace, py
851	Cobalt(II) chloride dihydrate	$CoCl_2 \cdot 2H_2O$	16544-92-6	165.870	viol-blue cry			2.477	56.2^{25}	

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
852	Cobalt(II) chloride hexahydrate	$CoCl_2 \cdot 6H_2O$	7791-13-1	237.930	pink-red monocl cry	87 dec		1.924	56.2^{25}	s EtOH, ace, eth
853	Cobalt(II) chromate	$CoCrO_4$	24613-38-5	174.927	yel-brn orth cry			≈4.0		i H_2O; s acid
854	Cobalt(II) chromite	$CoCr_2O_4$	13455-25-9	226.923	blue-grn cub cry			5.14		i H_2O, conc acid
855	Cobalt(II) citrate dihydrate	$Co_3(C_6H_5O_7)_2 \cdot 2H_2O$	18727-04-3	265.170	rose red cry	dec 150			0.8^{15}	
856	Cobalt(II) cyanide	$Co(CN)_2$	542-84-7	110.967	blue hyg cry			1.872		i H_2O
857	Cobalt(II) cyanide dihydrate	$Co(CN)_2 \cdot 2H_2O$	20427-11-6	146.998	pink-brn needles					i H_2O, acid
858	Cobalt(II) diiron tetroxide	$CoFe_2O_4$	12052-28-7	234.621	blk solid					s hot HCl
859	Cobalt(II) ferricyanide	$Co_3[Fe(CN)_6]_2$	14049-81-1	600.698	red needles					i H_2O, HCl; s NH_4OH
860	Cobalt(II) fluoride	CoF_2	10026-17-2	96.930	red tetr cry	1127	≈1400	4.46	1.4^{25}	s acid
861	Cobalt(II) fluoride tetrahydrate	$CoF_2 \cdot 4H_2O$	13817-37-3	168.992	red orth cry	dec		2.22	1.4^{25}	
862	Cobalt(II) formate dihydrate	$Co(CHO_2)_2 \cdot 2H_2O$	6424-20-0	184.998	red cry powder	140 dec		2.13	5.03^{20}	i EtOH
863	Cobalt(II) hexafluoro-2,4-pentanedioate	$Co(CF_3COCHCOCF_3)_2$	19648-83-0	473.035	powder	197				
864	Cobalt(II) hexafluorosilicate hexahydrate	$CoSiF_6 \cdot 6H_2O$	12021-68-0	309.100	pale red cry			2.087	76.8^{22}	
865	Cobalt(II) hydroxide	$Co(OH)_2$	21041-93-0	92.948	blue-grn cry	≈160 dec		3.60		sl H_2O; s acid
866	Cobalt(II) hydroxide monohydrate	$Co(OH)_2 \cdot H_2O$	35340-84-2	110.963	blue solid	136 dec				
867	Cobalt(II) iodate	$Co(IO_3)_2$	13455-28-2	408.738	blk-viol needles	200 dec		5.09	0.46^{20}	
868	Cobalt(II) iodide	CoI_2	15238-00-3	312.742	blk hex cry; hyg	520		5.60	203^{25}	
869	Cobalt(II) iodide dihydrate	$CoI_2 \cdot 2H_2O$	13455-29-3	348.773	hyg grn cry	dec 100				
870	Cobalt(II) iodide hexahydrate	$CoI_2 \cdot 6H_2O$	15238-00-3*	420.833	red hex prisms	130 dec		2.90	203^{25}	s EtOH, eth, ace
871	Cobalt(II) molybdate	$CoMoO_4$	13762-14-6	218.87	blk monocl cry	1040		4.7		
872	Cobalt(II) molybdate monohydrate	$CoMoO_4 \cdot H_2O$	18601-87-1	236.89	blk pow					
873	Cobalt(II) nitrate	$Co(NO_3)_2$	10141-05-6	182.942	pale red powder	100 dec		2.49	103^{25}	
874	Cobalt(II) nitrate hexahydrate	$Co(NO_3)_2 \cdot 6H_2O$	10026-22-9	291.034	red monocl cry; hyg	≈55		1.88	103^{25}	s EtOH
875	Cobalt(II) nitrite	$Co(NO_2)_2$	18488-96-5	150.944					0.49^{25}	
876	Cobalt(II) oleate	$Co(C_{18}H_{33}O_2)_2$	14666-94-5	621.840	brn amorp pow					i H_2O; s EtOH, eth
877	Cobalt(II) orthosilicate	Co_2SiO_4	12017-08-2	209.950	red-viol orth cry	1345		4.63		i H_2O; s dil HCl
878	Cobalt(II) oxalate	CoC_2O_4	814-89-1	146.952	pink powder	250 dec		3.02	0.0037^{20}	s acid, NH_4OH
879	Cobalt(II) oxalate dihydrate	$CoC_2O_4 \cdot 2H_2O$	5965-38-8	182.982	pink needles	dec			0.0037	sl acid; s NH_4OH
880	Cobalt(II) oxide	CoO	1307-96-6	74.932	gray cub cry	1830		6.44		i H_2O; s acid
881	Cobalt(II) 2,4-pentanedioate	$Co(CH_3COCHCOCH_3)_2$	14024-48-7	257.149	bl-viol cry	167				
882	Cobalt(II) perchlorate	$Co(ClO_4)_2$	13455-31-7	257.834	red needles			3.33	113^{25}	i EtOH, ace
883	Cobalt(II) perchlorate hexahydrate	$Co(ClO_4)_2 \cdot 6H_2O$	13478-33-6	365.926	dark red cry	dec 170		3.33		vs H_2O
884	Cobalt(II) phosphate octahydrate	$Co_3(PO_4)_2 \cdot 8H_2O$	10294-50-5	510.865	pink amorp powder			2.77		i H_2O; s acid
885	Cobalt(II) potassium sulfate hexahydrate	$CoK_2(SO_4)_2 \cdot 6H_2O$	10026-20-7	437.347	red monocl cry	75 dec		2.22		vs H_2O
886	Cobalt(II) selenate pentahydrate	$CoSeO_4 \cdot 5H_2O$	14590-19-3	291.97	red tricl cry	dec		2.51	55^{15}	
887	Cobalt(II) selenide	CoSe	1307-99-9	137.89	yel hex cry	1055		7.65		i H_2O, alk; s aqua regia
888	Cobalt(II) selenite dihydrate	$CoSeO_3 \cdot 2H_2O$	19034-13-0	221.92	blue-red powder					i H_2O
889	Cobalt(II) stannate	Co_2SnO_4	12139-93-4	300.574	grn-blue cub cry			6.30		i H_2O; s alk
890	Cobalt(II) stearate	$Co(C_{18}H_{35}O_2)_2$	1002-88-6	625.872	purp solid	74		1.13		
891	Cobalt(II) sulfate	$CoSO_4$	10124-43-3	154.996	red orth cry	>700		3.71	38.3^{25}	
892	Cobalt(II) sulfate monohydrate	$CoSO_4 \cdot H_2O$	13455-34-0	173.011	red monocl cry			3.08	38.3^{25}	
893	Cobalt(II) sulfate heptahydrate	$CoSO_4 \cdot 7H_2O$	10026-24-1	281.102	pink monocl cry	41 dec		2.03	38.3^{25}	sl EtOH, MeOH
894	Cobalt(II) sulfide	CoS	1317-42-6	90.998	blk amorp powder	1117		5.45		i H_2O; s acid
895	Cobalt(II) telluride	CoTe	12017-13-9	186.53	hex cry			≈8.8		
896	Cobalt(II) thiocyanate	$Co(SCN)_2$	3017-60-5	175.097	yel-brn powder				103^{25}	s EtOH, MeOH, ace, eth
897	Cobalt(II) thiocyanate trihydrate	$Co(SCN)_2 \cdot 3H_2O$	97126-35-7	229.143	viol rhomb cry				103^{25}	s EtOH, eth, ace
898	Cobalt(II) titanate	$CoTiO_3$	12017-01-5	154.798	grn rhomb cry			5.0		
899	Cobalt(II) tungstate	$CoWO_4$	12640-47-0	306.77	blue monocl cry			≈7.8		i H_2O; s hot conc acid
900	Cobalt(II,III) oxide	Co_3O_4	1308-06-1	240.798	blk cub cry	900 dec		6.11		i H_2O; s acid, alk
901	Cobalt(III) acetate	$Co(C_2H_3O_2)_3$	917-69-1	236.064	grn hyg cry	100 dec				s H_2O, EtOH
902	Cobalt(III) ammonium tetranitrodiammine	$NH_4[Co(NH_3)_2(NO_2)_4]$	13600-89-0	295.054	red-brn orth cry			1.97		s H_2O
903	Cobalt(III) fluoride	CoF_3	10026-18-3	115.928	brn hex cry	927		3.88		reac H_2O; s EtOH, eth, bz
904	Cobalt(III) fluoride dihydrate	$Co_2F_6 \cdot 2H_2O$	54496-71-8	267.887	red rhomb cry			2.19		s H_2O; i EtOH
905	Cobalt(III) hexammine chloride	$Co(NH_3)_6Cl_3$	10534-89-1	267.475	red monocl cry			1.71		s H_2O; i EtOH
906	Cobalt(III) hydroxide	$Co(OH)_3$	1307-86-4	109.955	brn powder	dec		≈4		i H_2O; s acid
907	Cobalt(III) nitrate	$Co(NO_3)_3$	15520-84-0	244.948	grn cub cry; hyg			≈3.0		s H_2O; reac os
908	Cobalt(III) oxide	Co_2O_3	1308-04-9	165.864	gray-blk powder	895 dec		5.18		i H_2O; s conc acid

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
909	Cobalt(III) oxide monohydrate	$Co_2O_3 \cdot H_2O$	12016-80-7	183.880	brn-blk hex cry	150 dec				i H_2O; s acid
910	Cobalt(III) 2,4-pentanedioate	$Co(CH_3COCHCOCH_3)_3$	21679-46-9	356.257	dark grn cry	213				s bz, ace
911	Cobalt(III) potassium nitrite sesquihydrate	$CoK_3(NO_2)_6 \cdot 1.5H_2O$	13782-01-9*	479.284	yel cub cry			2.6		sl H_2O; reac acid; i EtOH
912	Cobalt(III) sulfide	Co_2S_3	1332-71-4	214.061	blk cub cry			4.8		reac acid
913	Cobalt(III) titanate	Co_2TiO_4	12017-38-8	229.731	grn-blk cub cry			5.1		s conc HCl
914	Copper	Cu	7440-50-8	63.546	red metal; cub	1084.62	2562	8.96		sl dil acid
915	Copper arsenide	Cu_3As	12005-75-3	265.560	dark gray solid	827				
916	Copper nitride	Cu_3N	1308-80-1	204.645	cub cry	300 dec		5.84		
917	Copper phosphide	CuP_2	12019-11-3	125.494	monocl cry	≈900		4.20		
918	Copper silicide	Cu_5Si	12159-07-8	345.816	solid	825				
919	Copper(I) acetate	$CuC_2H_3O_2$	598-54-9	122.590	col cry	dec	subl			reac H_2O
920	Copper(I) acetylide	Cu_2C_2	1117-94-8	151.113	red amorp powder; exp					
921	Copper(I) azide	CuN_3	14336-80-2	105.566	tetr cry; exp					
922	Copper(I) bromide	CuBr	7787-70-4	143.450	wh cub cry; hyg	483	1345	4.98	0.0012[20]	i ace
923	Copper(I) chloride	CuCl	7758-89-6	98.999	wh cub cry	423	1490	4.14	0.0047[20]	i EtOH, ace
924	Copper(I) cyanide	CuCN	544-92-3	89.564	wh powder or grn orth cry	474	dec	2.9		i H_2O, EtOH; s KCN soln
925	Copper(I) fluoride	CuF	13478-41-6	82.544	cub cry			7.1		
926	Copper(I) hydride	CuH	13517-00-5	64.554	red-brn solid	60 dec				
927	Copper(I) iodide	CuI	7681-65-4	190.450	wh cub cry	591	≈1290	5.67	0.000020[20]	i dil acid
928	Copper(I) mercury iodide	Cu_2HgI_4	13876-85-2	835.30	red cry powder	trans ≈60 (brn)				i H_2O, EtOH
929	Copper(I) oxide	Cu_2O	1317-39-1	143.091	red-brn cub cry	1244	1800 dec	6.0		i H_2O
930	Copper(I) selenide	Cu_2Se	20405-64-5	206.05	blue-blk tetr cry	1113		6.84		i H_2O; s acid
931	Copper(I) sulfide	Cu_2S	22205-45-4	159.157	blue-blk orth cry	1129		5.6		i H_2O; sl acid
932	Copper(I) sulfite hemihydrate	$Cu_2SO_3 \cdot 0.5H_2O$	13982-53-1*	216.164	wh-yel hex cry					sl H_2O; s acid, alk; i EtOH, eth
933	Copper(I) sulfite monohydrate	$Cu_2SO_3 \cdot H_2O$	35788-00-2	225.171	cry			3.83		sl H_2O; s HCl
934	Copper(I) telluride	Cu_2Te	12019-52-2	254.69	blue hex cry	1127		4.6		
935	Copper(I) thiocyanate	CuSCN	1111-67-7	121.629	wh-yel amorp powder	1084		2.85		i H_2O, dil acid, EtOH, ace; s eth
936	Copper(I,II) sulfite dihydrate	$Cu_2SO_3 \cdot CuSO_3 \cdot 2H_2O$	13814-81-8	386.795	red prisms or powder					i H_2O, EtOH; s HCl
937	Copper(II) acetate	$Cu(C_2H_3O_2)_2$	142-71-2	181.635	blue-grn hyg powder					
938	Copper(II) acetate monohydrate	$Cu(C_2H_3O_2)_2 \cdot H_2O$	6046-93-1	199.650	grn monocl cry	115	240 dec	1.88		s H_2O, EtOH; sl eth
939	Copper(II) acetate metaarsenite	$Cu(C_2H_3O_2)_2 \cdot 3Cu(AsO_2)_2$	12002-03-8	1013.795	grn cry powder					i H_2O; reac acid
940	Copper(II) basic acetate	$Cu(C_2H_3O_2)_2 \cdot CuO \cdot 6H_2O$	52503-64-7	369.271	blue-grn cry or powder					sl H_2O, EtOH; s dil acid, NH_4OH
941	Copper(II) acetylide	CuC_2	12540-13-5	87.567	brn-blk solid; exp	exp 100				
942	Copper(II) arsenate	$Cu_3(AsO_4)_2$	7778-41-8	468.476	blue-grn cry					i H_2O, EtOH; s dil acid
943	Copper(II) arsenite	$CuHAsO_3$	10290-12-7	187.474	yel-grn powder					i H_2O, EtOH; s acid
944	Copper(II) azide	$Cu(N_3)_2$	14215-30-6	147.586	brn orth cry; exp			≈2.6		
945	Copper(II) borate	$Cu(BO_2)_2$	39290-85-2	149.166	blue-grn powder			3.859		i H_2O; s acid
946	Copper(II) bromide	$CuBr_2$	7789-45-9	223.354	blk monocl cry; hyg	498	900	4.710	126[25]	vs H_2O; s EtOH, ace; i bz, eth
947	Copper(II) butanoate monohydrate	$Cu(C_4H_7O_2)_2 \cdot H_2O$	540-16-9	255.756	grn monocl plates					s H_2O, diox, bz; sl EtOH
948	Copper(II) carbonate	$CuCO_3$	1184-64-1	123.555	cry					i H_2O
949	Copper(II) carbonate hydroxide	$CuCO_3 \cdot Cu(OH)_2$	12069-69-1	221.116	grn monocl cry	200 dec		4.0		i H_2O, EtOH; s dil acid
950	Copper(II) chlorate hexahydrate	$Cu(ClO_3)_2 \cdot 6H_2O$	14721-21-2	338.540	blue-grn hyg cry	65	100 dec		164[18]	vs EtOH
951	Copper(II) chloride	$CuCl_2$	7447-39-4	134.452	yel-brn monocl cry; hyg	598	993	3.4	75.7[25]	s EtOH, ace
952	Copper(II) chloride dihydrate	$CuCl_2 \cdot 2H_2O$	10125-13-0	170.483	grn-blue orth cry; hyg	100 dec		2.51	75.7[20]	vs EtOH, MeOH; s ace; i eth
953	Copper(II) chloride hydroxide	$Cu_2(OH)_3Cl$	1332-65-6	213.567	pale grn cry					i H_2O; s acid
954	Copper(II) chromate	$CuCrO_4$	13548-42-0	179.540	red-brn cry					i H_2O; s EtOH
955	Copper(II) basic chromate	$CuCrO_4 \cdot 2Cu(OH)_2$	12433-14-6	374.661	brn pow	dec 260				i H_2O; s HNO_3
956	Copper(II) chromite	$CuCr_2O_4$	12018-10-9	231.536	gray-blk tetr cry			5.4		i H_2O, dil acid
957	Copper(II) citrate hemipentahydrate	$Cu_2C_6H_4O_7 \cdot 2.5H_2O$	10402-15-0	360.221	blue-grn cry	100 dec				sl H_2O; s dil acid
958	Copper(II) cyanide	$Cu(CN)_2$	14763-77-0	115.580	grn powder					i H_2O; s acid, alk
959	Copper(II) cyclohexanebutanoate	$Cu(C_{10}H_{17}O_2)_2$	2218-80-6	402.028	powder	126 dec				
960	Copper(II) dichromate dihydrate	$CuCr_2O_7 \cdot 2H_2O$	13675-47-3	315.565	red-brn tricl cry			2.286		vs H_2O
961	Copper(II) ethanolate	$Cu(C_2H_5O)_2$	2850-65-9	153.667	blue hyg solid	120 dec				i os
962	Copper(II) ethylacetoacetate	$Cu(C_2H_5CO_2CHCOCH_3)_2$	14284-06-1	321.813	grn cry	192				s EtOH, chl

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
963	Copper(II) 2-ethylhexanoate	$Cu(C_8H_{15}O_2)_2$	149-11-1	349.953	powder	252 dec				
964	Copper(II) ferrate	$CuFe_2O_4$	12018-79-0	239.234	blk cry					
965	Copper(II) ferrocyanide	$Cu_2Fe(CN)_6$	13601-13-3	339.041	red-br cub cry or powder			2.2		i H_2O, acid, os
966	Copper(II) ferrous sulfide	$CuFeS_2$	1308-56-1	183.521	yel tetr cry	950		4.2		i H_2O, HCl; s HNO_3
967	Copper(II) fluoride	CuF_2	7789-19-7	101.543	wh monocl cry	836	1676	4.23	0.075^{25}	
968	Copper(II) fluoride dihydrate	$CuF_2 \cdot 2H_2O$	13454-88-1	137.574	blue monocl cry	130 dec		2.934	0.075^{25}	
969	Copper(II) formate	$Cu(CHO_2)_2$	544-19-4	153.581	blue cry				12.5^{20}	i os
970	Copper(II) formate tetrahydrate	$Cu(CHO_2)_2 \cdot 4H_2O$	5893-61-8	225.641	blue monocl cry				12.5	sl EtOH; i os
971	Copper(II) gluconate	$CuC_{12}H_{22}O_{14}$	527-09-3	453.841	bl-grn cry	156				sl EtOH; i os
972	Copper(II) hexafluoro-2,4-pentanedioate	$Cu(CF_3COCHCOCF_3)_2$	14781-45-4	477.648	cry	98	220 dec			s MeOH, ace, tol
973	Copper(II) hexafluorosilicate tetrahydrate	$CuSiF_6 \cdot 4H_2O$	12062-24-7	277.684	blue monocl cry	dec		2.56	99.7^{17}	sl EtOH
974	Copper(II) hydroxide	$Cu(OH)_2$	20427-59-2	97.561	blue-grn powder			3.37		i H_2O; s acid, conc alk
975	Copper(II) iodate	$Cu(IO_3)_2$	13454-89-2	413.351	grn monocl cry	dec		5.241	0.15^{20}	s dil acid
976	Copper(II) iodate monohydrate	$Cu(IO_3)_2 \cdot H_2O$	13454-90-5	431.367	blue tricl cry	248 dec		4.872	0.15^{20}	s dil H_2SO_4
977	Copper(II) molybdate	$CuMoO_4$	13767-34-5	223.48	grn cry	≈500		3.4	0.038	
978	Copper(II) nitrate	$Cu(NO_3)_2$	3251-23-8	187.555	blue-grn orth cry; hyg	255	subl		145^{25}	s diox; reac eth
979	Copper(II) nitrate trihydrate	$Cu(NO_3)_2 \cdot 3H_2O$	10031-43-3	241.602	blue rhomb cry	114	170 dec	2.32	145^{25}	vs EtOH
980	Copper(II) nitrate hexahydrate	$Cu(NO_3)_2 \cdot 6H_2O$	13478-38-1	295.647	blue rhomb cry; hyg			2.07	145^{25}	s EtOH
981	Copper(II) oleate	$Cu(C_{18}H_{33}O_2)_2$	1120-44-1	626.453	blue-grn solid					i H_2O; sl EtOH; s eth
982	Copper(II) oxalate	CuC_2O_4	814-91-5	151.565	blue-wh powder	310 dec			0.0026^{20}	i EtOH, eth; s NH_4OH
983	Copper(II) oxalate hemihydrate	$CuC_2O_4 \cdot 0.5H_2O$	814-91-5*	144.573	blue-wh cry	200 dec			0.0026^{20}	s NH_4OH
984	Copper(II) oxide	CuO	1317-38-0	79.545	blk powder or monocl cry	1227		6.31		i H_2O, EtOH; s dil acid
985	Copper(II) 2,4-pentanedioate	$Cu(CH_3COCHCOCH_3)_2$	13395-16-9	261.762	blue powder	284 dec	subl			sl H_2O; s chl
986	Copper(II) oxychloride hemiheptahydrate	$CuCl_2 \cdot 3CuO \cdot 3.5H_2O$	1332-40-7		blue-grn pow	dec 140				i H_2O; s acid, NH_4OH
987	Copper(II) perchlorate	$Cu(ClO_4)_2$	13770-18-8	262.447	grn hyg cry	130 dec			146^{30}	s eth, diox; i bz, ctc
988	Copper(II) perchlorate hexahydrate	$Cu(ClO_4)_2 \cdot 6H_2O$	10294-46-9	370.539	blue monocl cry; hyg	82	120 dec	2.22	146^{30}	vs EtOH, HOAc, ace; sl eth
989	Copper(II) phosphate	$Cu_3(PO_4)_2$	7798-23-4	380.581	blue-grn tricl cry					i H_2O; s acid, NH_4OH
990	Copper(II) phosphate trihydrate	$Cu_3(PO_4)_2 \cdot 3H_2O$	10031-48-8	434.627	blue-grn orth cry					i H_2O; s acid, NH_4OH
991	Copper(II) phthalocyanine	$CuC_{32}H_{16}N_8$	147-14-8	576.069	bl-purp cry					i H_2O, EtOH; s conc H_2SO_4
992	Copper(II) selenate pentahydrate	$CuSeO_4 \cdot 5H_2O$	10031-45-5	296.58	blue tricl cry	80 dec		2.56	27.4^{25}	s acid, NH_4OH; sl ace; i EtOH
993	Copper(II) selenide	CuSe	1317-41-5	142.51	blue-blk needles or plates	550 dec		5.99		reac acid
994	Copper(II) selenite dihydrate	$CuSeO_3 \cdot 2H_2O$	15168-20-4	226.54	blue orth cry			3.31		i H_2O; s acid, NH_4OH
995	Copper(II) silicate dihydrate	$CuSiO_3 \cdot 2H_2O$	26318-99-0	175.661	grn-blue orth cry					
996	Copper(II) stannate	$CuSnO_3$	12019-07-7	230.254	blue pow					
997	Copper(II) stearate	$Cu(C_{18}H_{35}O_2)_2$	660-60-6	630.485	blue-grn amorp powder	≈250				i H_2O, EtOH, eth; s py
998	Copper(II) sulfate	$CuSO_4$	7758-98-7	159.609	wh-grn amorp powder or rhomb cry	560 dec		3.60	22.0^{25}	i EtOH
999	Copper(II) sulfate pentahydrate	$CuSO_4 \cdot 5H_2O$	7758-99-8	249.685	blue tricl cry	110 dec		2.286	22.0^{25}	s MeOH; sl EtOH
1000	Copper(II) sulfate, basic	$Cu_3(OH)_4SO_4$	1332-14-5	354.730	grn rhomb cry			3.88		i H_2O
1001	Copper(II) sulfide	CuS	1317-40-4	95.611	blk hex cry	trans 507		4.76		i H_2O, EtOH, dil acid, alk
1002	Copper(II) tartrate trihydrate	$CuC_4H_4O_6 \cdot 3H_2O$	815-82-7	265.663	blue-grn powder					sl H_2O; s acid, alk
1003	Copper(II) telluride	CuTe	12019-23-7	191.15	yel orth cry	trans ≈400		7.09		
1004	Copper(II) tellurite	$CuTeO_3$	13812-58-3	239.14	blk glassy solid					i H_2O
1005	Copper(II) tetrafluoroborate	$Cu(BF_4)_2$	14735-84-3	237.155	solid					s H_2O
1006	Copper(II) titanate	$CuTiO_3$	12019-08-8	159.411	gray pow					
1007	Copper(II) 1,1,1-trifluoro-2,4-pentanedioate	$Cu(CF_3COCHCOCH_3)_2$	14324-82-4	369.705	blue-purp cry	197	dec 260			s EtOH, tol
1008	Copper(II) tungstate	$CuWO_4$	13587-35-4	311.38	yel-brn powder			7.5		
1009	Copper(II) tungstate dihydrate	$CuWO_4 \cdot 2H_2O$	13587-35-4*	347.41	grn powder					i H_2O; sl HOAc; reac conc acid
1010	Copper(II) vanadate	$Cu(VO_3)_2$	12789-09-2	261.425	powder					

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1011	Curium	Cm	7440-51-9	247	silv metal; hex or cub	1345	≈3100	13.51		
1012	Dysprosium	Dy	7429-91-6	162.500	silv metal; hex	1412	2567	8.55		s dil acid
1013	Dysprosium boride	DyB_4	12310-43-9	205.744	tetr cry	2500		6.98		
1014	Dysprosium nitride	DyN	12019-88-4	176.507	cub cry			9.93		
1015	Dysprosium silicide	$DySi_2$	12133-07-2	218.671	orth cry	1550		5.2		
1016	Dysprosium(II) bromide	$DyBr_2$	83229-05-4	322.308	blk solid					
1017	Dysprosium(II) chloride	$DyCl_2$	13767-31-2	233.406	blk cry	721 dec				reac H_2O
1018	Dysprosium(II) iodide	DyI_2	36377-94-3	416.309	purp cry	659				reac H_2O
1019	Dysprosium(III) acetate tetrahydrate	$Dy(C_2H_3O_2)_3 \cdot 4H_2O$	15280-55-4	411.693	yel needles	dec 120				s H_2O; sl EtOH
1020	Dysprosium(III) bromide	$DyBr_3$	14456-48-5	402.212	wh hyg cry	879				s H_2O
1021	Dysprosium(III) carbonate tetrahydrate	$Dy_2(CO_3)_3 \cdot 4H_2O$	38245-35-1	577.088	wh cry pow					i H_2O
1022	Dysprosium(III) chloride	$DyCl_3$	10025-74-8	268.859	wh or yel cry	718	1530	3.67		s H_2O, MeOH
1023	Dysprosium(III) chloride hexahydrate	$DyCl_3 \cdot 6H_2O$	15059-52-6	376.950	bright yel cry	dec 162				
1024	Dysprosium(III) fluoride	DyF_3	13569-80-7	219.495	grn cry	1157				
1025	Dysprosium(III) hydride	DyH_3	13537-09-2	165.524	hex cry			7.1		
1026	Dysprosium(III) hydroxide	$Dy(OH)_3$	1308-85-6	213.522	yel or wh needles	205 dec				i H_2O
1027	Dysprosium(III) iodide	DyI_3	15474-63-2	543.213	grn cry	978				
1028	Dysprosium(III) nitrate pentahydrate	$Dy(NO_3)_3 \cdot 5H_2O$	10143-38-1*	438.591	yel cry	88.6			208.4[25]	
1029	Dysprosium(III) oxide	Dy_2O_3	1308-87-8	372.998	wh cub cry	2228	3900	7.81		s acid
1030	Dysprosium(III) sulfate octahydrate	$Dy_2(SO_4)_3 \cdot 8H_2O$	10031-50-2	757.310	pale yel cry	110				sl H_2O
1031	Dysprosium(III) sulfide	Dy_2S_3	12133-10-7	421.195	red-brn monocl cry			6.08		
1032	Dysprosium(III) telluride	Dy_2Te_3	12159-43-2	707.80	solid	≈1550				
1033	Einsteinium	Es	7429-92-7	252	metal; cub	860				
1034	Erbium	Er	7440-52-0	167.259	silv metal; hex	1529	2868	9.07		i H_2O; s acid
1035	Erbium boride	ErB_4	12310-44-0	210.503	tetr cry	2450		7.0		
1036	Erbium acetate tetrahydrate	$Er(C_2H_3O_2)_3 \cdot 4H_2O$	15280-57-6	416.452	pink or wh cry			2.11		s H_2O
1037	Erbium bromide	$ErBr_3$	13536-73-7	406.971	viol hyg cry	950	≈1460			s H_2O, thf
1038	Erbium bromide hexahydrate	$ErBr_3 \cdot 6H_2O$	14890-44-9	515.062	pink cry					s H_2O
1039	Erbium chloride	$ErCl_3$	10138-41-7	273.618	viol monocl cry; hyg	776		4.1		s H_2O
1040	Erbium chloride hexahydrate	$ErCl_3 \cdot 6H_2O$	10025-75-9	381.709	pink hyg cry	dec				s H_2O; sl EtOH
1041	Erbium fluoride	ErF_3	13760-83-3	224.254	pink orth cry	1146		7.8		i H_2O
1042	Erbium hydride	ErH_3	13550-53-3	170.283	hex cry			≈7.6		
1043	Erbium hydroxide	$Er(OH)_3$	14646-16-3	218.281	pink solid					i H_2O
1044	Erbium iodide	ErI_3	13813-42-8	547.972	viol hex cry; hyg	1014		≈5.5		s H_2O
1045	Erbium nitrate pentahydrate	$Er(NO_3)_3 \cdot 5H_2O$	10168-80-6*	443.350	red cry	130 dec			240.8[25]	s EtOH, ace
1046	Erbium nitride	ErN	12020-21-2	181.266	cub cry			10.6		
1047	Erbium oxide	Er_2O_3	12061-16-4	382.516	pink powder	2344	3920	8.64		i H_2O; s acid
1048	Erbium silicide	$ErSi_2$	12020-28-9	223.430	orth cry			7.26		
1049	Erbium sulfate	$Er_2(SO_4)_3$	13478-49-4	622.706	hyg powder	dec		3.68	13[20]	
1050	Erbium sulfate octahydrate	$Er_2(SO_4)_3 \cdot 8H_2O$	10031-52-4	766.828	pink monocl cry	dec		3.20	13[20]	
1051	Erbium sulfide	Er_2S_3	12159-66-9	430.713	red-brn monocl cry	1730		6.07		
1052	Erbium telluride	Er_2Te_3	12020-39-2	717.32	orth cry	1213		7.11		
1053	Europium	Eu	7440-53-1	151.964	soft silv metal; cub	822	1529	5.24		reac H_2O
1054	Europium boride	EuB_6	12008-05-8	216.830	cub cry	≈2600		4.91		
1055	Europium nitride	EuN	12020-58-5	165.971	cub cry			8.7		
1056	Europium silicide	$EuSi_2$	12434-24-1	208.135	tetr cry	1500		5.46		
1057	Europium(II) bromide	$EuBr_2$	13780-48-8	311.772	wh cry	683				s H_2O
1058	Europium(II) chloride	$EuCl_2$	13769-20-5	222.870	wh orth cry	731		4.9		s H_2O
1059	Europium(II) fluoride	EuF_2	14077-39-5	189.961	grn-yel cub cry	≈1380		6.5		
1060	Europium(II) iodide	EuI_2	22015-35-6	405.773	grn cry	580				s H_2O
1061	Europium(II) selenide	EuSe	12020-66-5	230.92	brn cub cry			6.45		
1062	Europium(II) sulfate	$EuSO_4$	10031-54-6	248.027	col orth cry			4.99		i H_2O
1063	Europium(II) sulfide	EuS	12020-65-4	184.029	cub cry			5.7		
1064	Europium(II) telluride	EuTe	12020-69-8	279.56	blk cub cry	1526		6.48		
1065	Europium(III) bromide	$EuBr_3$	13759-88-1	391.676	gray cry	dec				s H_2O
1066	Europium(III) chloride	$EuCl_3$	10025-76-0	258.323	grn-yel needles	623		4.89		
1067	Europium(III) chloride hexahydrate	$EuCl_3 \cdot 6H_2O$	13759-92-7	366.414	wh-yel hyg cry	850		4.89		s H_2O
1068	Europium(III) fluoride	EuF_3	13765-25-8	208.959	wh hyg cry	1276				i H_2O
1069	Europium(III) iodide	EuI_3	13759-90-5	532.677	col cry; unstab	≈875				
1070	Europium(III) nitrate hexahydrate	$Eu(NO_3)_3 \cdot 6H_2O$	10031-53-5	446.070	wh-pink hyg cry	85 dec			193[25]	
1071	Europium(III) oxalate	$Eu_2(C_2O_4)_3$	3269-12-3	567.985	wh solid					i H_2O; s acid

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1072	Europium(III) oxide	Eu_2O_3	1308-96-9	351.926	pink powder	2291	3790	7.42		i H_2O; s acid
1073	Europium(III) perchlorate hexahydrate	$Eu(ClO_4)_3 \cdot 6H_2O$	36907-40-1	558.407	wh or pink cry					s H_2O, EtOH
1074	Europium(III) sulfate	$Eu_2(SO_4)_3$	13537-15-0	592.116	pale pink cry			4.99	2.1^{20}	
1075	Europium(III) sulfate octahydrate	$Eu_2(SO_4)_3 \cdot 8H_2O$	10031-52-4	736.238	pink cry	375 dec			2.1^{20}	
1076	Fermium	Fm	7440-72-4	257	metal	1527				
1077	Fluorine	F_2	7782-41-4	37.997	pale yel gas	-219.67 tp	-188.12	1.553 g/L		reac H_2O
1078	Fluorine monoxide	F_2O	7783-41-7	53.996	col gas	-223.8	-144.3	2.207 g/L		sl H_2O
1079	Difluorine dioxide	F_2O_2	7783-44-0	69.996	red-oran solid, unstab gas	-163.5	-57 (extrap)	2.861 g/L		
1080	Fluorine tetroxide	F_2O_4	107782-11-6	101.995	red-brn solid	-191	dec -185			
1081	Fluorine nitrate	FNO_3	7789-26-6	81.003	col gas	-175	-46	3.311 g/L		reac H_2O, EtOH, eth; s ace
1082	Fluorine perchlorate	$FOClO_3$	10049-03-3	118.449	col gas; exp	-167.3	-16	4.841 g/L		reac H_2O
1083	Francium	Fr	7440-73-5	223.000	short-lived alkali metal	27				
1084	Gadolinium	Gd	7440-54-2	157.25	silv metal; hex	1313	3273	7.90		s dil acid
1085	Gadolinium boride	GdB_6	12008-06-9	222.12	blk-brn cub cry	2510		5.31		
1086	Gadolinium nitride	GdN	25764-15-2	171.26	cub cry			9.10		
1087	Gadolinium silicide	$GdSi_2$	12134-75-7	213.42	orth cry	1540		5.9		
1088	Gadolinium(II) iodide	GdI_2	13814-72-7	411.06	bronze cry	831				
1089	Gadolinium(II) selenide	GdSe	12024-81-6	236.21	cub cry	2170		8.1		
1090	Gadolinium(III) acetate tetrahydrate	$Gd(C_2H_3O_2)_2 \cdot 4H_2O$	15280-53-2	406.44	wh tricl cry	dec		1.61		s H_2O
1091	Gadolinium(III) bromide	$GdBr_3$	13818-75-2	396.96	wh monocl cry; hyg	770		4.56		
1092	Gadolinium(III) chloride	$GdCl_3$	10138-52-0	263.61	wh monocl cry; hyg	602		4.52		s H_2O
1093	Gadolinium(III) chloride hexahydrate	$GdCl_3 \cdot 6H_2O$	19423-81-5	371.70	col hyg cry			2.424		s H_2O
1094	Gadolinium(III) fluoride	GdF_3	13765-26-9	214.25	wh cry	1232				
1095	Gadolinium(III) iodide	GdI_3	13572-98-0	537.96	yel cry	930				
1096	Gadolinium(III) nitrate pentahydrate	$Gd(NO_3)_3 \cdot 5H_2O$	52788-53-1	433.34	wh cry	92 dec		2.41	190^{25}	
1097	Gadolinium(III) nitrate hexahydrate	$Gd(NO_3)_3 \cdot 6H_2O$	19598-90-4	451.36	hyg tricl cry	91 dec		2.33	190^{25}	s EtOH
1098	Gadolinium(III) oxalate decahydrate	$Gd_2(C_2O_4)_3 \cdot 10H_2O$	22992-15-0	758.71	wh monocl pow	dec 110				i H_2O; sl acid
1099	Gadolinium(III) oxide	Gd_2O_3	12064-62-9	362.50	wh hyg powder	2339	3900	7.07		i H_2O; s acid
1100	Gadolinium(III) sulfate	$Gd_2(SO_4)_3$	13628-54-1	602.69	col cry	500 dec		4.1	2.60^{20}	sl H_2O
1101	Gadolinium(III) sulfate octahydrate	$Gd_2(SO_4)_3 \cdot 8H_2O$	13450-87-8	746.81	col monocl cry	400 dec		4.14	2.3^{20}	sl H_2O
1102	Gadolinium(III) sulfide	Gd_2S_3	12134-77-9	410.70	yel cub cry			6.1		
1103	Gadolinium(III) telluride	Gd_2Te_3	12160-99-5	697.30	orth cry	1255		7.7		
1104	Gallium	Ga	7440-55-3	69.723	silv liq or gray orth cry	29.7666 tp	2204	5.91		reac alk
1105	Gallium antimonide	GaSb	12064-03-8	191.483	brn cub cry	712		5.6137		
1106	Gallium arsenide	GaAs	1303-00-0	144.645	gray cub cry	1238		5.3176		
1107	Gallium nitride	GaN	25617-97-4	83.730	gray hex cry	>2500		6.1		
1108	Gallium phosphide	GaP	12063-98-8	100.697	yel cub cry	1457		4.138		
1109	Gallium suboxide	Ga_2O	12024-20-3	155.445	brn powder	>660	>800 dec	4.77		
1110	Gallium(II) chloride	$GaCl_2$	24597-12-4	140.629	wh orth cry	172.4	535	2.74		
1111	Gallium(II) selenide	GaSe	12024-11-2	148.68	hex cry	960		5.03		
1112	Gallium(II) sulfide	GaS	12024-10-1	101.788	hex cry	965		3.86		
1113	Gallium(II) telluride	GaTe	12024-14-5	197.32	monocl cry	824		5.44		
1114	Gallium(III) bromide	$GaBr_3$	13450-88-9	309.435	wh orth cry	123	279	3.69		
1115	Gallium(III) chloride	$GaCl_3$	13450-90-3	176.082	col needles or gl solid	77.9	201	2.47		
1116	Gallium(III) fluoride	GaF_3	7783-51-9	126.718	wh powder or col needles	>1000		4.47		i H_2O
1117	Gallium(III) fluoride trihydrate	$GaF_3 \cdot 3H_2O$	22886-66-4	180.764	wh cry	>140 dec				sl H_2O
1118	Gallium(III) hydride	GaH_3	13572-93-5	72.747	visc liq	-15	≈0 dec			
1119	Gallium(III) hydroxide	$Ga(OH)_3$	12023-99-3	120.745	unstab prec					
1120	Gallium(III) iodide	GaI_3	13450-91-4	450.436	monocl cry	212	340	4.5		
1121	Gallium(III) nitrate	$Ga(NO_3)_3$	13494-90-1	255.738	wh cry powder					s H_2O, EtOH, eth
1122	Gallium(III) oxide	Ga_2O_3	12024-21-4	187.444	wh cry	1807		≈6.0		s hot acid
1123	Gallium(III) oxide hydroxide	GaOOH	20665-52-5	102.730	orth cry			5.23		
1124	Gallium(III) 2,4-pentanedioate	$Ga(CH_3COCHCOCH_3)_3$	14405-43-7	367.047	wh powder	193	subl	1.42		
1125	Gallium(III) perchlorate hexahydrate	$Ga(ClO_4)_3 \cdot 6H_2O$	17835-81-3	476.166	cry	dec 175				
1126	Gallium(III) selenide	Ga_2Se_3	12024-24-7	376.33	cub cry	937		4.92		
1127	Gallium(III) sulfate	$Ga_2(SO_4)_3$	13494-91-2	427.634	hex cry					
1128	Gallium(III) sulfate octadecahydrate	$Ga_2(SO_4)_3 \cdot 18H_2O$	13780-42-2	751.909	octahed cry					s H_2O, EtOH
1129	Gallium(III) sulfide	Ga_2S_3	12024-22-5	235.641	monocl cry	1090		3.7		

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1130	Gallium(III) telluride	Ga_2Te_3	12024-27-0	522.25	cub cry	790		5.57		
1131	Germanium	Ge	7440-56-4	72.64	gray-wh cub cry	938.25	2833	5.3234		i H_2O, dil acid, alk
1132	Germane	GeH_4	7782-65-2	76.67	col gas; flam	-165	-88.1	3.133 g/L		i H_2O
1133	Digermane	Ge_2H_6	13818-89-8	151.33	col liq; flam	-109	29	1.98^{-109}		
1134	Trigermane	Ge_3H_8	14691-44-2	225.98	col liq	-105.6	110.5	2.20^{-105}		i H_2O
1135	Tetragermane	Ge_4H_{10}	14691-47-5	300.64	col liq		176.9			i H_2O
1136	Pentagermane	Ge_5H_{12}	15587-39-0	375.30	col liq		234			i H_2O
1137	Bromogermane	GeH_3Br	13569-43-2	155.57	col liq	-32	52	2.34		reac H_2O
1138	Chlorogermane	GeH_3Cl	13637-65-5	111.12	col liq	-52	28	1.75		reac H_2O
1139	Chlorotrifluorogermane	GeF_3Cl	14188-40-0	165.09	gas	-66.2	-20.3	6.747 g/L		
1140	Dibromogermane	GeH_2Br_2	13769-36-3	234.46	col liq	-15	89	2.80		reac H_2O
1141	Dichlorogermane	GeH_2Cl_2	15230-48-5	145.56	col liq	-68	69.5	1.90		reac H_2O
1142	Dichlorodifluorogermane	GeF_2Cl_2	24422-21-7	181.54	col gas	-51.8	-2.8	7.419 g/L		
1143	Dichlorodimethylgermane	$Ge(CH_3)_2Cl_2$	1529-48-2	173.62	liq	-22	124	1.49		
1144	Fluorogermane	GeH_3F	13537-30-9	94.66	col gas			3.868 g/L		reac H_2O
1145	Iodogermane	GeH_3I	13573-02-9	202.57	liq	-15	≈90			reac H_2O
1146	Methylgermane	GeH_3CH_3	1449-65-6	90.70	col gas	-158	-23	3.706 g/L		
1147	Tribromogermane	$GeHBr_3$	14779-70-5	313.36	col liq	-25	dec			reac H_2O
1148	Trichlorogermane	$GeHCl_3$	1184-65-2	180.01	liq	-71	75.3	1.93		reac H_2O
1149	Trichlorofluorogermane	$GeCl_3F$	24422-20-6	198.00	liq	-49.8	37.5			
1150	Germanium(II) bromide	$GeBr_2$	24415-00-7	232.45	yel monocl cry	122	150 dec			reac H_2O
1151	Germanium(II) chloride	$GeCl_2$	10060-11-4	143.55	wh-yel hyg powder	dec				reac H_2O; s eth, bz
1152	Germanium(II) fluoride	GeF_2	13940-63-1	110.64	wh orth cry; hyg	110	130 dec	3.64		reac H_2O
1153	Germanium(II) iodide	GeI_2	13573-08-5	326.45	oran-yel hex cry	428	550 dec	5.4		reac H_2O
1154	Germanium(II) oxide	GeO	20619-16-3	88.64	blk solid	700 dec				
1155	Germanium(II) selenide	GeSe	12065-10-0	151.60	gray orth cry or brn powder	675		5.6		
1156	Germanium(II) sulfide	GeS	12025-32-0	104.71	gray orth cry	658		4.1		
1157	Germanium(II) telluride	GeTe	12025-39-7	200.24	cub cry	724		6.16		i H_2O; s conc HNO_3
1158	Germanium(IV) bromide	$GeBr_4$	13450-92-5	392.26	wh cry	26.1	186.35	3.132		reac H_2O
1159	Germanium(IV) chloride	$GeCl_4$	10038-98-9	214.45	col liq	-51.50	86.55	1.88		reac H_2O; s bz, eth, EtOH, ctc
1160	Germanium(IV) fluoride	GeF_4	7783-58-6	148.63	col gas	-15 tp	-36.5 sp	6.074 g/L		reac H_2O
1161	Germanium(IV) iodide	GeI_4	13450-95-8	580.26	red-oran cub cry	146	348	4.322		reac H_2O
1162	Germanium(IV) nitride	Ge_3N_4	12065-36-0	273.95	orth cry	900 dec				i H_2O, acid, aqua regia
1163	Germanium(IV) oxide	GeO_2	1310-53-8	104.64	wh hex cry	1116		4.25		i H_2O
1164	Germanium(IV) selenide	$GeSe_2$	12065-11-1	230.56	yel-oran orth cry	707 dec		4.56		
1165	Germanium(IV) sulfide	GeS_2	12025-34-2	136.77	blk orth cry	530		3.01		
1166	Gold	Au	7440-57-5	196.967	soft yel metal	1064.18	2856	19.3		s aqua regia
1167	Bromoauric(III) acid pentahydrate	$HAuBr_4 \cdot 5H_2O$	17083-68-0	607.667	red-brn hyg cry	27				s H_2O, EtOH
1168	Chloroauric(III) acid tetrahydrate	$HAuCl_4 \cdot 4H_2O$	16903-35-8	411.848	yel monocl cry; hyg			≈3.9		vs H_2O, EtOH; s eth
1169	Gold(I) bromide	AuBr	10294-27-6	276.871	yel-gray tetr cry	165 dec		8.20		i H_2O
1170	Gold(I) chloride	AuCl	10294-29-8	232.420	yel orth cry	289 dec		7.6	0.000031^{20}	
1171	Gold(I) cyanide	AuCN	506-65-0	222.985	yel hex cry	dec		7.2		i H_2O, EtOH, eth, dil acid
1172	Gold(I) iodide	AuI	10294-31-2	323.871	yel-grn powder; tetr	120 dec		8.25		i H_2O; s CN soln
1173	Gold(I) sulfide	Au_2S	1303-60-2	425.998	brn-blk cub cry; unstab	240 dec		≈11		i H_2O, acid; s aqua regia
1174	Gold(III) bromide	$AuBr_3$	10294-28-7	436.679	red-br monocl cry	≈160 dec				s H_2O, EtOH
1175	Gold(III) chloride	$AuCl_3$	13453-07-1	303.326	red monocl cry	>160 dec		4.7	68^{20}	
1176	Gold(III) cyanide trihydrate	$Au(CN)_3 \cdot 3H_2O$	535-37-5*	329.065	wh hyg cry	50 dec				vs H_2O; sl EtOH
1177	Gold(III) fluoride	AuF_3	14720-21-9	253.962	oran-yel hex cry	>300	subl	6.75		
1178	Gold(III) hydroxide	$Au(OH)_3$	1303-52-2	247.989	brn powder	≈100 dec				i H_2O; s acid
1179	Gold(III) iodide	AuI_3	31032-13-0	577.680	unstab grn powder	20 dec				
1180	Gold(III) oxide	Au_2O_3	1303-58-8	441.931	brn powder	≈150 dec				i H_2O; s acid
1181	Gold(III) selenate	$Au_2(SeO_4)_3$	10294-32-3	822.81	yel cry					i H_2O; s acid
1182	Gold(III) selenide	Au_2Se_3	1303-62-4	630.81	blk amorp solid	dec		4.65		s aqua regia
1183	Gold(III) sulfide	Au_2S_3	1303-61-3	490.128	unstab blk powder	200 dec				
1184	Hafnium	Hf	7440-58-6	178.49	gray metal; hex	2233	4603	13.3		s HF
1185	Hafnium boride	HfB_2	12007-23-7	200.11	gray hex cry	3100		10.5		
1186	Hafnium carbide	HfC	12069-85-1	190.50	refrac cub cry	≈3000		12.2		
1187	Hafnium hydride	HfH_2	12770-26-2	180.51	refrac tetr cry			11.4		
1188	Hafnium nitride	HfN	25817-87-2	192.50	yel-brn cub cry	3310		13.8		
1189	Hafnium phosphide	HfP	12325-59-6	209.46	hex cry			9.78		

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1190	Hafnium silicide	$HfSi_2$	12401-56-8	234.66	gray orth cry	≈1700		7.6		
1191	Hafnocene dichloride	$Hf(C_5H_5)_2Cl_2$	12116-66-4	379.58	col hyg cry	235				s bz, chl; sl thf, eth; i hex
1192	Hafnium(II) bromide	$HfBr_2$	13782-95-1	338.30	blue-blk cry	dec 400				
1193	Hafnium(II) chloride	$HfCl_2$	13782-92-8	249.40	blk solid	dec 400				
1194	Hafnium(III) bromide	$HfBr_3$	13782-96-2	418.20	blue-blk cry	dec 350				
1195	Hafnium(III) chloride	$HfCl_3$	13782-93-9	284.85	blk solid	dec				
1196	Hafnium(III) iodide	HfI_3	13779-73-2	559.20	blk cry	dec				
1197	Hafnium(IV) bromide	$HfBr_4$	13777-22-5	498.11	wh cub cry	424 tp	323 sp	4.90		reac H_2O
1198	Hafnium(IV) chloride	$HfCl_4$	13499-05-3	320.30	wh monocl cry	432 tp	317 sp			reac H_2O
1199	Hafnium(IV) fluoride	HfF_4	13709-52-9	254.48	wh monocl cry	1025	970 sp	7.1		reac H_2O
1200	Hafnium(IV) iodide	HfI_4	13777-23-6	686.11	yel-oran cub cry	449 tp	394 sp	5.6		reac H_2O
1201	Hafnium(IV) oxide	HfO_2	12055-23-1	210.49	wh cub cry	2800	≈5400	9.68		i H_2O
1202	Hafnium(IV) oxychloride octahydrate	$HfOCl_2 \cdot 8H_2O$	14456-34-9	409.52	wh tetr cry	dec				s H_2O
1203	Hafnium(IV) selenide	$HfSe_2$	12162-21-9	336.41	brn hex cry			7.46		
1204	Hafnium(IV) silicate	$HfSiO_4$	13870-13-8	270.57	tetr cry	2758		7.0		
1205	Hafnium(IV) sulfate	$Hf(SO_4)_2$	15823-43-5	370.62	wh cry	>500 dec				
1206	Hafnium(IV) sulfide	HfS_2	18855-94-2	242.62	purp-brn hex cry			6.03		
1207	Hafnium(IV) titanate	$HfTiO_4$	12055-24-2	290.36	wh pow	1980 dec				
1208	Helium	He	7440-59-7	4.003	col gas		-268.93	0.164 g/L		sl H_2O; i EtOH
1209	Holmium	Ho	7440-60-0	164.930	silv metal; hex	1472	2700	8.80		s dil acid
1210	Holmium acetate	$Ho(C_2H_3O_2)_3$	25519-09-9	342.062	yel cry	dec 327				s H_2O
1211	Holmium bromide	$HoBr_3$	13825-76-8	404.642	yel hyg cry	919	1470			
1212	Holmium chloride	$HoCl_3$	10138-62-2	271.289	yel monocl cry; hyg	720	1500	3.7		s H_2O
1213	Holmium chloride hexahydrate	$HoCl_3 \cdot 6H_2O$	14914-84-2	379.381	hyg yel cry	160 dec				s H_2O
1214	Holmium fluoride	HoF_3	13760-78-6	221.925	pink-yel orth cry; hyg	1143	>2200	7.664		s H_2O
1215	Holmium iodide	HoI_3	13813-41-7	545.643	yel hex cry	994		5.4		
1216	Holmium nitrate pentahydrate	$Ho(NO_3)_3 \cdot 5H_2O$	14483-18-2	441.022	hyg oran cry					s H_2O, EtOH, ace
1217	Holmium nitride	HoN	12029-81-1	178.937	cub cry			10.6		
1218	Holmium oxalate decahydrate	$Ho_2(C_2O_4)_2 \cdot 10H_2O$	28965-57-3	774.070	yel solid	dec 40				
1219	Holmium oxide	Ho_2O_3	12055-62-8	377.859	yel cub cry	2330	3900	8.41		s acid
1220	Holmium silicide	$HoSi_2$	12136-24-2	221.101	hex cry			7.1		
1221	Holmium sulfide	Ho_2S_3	12162-59-3	426.056	yel-oran monocl cry			5.92		
1222	Hydrazine	N_2H_4	302-01-2	32.045	col oily liq	1.54	113.55	1.0036		vs H_2O, EtOH, MeOH
1223	Hydrazine acetate	$N_2H_4 \cdot CH_3COOH$	13255-48-6	92.097	cry	100				
1224	Hydrazine azide	$N_2H_4 \cdot HN_3$	14662-04-5	75.074	hyg wh prism	75 exp				vs H_2O
1225	Hydrazine monohydrate	$N_2H_4 \cdot H_2O$	7803-57-8	50.060	fuming liq	-51.7	119	1.030		vs H_2O, EtOH; i chl, eth
1226	Hydrazine hydrobromide	$N_2H_4 \cdot HBr$	13775-80-9	112.957	wh monocl cry flakes	84	≈190 dec	2.3		s H_2O, EtOH
1227	Hydrazine hydrochloride	$N_2H_4 \cdot HCl$	2644-70-4	68.506	wh orth cry	89	240 dec	1.5		s H_2O; i os
1228	Hydrazine dihydrochloride	$N_2H_4 \cdot 2HCl$	5341-61-7	104.967	wh orth cry	198 dec		1.42		s H_2O; sl EtOH
1229	Hydrazine hydroiodide	$N_2H_4 \cdot HI$	10039-55-1	159.957	hyg cry	125				s H_2O
1230	Hydrazine nitrate	$N_2H_4 \cdot HNO_3$	13464-97-6	95.058	monocl cry; exp	70				vs H_2O
1231	Hydrazine dinitrate	$N_2H_4 \cdot 2HNO_3$	13464-98-7	158.071	needles	104 dec			20^{35}	s H_2O
1232	Hydrazine perchlorate hemihydrate	$N_2H_4 \cdot HClO_4 \cdot 0.5H_2O$	13762-65-7		solid	137	exp	1.94		reac H_2O, s EtOH; i eth, bz
1233	Hydrazine sulfate	$N_2H_4 \cdot H_2SO_4$	10034-93-2	130.124	col orth cry	254		1.378		sl H_2O; i EtOH
1234	Dihydrazine sulfate	$(N_2H_4)_2 \cdot H_2SO_4$	13464-80-7	162.169	hyg wh cry flakes	104	dec >180		200^{25}	vs H_2O; i os
1235	Hydrazoic acid	HN_3	7782-79-8	43.028	col liq; exp	-80	35.7			s H_2O
1236	Hydroxylamine	H_2NOH	7803-49-8	33.030	wh orth flakes or needles	33.1	58	1.21		vs H_2O, MeOH
1237	Hydroxylamine hydrobromide	$H_2NOH \cdot HBr$	41591-55-3	113.942	monocl cry			2.35		s H_2O
1238	Hydroxylamine hydrochloride	$H_2NOH \cdot HCl$	5470-11-1	69.491	col monocl cry	159 dec	exp	1.68	94^{25}	vs H_2O
1239	Hydroxylamine perchlorate	$H_2NOH \cdot HClO_4$		133.489	orth cry	88	dec 120			
1240	Hydroxylamine sulfate	$2(H_2NOH) \cdot H_2SO_4$	10039-54-0	164.138	cry	170				vs H_2O
1241	Hydrogen	H_2	1333-74-0	2.016	col gas; flam	-259.198 tp	-252.762	0.082 g/L		sl H_2O
1242	Hydrogen-*d*$_2$	D_2	7782-39-0	4.028	col gas	-254.42	-249.48	0.164 g/L		
1243	Hydrogen-*t*$_2$	T_2	10028-17-8	6.032	col gas	-252.53	-248.11	0.246 g/L		
1244	Hydrogen-*d*$_1$	HD	13983-20-5	3.022	col gas	-256.55	-251.02	0.123 g/L		
1245	Hydrogen-*t*$_1$	HT	14885-60-0	4.024	col gas	-254.7	-249.6			
1246	Hydrogen-*d*$_1$,*t*$_1$	DT	14885-61-1	5.030	col gas	-253.5	-238.9			
1247	Hydrogen bromide	HBr	10035-10-6	80.912	col gas	-86.80	-66.38	3.307 g/L		vs H_2O; s EtOH
1248	Hydrogen bromide-*d*	DBr	13536-59-9	81.918	col gas	-87.54	-66.9			s H_2O

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1249	Hydrogen chloride	HCl	7647-01-0	36.461	col gas	-114.17	-85	1.490 g/L		vs H_2O
1250	Hydrogen chloride dihydrate	$HCl \cdot 2H_2O$	13465-05-9	72.492	col liq	-17.7		1.46		
1251	Hydrogen chloride-*d*	DCl	7698-05-7	37.467	col gas	-114.72	-84.4			s H_2O
1252	Hydrogen cyanide	HCN	74-90-8	27.026	col liq or gas	-13.29	26	0.6876^{20}		vs H_2O, EtOH; sl eth
1253	Hydrogen fluoride	HF	7664-39-3	20.006	col gas	-83.36	20	0.818 g/L		vs H_2O, EtOH; sl eth
1254	Hydrogen iodide	HI	10034-85-2	127.912	col or yel gas	-50.76	-35.55	5.228 g/L		vs H_2O; s os
1255	Hydrogen iodide-*d*	DI	14104-45-1	128.918	col gas	-51.93	-36.2			s H_2O
1256	Hydrogen peroxide	H_2O_2	7722-84-1	34.015	col liq	-0.43	150.2	1.44		vs H_2O
1257	Hydrogen selenide	H_2Se	7783-07-5	80.98	col gas; flam	-65.73	-41.25	3.310 g/L		s H_2O
1258	Hydrogen sulfide	H_2S	7783-06-4	34.081	col gas; flam	-85.5	-59.55	1.393 g/L		s H_2O
1259	Hydrogen disulfide	H_2S_2	13465-07-1	66.146	col liq		70.7	1.334		
1260	Hydrogen telluride	H_2Te	7783-09-7	129.62	col gas	-49	-2	5.298 g/L		s H_2O, EtOH, alk
1261	Indium	In	7440-74-6	114.818	soft wh metal	156.60	2072	7.31		s acid
1262	Indium antimonide	InSb	1312-41-0	236.578	blk cub cry	524		5.7747		
1263	Indium arsenide	InAs	1303-11-3	189.740	gray cub cry	942		5.67		i acid
1264	Indium nitride	InN	25617-98-5	128.825	brn hex cry	1100		6.88		
1265	Indium phosphide	InP	22398-80-7	145.792	blk cub cry	1062		4.81		sl acid
1266	Indium(I) bromide	InBr	14280-53-6	194.722	oran-red orth cry	285	656	4.96		reac H_2O
1267	Indium(I) chloride	InCl	13465-10-6	150.271	yel cub cry	225	608	4.19		reac H_2O
1268	Indium(I) iodide	InI	13966-94-4	241.722	orth cry	364.4	712	5.32		
1269	Indium(II) bromide	$InBr_2$	21264-43-7	274.626	orth cry			4.22		reac H_2O
1270	Indium(II) chloride	$InCl_2$	13465-11-7	185.724	col orth cry	235		3.64		reac H_2O
1271	Indium(II) sulfide	InS	12030-14-7	146.883	red-brn orth cry	692		5.2		
1272	Indium(III) bromide	$InBr_3$	13465-09-3	354.530	hyg yel-wh monocl cry	420		4.74	414^{20}	
1273	Indium(III) chloride	$InCl_3$	10025-82-8	221.177	yel monocl cry; hyg	583		4.0	195.1^{22}	s EtOH
1274	Indium(III) chloride tetrahydrate	$InCl_3 \cdot 4H_2O$	22519-64-8	293.239	wh cry					s H_2O
1275	Indium(III) fluoride	InF_3	7783-52-0	171.813	wh hex cry; hyg	1172	>1200	4.39		sl H_2O; s dil acid
1276	Indium(III) fluoride trihydrate	$InF_3 \cdot 3H_2O$	14166-78-0	225.859	wh cry	100 dec				s H_2O
1277	Indium(III) hydroxide	$In(OH)_3$	20661-21-6	165.840	cub cry			4.4		
1278	Indium(III) iodide	InI_3	13510-35-5	495.531	yel-red monocl cry; hyg	207		4.69	1308^{22}	
1279	Indium(III) nitrate trihydrate	$In(NO_3)_3 \cdot 3H_2O$	13770-61-1	354.879	col cry	dec 100				
1280	Indium(III) oxide	In_2O_3	1312-43-2	277.634	yel cub cry	1912		7.18		i H_2O; s hot acid
1281	Indium(III) perchlorate octahydrate	$In(ClO_4)_3 \cdot 8H_2O$	13465-15-1	557.292	wh cry	≈80	200 dec			
1282	Indium(III) phosphate	$InPO_4$	14693-82-4	209.789	wh orth cry			4.9		i H_2O
1283	Indium(III) selenide	In_2Se_3	1312-42-1	466.52	blk hex cry	660		5.8		
1284	Indium(III) sulfate	$In_2(SO_4)_3$	13464-82-9	517.824	hyg wh powder			3.44	117^{20}	
1285	Indium(III) sulfide	In_2S_3	12030-24-9	325.831	oran cub cry	1050		4.45		
1286	Indium(III) telluride	In_2Te_3	1312-45-4	612.44	blk cub cry	667		5.75		
1287	Iodine	I_2	7553-56-2	253.809	blue-blk plates	113.7	184.4	4.933	0.03^{20}	s bz, EtOH, eth, ctc, chl
1288	Iodic acid	HIO_3	7782-68-5	175.910	col orth cry	110 dec		4.63	308^{25}	i EtOH, eth
1289	Periodic acid dihydrate	$HIO_4 \cdot 2H_2O$	10450-60-9	227.940	monocl hyg cry	122 dec				s H_2O, EtOH; sl eth
1290	Iodine tetroxide	I_2O_4	12399-08-5	317.807	yel cry	130	dec >85	4.2		sl H_2O
1291	Iodine pentoxide	I_2O_5	12029-98-0	333.806	hyg wh cry	≈300 dec		4.98	253.4^{20}	i EtOH, eth, CS_2
1292	Iodine hexoxide	I_2O_6	65355-99-9	349.805	yel solid	dec 150				reac H_2O
1293	Iodine nonaoxide	I_4O_9	73560-00-6	651.613	hyg yel powder	75 dec				
1294	Iodine bromide	IBr	7789-33-5	206.808	blk orth cry	40	116 dec	4.3		s H_2O, EtOH, eth
1295	Iodine chloride	ICl	7790-99-0	162.357	red cry or oily liq	27.38	94.4 dec	3.24		reac H_2O; s EtOH
1296	Iodine trichloride	ICl_3	865-44-1	233.263	yel tricl cry; hyg	101 tp (16 atm)	64 sp dec	3.2		reac H_2O; s EtOH, bz
1297	Iodine fluoride	IF	13873-84-2	145.902	wh pow (-78°C)	-14 dec				
1298	Iodine trifluoride	IF_3	22520-96-3	183.899	yel solid, stable at low temp	-28 dec				
1299	Iodine pentafluoride	IF_5	7783-66-6	221.896	yel liq	9.43	100.5	3.19		reac H_2O
1300	Iodine heptafluoride	IF_7	16921-96-3	259.893	col gas	6.5 tp	4.8 sp	10.62 g/L		s H_2O
1301	Iodosyl trifluoride	IOF_3	19058-78-7	199.898	hyg col needles	dec >110				reac H_2O
1302	Iodosyl pentafluoride	IOF_5	16056-61-4	237.895	col liq	4.5				
1303	Iodyl trifluoride	IO_2F_3	25402-50-0	215.898	yel solid	41	subl			
1304	Periodyl fluoride	IO_3F	30708-86-2	193.900	col cry	dec >100				
1305	Iridium	Ir	7439-88-5	192.217	silv-wh metal; cub	2446	4428	22.562^{20}		s aqua regia
1306	Iridium carbonyl	$Ir_4(CO)_{12}$	11065-24-0	1104.989	yel cry	210 dec				
1307	Iridium(III) bromide	$IrBr_3$	10049-24-8	431.929	red-brn monocl cry			6.82		i H_2O, acid, alk

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1308	Iridium(III) bromide tetrahydrate	$IrBr_3 \cdot 4H_2O$	10049-24-8*	503.991	grn-brn cry					s H_2O; i EtOH
1309	Iridium(III) chloride	$IrCl_3$	10025-83-9	298.576	brn monocl cry	763 dec		5.30		i H_2O, acid, alk
1310	Iridium(III) fluoride	IrF_3	23370-59-4	249.212	blk hex cry	250 dec		≈8.0		i H_2O, dil acid
1311	Iridium(III) iodide	IrI_3	7790-41-2	572.930	dark brn monocl cry			≈7.4		i H_2O, acid, bz, chl; s alk
1312	Iridium(III) oxide	Ir_2O_3	1312-46-5	432.432	blue-blk cry	1000 dec				i H_2O; sl hot HCl
1313	Iridium(III) 2,4-pentanedioate	$Ir(CH_3COCHCOCH_3)_3$	15635-87-7	489.541	oran-yel cry	270	subl			sl H_2O; s tol, chl, ace, MeOH
1314	Iridium(III) sulfide	Ir_2S_3	12136-42-4	480.629	orth cry			10.2		
1315	Iridium(IV) chloride	$IrCl_4$	10025-97-5	334.029	brn hyg solid	≈700 dec				s H_2O, EtOH
1316	Iridium(IV) oxide	IrO_2	12030-49-8	224.216	brn tetr cry	1100 dec		11.7		
1317	Iridium(IV) sulfide	IrS_2	12030-51-2	256.347	orth cry			9.3		
1318	Iridium(VI) fluoride	IrF_6	7783-75-7	306.207	yel cub cry; hyg	44	53.6	4.8		reac H_2O
1319	Iron	Fe	7439-89-6	55.845	silv-wh or gray met	1538	2861	7.87		s dil acid
1320	Ferrocene	$Fe(C_5H_5)_2$	102-54-5	186.031	oran needles	172.5	249			i H_2O; s EtOH, eth, bz, dil HNO_3
1321	Tetracarbonyldihydroiron	$Fe(CO)_4H_2$	12002-28-7	169.902	col liq, stab low temp	-70	dec -20			s alk
1322	Iron pentacarbonyl	$Fe(CO)_5$	13463-40-6	195.896	yel oily liq; flam	-20.5	103	1.46		i H_2O; s eth, bz, ace
1323	Iron nonacarbonyl	$Fe_2(CO)_9$	15321-51-4	363.781	oran-yel cry	100 dec		2.85		
1324	Iron dodecacarbonyl	$Fe_3(CO)_{12}$	12088-65-2	503.656	dark grn cry	140		2.00		
1325	Iron arsenide	FeAs	12044-16-5	130.767	gray orth cry	1030		7.85		
1326	Iron boride (FeB)	FeB	12006-84-7	66.656	refrac solid; orth	1658		≈7		
1327	Iron boride (Fe_2B)	Fe_2B	12006-86-9	122.501	refrac solid; tetr	1389		7.3		
1328	Iron carbide	Fe_3C	12011-67-5	179.546	gray cub cry	1227		7.694		
1329	Iron phosphide (FeP)	FeP	26508-33-8	86.819	rhom cry			6.07		
1330	Iron phosphide (Fe_2P)	Fe_2P	1310-43-6	142.664	gray hex needles	1370		6.8		i H_2O, dil acid, alk
1331	Iron phosphide (Fe_3P)	Fe_3P	12023-53-9	198.509	gray solid	1100		6.74		i H_2O
1332	Iron disulfide	FeS_2	1317-66-4	119.975	blk cub cry	>600 dec		5.02		i H_2O
1333	Iron silicide	FeSi	12022-95-6	83.931	gray cub cry	1410		6.1		
1334	Iron disilicide	$FeSi_2$	12022-99-0	112.016	gray tetr cry	1220		4.74		
1335	Iron(II) acetate	$Fe(C_2H_3O_2)_2$	3094-87-9	173.934	wh cry	190 dec				s H_2O
1336	Iron(II) acetate tetrahydrate	$Fe(C_2H_3O_2)_2 \cdot 4H_2O$	3094-87-9*	245.994	grn cry	dec				s H_2O, EtOH
1337	Iron(II) aluminate	$Fe(AlO_2)_2$	12068-49-4	173.806	blk cub cry			4.3		
1338	Iron(II) arsenate	$Fe_3(AsO_4)_2$	10102-50-8	445.373	grn powder					i H_2O
1339	Iron(II) arsenate hexahydrate	$Fe_3(AsO_4)_2 \cdot 6H_2O$	10102-50-8*	553.465	grn amorp powder	dec				i H_2O; s acid
1340	Iron(II) bromide	$FeBr_2$	7789-46-0	215.653	yel-brn hex cry; hyg	691	dec	4.636	120^{25}	vs EtOH
1341	Iron(II) bromide hexahydrate	$FeBr_2 \cdot 6H_2O$	13463-12-2	323.744	grn hyg cry	27 dec		4.64	120^{25}	s EtOH
1342	Iron(II) carbonate	$FeCO_3$	563-71-3	115.854	gray-brn hex cry			3.944	0.000062^{20}	
1343	Iron(II) chloride	$FeCl_2$	7758-94-3	126.751	wh hex cry; hyg	677	1023	3.16	65.0^{25}	vs EtOH, ace; sl bz
1344	Iron(II) chloride dihydrate	$FeCl_2 \cdot 2H_2O$	16399-77-2	162.782	wh-grn monocl cry	120 dec		2.39	65.0^{25}	
1345	Iron(II) chloride tetrahydrate	$FeCl_2 \cdot 4H_2O$	13478-10-9	198.813	grn monocl cry	105 dec		1.93	65.0^{25}	s EtOH
1346	Iron(II) chromite	$FeCr_2O_4$	1308-31-2	223.835	blk cub cry			5.0		
1347	Iron(II) fluoride	FeF_2	7789-28-8	93.842	wh tetr cry	1100		4.09		sl H_2O; s dil HF; i EtOH, eth
1348	Iron(II) fluoride tetrahydrate	$FeF_2 \cdot 4H_2O$	13940-89-1	165.904	col hex cry			2.20		
1349	Iron(II) hydroxide	$Fe(OH)_2$	18624-44-7	89.860	wh-grn hex cry			3.4	0.000052^{20}	
1350	Iron(II) iodide	FeI_2	7783-86-0	309.654	red-viol hex cry; hyg	594		5.3		s H_2O, EtOH, eth
1351	Iron(II) iodide tetrahydrate	$FeI_2 \cdot 4H_2O$	7783-86-0*	381.716	blk hyg leaflets	90 dec		2.87		s H_2O, EtOH
1352	Iron(II) molybdate	$FeMoO_4$	13718-70-2	215.78	brn-yel monocl cry	1115		5.6		i H_2O
1353	Iron(II) nitrate	$Fe(NO_3)_2$	14013-86-6	179.854	grn solid				87.5^{25}	
1354	Iron(II) nitrate hexahydrate	$Fe(NO_3)_2 \cdot 6H_2O$	14013-86-6*	287.946	grn solid	60 dec			87.5^{25}	
1355	Iron(II) orthosilicate	Fe_2SiO_4	10179-73-4	203.774	brn orth cry			4.30		
1356	Iron(II) oxalate dihydrate	$FeC_2O_4 \cdot 2H_2O$	6047-25-2	179.894	yel cry	150 dec		2.28	0.078^{25}	s acid
1357	Iron(II) oxide	FeO	1345-25-1	71.844	blk cub cry	1377		6.0		i H_2O, alk; s acid
1358	Iron(II) 2,4-pentanedioate	$Fe(CH_3COCHCOCH_3)_2$	14024-17-0	254.061	oran-brn cry	170	subl			sl bz, tol
1359	Iron(II) perchlorate	$Fe(ClO_4)_2$	13933-23-8	254.746	grn-wh hyg needles	>100 dec			210^{25}	
1360	Iron(II) phosphate octahydrate	$Fe_3(PO_4)_2 \cdot 8H_2O$	14940-41-1	501.600	gray-blue monocl cry; hyg			2.58		i H_2O; s acid
1361	Iron(II) selenide	FeSe	1310-32-3	134.81	blk hex cry			6.7		i H_2O
1362	Iron(II) sulfate	$FeSO_4$	7720-78-7	151.908	wh orth cry; hyg			3.65	29.5^{25}	
1363	Iron(II) sulfate monohydrate	$FeSO_4 \cdot H_2O$	17375-41-6	169.923	wh-yel monocl cry	300 dec		3.0	29.5^{25}	

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1364	Iron(II) sulfate heptahydrate	$FeSO_4 \cdot 7H_2O$	7782-63-0	278.014	blue-grn monocl cry	≈60 dec		1.895	29.5^{25}	i EtOH
1365	Iron(II) sulfide	FeS	1317-37-9	87.910	col hex or tetr cry; hyg	1188	dec	4.7		i H_2O; reac acid
1366	Iron(II) tantalate	$Fe(TaO_3)_2$	12140-41-9	513.737	brn tetr cry			7.33		
1367	Iron(II) tartrate	$FeC_4H_4O_6$		203.916	wh cry				0.88	vs acid; s NH_4OH
1368	Iron(II) telluride	FeTe	12125-63-2	183.45	tetr cry	914		6.8		
1369	Iron(II) thiocyanate trihydrate	$Fe(SCN)_2 \cdot 3H_2O$	6010-09-9	226.055	grn monocl cry					s H_2O, EtOH, eth
1370	Iron(II) titanate	$FeTiO_3$	12168-52-4	151.710	blk rhomb cry	≈1470		4.72		
1371	Iron(II) tungstate	$FeWO_4$	13870-24-1	303.68	monocl cry			7.51		
1372	Iron(II,III) oxide	Fe_3O_4	1317-61-9	231.533	blk cub cry or amorp powder	1597		5.17		i H_2O; s acid
1373	Iron(III) acetate, basic	$FeOH(C_2H_3O_2)_2$	10450-55-2	190.941	brn-red amorp powder					i H_2O; s EtOH, acid
1374	Iron(III) ammonium citrate	$Fe(NH_4)_3(C_6H_5O_7)_2$	1185-57-5	488.160	red or brn pow; hyg					s H_2O; i EtOH
1375	Iron(III) arsenate dihydrate	$FeAsO_4 \cdot 2H_2O$	10102-49-5	230.795	grn-brn powder	dec		3.18		i H_2O; s dil acid
1376	Iron(III) bromide	$FeBr_3$	10031-26-2	295.557	dark red hex cry; hyg	dec		4.5	455^{25}	s EtOH, eth
1377	Iron(III) chloride	$FeCl_3$	7705-08-0	162.204	grn hex cry; hyg	307.6	≈316	2.90	91.2^{25}	s EtOH, eth, ace
1378	Iron(III) chloride hexahydrate	$FeCl_3 \cdot 6H_2O$	10025-77-1	270.295	yel-oran monocl cry; hyg	37 dec		1.82	91.2^{25}	s EtOH, eth, ace
1379	Iron(III) chromate	$Fe_2(CrO_4)_3$	10294-52-7	459.671	yel powder					i H_2O, EtOH; s acid
1380	Iron(III) citrate pentahydrate	$FeC_6H_5O_7 \cdot 5H_2O$	3522-50-7	335.021	red-brn cry					s H_2O; i EtOH
1381	Iron(III) dichromate	$Fe_2(Cr_2O_7)_3$	10294-53-8	759.654	red-brn solid					s H_2O, acid
1382	Iron(III) ferrocyanide	$Fe_4[Fe(CN)_6]_3$	14038-43-8	859.229	dark blue powder			1.80		i H_2O, dil acid, os
1383	Iron(III) fluoride	FeF_3	7783-50-8	112.840	grn hex cry	>1000		3.87	5.92^{25}	i EtOH, eth, bz
1384	Iron(III) fluoride trihydrate	$FeF_3 \cdot 3H_2O$	15469-38-2	166.886	yel-brn tetr cry			2.3	5.92^{25}	
1385	Iron(III) formate	$Fe(CHO_2)_3$	555-76-0	190.897	red-yel cry powder					s H_2O; sl EtOH
1386	Iron(III) hydroxide	$Fe(OH)_3$	1309-33-7	106.867	yel monocl cry			3.12		
1387	Iron(III) hydroxide oxide	FeO(OH)	20344-49-4	88.852	red-brn orth cry			4.26		i H_2O; s acid
1388	Iron(III) metavanadate	$Fe(VO_3)_3$	65842-03-7	352.665	gray-brn powder					i H_2O, EtOH; s acid
1389	Iron(III) nitrate	$Fe(NO_3)_3$	10421-48-4	241.860	cry				82.5^{20}	
1390	Iron(III) nitrate hexahydrate	$Fe(NO_3)_3 \cdot 6H_2O$	13476-08-9	349.951	viol cub cry	35 dec			82.5^{20}	
1391	Iron(III) nitrate nonahydrate	$Fe(NO_3)_3 \cdot 9H_2O$	7782-61-8	403.997	viol-gray hyg cry	47 dec		1.68	82.5^{20}	vs EtOH, ace
1392	Iron(III) oxalate	$Fe_2(C_2O_4)_3$	19469-07-9	375.747	yel amorp powder	100 dec				s H_2O, acid; i alk
1393	Iron(III) oxide	Fe_2O_3	1309-37-1	159.688	red-brn hex cry	1539		5.25		i H_2O; s acid
1394	Iron(III) 2,4-pentanedioate	$Fe(CH_3COCHCOCH_3)_3$	14024-18-1	353.169	red-oran cry	179		5.24		sl H_2O; s os
1395	Iron(III) perchlorate hexahydrate	$Fe(ClO_4)_3 \cdot 6H_2O$	32963-81-8	462.288	viol cry					
1396	Iron(III) phosphate dihydrate	$FePO_4 \cdot 2H_2O$	10045-86-0	186.847	gray-wh orth cry			2.87		i H_2O; s HCl
1397	Iron(III) pyrophosphate nonahydrate	$Fe_4(P_2O_7)_3 \cdot 9H_2O$	10058-44-3	907.348	yel powder					i H_2O; s acid
1398	Iron(III) hypophosphite	$Fe(H_2PO_2)_3$	7783-84-8	250.811	wh-gray powder					i H_2O
1399	Iron(III) sodium pyrophosphate	$FeNaP_2O_7$	10045-87-1	252.778	wh powder			1.5		i H_2O; s HCl
1400	Iron(III) sulfate	$Fe_2(SO_4)_3$	10028-22-5	399.878	gray-wh rhomb cry; hyg			3.10	440^{20}	sl EtOH; i ace
1401	Iron(III) sulfate nonahydrate	$Fe_2(SO_4)_3 \cdot 9H_2O$	13520-56-4	562.015	yel hex cry	400 dec		2.1	440^{20}	
1402	Iron(III) thiocyanate	$Fe(SCN)_3$	4119-52-2	230.092	red-viol hyg cry	dec				s H_2O, EtOH, ace; i tol, chl
1403	Krypton	Kr	7439-90-9	83.798	col gas	-157.38 tp (73.2 kPa)	-153.34	3.425 g/L		sl H_2O
1404	Krypton difluoride	KrF_2	13773-81-4	121.795	col tetr cry	≈25 dec		3.24		reac H_2O
1405	Krypton fluoride hexafluoroantimonate	$KrFSb_2F_{11}$	39578-36-4	555.299	wh solid	dec 45				
1406	Lanthanum	La	7439-91-0	138.905	silv metal; hex	920	3464	6.15		s dil acid
1407	Lanthanum aluminum oxide	$LaAlO_3$	12003-65-5	213.885	wh rhom cry	trans cub 500				
1408	Lanthanum boride	LaB_6	12008-21-8	203.771	blk cub cry; refrac	2715		4.76		
1409	Lanthanum bromate nonahydrate	$LaBrO_3 \cdot 9H_2O$		684.749	hex cry	dec 100		5.06		vs H_2O
1410	Lanthanum bromide	$LaBr_3$	13536-79-3	378.617	wh hex cry; hyg	788		5.1		s H_2O
1411	Lanthanum carbide	LaC_2	12071-15-7	162.926	tetr cry	2360		5.29		
1412	Lanthanum carbonate octahydrate	$La_2(CO_3)_3 \cdot 8H_2O$	6487-39-4	601.960	wh cry powder			2.6		i H_2O; s dil acid
1413	Lanthanum chloride	$LaCl_3$	10099-58-8	245.264	wh hex cry; hyg	858		3.84	95.7^{25}	
1414	Lanthanum chloride heptahydrate	$LaCl_3 \cdot 7H_2O$	20211-76-1	371.371	wh tricl cry; hyg	91 dec			95.7^{25}	s EtOH
1415	Lanthanum fluoride	LaF_3	13709-38-1	195.900	wh hex cry; hyg	1493		5.9		i H_2O, acid
1416	Lanthanum hydride	LaH_3	13864-01-2	141.929	blk cub cry			5.36		
1417	Lanthanum hydroxide	$La(OH)_3$	14507-19-8	189.927	wh amorp solid	dec			0.000020^{20}	
1418	Lanthanum iodate	$La(IO_3)_3$	13870-19-4	663.614	col cry				1.7	
1419	Lanthanum iodide	LaI_3	13813-22-4	519.619	wh orth cry; hyg	778		5.6		s H_2O
1420	Lanthanum nitrate hexahydrate	$La(NO_3)_3 \cdot 6H_2O$	10277-43-7	433.011	wh hyg tricl cry	≈40 dec			200^{25}	vs EtOH; s ace

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1421	Lanthanum nitride	LaN	25764-10-7	152.912	cub cry			6.73		
1422	Lanthanum oxide	La_2O_3	1312-81-8	325.809	wh amorp powder	2304	3620	6.51		i H_2O; s dil acid
1423	Lanthanum perchlorate hexahydrate	$La(ClO_4)_3 \cdot 6H_2O$	36907-37-6	475.021	hyg col cry	dec 100				vs H_2O; s EtOH
1424	Lanthanum silicide	$LaSi_2$	12056-90-5	195.076	gray tetr cry	1520		5.0		
1425	Lanthanum sulfate	$La_2(SO_4)_3$	10099-60-2	565.999	hyg wh pow	1150				sl H_2O
1426	Lanthanum sulfate octahydrate	$La_2(SO_4)_3 \cdot 8H_2O$	57804-25-8	702.058	col cry	dec		2.82		sl H_2O
1427	Lanthanum sulfate nonahydrate	$La_2(SO_4)_3 \cdot 9H_2O$	10294-62-9	728.136	hex cry			2.82	2.7^{20}	i EtOH
1428	Lanthanum monosulfide	LaS	12031-30-0	170.970	yel cub cry	2300		5.61		
1429	Lanthanum sulfide	La_2S_3	12031-49-1	374.006	red cub cry	2110		4.9		
1430	Lawrencium	Lr	22537-19-5	262	metal	1627				
1431	Lead	Pb	7439-92-1	207.2	soft silv-gray metal; cub	327.462	1749	11.3		s conc acid
1432	Plumbane	PbH_4	15875-18-0	211.2	unstab col gas		-13			
1433	Lead(II) acetate	$Pb(C_2H_3O_2)_2$	301-04-2	325.3	wh cry	280	dec	3.25	44.3^{20}	
1434	Lead(II) acetate trihydrate	$Pb(C_2H_3O_2)_2 \cdot 3H_2O$	6080-56-4	427.3	col cry	75 dec		2.55		vs H_2O; sl EtOH
1435	Lead(II) acetate, basic	$Pb(C_2H_3O_2)_2 \cdot 2Pb(OH)_2$	1335-32-6	807.7	wh powder	dec			6.3^{0}	
1436	Lead(II) antimonate	$Pb_3(SbO_4)_2$	13510-89-9	993.1	oran-yel powder			6.58		i H_2O, dil acid
1437	Lead(II) arsenate	$Pb_3(AsO_4)_2$	3687-31-8	899.4	wh cry	1042 dec		5.8		i H_2O; s HNO_3
1438	Lead(II) arsenite	$Pb(AsO_2)_2$	10031-13-7	421.0	wh powder			5.85		i H_2O; s dil HNO_3
1439	Lead(II) azide	$Pb(N_3)_2$	13424-46-9	291.2	col orth needles; exp	exp ≈350		4.7	0.023^{18}	vs HOAc
1440	Lead(II) borate monohydrate	$Pb(BO_2)_2 \cdot H_2O$	10214-39-8	310.8	wh powder	500 dec		5.6		i H_2O; s dil HNO_3
1441	Lead(II) bromate monohydrate	$Pb(BrO_3)_2 \cdot H_2O$	10031-21-7	481.0	col cry	≈180 dec		5.53	1.33^{20}	
1442	Lead(II) bromide	$PbBr_2$	10031-22-8	367.0	wh orth cry	371	892	6.69	0.975^{25}	i EtOH
1443	Lead(II) butanoate	$Pb(C_4H_7O_2)_2$	819-73-8	381.4	col solid	≈90				i H_2O; s dil HNO_3
1444	Lead(II) carbonate	$PbCO_3$	598-63-0	267.2	col orth cry	≈315 dec		6.582		i H_2O
1445	Lead(II) carbonate, basic	$Pb(OH)_2 \cdot 2PbCO_3$	1319-46-6	775.6	wh hex cry	400 dec		≈6.5		i H_2O, EtOH; s acid
1446	Lead(II) chlorate	$Pb(ClO_3)_2$	10294-47-0	374.1	col hyg cry	230 dec		3.9	144^{18}	vs EtOH
1447	Lead(II) chloride	$PbCl_2$	7758-95-4	278.1	wh orth needles or powder	501	951	5.98	1.08^{25}	s alk
1448	Lead(II) chloride fluoride	PbClF	13847-57-9	261.7	tetr cry			7.05	0.035^{20}	
1449	Lead(II) chlorite	$Pb(ClO_2)_2$	13453-57-1	342.1	yel monocl cry	dec 126			0.2^{25}	sl H_2O; s alk
1450	Lead(II) chromate	$PbCrO_4$	7758-97-6	323.2	yel-oran monocl cry	844		6.12	0.000017^{20}	s alk, dil acid
1451	Lead(II) chromate(VI) oxide	$PbCrO_4 \cdot PbO$	18454-12-1	546.4	red powder					i H_2O
1452	Lead(II) citrate trihydrate	$Pb_3(C_6H_5O_7)_2 \cdot 3H_2O$	512-26-5	1053.8	wh cry powder					s H_2O; sl EtOH
1453	Lead(II) cyanide	$Pb(CN)_2$	592-05-2	259.2	wh-yel powder					sl H_2O; reac acid
1454	Lead(II) 2-ethylhexanoate	$Pb(C_7H_{15}CO_2)_2$	301-08-6	493.6	visc liq			1.56		
1455	Lead(II) fluoride	PbF_2	7783-46-2	245.2	wh orth cry	830	1293	8.44	0.0670^{25}	
1456	Lead(II) fluoroborate	$Pb(BF_4)_2$	13814-96-5	380.8	stab only in aq soln					s H_2O
1457	Lead(II) formate	$Pb(CHO_2)_2$	811-54-1	297.2	wh prisms or needles	190 dec		4.63	1.6^{16}	i EtOH
1458	Lead(II) hexafluoro-2,4-pentanedioate	$Pb(CF_3COCHCOCF_3)_2$	19648-88-5	621.3	cry	155	210			
1459	Lead(II) hydrogen arsenate	$PbHAsO_4$	7784-40-9	347.1	wh monocl cry	280 dec		5.943		i H_2O; s HNO_3, alk
1460	Lead(II) hydrogen phosphate	$PbHPO_4$	15845-52-0	303.2	wh monocl cry	dec		5.66		
1461	Lead(II) hydroxide	$Pb(OH)_2$	19783-14-3	241.2	wh powder	145 dec		7.59	0.00012^{20}	s acid
1462	Lead(II) iodate	$Pb(IO_3)_2$	25659-31-8	557.0	wh orth cry			6.50	0.0025^{25}	
1463	Lead(II) iodide	PbI_2	10101-63-0	461.0	yel hex cry or powder	410	872 dec	6.16	0.076^{25}	i EtOH
1464	Lead(II) lactate	$Pb(C_3H_5O_3)_2$	18917-82-3	385.3	wh cry powder					s H_2O, hot EtOH
1465	Lead(II) molybdate	$PbMoO_4$	10190-55-3	367.1	yel tetr cry	≈1060		6.7		i H_2O; s HNO_3, NaOH
1466	Lead(II) niobate	$Pb(NbO_3)_2$	12034-88-7	489.0	rhom or tetr cry	1343		6.6		i H_2O
1467	Lead(II) nitrate	$Pb(NO_3)_2$	10099-74-8	331.2	col cub cry	470		4.53	59.7^{25}	sl EtOH
1468	Lead(II) oleate	$Pb(C_{18}H_{33}O_2)_2$	1120-46-3	770.1	wax-like solid					i H_2O; s EtOH, bz, eth
1469	Lead(II) oxalate	PbC_2O_4	814-93-7	295.2	wh powder	300 dec		5.28	0.00025^{20}	s dil HNO_3
1470	Lead(II) oxide (litharge)	PbO	1317-36-8	223.2	red tetr cry	trans to massicot 489		9.35		i H_2O, EtOH; s dil HNO_3
1471	Lead(II) oxide (massicot)	PbO	1317-36-8	223.2	yel orth cry	887		9.64		i H_2O, EtOH; s dil HNO_3
1472	Lead(II) oxide hydrate	$3PbO \cdot H_2O$	1311-11-1	687.6	wh powder			7.41		i H_2O; s dil acid
1473	Lead(II) 2,4-pentanedioate	$Pb(CH_3COCHCOCH_3)_2$	15282-88-9	405.4	cry	143				
1474	Lead(II) perchlorate	$Pb(ClO_4)_2$	13453-62-8	406.1	wh cry				441^{25}	
1475	Lead(II) perchlorate trihydrate	$Pb(ClO_4)_2 \cdot 3H_2O$	13637-76-8	460.1	wh cry	100 dec		2.6	441^{25}	s EtOH

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1476	Lead(II) phosphate	$Pb_3(PO_4)_2$	7446-27-7	811.5	wh hex cry	1014		7.01		i H_2O, EtOH
1477	Lead(II) hypophosphite	$Pb(H_2PO_2)_2$	10294-58-3	337.2	hyg cry powder	dec				sl H_2O; i EtOH
1478	Lead(II) metasilicate	$PbSiO_3$	10099-76-0	283.3	wh monocl cry powder	764		6.49		i H_2O, os
1479	Lead(II) orthosilicate	Pb_2SiO_4	13566-17-1	506.5	monocl cry	743		7.60		
1480	Lead(II) hexafluorosilicate dihydrate	$PbSiF_6 \cdot 2H_2O$	1310-03-8	385.3	col cry	dec				vs H_2O
1481	Lead(II) selenate	$PbSeO_4$	7446-15-3	350.2	orth cry			6.37	0.013^{25}	s conc acid
1482	Lead(II) selenide	PbSe	12069-00-0	286.2	gray cub cry	1078		8.1		i H_2O; s HNO_3
1483	Lead(II) selenite	$PbSeO_3$	7488-51-9	334.2	wh monocl cry	≈500		7.0		i H_2O
1484	Lead(II) sodium thiosulfate	$Na_4Pb(S_2O_3)_3$	10101-94-7	635.5	wh cry					sl H_2O
1485	Lead(II) stearate	$Pb(C_{18}H_{35}O_2)_2$	1072-35-1	774.1	wh powder	≈100		1.4		i H_2O; s hot EtOH
1486	Lead(II) sulfate	$PbSO_4$	7446-14-2	303.3	orth cry	1087		6.29	0.0044^{25}	i acid; sl alk
1487	Lead(II) sulfide	PbS	1314-87-0	239.3	blk powder or silv cub cry	1113		7.60		i H_2O; s acid
1488	Lead(II) sulfite	$PbSO_3$	7446-10-8	287.3	wh powder	dec				i H_2O; s HNO_3
1489	Lead(II) tantalate	$Pb(TaO_3)_2$	12065-68-8	665.1	orth cry			7.9		i H_2O
1490	Lead(II) telluride	PbTe	1314-91-6	334.8	gray cub cry	924		8.164		i H_2O, acid
1491	Lead(II) thiocyanate	$Pb(SCN)_2$	592-87-0	323.4	wh-yel powder			3.82	0.05^{20}	
1492	Lead(II) thiosulfate	PbS_2O_3	13478-50-7	319.3	wh cry	dec		5.18		i H_2O; s acid
1493	Lead(II) titanate	$PbTiO_3$	12060-00-3	303.1	yel tetr cry			7.9		i H_2O; reac HCl
1494	Lead(II) tungstate (stolzite)	$PbWO_4$	7759-01-5	455.0	yel tetr cry	1130		8.24	0.03^{20}	s alk
1495	Lead(II) tungstate (raspite)	$PbWO_4$	7759-01-5	455.0	monocl cry	trans 400		8.46	0.03^{20}	s alk
1496	Lead(II) metavanadate	$Pb(VO_3)_2$	10099-79-3	405.1	yel powder					i H_2O; reac HNO_3
1497	Lead(II) zirconate	$PbZrO_3$	12060-01-4	346.4	col orth cry			≈8		i H_2O, alk; s acid
1498	Lead(II,IV) oxide	Pb_2O_3	1314-27-8	462.4	blk monocl cry or red amorp powder	530 dec		10.05		i H_2O; s alk; reac conc HCl
1499	Lead(II,II,IV) oxide	Pb_3O_4	1314-41-6	685.6	red tetr cry	830		8.92		i H_2O, EtOH; s hot HCl
1500	Lead(IV) acetate	$Pb(C_2H_3O_2)_4$	546-67-8	443.4	col monocl cry	≈175		2.23		reac H_2O, EtOH; s bz, chl
1501	Lead(IV) bromide	$PbBr_4$	13701-91-2	526.8	unstab liq					
1502	Lead(IV) chloride	$PbCl_4$	13463-30-4	349.0	yel oily liq	-15	≈50 dec			
1503	Lead(IV) fluoride	PbF_4	7783-59-7	283.2	wh tetr cry; hyg	≈600		6.7		
1504	Lead(IV) oxide	PbO_2	1309-60-0	239.2	red tetr cry or brn powder	290 dec		9.64		
1505	Lithium	Li	7439-93-2	6.941	soft silv-wh metal	180.50	1342	0.534		reac H_2O
1506	Lithium acetate	$LiC_2H_3O_2$	546-89-4	65.985	cry	286			45.0^{25}	vs EtOH
1507	Lithium acetate dihydrate	$LiC_2H_3O_2 \cdot 2H_2O$	6108-17-4	102.016	wh rhomb cry	58 dec		1.3	45.0^{25}	s EtOH
1508	Lithium aluminum hydride	$LiAlH_4$	16853-85-3	37.955	gray-wh monocl cry	>125 dec		0.917		reac H_2O, EtOH; s eth, thf
1509	Lithium aluminum silicate	$LiAlSi_2O_6$	12068-40-5	186.090	wh monocl cry	1430		3.188		
1510	Lithium amide	$LiNH_2$	7782-89-0	22.964	tetr cry	380		1.18		reac H_2O
1511	Lithium arsenate	Li_3AsO_4	13478-14-3	159.743	col orth cry			3.07		sl H_2O; s HOAc
1512	Lithium azide	LiN_3	19597-69-4	48.961	hyg monocl cry; exp			1.83		vs H_2O
1513	Lithium borohydride	$LiBH_4$	16949-15-8	21.784	wh-gray orth cry or powder	268	380 dec	0.66		s alk, eth, thf
1514	Lithium bromate	$LiBrO_3$	13550-28-2	134.843	hyg col orth cry	260			65.4^{25}	vs H_2O
1515	Lithium bromide	LiBr	7550-35-8	86.845	wh cub cry; hyg	550	≈1300	3.464	181^{25}	s EtOH, eth
1516	Lithium bromide monohydrate	$LiBr \cdot H_2O$	23303-71-1	104.860	wh orth cry	trans cub 33		3.46	145^{4}	vs H_2O
1517	Lithium carbide	Li_2C_2	1070-75-3	37.903	wh hyg cry			1.65		reac H_2O; i os
1518	Lithium carbonate	Li_2CO_3	554-13-2	73.891	wh monocl cry	732	1300 dec	2.11	1.30^{25}	s acid; i EtOH
1519	Lithium chlorate	$LiClO_3$	13453-71-9	90.392	col hyg rhom needles	127.6	300 dec	1.119	459^{25}	vs EtOH; sl ace
1520	Lithium chloride	LiCl	7447-41-8	42.394	wh cub cry or powder; hyg	610	1383	2.07	84.5^{25}	s EtOH, ace, py
1521	Lithium chloride monohydrate	$LiCl \cdot H_2O$	16712-20-2	60.409	hyg wh tetr cry	dec 98		1.78	45.9^{25}	vs H_2O
1522	Lithium chromate dihydrate	$Li_2CrO_4 \cdot 2H_2O$	7789-01-7	165.906	yel orth cry; hyg	75 dec		2.15		vs H_2O; s EtOH
1523	Lithium citrate tetrahydrate	$Li_3C_6H_5O_7 \cdot 4H_2O$	6680-58-6	281.983	wh cry	210 (anh)			75^{25}	vs H_2O; sl EtOH
1524	Lithium cobaltite	$LiCoO_2$	12190-79-3	97.873	dark gray pow					i H_2O
1525	Lithium cyanide	LiCN	2408-36-8	32.959	wh orth cry	160				
1526	Lithium hydride-*d*	LiD	13587-16-1	8.955	hyg wh cry	680		0.82		reac H_2O
1527	Lithium dichromate dihydrate	$Li_2Cr_2O_7 \cdot 2H_2O$	10022-48-7	265.901	yel-red hyg cry	130 dec		2.34		vs H_2O
1528	Lithium dihydrogen phosphate	LiH_2PO_4	13453-80-0	103.928	col hyg cry	>100		2.461	126^{0}	
1529	Lithium diisopropylamide	$LiN(C_3H_7)_2$	4111-54-0	107.123	hyg col cry	dec				s eth; i hc
1530	Lithium ferrosilicon	LiFeSi	64082-35-5	90.872	dark brittle cry					reac H_2O

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1531	Lithium fluoride	LiF	7789-24-4	25.939	wh cub cry or powder	848.2	1673	2.640	0.134^{25}	s acid
1532	Lithium formate monohydrate	$Li(CHO_2) \cdot H_2O$	6108-23-2	69.974	col-wh cry			1.46		s H_2O
1533	Lithium hexafluoroantimonate	$LiSbF_6$	18424-17-4	242.691	hyg pow	dec				
1534	Lithium hexafluoroarsenate	$LiAsF_6$	29935-35-1	310.672	rhom wh cry; hyg					
1535	Lithium hexafluorophosphate	$LiPF_6$	21324-40-3	151.905	wh pow					
1536	Lithium hexafluorosilicate	Li_2SiF_6	17347-95-4	155.958	col hex cry	dec 350				sl ace
1537	Lithium hexafluorostannate	Li_2SnF_6	17029-16-2	246.582	wh pow					
1538	Lithium hydride	LiH	7580-67-8	7.949	gray cub cry or powder; hyg	692		0.78		reac H_2O, EtOH
1539	Lithium hydrogen carbonate	$LiHCO_3$	5006-97-3	67.958	wh pow					sl H_2O
1540	Lithium hydroxide	LiOH	1310-65-2	23.948	col tetr cry	473	1626	1.45	12.5^{25}	sl EtOH
1541	Lithium hydroxide monohydrate	$LiOH \cdot H_2O$	1310-66-3	41.964	wh monocl cry or powder			1.51	12.5^{25}	sl EtOH
1542	Lithium hypochlorite	LiOCl	13840-33-0	58.393	wh pow					vs H_2O
1543	Lithium iodate	$LiIO_3$	13765-03-2	181.843	wh hyg hex cry	450		4.502	77.9^{25}	i EtOH
1544	Lithium iodide	LiI	10377-51-2	133.845	wh cub cry; hyg	469	1171	4.06	165^{25}	
1545	Lithium iodide trihydrate	$LiI \cdot 3H_2O$	7790-22-9	187.891	wh hyg cry	73		2.38	165^{25}	vs EtOH, ace
1546	Lithium manganate	Li_2MnO_3	12163-00-7	116.818	red-brn monocl cry			3.90		i H_2O
1547	Lithium metaborate	$LiBO_2$	13453-69-5	49.751	wh monocl cry; hyg	844		2.18	2.6^{20}	sl H_2O; s EtOH
1548	Lithium metaborate dihydrate	$LiBO_2 \cdot 2H_2O$	15293-74-0	85.782	wh cry pow			1.8		s H_2O
1549	Lithium metaphosphate	$LiPO_3$	13762-75-9	85.913	wh cry or gl solid			1.8		i H_2O
1550	Lithium metasilicate	Li_2SiO_3	10102-24-6	89.966	wh orth needles	1201		2.52		i cold H_2O; reac dil acid
1551	Lithium molybdate	Li_2MoO_4	13568-40-6	173.82	hyg wh cry	702		2.66	44.8^{25}	s H_2O
1552	Lithium niobate	$LiNbO_3$	12031-63-9	147.845	wh hex cry	1240		4.30		
1553	Lithium nitrate	$LiNO_3$	7790-69-4	68.946	col hex cry; hyg	253		2.38	102^{25}	s EtOH
1554	Lithium nitride	Li_3N	26134-62-3	34.830	red hex cry	813		1.27		reac H_2O
1555	Lithium nitrite	$LiNO_2$	13568-33-7	52.947	wh hyg cry	222				vs H_2O
1556	Lithium nitrite monohydrate	$LiNO_2 \cdot H_2O$	13568-33-7*	70.962	col needles	>100		1.615	139.5^{25}	vs H_2O, EtOH
1557	Lithium orthosilicate	$LiSiO_4$	13453-84-4	99.025	wh rhom cry	1256		2.39		
1558	Lithium oxalate	$Li_2C_2O_4$	30903-87-8	101.901	col cry	dec		2.121^{17}		s H_2O; i EtOH, eth
1559	Lithium phosphate	Li_3PO_4	10377-52-3	115.794	wh orth cry	1205		2.46	0.027^{25}	
1560	Lithium oxide	Li_2O	12057-24-8	29.881	wh cub cry	1437		2.013		
1561	Lithium perchlorate	$LiClO_4$	7791-03-9	106.392	wh orth cry or powder	236	430 dec	2.428	58.7^{25}	s EtOH, ace, eth
1562	Lithium perchlorate trihydrate	$LiClO_4 \cdot 3H_2O$	13453-78-6	160.438	wh hex cry	95 dec		1.84		vs H_2O, EtOH, ace; i eth
1563	Lithium peroxide	Li_2O_2	12031-80-0	45.881	wh hex cry			2.31		s H_2O; i EtOH
1564	Lithium selenate monohydrate	$Li_2SeO_4 \cdot H_2O$	7790-71-8	174.86	monocl cry			2.56		vs H_2O
1565	Lithium selenite monohydrate	$Li_2SeO_3 \cdot H_2O$	15593-51-8	158.86	hyg cry					
1566	Lithium stearate	$LiC_{18}H_{35}O_2$	4485-12-5	290.411	cry	≈220				
1567	Lithium sulfate	Li_2SO_4	10377-48-7	109.945	wh monocl cry; hyg	860		2.21	34.2^{25}	
1568	Lithium sulfate monohydrate	$Li_2SO_4 \cdot H_2O$	10102-25-7	127.960	col cry	130 dec		2.06	34.2^{25}	sl EtOH
1569	Lithium sulfide	Li_2S	12136-58-2	45.947	wh cub cry; hyg	1372		1.64		
1570	Lithium tantalate	$LiTaO_3$	12031-66-2	235.887	wh pow	1650				
1571	Lithium tetraborate	$Li_2B_4O_7$	12007-60-2	169.122	wh tetr cry	917			2.9^{20}	sl H_2O
1572	Lithium tetraborate pentahydrate	$Li_2B_4O_7 \cdot 5H_2O$	1303-94-2	259.198	wh cry pow	dec 200				vs H_2O; i EtOH
1573	Lithium tetracyanoplatinate pentahydrate	$Li_2Pt(CN)_4 \cdot 5H_2O$	14402-73-4	403.112	grn-yel cry					sl H_2O
1574	Lithium tetrafluoroborate	$LiBF_4$	14283-07-9	93.746	hyg wh pow	dec				vs H_2O
1575	Lithium thiocyanate	LiSCN	556-65-0	65.024	wh hyg cry				120^{25}	
1576	Lithium titanate	Li_2TiO_3	12031-82-2	109.747	wh pow	1325				i H_2O
1577	Lithium tungstate	Li_2WO_4	13568-45-1	261.72	wh trig pow	740		3.71		s H_2O
1578	Lithium vanadate	$LiVO_3$	15060-59-0	105.881	yel pow		subl 1400			
1579	Lithium zirconate	Li_2ZrO_3	12031-83-3	153.104	wh solid					
1580	Lutetium	Lu	7439-94-3	174.967	silv metal; hex	1663	3402	9.84		s dil acid
1581	Lutetium boride	LuB_4	12688-52-7	218.211	tetr cry	2600		≈7.0		
1582	Lutetium bromide	$LuBr_3$	14456-53-2	414.679	wh hyg cry	1025				vs H_2O
1583	Lutetium chloride	$LuCl_3$	10099-66-8	281.326	wh monocl cry; hyg	925		3.98		s H_2O
1584	Lutetium chloride hexahydrate	$LuCl_3 \cdot 6H_2O$	15230-79-2	389.417	col cry	dec 150				s H_2O, EtOH
1585	Lutetium fluoride	LuF_3	13760-81-1	231.962	orth cry	1182	2200	8.3		i H_2O
1586	Lutetium iodide	LuI_3	13813-45-1	555.680	brn hex cry; hyg	1050		≈5.6		vs H_2O
1587	Lutetium iron oxide	$Lu_3Fe_5O_{12}$	12023-71-1	996.119	cry					
1588	Lutetium nitrate	$Lu(NO_3)_3$	10099-67-9	360.982	hyg col solid					s H_2O, EtOH

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1589	Lutetium nitride	LuN	12125-25-6	188.974	cub cry			11.6		
1590	Lutetium oxide	Lu_2O_3	12032-20-1	397.932	wh cub cry or powder	2427	3980	9.41		
1591	Lutetium perchlorate hexahydrate	$Lu(ClO_4)_3 \cdot 6H_2O$	14646-29-8	581.410	col cry	dec 350 (anh)				s H_2O, MeOH
1592	Lutetium sulfate	$Lu_2(SO_4)_3$	14986-89-1	638.122	wh pow	dec >850				vs H_2O
1593	Lutetium sulfate octahydrate	$Lu_2(SO_4)_3 \cdot 8H_2O$	13473-77-3	782.244	col cry				42.3^{20}	s H_2O
1594	Lutetium sulfide	Lu_2S_3	12163-20-1	446.129	gray rhomb cry	1750 dec		6.26		
1595	Lutetium telluride	Lu_2Te_3	12163-22-3	732.73	orth cry			7.8		
1596	Magnesium	Mg	7439-95-4	24.305	silv-wh metal	650	1090	1.74		s dil acid
1597	Magnesium acetate	$Mg(C_2H_3O_2)_2$	142-72-3	142.394	wh orth/monocl cry	323 dec		1.50	65.6^{25}	
1598	Magnesium acetate monohydrate	$Mg(C_2H_3O_2)_2 \cdot H_2O$	60582-92-5	160.409	orth cry			1.55		
1599	Magnesium acetate tetrahydrate	$Mg(C_2H_3O_2)_2 \cdot 4H_2O$	16674-78-5	214.454	col monocl cry; hyg	80 dec		1.45	65.6^{25}	vs EtOH
1600	Magnesium aluminate	$Mg(AlO_2)_2$	12068-51-8	142.266	col cub cry	2105		3.55		i H_2O
1601	Magnesium aluminum silicate	$Mg_2Al_3(AlSi_5O_{18})$	1302-88-1	584.953	blue cry			2.6		
1602	Magnesium amide	$Mg(NH_2)_2$	7803-54-5	56.350	wh powder; flam	dec		1.39		reac H_2O
1603	Magnesium ammonium phosphate hexahydrate	$MgNH_4PO_4 \cdot 6H_2O$	13478-16-5	245.407	wh pow	dec		1.71		i H_2O, EtOH; s acid
1604	Magnesium antimonide	Mg_3Sb_2	12057-75-9	316.435	hex cry	1245		3.99		
1605	Magnesium arsenide	Mg_3As_2	12044-49-4	222.758	solid	≈1200		3.15		i H_2O
1606	Magnesium diboride	MgB_2	12007-25-9	45.927	hex cry	800 dec		2.57		
1607	Magnesium hexaboride	MgB_6	12008-22-9	89.171	refrac solid	1100 dec				i H_2O
1608	Magnesium dodecaboride	MgB_{12}	12230-32-9	154.037	refrac solid	1300 dec				
1609	Magnesium bromate hexahydrate	$Mg(BrO_3)_2 \cdot 6H_2O$	7789-36-8	388.201	col cub cry	200 dec		2.29	98^{25}	
1610	Magnesium bromide	$MgBr_2$	7789-48-2	184.113	wh hex cry; hyg	711		3.72	102^{25}	
1611	Magnesium bromide hexahydrate	$MgBr_2 \cdot 6H_2O$	13446-53-2	292.204	col monocl cry	165 dec		2.0	102^{25}	s EtOH
1612	Magnesium carbonate	$MgCO_3$	546-93-0	84.314	wh hex cry	990		3.010	0.18^{20}	i EtOH; s acid
1613	Magnesium carbonate dihydrate	$MgCO_3 \cdot 2H_2O$	5145-48-2	120.345	col tricl cry			2.8		i H_2O, ace, NH_4OH
1614	Magnesium carbonate trihydrate	$MgCO_3 \cdot 3H_2O$	14457-83-1	138.360	col monocl cry	165		1.8	0.18^{16}	
1615	Magnesium carbonate pentahydrate	$MgCO_3 \cdot 5H_2O$	61042-72-6	174.390	wh monocl cry	dec >400		3.04	0.38^{16}	
1616	Magnesium carbonate hydroxide tetrahydrate	$4MgCO_3 \cdot Mg(OH)_2 \cdot 4H_2O$	39409-82-0	467.636	wh monocl cry			2.3		
1617	Magnesium carbonate hydroxide pentahydrate	$4MgCO_3 \cdot Mg(OH)_2 \cdot 5H_2O$	56378-72-4	485.652	wh pow	dec 700				i H_2O; s dil acid; i EtOH
1618	Magnesium carbonate dihydroxide trihydrate	$MgCO_3 \cdot Mg(OH)_2 \cdot 3H_2O$	12143-96-3	196.680	wh monocl cry	dec		2.04		
1619	Magnesium chlorate hexahydrate	$Mg(ClO_3)_2 \cdot 6H_2O$	13446-19-0	299.299	wh hyg cry	≈35 dec		1.80	142^{25}	sl EtOH
1620	Magnesium chloride	$MgCl_2$	7786-30-3	95.211	wh hex leaflets; hyg	714	1412	2.325	56.0^{25}	
1621	Magnesium chloride hexahydrate	$MgCl_2 \cdot 6H_2O$	7791-18-6	203.302	wh hyg cry	≈100 dec		1.56	56.0^{25}	s EtOH
1622	Magnesium chromate heptahydrate	$MgCrO_4 \cdot 7H_2O$	13423-61-5*	266.405	yel rhom cry			1.695	54.8^{25}	
1623	Magnesium chromite	$MgCr_2O_4$	12053-26-8	192.295	deep grn cry	2390		4.4		
1624	Magnesium citrate	$Mg_3(C_6H_5O_7)_2$	3344-18-1	451.114	wh cry					sl H_2O
1625	Magnesium citrate tetradecahydrate	$Mg_3(C_6H_5O_7)_2 \cdot 14H_2O$	3344-18-1*	703.328	wh cry pow					sl H_2O; s acid
1626	Magnesium fluoride	MgF_2	7783-40-6	62.302	wh tetr cry	1263	2227	3.148	0.013^{25}	
1627	Magnesium formate dihydrate	$Mg(CHO_2)_2 \cdot 2H_2O$	6150-82-9	150.370	wh cry	dec				s H_2O; i EtOH
1628	Magnesium germanate	Mg_2GeO_4	12025-13-7	185.25	wh prec					i H_2O
1629	Magnesium germanide	Mg_2Ge	1310-52-7	121.25	cub cry	1117		3.09		
1630	Magnesium hydride	MgH_2	7693-27-8	26.321	wh tetr cry	327		1.45		reac H_2O
1631	Magnesium hydrogen phosphate trihydrate	$MgHPO_4 \cdot 3H_2O$	7757-86-0	174.331	wh powder	550 dec		2.13		sl H_2O; s dil acid
1632	Magnesium hydroxide	$Mg(OH)_2$	1309-42-8	58.320	wh hex cry	350		2.37	0.00069^{20}	s dil acid
1633	Magnesium iodate tetrahydrate	$Mg(IO_3)_2 \cdot 4H_2O$	7790-32-1*	446.172	col monocl cry	210 dec		3.3	11.1^{25}	
1634	Magnesium iodide	MgI_2	10377-58-9	278.114	wh hex cry; hyg	634		4.43	146^{25}	
1635	Magnesium iodide hexahydrate	$MgI_2 \cdot 6H_2O$	66778-21-0	386.205	wh monocl cry			2.35		
1636	Magnesium iodide octahydrate	$MgI_2 \cdot 8H_2O$	7790-31-0	422.236	wh orth cry; hyg	41 dec		2.10	146^{25}	s EtOH
1637	Magnesium metaborate octahydrate	$Mg(BO_2)_2 \cdot 8H_2O$	13703-82-7*	254.047	wh pow	988 (anh)				sl H_2O
1638	Magnesium metasilicate	$MgSiO_3$	13776-74-4	100.389	wh monocl cry	≈1550 dec		3.19		i H_2O; sl HF
1639	Magnesium metatitanate	$MgTiO_3$	12032-30-3	120.170	col hex cry	1565		3.85		
1640	Magnesium molybdate	$MgMoO_4$	12013-21-7	184.24	wh pow	≈1060		2.2	15.9^{25}	s H_2O
1641	Magnesium nitrate	$Mg(NO_3)_2$	10377-60-3	148.314	wh cub cry			≈2.3	71.2^{25}	
1642	Magnesium nitrate dihydrate	$Mg(NO_3)_2 \cdot 2H_2O$	15750-45-5	184.345	wh cry	≈100 dec		1.45	71.2^{25}	s EtOH
1643	Magnesium nitrate hexahydrate	$Mg(NO_3)_2 \cdot 6H_2O$	13446-18-9	256.406	col monocl cry; hyg	≈95 dec		1.46	71.2^{25}	s EtOH
1644	Magnesium nitride	Mg_3N_2	12057-71-5	100.928	yel cub cry	≈1500 dec		2.71		
1645	Magnesium nitrite trihydrate	$Mg(NO_2)_2 \cdot 3H_2O$	15070-34-5	170.362	wh hyg prisms	100 dec			129.9^{25}	s EtOH

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1646	Magnesium orthosilicate	Mg_2SiO_4	26686-77-1	140.694	wh orth cry	1897		3.21		i H_2O
1647	Magnesium orthotitanate	Mg_2TiO_4	12032-52-9	160.475	wh cub cry	1840		3.53		
1648	Magnesium oxalate	MgC_2O_4	547-66-0	112.324	wh powder				0.038^{25}	
1649	Magnesium oxalate dihydrate	$MgC_2O_4 \cdot 2H_2O$	6150-88-5	148.354	wh powder				0.038^{25}	i EtOH; s dil acid
1650	Magnesium oxide	MgO	1309-48-4	40.304	wh cub cry	2825	3600	3.6		sl H_2O; i EtOH
1651	Magnesium perborate heptahydrate	$Mg(BO_3)_2 \cdot 7H_2O$	14635-87-1	268.030	wh pow					sl H_2O
1652	Magnesium perchlorate	$Mg(ClO_4)_2$	10034-81-8	223.206	wh hyg powder	250 dec		2.2	100^{25}	
1653	Magnesium perchlorate hexahydrate	$Mg(ClO_4)_2 \cdot 6H_2O$	13446-19-0	331.298	wh hyg cry	190 dec		1.98	100^{25}	s EtOH
1654	Magnesium permanganate hexahydrate	$Mg(MnO_4)_2 \cdot 6H_2O$	10377-62-5	370.268	blue-blk cry	dec		2.18		s H_2O
1655	Magnesium peroxide	MgO_2	1335-26-8	56.304	wh cub cry	100 dec		≈3.0		i H_2O; s dil acid
1656	Magnesium phosphate pentahydrate	$Mg_3(PO_4)_2 \cdot 5H_2O$	7757-87-1*	352.934	wh cry	400 dec			0.00009^{20}	s dil acid
1657	Magnesium phosphate octahydrate	$Mg_3(PO_4)_2 \cdot 8H_2O$	13446-23-6	406.980	wh monocl cry			2.17	0.00009^{20}	s acid
1658	Magnesium pyrophosphate	$Mg_2P_2O_7$	13446-24-7	222.553	col monocl plates	1395		2.56		
1659	Magnesium pyrophosphate trihydrate	$Mg_2P_2O_7 \cdot 3H_2O$	10102-34-8	276.600	wh powder	100 dec		2.56		i H_2O; s acid
1660	Magnesium phosphide	Mg_3P_2	12057-74-8	134.863	yel cub cry			2.06		reac H_2O
1661	Magnesium selenate hexahydrate	$MgSeO_4 \cdot 6H_2O$	13446-28-1	275.35	wh monocl cry			1.928	55.5^{25}	
1662	Magnesium selenide	MgSe	1313-04-8	103.27	brn cub cry			4.2		reac H_2O
1663	Magnesium selenite hexahydrate	$MgSeO_3 \cdot 6H_2O$	15593-61-0	259.36	col hex cry			2.09		i H_2O; s dil acid
1664	Magnesium hexafluorosilicate hexahydrate	$MgSiF_6 \cdot 6H_2O$	60950-56-3	274.472	wh cry	120 dec		1.79	39.3^{18}	i EtOH
1665	Magnesium silicide	Mg_2Si	22831-39-6	76.696	gray cub cry	1102		1.99		reac H_2O
1666	Magnesium stannide	Mg_2Sn	1313-08-2	167.320	blue cub cry	771		3.60		s H_2O, dil HCl
1667	Magnesium sulfate	$MgSO_4$	7487-88-9	120.368	col orth cry	1137		2.66	35.7^{25}	
1668	Magnesium sulfate monohydrate	$MgSO_4 \cdot H_2O$	14168-73-1	138.383	col monocl cry	150 dec		2.57	35.7^{25}	
1669	Magnesium sulfate heptahydrate	$MgSO_4 \cdot 7H_2O$	10034-99-8	246.474	col orth cry	150 dec		1.67	35.7^{25}	sl EtOH
1670	Magnesium sulfide	MgS	12032-36-9	56.370	red-brn cub cry	2226		2.68		reac H_2O
1671	Magnesium sulfite trihydrate	$MgSO_3 \cdot 3H_2O$	19086-20-5	158.414	col orth cry			2.12	0.79^{25}	
1672	Magnesium sulfite hexahydrate	$MgSO_3 \cdot 6H_2O$	13446-29-2	212.460	wh hex cry	200 dec		1.72	0.79^{25}	i EtOH
1673	Magnesium tetrahydrogen phosphate dihydrate	$Mg(H_2PO_4)_2 \cdot 2H_2O$	15609-80-0	254.311	wh hyg cry	dec 90				s H_2O; i EtOH
1674	Magnesium thiocyanate tetrahydrate	$Mg(SCN)_2 \cdot 4H_2O$	306-61-6	212.531	wh hyg cry					vs H_2O, EtOH
1675	Magnesium thiosulfate hexahydrate	$MgS_2O_3 \cdot 6H_2O$	13446-30-5	244.525	col cry	170 dec		1.82	93^{25}	i EtOH
1676	Magnesium trisilicate	$Mg_2Si_3O_8$	14987-04-3	260.862	wh powder					i H_2O, EtOH
1677	Magnesium tungstate	$MgWO_4$	13573-11-0	272.14	wh monocl cry			6.89	0.016^{20}	i EtOH
1678	Magnesium vanadate	$Mg_2V_2O_7$	13568-63-3	262.489	tricl cry			3.1		
1679	Magnesium zirconate	$MgZrO_3$	12032-31-4	163.527	col cry	2060		4.23		
1680	Magnesium zirconium silicate	$MgO \cdot ZrO_2 \cdot SiO_2$	52110-05-1	223.612	wh solid					i H_2O, alk; sl acid
1681	Manganese	Mn	7439-96-5	54.938	hard gray metal	1246	2061	7.3		s dil acids
1682	Manganocene	$Mn(C_5H_5)_2$	1271-27-8	185.124	yel-brn cry	173				s py, thf; sl bz
1683	Manganese antimonide (MnSb)	MnSb	12032-82-5	176.698	hex cry	840		6.9		
1684	Manganese antimonide (Mn_2Sb)	Mn_2Sb	12032-97-2	231.636	tetr cry	948		7.0		
1685	Manganese boride (MnB)	MnB	12045-15-7	65.749	orth cry	1890		6.45		
1686	Manganese boride (MnB_2)	MnB_2	12228-50-1	76.560	hex cry	1827		5.3		
1687	Manganese boride (Mn_2B)	Mn_2B	12045-16-8	120.687	red-brn tetr cry	1580		7.20		
1688	Manganese carbide	Mn_3C	12266-65-8	176.825	refrac solid	1520		6.89		
1689	Manganese carbonyl	$Mn_2(CO)_{10}$	10170-69-1	389.977	yel monocl cry	154		1.75		i H_2O; s os
1690	Manganese pentacarbonyl bromide	$Mn(CO)_5Br$	14516-54-2	274.893	oran-yel cry					s os
1691	Manganese phosphide (MnP)	MnP	12032-78-9	85.912	orth cry	1147		5.49		
1692	Manganese phosphide (Mn_2P)	Mn_2P	12333-54-9	140.850	hex cry	1327		6.0		
1693	Manganese silicide	$MnSi_2$	12032-86-9	111.109	gray solid	1152 dec				
1694	Manganese(II) acetate tetrahydrate	$Mn(C_2H_3O_2)_2 \cdot 4H_2O$	6156-78-1	245.087	red monocl cry	80		1.59		s H_2O, EtOH
1695	Manganese(II) bromide	$MnBr_2$	13446-03-2	214.746	pink hex cry	698		4.385	151^{25}	
1696	Manganese(II) bromide tetrahydrate	$MnBr_2 \cdot 4H_2O$	10031-20-6	286.808	red hyg cry	64 dec			151^{25}	
1697	Manganese(II) carbonate	$MnCO_3$	598-62-9	114.947	pink hex cry	>200 dec		3.70	0.00008^{20}	s dil acid
1698	Manganese(II) chloride	$MnCl_2$	7773-01-5	125.844	pink trig cry; hyg	650	1190	2.977	77.3^{25}	s py, EtOH; i eth
1699	Manganese(II) chloride tetrahydrate	$MnCl_2 \cdot 4H_2O$	13446-34-9	197.906	red monocl cry; hyg	87.5		1.913	77.3^{25}	s EtOH; i eth
1700	Manganese(II) dihydrogen phosphate dihydrate	$Mn(H_2PO_4)_2 \cdot 2H_2O$	18718-07-5	284.944	col hyg cry					s H_2O; i EtOH
1701	Manganese(II) fluoride	MnF_2	7782-64-1	92.935	red tetr cry	900		3.98	1.02^{25}	i EtOH
1702	Manganese(II) hydroxide	$Mn(OH)_2$	18933-05-6	88.953	pink hex cry	dec		3.26	0.00034^{20}	
1703	Manganese(II) hypophosphite monohydrate	$Mn(H_2PO_2)_2 \cdot H_2O$	10043-84-2	202.931	pink cry	>250			15^{20}	s H_2O

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1704	Manganese(II) iodide	MnI_2	7790-33-2	308.747	wh hex cry; hyg	638		5.04		s H_2O, EtOH
1705	Manganese(II) iodide tetrahydrate	$MnI_2 \cdot 4H_2O$	7790-33-2*	380.809	red cry					vs H_2O; s EtOH
1706	Manganese(II) metasilicate	$MnSiO_3$	7759-00-4	131.022	red orth cry	1291		3.48		i H_2O
1707	Manganese(II) molybdate	$MnMoO_4$	14013-15-1	214.88	yel monocl cry			4.05		
1708	Manganese(II) nitrate	$Mn(NO_3)_2$	10377-93-2	178.947	col orth cry; hyg			2.2	161^{25}	s diox, thf
1709	Manganese(II) nitrate tetrahydrate	$Mn(NO_3)_2 \cdot 4H_2O$	20694-39-7	251.009	pink hyg cry	37.1 dec		2.13	161^{25}	s EtOH
1710	Manganese(II) nitrate hexahydrate	$Mn(NO_3)_2 \cdot 6H_2O$	10377-66-9	287.039	rose monocl cry	28 dec		1.8	161^{25}	vs EtOH
1711	Manganese(II) orthosilicate	Mn_2SiO_4	13568-32-6	201.960	orth cry			4.11		i H_2O
1712	Manganese(II) oxalate dihydrate	$MnC_2O_4 \cdot 2H_2O$	6556-16-7	178.987	wh cry powder	150 dec		2.45	0.032^{20}	s acid
1713	Manganese(II) oxide	MnO	1344-43-0	70.937	grn cub cry or powder	1842		5.37		i H_2O; s acid
1714	Manganese(II) perchlorate hexahydrate	$Mn(ClO_4)_2 \cdot 6H_2O$	15364-94-0	361.931	pink hex cry			2.10		
1715	Manganese(II) pyrophosphate	$Mn_2P_2O_7$	53731-35-4	283.819	wh monocl cry	1196		3.71		i H_2O
1716	Manganese(II) selenide	MnSe	1313-22-0	133.90	gray cub cry	1460		5.45		i H_2O
1717	Manganese(II) sulfate	$MnSO_4$	7785-87-7	151.001	wh orth cry	700	850 dec	3.25	63.7^{25}	
1718	Manganese(II) sulfate monohydrate	$MnSO_4 \cdot H_2O$	10034-96-5	169.016	red monocl cry			2.95	63.7^{25}	i EtOH
1719	Manganese(II) sulfate tetrahydrate	$MnSO_4 \cdot 4H_2O$	10101-68-5	223.062	red monocl cry	38 dec		2.26	63.7^{25}	i EtOH
1720	Manganese(II) sulfide (α form)	MnS	18820-29-6	87.003	grn cub cry	1610		4.0		i H_2O; s dil acid
1721	Manganese(II) sulfide (β form)	MnS	18820-29-6	87.003	red cub cry			3.3		i H_2O; s dil acid
1722	Manganese(II) sulfide (γ form)	MnS	18820-29-6	87.003	red hex cry			≈3.3		i H_2O; s dil acid
1723	Manganese(II) telluride	MnTe	12032-88-1	182.54	hex cry	≈1150		6.0		
1724	Manganese(II) tetraborate octahydrate	$MnB_4O_7 \cdot 8H_2O$	12228-91-0	354.300	red solid					i H_2O, EtOH; s dil acid
1725	Manganese(II) titanate	$MnTiO_3$	12032-74-5	150.803	red hex cry	1360		4.55		
1726	Manganese(II) tungstate	$MnWO_4$	13918-22-4	302.78	wh monocl cry			7.2	0.0054^{20}	
1727	Manganese(II,III) oxide	Mn_3O_4	1317-35-7	228.812	brn tetr cry	1567		4.84		i H_2O; s HCl
1728	Manganese(III) acetate dihydrate	$Mn(C_2H_3O_2)_3 \cdot 2H_2O$	19513-05-4	268.100	brn cry					s eth, HOAc
1729	Manganese(III) fluoride	MnF_3	7783-53-1	111.933	red monocl cry; hyg	>600 dec		3.54		reac H_2O
1730	Manganese(III) hydroxide	MnO(OH)	1332-63-4	87.945	blk monocl cry	250 dec		≈4.3		i H_2O
1731	Manganese(III) oxide	Mn_2O_3	1317-34-6	157.874	blk cub cry	1080 dec		≈5.0		i H_2O
1732	Manganese(IV) oxide	MnO_2	1313-13-9	86.937	blk tetr cry	535 dec		5.08		i H_2O, HNO_3
1733	Manganese(VII) oxide	Mn_2O_7	12057-92-0	221.872	grn oil; exp	5.9	95 exp	2.40		vs H_2O
1734	Mendelevium	Md	7440-11-1	258	metal	827				
1735	Mercury	Hg	7439-97-6	200.59	heavy silv liq	-38.8290	356.62	13.5336		i H_2O
1736	Dimethyl mercury	$Hg(CH_3)_2$	593-74-8	230.66	liq		93	3.17		i H_2O; vs EtOH, eth
1737	Mercury(I) acetate	$Hg_2(C_2H_3O_2)_2$	631-60-7	519.27	col scales	dec				sl H_2O; i EtOH, eth
1738	Mercury(I) bromate	$Hg_2(BrO_3)_2$	13465-33-3	656.98	col cry	dec				i H_2O; sl acid
1739	Mercury(I) bromide	Hg_2Br_2	15385-58-7	560.99	wh tetr cry or powder	345 dec		7.307		i H_2O, EtOH, eth
1740	Mercury(I) carbonate	Hg_2CO_3	6824-78-8	461.19	yel-brn cry	130 dec			0.0000045	i EtOH
1741	Mercury(I) chlorate	$Hg_2(ClO_3)_2$	10294-44-7	568.08	wh rhom cry	≈250 dec		6.409		sl H_2O; s EtOH
1742	Mercury(I) chloride	Hg_2Cl_2	10112-91-1	472.09	wh tetr cry	525 tp	383 sp	7.16	0.0004^{25}	i EtOH, eth
1743	Mercury(I) chromate	Hg_2CrO_4	13465-34-4	517.17	brn-red solid					i H_2O EtOH; s conc HNO_3
1744	Mercury(I) fluoride	Hg_2F_2	13967-25-4	439.18	yel cub cry	570 dec	subl	8.73		reac H_2O
1745	Mercury(I) iodate	$Hg_2(IO_3)_2$	13465-35-5	750.99	yel-wh pow	dec 175			0.0032^{20}	
1746	Mercury(I) iodide	Hg_2I_2	15385-57-6	654.99	yel amorp powder	290		7.70		i H_2O, EtOH, eth
1747	Mercury(I) nitrate	$Hg_2(NO_3)_2$	10415-75-5	525.19	cry					sl H_2O
1748	Mercury(I) nitrate dihydrate	$Hg_2(NO_3)_2 \cdot 2H_2O$	14836-60-3	561.22	col cry	70 dec		4.8		sl H_2O
1749	Mercury(I) nitrite	$Hg_2(NO_2)_2$	13492-25-6	493.19	yel cry	100 dec		7.3		reac H_2O
1750	Mercury(I) oxalate	$Hg_2C_2O_4$	2949-11-3	489.20	cry					i H_2O; sl HNO_3
1751	Mercury(I) oxide	Hg_2O	15829-53-5	417.18	prob mixture of HgO+Hg	100 dec		9.8		i H_2O; s HNO_3
1752	Mercury(I) perchlorate tetrahydrate	$Hg_2(ClO_4)_2 \cdot 4H_2O$	65202-12-2	672.14	cry	64			442^{25}	
1753	Mercury(I) sulfate	Hg_2SO_4	7783-36-0	497.24	wh-yel cry powder			7.56	0.051^{25}	s dil HNO_3
1754	Mercury(I) sulfide	Hg_2S	51595-71-2	433.25	unstab blk pow	dec				i H_2O
1755	Mercury(I) thiocyanate	$Hg_2(SCN)_2$	13465-37-7	517.34	col powder	dec			0.03^{25}	s HCl, KCNS
1756	Mercury(I) tungstate	Hg_2WO_4	38705-19-0	649.02	yel amorp solid	dec				i H_2O, EtOH
1757	Mercury(II) acetate	$Hg(C_2H_3O_2)_2$	1600-27-7	318.68	wh-yel cry or powder	179 dec		3.28	25^{10}	s EtOH
1758	Mercury(II) amide chloride	$Hg(NH_2)Cl$	10124-48-8	252.07	wh solid		subl	5.38		i H_2O, EtOH; s warm acid
1759	Mercury(II) benzoate monohydrate	$Hg(C_7H_5O_2)_2 \cdot H_2O$	32839-04-6	460.83	wh cry	165			1.2^{15}	sl EtOH
1760	Mercury(II) bromate	$Hg(BrO_3)_2$	26522-91-8	456.39	cry	130 dec			0.15	s acid
1761	Mercury(II) bromide	$HgBr_2$	7789-47-1	360.40	wh rhomb cry or powder	241	318	6.05	0.61^{25}	sl chl; s EtOH, MeOH
1762	Mercury(II) chlorate	$Hg(ClO_3)_2$	13465-30-0	367.49	wh needles	dec		4.998	25	

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1763	Mercury(II) chloride	$HgCl_2$	7487-94-7	271.50	wh orth cry	277	304	5.6	7.31^{25}	sl bz; s EtOH, MeOH, ace, eth
1764	Mercury(II) chromate	$HgCrO_4$	13444-75-2	316.58	red monocl cry			6.06		sl H_2O
1765	Mercury(II) cyanide	$Hg(CN)_2$	592-04-1	252.62	col tetr cry	320 dec		4.00	11.4^{25}	s EtOH; sl eth
1766	Mercury(II) dichromate	$HgCr_2O_7$	7789-10-8	416.58	red cry powder					i H_2O; s acid
1767	Mercury(II) fluoride	HgF_2	7783-39-3	238.59	wh cub cry; hyg	645 dec		8.95		reac H_2O
1768	Mercury(II) fulminate	$Hg(CNO)_2$	628-86-4	284.62	gray cry	exp		4.42		sl H_2O; s EtOH, NH_4OH
1769	Mercury(II) hydrogen arsenate	$HgHAsO_4$	7784-37-4	340.52	yel powder					i H_2O; s acid
1770	Mercury(II) iodate	$Hg(IO_3)_2$	7783-32-6	550.40	wh powder	175 dec				i H_2O
1771	Mercury(II) iodide (yellow)	HgI_2	7774-29-0	454.40	yel tetr cry or powder	256	351	6.28	0.0055^{25}	sl EtOH, ace, eth
1772	Mercury(II) iodide (red)	HgI_2	7774-29-0	454.40	red pow	trans to yel 127			0.006^{25}	sl EtOH, ace, eth, chl
1773	Mercury(II) nitrate	$Hg(NO_3)_2$	10045-94-0	324.60	col hyg cry	79		4.3		s H_2O; i EtOH
1774	Mercury(II) nitrate monohydrate	$Hg(NO_3)_2 \cdot H_2O$	7783-34-8	342.62	wh-yel hyg cry			4.3		s H_2O, dil acid
1775	Mercury(II) nitrate dihydrate	$Hg(NO_3)_2 \cdot 2H_2O$	22852-67-1	360.63	monocl cry			4.78		s H_2O
1776	Mercury(II) oxalate	HgC_2O_4	3444-13-1	288.61	powder	165 dec				i H_2O
1777	Mercury(II) oxide	HgO	21908-53-2	216.59	red or yel orth cry	500 dec		11.14		i H_2O, EtOH; s dil acid
1778	Mercury(II) oxide sulfate	$(Hg_3O_2)SO_4$	1312-03-4	729.83	yel powder					i H_2O; s acid
1779	Mercury(II) oxycyanide	$Hg(CN)_2 \cdot HgO$	1335-31-5	469.21	wh orth cry	exp		4.44	11.4^{25}	
1780	Mercury(II) perchlorate trihydrate	$Hg(ClO_4)_2 \cdot 3H_2O$	7616-83-3	453.54	cry					
1781	Mercury(II) phosphate	$Hg_3(PO_4)_2$	7782-66-3	791.71	wh-yel powder					i H_2O, EtOH; s acid
1782	Mercury(II) selenide	HgSe	20601-83-6	279.55	gray cub cry		subl	8.21		i H_2O
1783	Mercury(II) sulfate	$HgSO_4$	7783-35-9	296.65	wh monocl cry			6.47		reac H_2O
1784	Mercury(II) sulfide (black)	HgS	1344-48-5	232.66	blk cub cry or powder	850		7.70		i H_2O; s acid, EtOH
1785	Mercury(II) sulfide (red)	HgS	1344-48-5	232.66	red hex cry	trans to blk HgS 344		8.17		i H_2O, acid; s aqua regia
1786	Mercury(II) telluride	HgTe	12068-90-5	328.19	gray cub cry	673		8.63		
1787	Mercury(II) thiocyanate	$Hg(SCN)_2$	592-85-8	316.75	monocl cry	≈165 dec		3.71	0.070^{25}	s dil HCl
1788	Mercury(II) tungstate	$HgWO_4$	37913-38-5	448.43	yel cry	dec				i H_2O, EtOH
1789	Molybdenum	Mo	7439-98-7	95.94	gray-blk metal; cub	2623	4639	10.2		i H_2O, dil acid, alk
1790	Molybdophosphoric acid	$H_3P(Mo_3O_{10})_4$	51429-74-4	1825.25	bright yel cry					
1791	Molybdenum boride (Mo_2B)	Mo_2B	12006-99-4	202.69	refrac tetr cry	2000		9.2		
1792	Molybdenum boride (Mo_2B_5)	Mo_2B_5	12007-97-5	245.94	refrac hex cry	1600		≈7.2		
1793	Molybdenum carbide (MoC)	MoC	12011-97-1	107.95	refrac solid; cub	2577				
1794	Molybdenum carbide (Mo_2C)	Mo_2C	12069-89-5	203.89	gray orth cry	2687		9.18		
1795	Molybdenum carbonyl	$Mo(CO)_6$	13939-06-5	264.00	wh orth cry	148	155 dec	1.96		i H_2O; s bz; sl eth
1796	Molybdenum nitride (MoN)	MoN	12033-19-1	109.95	hex cry	1750		9.20		
1797	Molybdenum nitride (Mo_2N)	Mo_2N	12033-31-7	205.89	gray cub cry	790 dec		9.46		
1798	Molybdenum phosphide	MoP	12163-69-8	126.91	blk hex cry			7.34		
1799	Molybdenum silicide ($MoSi_2$)	$MoSi_2$	12136-78-6	152.11	gray tetr cry	≈1900		6.2		i H_2O; s HF
1800	Molybdenum(II) bromide	$MoBr_2$	13446-56-5	255.75	yel-red cry	dec 700		4.88		i H_2O, EtOH
1801	Molybdenum(II) chloride	$MoCl_2$	13478-17-6	166.85	yel cry	dec 500		3.71		i H_2O
1802	Molybdenum(II) iodide	MoI_2	14055-74-4	349.75	blk hyg cry	700		5.28		i H_2O
1803	Molybdenum(III) bromide	$MoBr_3$	13446-57-6	335.65	grn hex cry	dec 500		4.89		i H_2O, EtOH
1804	Molybdenum(III) chloride	$MoCl_3$	13478-18-7	202.30	dark red monocl cry	dec 400		3.74		i H_2O, os
1805	Molybdenum(III) fluoride	MoF_3	20193-58-2	152.94	yel-brn hex cry	>600		4.64		i H_2O
1806	Molybdenum(III) iodide	MoI_3	14055-75-5	476.65	blk solid	927				i H_2O
1807	Molybdenum(III) oxide	Mo_2O_3	1313-29-7	239.88	gray-blk powder					i H_2O; sl acid
1808	Molybdenum(IV) bromide	$MoBr_4$	13520-59-7	415.56	blk cry	dec 110				reac H_2O
1809	Molybdenum(IV) chloride	$MoCl_4$	13320-71-3	237.75	blk cry	317				reac H_2O, sl chl; i eth, bz
1810	Molybdenum(IV) fluoride	MoF_4	23412-45-5	171.93	grn cry	dec				reac H_2O
1811	Molybdenum(IV) iodide	MoI_4	14055-76-6	603.56	blk cry	dec 100				i H_2O
1812	Molybdenum(IV) oxide	MoO_2	18868-43-4	127.94	brn-viol tetr cry	≈1800 dec		6.47		i H_2O, acid, alk
1813	Molybdenum(IV) selenide	$MoSe_2$	12058-18-3	253.86	gray hex cry	>1200		6.90		
1814	Molybdenum(IV) sulfide	MoS_2	1317-33-5	160.07	blk powder or hex cry	1750		5.06		i H_2O; s conc acid
1815	Molybdenum(IV) telluride	$MoTe_2$	12058-20-7	351.14	gray hex cry			7.7		
1816	Molybdenum(V) chloride	$MoCl_5$	10241-05-1	273.21	grn-blk monocl cry; hyg	194	268	2.93		reac H_2O; s EtOH, eth
1817	Molybdenum(V) fluoride	MoF_5	13819-84-6	190.93	yel monocl cry	67	213.6	3.5		reac H_2O
1818	Molybdenum(V) oxytrichloride	$MoOCl_3$	13814-74-9	218.30	blk monocl cry	310	subl			reac H_2O
1819	Molybdenum(VI) acid monohydrate	$H_2MoO_4 \cdot H_2O$	7782-91-4	179.97	wh powder			3.1		sl H_2O; s alk

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1820	Molybdenum(VI) dioxydichloride	MoO_2Cl_2	13637-68-8	198.85	yel-oran solid	176	250	3.31		reac H_2O
1821	Molybdenum(VI) dioxydifluoride	MoO_2F_2	13824-57-2	165.94	pale lilac cry	subl 270		3.5		i MeCN, chl; sl HF
1822	Molybdenum(VI) fluoride	MoF_6	7783-77-9	209.93	wh cub cry or col liq; hyg	17.5	34.0	2.54		reac H_2O; vs hex, ctc
1823	Molybdenum(VI) metaphosphate	$Mo(PO_3)_6$	133863-98-6	569.77	yel powder			3.28		i H_2O, acid
1824	Molybdenum(VI) oxide	MoO_3	1313-27-5	143.94	wh-yel rhomb cry	802	1155	4.70	0.14^{20}	sl H_2O; s alk, acid
1825	Molybdenum(VI) oxytetrachloride	$MoOCl_4$	13814-75-0	253.75	grn hyg powder	105	159			
1826	Molybdenum(VI) oxytetrafluoride	$MoOF_4$	14459-59-7	187.93	volatile solid	97.2	186.0			
1827	Molybdenum(VI) sulfide	MoS_3	12033-29-3	192.14	blk solid	350 dec				i H_2O, os
1828	Neodymium	Nd	7440-00-8	144.242	silv metal; hex	1016	3074	7.01		
1829	Neodymium boride	NdB_6	12008-23-0	209.108	blk cub cry	2610		4.93		
1830	Neodymium nitride	NdN	25764-11-8	158.249	blk cub cry			7.69		
1831	Neodymium(II) acetate	$Nd(C_2H_3O_2)_2$	6192-13-8	321.373	red-purp cry					s H_2O
1832	Neodymium(II) chloride	$NdCl_2$	25469-93-6	215.148	grn hyg solid	841				
1833	Neodymium(III) bromate nonahydrate	$Nd(BrO_3)_3 \cdot 9H_2O$	15162-92-2	690.086	red hex cry	66 dec				
1834	Neodymium(III) bromide	$NdBr_3$	13536-80-6	383.954	viol orth cry; hyg	682	1540	5.3		s H_2O
1835	Neodymium(III) chloride	$NdCl_3$	10024-93-8	250.601	viol hex cry	759	1600	4.13	100^{25}	vs EtOH; i eth, chl
1836	Neodymium(III) chloride hexahydrate	$NdCl_3 \cdot 6H_2O$	13477-89-9	358.692	purp cry	124 dec		2.3	100^{25}	s EtOH
1837	Neodymium(III) fluoride	NdF_3	13709-42-7	201.237	viol hex cry; hyg	1377	2300	6.51		i H_2O
1838	Neodymium(III) hydroxide	$Nd(OH)_3$	16469-17-3	195.264	blue solid	dec 210				i H_2O
1839	Neodymium(III) iodide	NdI_3	13813-24-6	524.955	grn orth cry; hyg	787		5.85		s H_2O
1840	Neodymium(III) nitrate	$Nd(NO_3)_3$	10045-95-1	330.257	viol hyg. cry				152^{25}	s EtOH
1841	Neodymium(III) nitrate hexahydrate	$Nd(NO_3)_3 \cdot 6H_2O$	14517-29-4	438.348	purp hyg cry				152^{25}	s EtOH, ace
1842	Neodymium(III) oxide	Nd_2O_3	1313-97-9	336.482	blue hex cry; hyg	2233	3760	7.24		i H_2O; s dil acid
1843	Neodymium(III) sulfate	$Nd_2(SO_4)_3$	13477-91-3	576.672	pink needles	≈700 dec			7.1^{20}	
1844	Neodymium(III) sulfate octahydrate	$Nd_2(SO_4)_3 \cdot 8H_2O$	13477-91-3	720.794	red cry	350 dec		2.85		sl H_2O
1845	Neodymium(III) sulfide	Nd_2S_3	12035-32-4	384.679	orth cry	2207		5.46		
1846	Neodymium(III) telluride	Nd_2Te_3	12035-35-7	671.28	gray orth cry	1377		7.0		
1847	Neodymium(III) tris(cyclopentadienyl)	$Nd(C_5H_5)_3$	1273-98-9	339.522	red-blue cry	380				s thf
1848	Neon	Ne	7440-01-9	20.180	col gas	-248.609 tp (43 kPa)	-246.053	0.825 g/L		sl H_2O
1849	Neptunium	Np	7439-99-8	237	silv metal	644		20.2		s HCl
1850	Neptunium(IV) oxide	NpO_2	12035-79-9	269	grn cub cry	2547		11.1		
1851	Nickel	Ni	7440-02-0	58.693	wh metal; cub	1455	2913	8.90		i H_2O; sl dil acid
1852	Nickelocene	$Ni(C_5H_5)_2$	1271-28-9	188.879	grn cry	173				
1853	Nickel aluminide (NiAl)	NiAl	12003-78-0	85.675	metallic solid	1638				
1854	Nickel antimonide	NiSb	12035-52-8	180.453	hex cry	1147		8.74		
1855	Nickel arsenide	NiAs	27016-75-7	133.615	hex cry	967		7.77		
1856	Nickel boride (NiB)	NiB	12007-00-0	69.504	grn refrac solid	1035		7.13		
1857	Nickel boride (Ni_2B)	Ni_2B	12007-01-1	128.198	refrac solid	1125		7.90		
1858	Nickel boride (Ni_3B)	Ni_3B	12007-02-2	186.891	refrac solid	1166		8.17		
1859	Nickel carbonyl	$Ni(CO)_4$	13463-39-3	170.734	col liq	-19.3	42.1 (exp ≈60)	1.31		i H_2O; s EtOH, bz, ace, ctc
1860	Nickel phosphide	Ni_2P	12035-64-2	148.361	hex cry	1100		7.33		
1861	Nickel silicide ($NiSi_2$)	$NiSi_2$	12201-89-7	114.864	cub cry	993		4.83		
1862	Nickel silicide (Ni_2Si)	Ni_2Si	12059-14-2	145.473	orth cry	1255		7.40		
1863	Nickel subsulfide	Ni_3S_2	12035-72-2	240.210	yel hex cry	789		5.87		
1864	Nickel(II) acetate tetrahydrate	$Ni(C_2H_3O_2)_2 \cdot 4H_2O$	6018-89-9	248.842	grn monocl cry	250 dec		1.74	16^{20}	s H_2O, EtOH
1865	Nickel(II) ammonium chloride hexahydrate	$NH_4NiCl_3 \cdot 6H_2O$	16122-03-5*	291.182	grn hyg cry			1.65		s H_2O
1866	Nickel(II) ammonium sulfate	$Ni(NH_4)_2(SO_4)_2$	15699-18-0	286.895	blue-grn cry	dec 250				sl H_2O
1867	Nickel(II) ammonium sulfate hexahydrate	$Ni(NH_4)_2(SO_4)_2 \cdot 6H_2O$	7785-20-8	394.987	blue-grn cry	dec 130		1.92	6.5^{20}	s H_2O; i EtOH
1868	Nickel(II) arsenate octahydrate	$Ni_3(AsO_4)_2 \cdot 8H_2O$	7784-48-7	598.040	yel-grn powder	dec		4.98		i H_2O; s acid
1869	Nickel(II) bromide	$NiBr_2$	13462-88-9	218.501	yel hex cry; hyg	963	subl	5.10	131^{20}	
1870	Nickel(II) bromide trihydrate	$NiBr_2 \cdot 3H_2O$	13462-88-9*	272.547	yel-grn hyg cry	200 dec				vs H_2O; s EtOH, eth
1871	Nickel(II) carbonate	$NiCO_3$	3333-67-3	118.702	grn rhomb cry			4.389	0.0043^{20}	s dil acid
1872	Nickel(II) chlorate hexahydrate	$Ni(ClO_3)_2 \cdot 6H_2O$	13477-94-6	333.687	grn cub cry	dec 80		2.07		vs H_2O
1873	Nickel(II) chloride	$NiCl_2$	7718-54-9	129.599	yel hex cry; hyg	1031	985 sp	3.51	67.5^{25}	s EtOH
1874	Nickel(II) chloride hexahydrate	$NiCl_2 \cdot 6H_2O$	7791-20-0	237.690	grn monocl cry				67.5^{25}	s EtOH
1875	Nickel(II) chromate	$NiCrO_4$	14721-18-7	174.687	red solid					sl H_2O
1876	Nickel(II) cyanide tetrahydrate	$Ni(CN)_2 \cdot 4H_2O$	13477-95-7	182.789	grn plates	200 dec				i H_2O; sl dil acid; s NH_4OH
1877	Nickel(II) fluoride	NiF_2	10028-18-9	96.690	yel tetr cry	1380		4.7	2.56^{25}	i EtOH, eth
1878	Nickel(II) fluoride tetrahydrate	$NiF_2 \cdot 4H_2O$	13940-83-5	168.752	grn pow					sl H_2O

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1879	Nickel(II) hydroxide	$Ni(OH)_2$	12054-48-7	92.708	grn hex cry	230 dec		4.1	0.00015^{20}	
1880	Nickel(II) hydroxide monohydrate	$Ni(OH)_2 \cdot H_2O$	36897-37-7	110.723	grn powder				0.00015^{20}	s dil acid
1881	Nickel(II) iodate	$Ni(IO_3)_2$	13477-98-0	408.498	yel needles			5.07	1.1^{30}	sl H_2O
1882	Nickel(II) iodate tetrahydrate	$Ni(IO_3)_2 \cdot 4H_2O$	13477-99-1	480.560	yel hex cry	dec 100		5.07		sl H_2O
1883	Nickel(II) iodide	NiI_2	13462-90-3	312.502	blk hex cry; hyg	800	subl	5.22	154^{25}	
1884	Nickel(II) iodide hexahydrate	$NiI_2 \cdot 6H_2O$	7790-34-3	420.593	grn monocl cry; hyg				154^{25}	vs EtOH
1885	Nickel(II) nitrate	$Ni(NO_3)_2$	13138-45-9	182.702	grn cry				99.2^{25}	s EtOH
1886	Nickel(II) nitrate hexahydrate	$Ni(NO_3)_2 \cdot 6H_2O$	13478-00-7	290.794	grn monocl cry; hyg	56 dec		2.05	99.2^{25}	s EtOH
1887	Nickel(II) oxalate dihydrate	$NiC_2O_4 \cdot 2H_2O$	6018-94-6	182.742	grn-wh solid	dec 150			0.0012^{25}	i H_2O; s acid, NH_4OH
1888	Nickel(II) oxide	NiO	1313-99-1	74.692	grn cub cry	1957		6.72		i H_2O; s acid
1889	Nickel(II) perchlorate hexahydrate	$Ni(ClO_4)_2 \cdot 6H_2O$	13637-71-3*	365.686	grn hex needles	140			158.8^{25}	s EtOH, ace
1890	Nickel(II) phosphate octahydrate	$Ni_3(PO_4)_2 \cdot 8H_2O$	10381-36-9*	510.145	grn plates					s acid
1891	Nickel(II) selenate hexahydrate	$NiSeO_4 \cdot 6H_2O$	15060-62-5*	309.74	grn tetr cry			2.314	35.5^{20}	
1892	Nickel(II) selenide	NiSe	1314-05-2	137.65	yel-grn hex cry	980		7.2		
1893	Nickel(II) stannate dihydrate	$NiSnO_3 \cdot 2H_2O$	12035-38-0	261.432	grn pow	dec 120				
1894	Nickel(II) sulfate	$NiSO_4$	7786-81-4	154.756	grn-yel orth cry	840 dec		4.01	40.4^{25}	
1895	Nickel(II) sulfate hexahydrate	$NiSO_4 \cdot 6H_2O$	10101-97-0	262.847	blue-grn tetr cry	≈100 dec		2.07	40.4^{25}	sl EtOH
1896	Nickel(II) sulfate heptahydrate	$NiSO_4 \cdot 7H_2O$	10101-98-1	280.862	grn orth cry			1.98	40.4^{25}	s EtOH
1897	Nickel(II) sulfide	NiS	16812-54-7	90.758	yel hex cry	976		5.5		i H_2O
1898	Nickel(II) thiocyanate	$Ni(SCN)_2$	13689-92-4	174.857	grn pwd				55.0^{25}	
1899	Nickel(II) titanate	$NiTiO_3$	12035-39-1	154.558	brn hex cry			5.0		
1900	Nickel(II,III) sulfide	Ni_3S_4	12137-12-1	304.340	cub cry	995		4.77		
1901	Nickel(III) oxide	Ni_2O_3	1314-06-3	165.385	gray-blk cub cry	≈600 dec				i H_2O; s hot acid
1902	Niobium	Nb	7440-03-1	92.906	gray metal; cub	2477	4744	8.57		i acid
1903	Niobocene dichloride	$Nb(C_5H_5)_2Cl_2$	12793-14-5	293.998	hyg blk cry					sl tol
1904	Niobium boride (NbB)	NbB	12045-19-1	103.717	gray orth cry	2270		7.5		
1905	Niobium boride (NbB_2)	NbB_2	12007-29-3	114.528	gray hex cry	3050		6.97		
1906	Niobium carbide (NbC)	NbC	12069-94-2	104.917	gray cub cry	3608	4300	7.82		i H_2O, acid
1907	Niobium carbide (Nb_2C)	Nb_2C	12011-99-3	197.824	refrac hex cry	3080		7.8		i H_2O
1908	Niobium nitride	NbN	24621-21-4	106.913	gray cry; cub	2300		8.47		i HCl, acid
1909	Niobium phosphide	NbP	12034-66-1	123.880	tetr cry			6.5		
1910	Niobium silicide	$NbSi_2$	12034-80-9	149.077	gray hex cry	1950		5.7		
1911	Niobium(II) oxide	NbO	12034-57-0	108.905	gray cub cry	1937		7.30		
1912	Niobium(III) bromide	$NbBr_3$	15752-41-7	332.618	dark brn solid		subl 400			
1913	Niobium(III) chloride	$NbCl_3$	13569-59-0	199.265	blk solid					
1914	Niobium(III) fluoride	NbF_3	15195-53-6	149.901	blue cub cry			4.2		
1915	Niobium(III) iodide	NbI_3	13870-20-7	473.619	blk solid	dec 510				
1916	Niobium(IV) bromide	$NbBr_4$	13842-75-6	412.522	dark brn cry		subl 300	4.72		reac H_2O
1917	Niobium(IV) chloride	$NbCl_4$	13569-70-5	234.718	viol-blk monocl cry	dec 800	275 subl	3.2		reac H_2O
1918	Niobium(IV) fluoride	NbF_4	13842-88-1	168.900	blk tetr cry; hyg	>350 dec		4.01		
1919	Niobium(IV) iodide	NbI_4	13870-21-8	600.524	gray orth cry	503		5.6		reac H_2O
1920	Niobium(IV) oxide	NbO_2	12034-59-2	124.905	wh tetr cry or powder	1901		5.9		
1921	Niobium(IV) selenide	$NbSe_2$	12034-77-4	250.83	gray hex cry	>1300		6.3		
1922	Niobium(IV) sulfide	NbS_2	12136-97-9	157.036	blk rhomb cry			4.4		
1923	Niobium(IV) telluride	$NbTe_2$	12034-83-2	348.11	hex cry			7.6		
1924	Niobium(V) bromide	$NbBr_5$	13478-45-0	492.426	oran orth cry	265.2	361.6	4.36		s H_2O, EtOH
1925	Niobium(V) chloride	$NbCl_5$	10026-12-7	270.171	yel monocl cry; hyg	205.8	247.4	2.78		reac H_2O; s HCl, ctc
1926	Niobium(V) dioxyfluoride	NbO_2F	15195-33-2	143.903	wh cub cry			4.0		
1927	Niobium(V) ethoxide	$Nb(OC_2H_5)_5$	3236-82-6	318.209	col hyg liq	5	203	1.258		reac H_2O; s peth
1928	Niobium(V) fluoride	NbF_5	7783-68-8	187.898	col monocl cry; hyg	80	234	2.70		reac H_2O; sl CS_2, chl
1929	Niobium(V) iodide	NbI_5	13779-92-5	727.428	yel-blk monocl cry	327		5.32		reac H_2O
1930	Niobium(V) oxide	Nb_2O_5	1313-96-8	265.810	wh orth cry	1500		4.47		i H_2O; s HF
1931	Niobium(V) oxybromide	$NbOBr_3$	14459-75-7	348.617	yel-brn cry	≈320 dec	subl			
1932	Niobium(V) oxychloride	$NbOCl_3$	13597-20-1	215.264	wh tetr cry		subl	3.72		
1933	Nitrogen	N_2	7727-37-9	28.013	col gas	-210.0	-195.798	1.145 g/L		sl H_2O; i EtOH
1934	Nitramide	NO_2NH_2	7782-94-7	62.028	unstab wh cry	72 dec				s H_2O, EtOH, ace, eth; i chl
1935	Nitric acid	HNO_3	7697-37-2	63.013	col liq; hyg	-41.6	83	1.5129^{20}		vs H_2O
1936	Nitrous acid	HNO_2	7782-77-6	47.014	stab only in soln					
1937	Nitrous oxide	N_2O	10024-97-2	44.012	col gas	-90.8	-88.48	1.799 g/L		sl H_2O; s EtOH, eth
1938	Nitric oxide	NO	10102-43-9	30.006	col gas	-163.6	-151.74	1.226 g/L		sl H_2O

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1939	Nitrogen dioxide	NO_2	10102-44-0	46.006	brn gas; equil with N_2O_4		see N_2O_4	1.880 g/L		reac H_2O
1940	Nitrogen trioxide	N_2O_3	10544-73-7	76.011	blue solid or liq (low temp)	-101.1	≈3 dec	1.4[2]		reac H_2O
1941	Nitrogen tetroxide	N_2O_4	10544-72-6	92.011	col liq; equil with NO_2	-9.3	21.15	1.45[20]		reac H_2O
1942	Nitrogen pentoxide	N_2O_5	10102-03-1	108.010	col hex cry		33 sp	2.0		s chl; sl ctc
1943	Nitrogen tribromide	NBr_3	15162-90-0	253.719	unstab solid	exp -100				
1944	Nitrogen trichloride	NCl_3	10025-85-1	120.366	yel oily liq; exp	-40	71	1.653		i H_2O; s CS_2, bz, ctc
1945	Nitrogen trifluoride	NF_3	7783-54-2	71.002	col gas	-206.79	-128.75	2.902 g/L		i H_2O
1946	Nitrogen triiodide	NI_3	13444-85-4	394.720	unstab blk cry; exp					
1947	Nitrogen chloride difluoride	$NClF_2$	13637-87-1	87.457	col gas	-195	-67	3.575 g/L		
1948	Chloramine	NH_2Cl	10599-90-3	51.476	yel liq	-66				s H_2O, EtOH, eth; sl bz, ctc
1949	Fluoramine	NH_2F	15861-05-9	35.021	unstab gas	≈-110		1.431 g/L		
1950	Difluoramine	NHF_2	10405-27-3	53.012	col gas	-116	-23	2.167 g/L		
1951	*cis*-Difluorodiazine	N_2F_2	13812-43-6	66.010	col gas	<-195	-105.75	2.698 g/L		
1952	*trans*-Difluorodiazine	N_2F_2	13776-62-0	66.010	col gas	-172	-111.45	2.698 g/L		
1953	Tetrafluorohydrazine	N_2F_4	10036-47-2	104.007	col gas	-164.5	-74	4.251 g/L		
1954	Nitrosyl bromide	NOBr	13444-87-6	109.910	red gas	-56	≈0	4.492 g/L		reac H_2O
1955	Nitrosyl chloride	NOCl	2696-92-6	65.459	yel gas	-59.6	-5.5	2.676 g/L		reac H_2O
1956	Nitrosyl fluoride	NOF	7789-25-5	49.004	col gas	-132.5	-59.9	2.003 g/L		
1957	Trifluoramine oxide	NOF_3	13847-65-9	87.001	col gas	-161	-87.5	3.556 g/L		
1958	Nitryl chloride	NO_2Cl	13444-90-1	81.459	col gas	-145	-15	3.330 g/L		
1959	Nitryl fluoride	NO_2F	10022-50-1	65.004	col gas	-166	-72.4	2.657 g/L		reac H_2O
1960	Nitrogen selenide	N_4Se_4	12033-88-4	371.87	red monocl cry; hyg	exp		4.2		i H_2O, eth, EtOH; sl bz, CS_2
1961	Nobelium	No	10028-14-5	259.000	metal	827				
1962	Osmium	Os	7440-04-2	190.23	blue-wh metal; hex	3033	5012	22.587[20]		s aqua regia
1963	Osmocene	$Os(C_5H_5)_2$	1273-81-0	320.42	col cry	229				
1964	Osmium carbonyl	$Os_3(CO)_{12}$	15696-40-9	906.81	yel cry	224		3.48		
1965	Osmium pentacarbonyl	$Os(CO)_5$	16406-49-8	330.28	col liq	-15	dec 100			s os
1966	Osmium nonacarbonyl	$Os_2(CO)_9$	28411-13-4	632.55	oran-yel cry	65 dec				s hc
1967	Osmium(II) chloride	$OsCl_2$	13444-92-3	261.14	hyg brn solid	dec >450				s EtOH, eth
1968	Osmium(III) bromide	$OsBr_3$	59201-51-3	429.94	dark gray cry	340 dec				i H_2O, os, acid
1969	Osmium(III) chloride	$OsCl_3$	13444-93-4	296.59	gray cub cry	450 dec				i H_2O, os; s conc acid
1970	Osmium(IV) chloride	$OsCl_4$	10026-01-4	332.04	red-blk orth cry	323 dec		4.38		reac H_2O; i os
1971	Osmium(IV) fluoride	OsF_4	54120-05-7	266.22	yel cry	230				reac H_2O
1972	Osmium(IV) oxide	OsO_2	12036-02-1	222.23	yel-brn tetr cry	dec 500		11.4		i H_2O, acid
1973	Osmium(V) fluoride	OsF_5	31576-40-6	285.22	hyg blue-grn cry	70	233			reac H_2O
1974	Osmium(VI) fluoride	OsF_6	13768-38-2	304.22	yel cub cry	33.4	47.5	4.1		reac H_2O
1975	Osmium(VI) tetrachloride oxide	$OsOCl_4$	36509-15-6	348.04	dark brn hyg cry	32	200			reac H_2O; s hc
1976	Osmium(VIII) oxide	OsO_4	20816-12-0	254.23	yel monocl cry	40.6	131.2	5.1	6.44[20]	sl H_2O; s ctc, bz, EtOH, eth
1977	Oxygen	O_2	7782-44-7	31.999	col gas	-218.79	-182.953	1.308 g/L		sl H_2O, EtOH, os
1978	Ozone	O_3	10028-15-6	47.998	blue gas	-193	-111.35	1.962 g/L		sl H_2O
1979	Palladium	Pd	7440-05-3	106.42	silv-wh metal; cub	1554.8	2963	12.0		s aqua regia
1980	Palladium(II) acetate	$Pd(C_2H_3O_2)_2$	3375-31-3	224.51	oran-brn cry	205 dec				i H_2O; s MeCN, eth, ace
1981	Palladium(II) bromide	$PdBr_2$	13444-94-5	266.23	red-blk monocl cry; hyg	250 dec		≈5.2		i H_2O
1982	Palladium(II) chloride	$PdCl_2$	7647-10-1	177.33	red rhomb cry; hyg	679		4.0		s H_2O, EtOH, ace
1983	Palladium(II) chloride dihydrate	$PdCl_2 \cdot 2H_2O$	7647-10-1*	213.36	brn cry					s H_2O, EtOH, ace
1984	Palladium(II) cyanide	$Pd(CN)_2$	2035-66-7	158.45	yel solid	dec				
1985	Palladium(II) fluoride	PdF_2	13444-96-7	144.42	viol tetr cry; hyg	952		5.76		reac H_2O
1986	Palladium(II) iodide	PdI_2	7790-38-7	360.23	blk cry	360 dec		6.0		i H_2O, EtOH, eth
1987	Palladium(II) nitrate	$Pd(NO_3)_2$	10102-05-3	230.43	brn hyg cry	dec				sl H_2O; s dil HNO_3
1988	Palladium(II) oxide	PdO	1314-08-5	122.42	grn-blk tetr cry	750 dec		8.3		i H_2O, acid; sl aqua regia
1989	Palladium(II) 2,4-pentanedioate	$Pd(CH_3COCHCOCH_3)_2$	14024-61-4	304.64	oran-yel cry	205 dec				s bz, chl
1990	Palladium(II) sulfate dihydrate	$PdSO_4 \cdot 2H_2O$	13566-03-5	238.51	grn-brn cry	dec				
1991	Palladium(II) sulfide	PdS	12125-22-3	138.49	gray tetr cry			6.7		
1992	*cis*-Dichlorodiamminepalladium(II)	$Pd(NH_3)_2Cl_2$	15684-18-1	211.39	yel pow				0.025[25]	
1993	*trans*-Dichlorodiamminepalladium (II)	$Pd(NH_3)_2Cl_2$	13782-33-7	211.39	yel solid			2.50		

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
1994	Phosphorus (white)	P	7723-14-0	30.974	col waxlike cub cry	44.15	280.5	1.823		i H_2O; sl bz, EtOH, chl; s CS_2
1995	Phosphorus (red)	P	7723-14-0	30.974	red-viol amorp powder	579.2	431 sp	2.16		i H_2O, os
1996	Phosphorus (black)	P	7723-14-0	30.974	blk orth cry or amorp solid	610		2.69		i os
1997	Phosphine	PH_3	7803-51-2	33.998	col gas; flam	-133.8	-87.75	1.390 g/L		i H_2O; sl EtOH, eth
1998	Diphosphine	P_2H_4	13445-50-6	65.980	col liq	-99	63.5 dec			reac H_2O
1999	Diphosphorus tetrachloride	P_2Cl_4	13497-91-1	203.760	col oily liq	-28	≈180 dec			
2000	Diphosphorus tetrafluoride	P_2F_4	13824-74-3	137.942	col gas	-86.5	-6.2	5.638 g/L		
2001	Diphosphorus tetraiodide	P_2I_4	13455-00-0	569.566	red tricl needles	125.5	dec	3.89		
2002	Phosphonium chloride	PH_4Cl	24567-53-1	70.459	gas		-27 sp	2.880 g/L		reac H_2O
2003	Phosphonium iodide	PH_4I	12125-09-6	161.910	col tetr cry	18.5	62.5	2.86		reac H_2O, EtOH
2004	Phosphoric acid	H_3PO_4	7664-38-2	97.995	col visc liq	42.4	407		548^{20}	s EtOH
2005	Phosphotungstic acid	$H_3PW_{12}O_{40}$	12067-99-1	2880.05	wh-yel cry	89				vs H_2O; s EtOH, eth
2006	Phosphonic acid	H_3PO_3	13598-36-2	81.996	wh hyg cry	74.4	200	1.65	309^{0}	vs EtOH
2007	Phosphinic acid	HPH_2O_2	6303-21-5	65.997	hyg cry or col oily liq	26.5	130	1.49		vs H_2O, EtOH, eth
2008	Metaphosphoric acid	HPO_3	37267-86-0	79.980	gl solid; hyg					sl H_2O; s EtOH
2009	Hypophosphoric acid	$H_4P_2O_6$	7803-60-3	161.976	col orth cry	73 dec				vs H_2O
2010	Diphosphoric acid	$H_4P_2O_7$	2466-09-3	177.975	wh cry	71.5			709^{23}	
2011	Difluorophosphoric acid	HPO_2F_2	13779-41-4	101.978	col liq	≈-94	110 dec	1.583		reac H_2O
2012	Hexafluorophosphoric acid	HPF_6	16940-81-1	145.972	col oily liq	25 dec				reac H_2O
2013	Fluorophosphonic acid	H_2PFO_3	13537-32-1	99.986	col visc liq	<-70		1.82		vs H_2O
2014	Phosphorus nitride (P_3N_5)	P_3N_5	12136-91-3	162.955	yel-brn solid	800 dec				i H_2O; s os
2015	Phosphorus sesquisulfide	P_4S_3	1314-85-8	220.090	yel-grn orth cry	173	407	2.03		i H_2O; s bz; vs CS_2
2016	Phosphorus heptasulfide	P_4S_7	12037-82-0	348.350	pale yel monocl cry	308	523	2.19		sl CS_2
2017	Phosphonitrilic chloride trimer	$(PNCl_2)_3$	940-71-6	347.659	wh hyg cry	128.8		1.98		reac H_2O
2018	Phosphorus(III) bromide	PBr_3	7789-60-8	270.686	col liq	-41.5	173.2	2.8		reac H_2O, EtOH; s ace, CS_2
2019	Phosphorus(III) dibromide fluoride	PBr_2F	15597-39-4	209.780	col liq	-115	78.5			
2020	Phosphorus(III) bromide difluoride	$PBrF_2$	15597-40-7	148.875	col gas	-133.8	-16.1	6.085 g/L		
2021	Phosphorus(III) chloride	PCl_3	7719-12-2	137.333	col liq	-93	76	1.574		reac H_2O, EtOH; s bz, chl, eth
2022	Phosphorus(III) dichloride fluoride	PCl_2F	15597-63-4	120.878	col gas	-144	13.85	4.941 g/L		
2023	Phosphorus(III) chloride difluoride	$PClF_2$	14335-40-1	104.424	col gas	-164.8	-47.3	4.268 g/L		
2024	Phosphorus(III) fluoride	PF_3	7783-55-3	87.969	col gas	-151.5	-101.8	3.596 g/L		reac H_2O
2025	Phosphorus(III) iodide	PI_3	13455-01-1	411.687	red-oran hex cry; hyg	61.2	227 dec	4.18		reac H_2O; s EtOH
2026	Phosphorus(III) oxide	P_2O_3	1314-24-5	109.946	col monocl cry or liq	23.8	173	2.13		reac H_2O
2027	Tetraphosphorus(III) hexoxide	P_4O_6	12440-00-5	219.891	soft wh cry	23.8	175.4			
2028	Phosphorus(III) selenide	P_2Se_3	1314-86-9	298.83	oran-red cry	245	≈380	1.31		reac H_2O; s bz, ctc, CS_2, ace
2029	Phosphorus(III) sulfide	P_2S_3	12165-69-4	158.143	yel solid	290	490			reac H_2O; s EtOH, eth, CS_2
2030	Phosphorus(V) bromide	PBr_5	7789-69-7	430.494	yel orth cry, hyg	≈100 dec		3.61		reac H_2O, EtOH; s CS_2, ctc
2031	Phosphorus(V) tetrabromide fluoride	PBr_4F		369.588	pale yel cry	87 dec				
2032	Phosphorus(V) dibromide trifluoride	PBr_2F_3	13445-58-4	247.777	yel-red liq	-20	15 dec			
2033	Phosphorus(V) chloride	PCl_5	10026-13-8	208.239	wh-yel tetr cry; hyg	167 tp	160 sp	2.1		reac H_2O; s CS_2, ctc
2034	Phosphorus(V) tetrachloride fluoride	PCl_4F	13498-11-8	191.784	col liq	-59	30 dec			
2035	Phosphorus(V) trichloride difluoride	PCl_3F_2	13537-23-0	175.330	col liq	-63				
2036	Phosphorus(V) dichloride trifluoride	PCl_2F_3	13454-99-4	158.875	col gas	-125	7.1	6.494 g/L		
2037	Phosphorus(V) chloride tetrafluoride	$PClF_4$	13498-11-8	142.421	col gas	-132	-43.4	5.821 g/L		
2038	Phosphorus(V) fluoride	PF_5	7647-19-0	125.966	col gas	-93.8	-84.6	5.149 g/L		reac H_2O
2039	Phosphorus(V) oxide	P_2O_5	1314-56-3	141.945	wh orth cry; hyg	562	605	2.30		reac H_2O, EtOH
2040	Phosphorus(V) selenide	P_2Se_5	1314-82-5	456.75	blk-purp amorp solid					reac hot H_2O, ctc; i CS_2
2041	Phosphorus(V) sulfide	P_2S_5	1314-80-3	222.273	grn-yel hyg cry	285	515	2.03		reac H_2O; s CS_2
2042	Phosphonic difluoride	POF_2H	14939-34-5	85.978	volatile liq	>-120	≈60 (gas unstab)			

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2043	Phosphoryl bromide	$POBr_3$	7789-59-5	286.685	faint oran plates	55	191.7	2.822		reac H_2O; s bz, eth, chl
2044	Phosphoryl dibromide chloride	$POBr_2Cl$	13550-31-7	242.234	yel solid	31	165			
2045	Phosphoryl dibromide fluoride	$POBr_2F$	14014-19-8	225.779	col liq	-117.2	110.1			
2046	Phosphoryl bromide dichloride	$POBrCl_2$	13455-03-3	197.783	col liq	11	136.5	2.104[14]		
2047	Phosphoryl bromide difluoride	$POBrF_2$	14014-18-7	164.874	col liq	-84.8	31.6			
2048	Phosphoryl bromide chloride fluoride	POBrClF	14518-81-1	181.328	col liq		79			
2049	Phosphoryl chloride	$POCl_3$	10025-87-3	153.332	col liq	1.18	105.5	1.645		reac H_2O, EtOH
2050	Phosphoryl dichloride fluoride	$POCl_2F$	13769-76-1	136.877	col liq	-80.1	52.9			
2051	Phosphoryl chloride difluoride	$POClF_2$	13769-75-0	120.423	col gas	-96.4	3.1	4.922 g/L		
2052	Phosphoryl fluoride	POF_3	13478-20-1	103.968	col gas	-39.1 tp	-39.7 sp	4.250 g/L		reac H_2O
2053	Phosphoryl iodide	POI_3	13455-04-4	427.686	viol cry	53				
2054	Phosphorothioc tribromide	$PSBr_3$	3931-89-3	302.751	yel cry	37.8	212 dec	2.85		
2055	Phosphorothioc dibromide fluoride	$PSBr_2F$	13706-10-0	241.845	yel liq	-75.2	125.3			
2056	Phosphorothioc bromide difluoride	$PSBrF_2$	13706-09-7	180.940	yel liq	-136.9	35.5			
2057	Phosphorothioc trichloride	$PSCl_3$	3982-91-0	169.398	fuming liq	-36.2	125	1.635		reac H_2O; s bz, ctc, chl, CS_2
2058	Phosphorothioc dichloride fluoride	$PSCl_2F$	155698-29-6	152.943	col liq	-96.0	64.7			
2059	Phosphorothioc chloride difluoride	$PSClF_2$	2524-02-9	136.489	col gas	-155.2	6.3	5.579 g/L		
2060	Phosphorothioc trifluoride	PSF_3	2404-52-6	120.034	col gas	-148.8	-52.25	4.906 g/L		
2061	Phosphorothioc triiodide	PSI_3	63972-04-3	443.752	yel cry	48	dec			
2062	Platinum	Pt	7440-06-4	195.084	silv-gray metal; cub	1768.2	3825	21.5		i acid; s aqua regia
2063	Hexachloroplatinic acid	H_2PtCl_6	16941-12-1	409.818	hyg yel-brn cry	60				s H_2O, EtOH
2064	Hydrogen hexahydroxyplatinate(IV)	$H_2Pt(OH)_6$	51850-20-5	299.144	yel needles	dec 100				s H_2O, acid, dil alk
2065	Platinum(II) bromide	$PtBr_2$	13455-12-4	354.892	red-brn powder	250 dec		6.65		i H_2O
2066	Platinum(II) chloride	$PtCl_2$	10025-65-7	265.990	grn hex cry	581 dec		6.0		i H_2O, EtOH, eth; s HCl
2067	Platinum(II) cyanide	$Pt(CN)_2$	592-06-3	247.118	pale yel cry					i H_2O, acid, alk
2068	Platinum(II) iodide	PtI_2	7790-39-8	448.893	blk powder	325 dec		6.4		i H_2O
2069	Platinum(II) oxide	PtO	12035-82-4	211.083	blk tetr cry	325 dec		14.1		i H_2O, EtOH; s aqua regia
2070	Platinum(II) sulfide	PtS	12038-20-9	227.149	tetr cry			10.25		
2071	Platinum(III) bromide	$PtBr_3$	25985-07-3	434.796	grn-blk cry	200 dec				
2072	Platinum(III) chloride	$PtCl_3$	25909-39-1	301.443	grn-blk cry	435 dec		5.26		
2073	Platinum(IV) bromide	$PtBr_4$	68938-92-1	514.700	brn-blk cry	180 dec			0.41[20]	sl EtOH, eth
2074	Platinum(IV) chloride	$PtCl_4$	37773-49-2	336.896	red-brn cub cry	327 dec		4.30	142[25]	
2075	Platinum(IV) chloride pentahydrate	$PtCl_4 \cdot 5H_2O$	13454-96-1	426.972	red cry			2.43		s H_2O, EtOH
2076	Platinum(IV) fluoride	PtF_4	13455-15-7	271.078	red cry	600				
2077	Platinum(IV) iodide	PtI_4	7790-46-7	702.702	brn-blk powder	130 dec				s H_2O
2078	Platinum(IV) oxide	PtO_2	1314-15-4	227.083	blk hex cry	450		11.8		i H_2O; s conc acid, dil alk
2079	Platinum(IV) sulfide	PtS_2	12038-21-0	259.214	hex cry			7.85		
2080	Platinum(VI) fluoride	PtF_6	13693-05-5	309.074	red cub cry	61.3	69.1	≈4.0		
2081	*cis*-Diamminedichloroplatinum	$Pt(NH_3)_2Cl_2$	15663-27-1	300.051	yel solid	270 dec			0.253[25]	
2082	*trans*-Diamminedichloroplatinum	$Pt(NH_3)_2Cl_2$	14913-33-8	300.051	pale yel solid	270 dec			0.036[25]	s DMF, DMSO
2083	Hexachloroplatinic acid hexahydrate	$H_2PtCl_6 \cdot 6H_2O$	16941-12-1	517.909	brn-yel hyg cry	60		2.43	140[18]	vs EtOH
2084	Platinum silicide	PtSi	12137-83-6	223.170	orth cry	1229		12.4		
2085	Plutonium	Pu	7440-07-5	244	silv-wh metal; monocl	640	3228	19.7		
2086	Plutonium nitride	PuN	12033-54-4	258	gray cub cry	2550		14.4		
2087	Plutonium(II) oxide	PuO	12035-83-5	260	cub cry			14.0		
2088	Plutonium(III) bromide	$PuBr_3$	15752-46-2	484	grn orth cry	681		6.75		s H_2O
2089	Plutonium(III) chloride	$PuCl_3$	13569-62-5	350	grn hex cry	760		5.71		s H_2O
2090	Plutonium(III) fluoride	PuF_3	13842-83-6	301	purp hex cry	1396		9.33		i H_2O; sl acid
2091	Plutonium(III) iodide	PuI_3	13813-46-2	625	grn orth cry; hyg	777		6.92		s H_2O
2092	Plutonium(III) oxide	Pu_2O_3	12036-34-9	536	blk cub cry	2085		10.5		
2093	Plutonium(IV) fluoride	PuF_4	13709-56-3	320	red-brn monocl cry	1037		7.1		
2094	Plutonium(IV) oxide	PuO_2	12059-95-9	276	yel-brn cub cry	2390		11.5		
2095	Plutonium(VI) fluoride	PuF_6	13693-06-6	358	red-brn orth cry	51.6		5.08		
2096	Polonium	Po	7440-08-6	209	silv metal; cub	254	962	9.20		
2097	Polonium(IV) chloride	$PoCl_4$	10026-02-5	351	yel hyg cry	≈300	390			s H_2O, EtOH, ace
2098	Polonium(IV) oxide	PoO_2	7446-06-2	241	yel cub cry	500 dec		8.9		
2099	Potassium	K	7440-09-7	39.098	soft silv-wh metal; cub	63.5	759	0.89		reac H_2O
2100	Potassium acetate	$KC_2H_3O_2$	127-08-2	98.142	wh hyg cry	309		1.57	269[25]	s EtOH; i eth

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2101	Potassium aluminate trihydrate	$K_2Al_2O_4 \cdot 3H_2O$	12003-63-3*	250.204	wh orth cry			2.13		vs H_2O; i EtOH
2102	Potassium aluminum silicate	$KAlSi_3O_8$	1327-44-2	278.332	col monocl cry			2.56		i H_2O
2103	Potassium aluminum sulfate	$KAl(SO_4)_2$	10043-67-1	258.205	wh hyg powder				5.9^{20}	
2104	Potassium aluminum sulfate dodecahydrate	$KAl(SO_4)_2 \cdot 12H_2O$	7784-24-9	474.389	col cry	≈100 dec		1.72	5.9^{20}	
2105	Potassium amide	KNH_2	17242-52-3	55.121	wh/yel-grn hyg cry	335				reac H_2O, EtOH
2106	Potassium arsenate	K_3AsO_4	13464-36-3	256.215	col cry			2.8	125^{25}	
2107	Potassium arsenite	$KAsO_2$	13464-35-2	146.019	wh hyg powder					s H_2O; sl EtOH
2108	Potassium azide	KN_3	20762-60-1	81.118	tetr cry; exp			2.04	49.7^{17}	
2109	Potassium borohydride	KBH_4	13762-51-1	53.941	wh cub cry	≈500 dec		1.11		s H_2O
2110	Potassium bromate	$KBrO_3$	7758-01-2	167.000	wh hex cry	434 dec		3.27	8.17^{25}	i EtOH
2111	Potassium bromide	KBr	7758-02-3	119.002	col cub cry; hyg	734	1435	2.74	67.8^{25}	sl EtOH
2112	Potassium carbonate	K_2CO_3	584-08-7	138.206	wh monocl cry; hyg	899	dec	2.29	111^{25}	i EtOH
2113	Potassium carbonate sesquihydrate	$K_2CO_3 \cdot 1.5H_2O$	6381-79-9	165.229	granular cry				111^{20}	
2114	Potassium chlorate	$KClO_3$	3811-04-9	122.549	wh monocl cry	357	dec	2.34	8.61^{25}	
2115	Potassium chloride	KCl	7447-40-7	74.551	wh cub cry	771		1.988	35.5^{25}	i eth, ace
2116	Potassium chlorochromate	$KCrO_3Cl$	16037-50-6	174.545	oran cry			2.5		reac H_2O; s ace, acid
2117	Potassium chromate	K_2CrO_4	7789-00-6	194.191	yel orth cry	974		2.73	65.0^{25}	
2118	Potassium citrate monohydrate	$K_3C_6H_5O_7 \cdot H_2O$	6100-05-6	324.410	col hyg cry	180 dec		1.98	172^{20}	vs H_2O; sl EtOH
2119	Potassium cobalt(II) selenate hexahydrate	$K_2Co(SeO_4)_2 \cdot 6H_2O$	28041-86-3	531.14	red monocl cry			2.51		
2120	Potassium cyanate	KCNO	590-28-3	81.115	wh tetr cry	≈700 dec		2.05	75^{25}	sl EtOH
2121	Potassium cyanide	KCN	151-50-8	65.116	wh cub cry; hyg	622		1.55	69.9^{20}	sl EtOH
2122	Potassium cyanoaurite	$KAu(CN)_2$	13967-50-5	288.099	col cry			3.45	14^{20}	s H_2O; sl EtOH; i eth, ace
2123	Potassium dichromate	$K_2Cr_2O_7$	7778-50-9	294.185	oran-red tricl cry	398	≈500 dec	2.68	15.1^{25}	
2124	Potassium dihydrogen arsenate	KH_2AsO_4	7784-41-0	180.034	col cry	288		2.87	19^{6}	i EtOH
2125	Potassium dihydrogen phosphate	KH_2PO_4	7778-77-0	136.085	wh tetr cry	253		2.34	25.0^{25}	sl EtOH
2126	Potassium dihydrogen phosphonate	KH_2PO_3	13977-65-6	120.086	col monocl hyg cry					
2127	Potassium dithionate	$K_2S_2O_6$	13455-20-4	238.323	col hex cry	dec		2.27		sl H_2O; i EtOH
2128	Potassium ferricyanide	$K_3Fe(CN)_6$	13746-66-2	329.244	red cry	dec		1.89	48.8^{25}	
2129	Potassium ferrocyanide trihydrate	$K_4Fe(CN)_6 \cdot 3H_2O$	14459-95-1	422.388	yel monocl cry	60 dec		1.85	36.0^{25}	i EtOH, eth
2130	Potassium fluoride	KF	7789-23-3	58.096	wh cub cry	858	1502	2.48	102^{25}	
2131	Potassium fluoride dihydrate	$KF \cdot 2H_2O$	13455-21-5	94.127	monocl cry	41 dec		2.5	102^{25}	
2132	Potassium fluoroborate	KBF_4	14075-53-7	125.903	col orth cry	530		2.505	0.55^{25}	sl EtOH
2133	Potassium fluorotantalate	K_2TaF_7	16924-00-8	392.134	col cry	730		5.24	0.5^{0}	
2134	Potassium formate	$KCHO_2$	590-29-4	84.116	col hyg cry	167		1.91	331^{18}	
2135	Potassium hexachloroosmate(IV)	K_2OsCl_6	16871-60-6	481.15	red cub cry					vs H_2O; sl EtOH
2136	Potassium hexachloroplatinate	K_2PtCl_6	16921-30-5	485.999	yel-oran cub cry	250 dec		3.50	0.77^{20}	i EtOH
2137	Potassium hexacyanocobaltate	$K_3Co(CN)_6$	13963-58-1	332.332	yel monocl cry	dec		1.91		vs H_2O; i EtOH
2138	Potassium hexafluoromanganate(IV)	K_2MnF_6	16962-31-5	247.125	yel hex cry					reac H_2O
2139	Potassium hexafluorosilicate	K_2SiF_6	16871-90-2	220.273	wh cry	dec		2.27	0.084^{20}	i EtOH
2140	Potassium hexafluorozirconate(IV)	K_2ZrF_6	16923-95-8	283.411	col monocl cry			3.48	0.78^{2}	
2141	Potassium hydride	KH	7693-26-7	40.106	cub cry	619		1.43		reac H_2O
2142	Potassium hydrogen arsenate	K_2HAsO_4	21093-83-4	218.125	col monocl prisms	300 dec			18.7^{6}	i EtOH
2143	Potassium hydrogen carbonate	$KHCO_3$	298-14-6	100.115	col monocl cry	≈100 dec		2.17	36.2^{25}	i EtOH
2144	Potassium hydrogen fluoride	KHF_2	7789-29-9	78.103	col tetr cry	238.8		2.37	39.2^{20}	i EtOH
2145	Potassium hydrogen iodate	$KH(IO_3)_2$	13455-24-8	389.911	col cry	dec			1.3^{15}	sl H_2O; i EtOH
2146	Potassium hydrogen oxalate hemihydrate	$KHC_2O_4 \cdot 0.5H_2O$	127-95-7		wh cry	dec		2.09	2.5^{20}	sl EtOH
2147	Potassium hydrogen phosphate	K_2HPO_4	7758-11-4	174.176	wh hyg cry	dec			168^{25}	s EtOH
2148	Potassium hydrogen phosphite	K_2HPO_3	13492-26-7	158.177	wh hyg powder	dec			170^{20}	i EtOH
2149	Potassium hydrogen selenite	$KHSeO_3$	7782-70-9	167.06	hyg orth cry	>100 dec				s H_2O; sl EtOH
2150	Potassium hydrogen sulfate	$KHSO_4$	7646-93-7	136.169	wh monocl cry; hyg	≈200		2.32	50.6^{25}	
2151	Potassium hydrogen sulfide	KHS	1310-61-8	72.171	wh hex cry; hyg	≈450		1.69		s H_2O, EtOH
2152	Potassium hydrogen sulfide hemihydrate	$KHS \cdot 0.5H_2O$	1310-61-8*	81.179	wh-yel hyg cry	≈175		1.7		vs H_2O, EtOH
2153	Potassium hydrogen sulfite	$KHSO_3$	7773-03-7	120.169	wh cry powder	190 dec			49^{20}	i EtOH
2154	Potassium hydrogen tartrate	$KHC_4H_4O_6$	868-14-4	188.177	wh cry			1.98	0.57^{20}	s acid, alk; i EtOH
2155	Potassium hydroxide	KOH	1310-58-3	56.105	wh rhomb cry; hyg	406	1327	2.044	121^{25}	s EtOH; s MeOH
2156	Potassium hypochlorite	KOCl	7778-66-7	90.550	exists only in aq soln					
2157	Potassium phosphinate	KH_2PO_2	7782-87-8	104.087	wh hyg cry	dec				vs H_2O; s EtOH

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2158	Potassium iodate	KIO_3	7758-05-6	214.001	wh monocl cry	560 dec		3.89	9.22^{25}	
2159	Potassium iodide	KI	7681-11-0	166.003	col cub cry	681	1323	3.12	148^{25}	sl EtOH
2160	Potassium iron(III) oxalate trihydrate	$K_3Fe(C_2O_4)_3 \cdot 3H_2O$		491.243	grn monocl cry	100	230 dec	2.133	4.7^{0}	i EtOH
2161	Potassium manganate	K_2MnO_4	10294-64-1	197.133	grn cry	190 dec				s H_2O; reac HCl
2162	Potassium metaarsenate	$KAsO_3$	19197-73-0	162.018	wh solid	660				
2163	Potassium metabisulfite	$K_2S_2O_5$	16731-55-8	222.324	wh powder	≈150 dec		2.3	49.5^{25}	reac acid; i EtOH
2164	Potassium metaborate	KBO_2	13709-94-9	81.908	wh hex cry	947		≈2.3		
2165	Potassium molybdate	K_2MoO_4	13446-49-6	238.14	wh hyg cry	919		2.3	183^{25}	i EtOH
2166	Potassium niobate	$KNbO_3$	12030-85-2	180.002	wh rhomb cry	≈1100		4.64		i H_2O
2167	Potassium nitrate	KNO_3	7757-79-1	101.103	col orth cry or powder	334	400 dec	2.105	38.3^{25}	i EtOH
2168	Potassium nitrite	KNO_2	7758-09-0	85.104	wh hyg cry	438	537 exp	1.915	312^{25}	sl EtOH
2169	Potassium oxalate	$K_2C_2O_4$	583-52-8	166.216	wh pwd					sl H_2O
2170	Potassium oxalate monohydrate	$K_2C_2O_4 \cdot H_2O$	6487-48-5	184.231	col cry	160 dec		2.13	36.4^{20}	
2171	Potassium oxide	K_2O	12136-45-7	94.196	gray cub cry	740		2.35		s H_2O, EtOH, eth
2172	Potassium perbromate	$KBrO_4$	22207-96-1	183.000	wh cry	275 dec			4.21^{25}	
2173	Potassium percarbonate monohydrate	$K_2C_2O_6 \cdot H_2O$	589-97-9	216.230	oran or blue pow				6.5^{20}	
2174	Potassium perchlorate	$KClO_4$	7778-74-7	138.549	col orth cry; hyg	525		2.52	2.08^{25}	
2175	Potassium periodate	KIO_4	7790-21-8	230.001	col tetr cry	582	exp	3.618	0.51^{25}	
2176	Potassium permanganate	$KMnO_4$	7722-64-7	158.034	purp orth cry	dec		2.7	7.60^{25}	reac EtOH
2177	Potassium peroxide	K_2O_2	17014-71-0	110.196	yel amorp solid	490				reac H_2O
2178	Potassium persulfate	$K_2S_2O_8$	7727-21-1	270.322	col cry	≈100 dec		2.48	4.7^{20}	
2179	Potassium phosphate	K_3PO_4	7778-53-2	212.266	wh orth cry; hyg	1340		2.564	106^{25}	i EtOH
2180	Potassium pyrophosphate	$K_4P_2O_7$	7320-34-5		wh cry	dec 1300				s H_2
2181	Potassium pyrophosphate trihydrate	$K_4P_2O_7 \cdot 3H_2O$	7790-67-2	384.383	col hyg cry	dec 300		2.33		vs H_2O; i EtOH
2182	Potassium pyrosulfate	$K_2S_2O_7$	7790-62-7	254.323	col needles	≈325		2.28		s H_2O
2183	Potassium selenate	K_2SeO_4	7790-59-2	221.16	wh powder			3.07	114^{25}	
2184	Potassium selenide	K_2Se	1312-74-9	157.16	red cub cry; hyg	800		2.29		s H_2O
2185	Potassium selenite	K_2SeO_3	10431-47-7	205.16	wh hyg cry	875 dec			217^{25}	sl EtOH
2186	Potassium silver cyanide	$KAg(CN)_2$	506-61-6	199.000	wh cry					s H_2O
2187	Potassium sodium tartrate tetrahydrate	$KNaC_4H_4O_6 \cdot 4H_2O$	6381-59-5	282.220	wh cry	≈70 dec	anh at 130	1.79		vs H_2O; i EtOH
2188	Potassium stannate trihydrate	$K_2SnO_3 \cdot 3H_2O$	12142-33-5*	298.951	col cry			3.20		vs H_2O; i EtOH
2189	Potassium stearate	$KC_{18}H_{35}O_2$	593-29-3	322.568	wh pow					sl cold H_2O; s hot H_2O, EtOH
2190	Potassium sulfate	K_2SO_4	7778-80-5	174.260	wh orth cry	1069		2.66	12.0^{25}	i EtOH
2191	Potassium sulfide	K_2S	1312-73-8	110.262	red-yel cub cry; hyg	948		1.74		s H_2O, EtOH; i eth
2192	Potassium sulfide pentahydrate	$K_2S \cdot 5H_2O$	37248-34-3	200.338	col rhomb cry	60				vs H_2O, EtOH; i eth
2193	Potassium sulfite	K_2SO_3	10117-38-1	158.260	col hex cry				106^{25}	sl EtOH
2194	Potassium sulfite dihydrate	$K_2SO_3 \cdot 2H_2O$	7790-56-9	194.291	wh monocl cry	dec			107^{20}	sl EtOH; dec dil acid
2195	Potassium superoxide	KO_2	12030-88-5	71.097	yel tetr cry; hyg	380		2.16		reac H_2O
2196	Potassium tellurate(VI) trihydrate	$K_2TeO_4 \cdot 3H_2O$	15571-91-2*	323.84	wh cry powder					s H_2O
2197	Potassium tellurite	K_2TeO_3	7790-58-1	253.80	wh hyg cry	≈460 dec				vs H_2O
2198	Potassium tetraborate pentahydrate	$K_2B_4O_7 \cdot 5H_2O$	1332-77-0	323.513	wh cry powder				16.5^{30}	sl EtOH
2199	Potassium tetrachloroaurate dihydrate	$KAuCl_4 \cdot 2H_2O$	13682-61-6	413.908	yel monocl cry					s H_2O, EtOH, eth
2200	Potassium tetrachloroplatinate	K_2PtCl_4	10025-99-7	415.093	pink-red tetr cry	500 dec		3.38		s H_2O; i EtOH
2201	Potassium tetracyanocadmate	$K_2Cd(CN)_4$	14402-75-6	294.678	cub cry	≈450		1.85	25^{20}	sl EtOH
2202	Potassium tetracyanonickelate monohydrate	$K_2[Ni(CN)_4] \cdot H_2O$	14220-17-8*	258.975	red-oran cry	dec 100				
2203	Potassium tetracyanoplatinate(II) trihydrate	$K_2Pt(CN)_4 \cdot 3H_2O$	562-76-5*	431.397	col rhomb prisms					s H_2O
2204	Potassium tetracyanozincate	$K_2Zn(CN)_4$	14244-62-3	247.676	cry pow					vs H_2O
2205	Potassium tetraiodomercurate(II)	K_2HgI_4	7783-33-7	786.40	yel hyg cry			4.29		vs H_2O; s EtOH, eth, ace
2206	Potassium thiocyanate	KSCN	333-20-0	97.181	col tetr cry; hyg	173	500 dec	1.88	238^{25}	s EtOH
2207	Potassium thiosulfate	$K_2S_2O_3$	10294-66-3	190.325	col hyg cry				165^{25}	i EtOH
2208	Potassium titanate	K_2TiO_3	12030-97-6	174.062	wh orth cry	1515		3.1		reac H_2O
2209	Potassium triiodide monohydrate	$KI_3 \cdot H_2O$	7790-42-3	437.827	brn monocl cry; hyg	225 dec		3.5		s H_2O; reac EtOH, eth
2210	Potassium triiodozincate	$KZnI_3$	7790-43-4	485.221	hyg cry					vs H_2O
2211	Potassium thiocarbonate	K_2CS_3	26750-66-3	186.403	yel-red hyg cry					vs H_2O
2212	Potassium tungstate	K_2WO_4	7790-60-5	326.04	hyg cry	921		3.12		vs H_2O; i EtOH
2213	Potassium uranate	$K_2U_2O_7$	7790-63-8	666.251	oran cub cry			6.12		i H_2O; s acid

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2214	Potassium uranyl nitrate	$K(UO_2)(NO_3)_3$	18078-40-5	495.140	grn-yel cry pow					vs H_2O
2215	Potassium uranyl sulfate dihydrate	$K_2(UO_2)(SO_4)_2 \cdot 2H_2O$	27709-53-1	576.381	grn-yel cry pow	dec 120		3.36		vs H_2O
2216	Potassium zinc sulfate hexahydrate	$K_2Zn(SO_4)_2 \cdot 6H_2O$	13932-17-7	443.823	cry					s H_2O
2217	Potassium zirconium sulfate trihydrate	$K_4Zr(SO_4)_4 \cdot 3H_2O$	53608-79-0	685.914	wh cry pow					sl H_2O
2218	Praseodymium	Pr	7440-10-0	140.908	silv metal; hex	931	3520	6.77		
2219	Praseodymium boride	PrB_6	12008-27-4	205.774	blk cub cry	2610		4.84		
2220	Praseodymium nitride	PrN	25764-09-4	154.915	cub cry			7.46		
2221	Praseodymium silicide	$PrSi_2$	12066-83-0	197.079	tetr cry	1712		5.46		
2222	Praseodymium(II) iodide	PrI_2	65530-47-4	394.717	bronze solid	758				
2223	Praseodymium(III) bromate	$Pr(BrO_3)_3$	15162-93-3	524.615	grn cry					vs H_2O
2224	Praseodymium(III) bromide	$PrBr_3$	13536-53-3	380.620	grn hex cry; hyg	693		5.28		s H_2O
2225	Praseodymium(III) carbonate octahydrate	$Pr_2(CO_3)_3 \cdot 8H_2O$	14948-62-0	605.964	grn silky plates	dec 420 (anh)				i H_2O; s acid
2226	Praseodymium(III) chloride	$PrCl_3$	10361-79-2	247.267	grn hex needles; hyg	786		4.0	96.1^{25}	s EtOH
2227	Praseodymium(III) chloride heptahydrate	$PrCl_3 \cdot 7H_2O$	10025-90-8	373.374	grn cry	110 dec			96.1^{25}	s EtOH
2228	Praseodymium(III) fluoride	PrF_3	13709-46-1	197.903	grn hex cry	1399		6.3		
2229	Praseodymium(III) hydroxide	$Pr(OH)_3$	16469-16-2	191.930	grn solid	dec 220		3.7		i H_2O
2230	Praseodymium(III) iodide	PrI_3	13813-23-5	521.621	orth hyg cry	738		≈5.8		s H_2O
2231	Praseodymium(III) nitrate	$Pr(NO_3)_3$	10361-80-5	326.923	pale grn hyg cry				165^{25}	s EtOH
2232	Praseodymium(III) nitrate hexahydrate	$Pr(NO_3)_3 \cdot 6H_2O$	15878-77-0	435.014	grn needles				165^{25}	s EtOH, ace
2233	Praseodymium(III) oxide	Pr_2O_3	12036-32-7	329.813	wh hex cry	2183	3760	6.9		
2234	Praseodymium(III) perchlorate hexahydrate	$Pr(ClO_4)_3 \cdot 6H_2O$	13498-07-2*	547.351	hyg grn cry	dec 200				s H_2O, EtOH
2235	Praseodymium(III) sulfate octahydrate	$Pr_2(SO_4)_3 \cdot 8H_2O$	13510-41-3	714.125	grn monocl cry			2.83	17^{20}	s H_2O
2236	Praseodymium(III) sulfide	Pr_2S_3	12038-13-0	378.010	cub cry	1765		5.1		
2237	Praseodymium(III) telluride	Pr_2Te_3	12038-12-9	664.62	cub cry	1500		≈7.0		
2238	Praseodymium(IV) fluoride	PrF_4	15192-24-2	216.902	yel-wh solid	dec 90				
2239	Promethium	Pm	7440-12-2	145	silv metal; hex	1042	3000	7.26		
2240	Promethium(III) bromide	$PmBr_3$	14325-78-1	385	red cry	625				s H_2O
2241	Promethium(III) chloride	$PmCl_3$	13779-10-7	251	pale blue hyg cry	655				s H_2O
2242	Promethium(III) fluoride	PmF_3	13709-45-0	202	pink solid	1338				s H_2O
2243	Promethium(III) iodide	PmI_3	13818-73-0	526	red solid	695				
2244	Protactinium	Pa	7440-13-3	231.036	shiny metal; tetr or cub	1572		15.4		
2245	Protactinium(V) chloride	$PaCl_5$	13760-41-3	408.301	yel monocl cry	306		3.74		
2246	Radium	Ra	7440-14-4	226	wh metal; cub	696		5		
2247	Radium bromide	$RaBr_2$	10031-23-9	386	wh orth cry	728		5.79	70.6^{20}	s EtOH
2248	Radium carbonate	$RaCO_3$	7116-98-5	286	wh orth cry					i H_2O
2249	Radium chloride	$RaCl_2$	10025-66-8	297	wh orth cry	1000		4.9	24.5^{20}	s EtOH
2250	Radium fluoride	RaF_2	20610-49-5	264	wh cub cry			6.7		
2251	Radium nitrate	$Ra(NO_3)_2$	10213-12-4	350	cry				13.9	
2252	Radium sulfate	$RaSO_4$	7446-16-4	322	wh cry					i H_2O, acid
2253	Radon	Rn	10043-92-2	222	col gas	-71	-61.7	9.074 g/L		sl H_2O
2254	Rhenium	Re	7440-15-5	186.207	silv-gray metal	3185	5596	20.8		i HCl
2255	Perrhenic acid	$HReO_4$	13768-11-1	251.213	exists only in soln					vs H_2O, os
2256	Rhenium carbonyl	$Re_2(CO)_{10}$	14285-68-8	652.515	yel-wh cry	170 dec		2.87		s os
2257	Rhenium pentacarbonyl bromide	$Re(CO)_5Br$	14220-21-4	406.162	wh cry	90				
2258	Rhenium pentacarbonyl chloride	$Re(CO)_5Cl$	14099-01-5	361.711	wh cry		subl 140			
2259	Rhenium(III) bromide	$ReBr_3$	13569-49-8	425.919	red-brn monocl cry		500 subl	6.10		s ace, MeOH, EtOH
2260	Rhenium(III) chloride	$ReCl_3$	13569-63-6	292.566	red-blk hyg cry	500 dec		4.81		s H_2O
2261	Rhenium(III) iodide	ReI_3	15622-42-1	566.920	blk solid	dec				
2262	Rhenium(IV) chloride	$ReCl_4$	13569-71-6	328.019	purp-blk cry; hyg	300 dec		4.9		
2263	Rhenium(IV) fluoride	ReF_4	15192-42-4	262.201	blue tetr cry		>300 subl	7.49		
2264	Rhenium(IV) oxide	ReO_2	12036-09-8	218.206	gray orth cry	900 dec		11.4		
2265	Rhenium(IV) selenide	$ReSe_2$	12038-64-1	344.13	tricl cry					
2266	Rhenium(IV) silicide	$ReSi_2$	12038-66-3	242.378	refrac solid	2000				
2267	Rhenium(IV) sulfide	ReS_2	12038-63-0	250.337	tricl cry			7.6		
2268	Rhenium(IV) telluride	$ReTe_2$	12067-00-4	441.41	orth cry			8.50		
2269	Rhenium(V) bromide	$ReBr_5$	30937-53-2	585.727	brn solid	110 dec				
2270	Rhenium(V) chloride	$ReCl_5$	39368-69-9	363.472	brn-blk solid	220		4.9		reac H_2O
2271	Rhenium(V) fluoride	ReF_5	30937-52-1	281.199	yel-grn solid	48	221.3			
2272	Rhenium(V) oxide	Re_2O_5	12165-05-8	452.411	blue-blk tetr cry			≈7		

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2273	Rhenium(VI) chloride	$ReCl_6$	31234-26-1	398.925	red-grn solid	29				
2274	Rhenium(VI) dioxydifluoride	ReO_2F_2	81155-18-2	256.203	col cry	156				
2275	Rhenium(VI) fluoride	ReF_6	10049-17-9	300.197	yel liq or cub cry	18.5	33.8	4.06(cry)		s HNO_3
2276	Rhenium(VI) oxide	ReO_3	1314-28-9	234.205	red cub cry	400 dec		6.9		i H_2O, acid, alk
2277	Rhenium(VI) oxytetrachloride	$ReOCl_4$	13814-76-1	344.018	brn cry	29.3	223			reac H_2O
2278	Rhenium(VI) oxytetrafluoride	$ReOF_4$	17026-29-8	278.200	blue solid	108	171.7			
2279	Rhenium(VII) fluoride	ReF_7	17029-21-9	319.196	yel cub cry	48.3	73.7	4.32		
2280	Rhenium(VII) oxide	Re_2O_7	1314-68-7	484.410	yel hyg cry	327	360	6.10		s H_2O, EtOH, eth, diox, py
2281	Rhenium(VII) trioxychloride	ReO_3Cl	7791-09-5	269.658	col liq	4.5	128	3.87		reac H_2O
2282	Rhenium(VII) trioxyfluoride	ReO_3F	42246-24-2	253.203	yel solid	147	164			
2283	Rhenium(VII) dioxytrifluoride	ReO_2F_3	57246-89-6	275.201	yel solid	90	185.4			reac H_2O
2284	Rhenium(VII) oxypentafluoride	$ReOF_5$	23377-53-9	297.198	cream solid	43.8	73.0			
2285	Rhenium(VII) sulfide	Re_2S_7	12038-67-4	596.869	brn-blk tetr cry			4.87		i H_2O
2286	Rhodium	Rh	7440-16-6	102.906	silv-wh metal; cub	1964	3695	12.4		i acid, sl aqua regia
2287	Rhodium carbonyl	$Rh_6(CO)_{16}$	28407-51-4	1065.594	red-brn cry	220 dec				
2288	Rhodium carbonyl chloride	$[Rh(CO)_2Cl]_2$	14523-22-9	388.758	red-oran cry	124				s os
2289	Rhodium dodecacarbonyl	$Rh_4(CO)_{12}$	19584-30-6	747.743	red hyg cry	150 dec		2.52		reac H_2O
2290	Rhodium(III) bromide	$RhBr_3$	15608-29-4	342.618	dark brn plates	800 dec		5.56		s H_2O; i acid, os
2291	Rhodium(III) chloride	$RhCl_3$	10049-07-7	209.265	red monocl cry		717	5.38		i H_2O; s alk
2292	Rhodium(III) fluoride	RhF_3	60804-25-3	159.901	red hex cry			5.4		
2293	Rhodium(III) iodide	RhI_3	15492-38-3	483.619	blk monocl cry; hyg			6.4		
2294	Rhodium(III) nitrate	$Rh(NO_3)_3$	10139-58-9	288.921	hyg brn solid	600 dec				i H_2O
2295	Rhodium(III) nitrate dihydrate	$Rh(NO_3)_3 \cdot 2H_2O$	13465-43-5	324.951	blk solid	dec				i H_2O; s aqua regia
2296	Rhodium(III) oxide	Rh_2O_3	12036-35-0	253.809	gray hex cry	1100 dec		8.2		
2297	Rhodium(III) oxide pentahydrate	$Rh_2O_3 \cdot 5H_2O$	39373-27-8	309.010	yel pow	dec				sl H_2O; s acid
2298	Rhodium(III) sulfate	$Rh_2(SO_4)_3$	10489-46-0	493.999	red-yel solid	>500 dec				
2299	Rhodium(IV) oxide	RhO_2	12137-27-8	134.905	blk tetr cry			7.2		
2300	Rhodium(IV) oxide dihydrate	$RhO_2 \cdot 2H_2O$	12137-27-8	170.936	grn solid	dec		8.20		i H_2O, sol HCl, alk
2301	Rhodium(VI) fluoride	RhF_6	13693-07-7	216.896	blk cub cry	≈70		3.1		
2302	Rubidium	Rb	7440-17-7	85.468	soft silv metal; cub	39.30	688	1.53		reac H_2O
2303	Rubidium acetate	$RbC_2H_3O_2$	563-67-7	144.512	wh hyg cry	246				vs H_2O
2304	Rubidium aluminum sulfate	$RbAl(SO_4)_2$	13530-57-9	304.575	hex cry			≈3.1	1.60^{20}	i EtOH
2305	Rubidium aluminum sulfate dodecahydrate	$RbAl(SO_4)_2 \cdot 12H_2O$	7784-29-4	520.759	col cub cry	≈100 dec		≈1.9		s H_2O; i EtOH
2306	Rubidium azide	RbN_3	22756-36-1	127.488	tetr cry; exp	317		2.79	107^{16}	
2307	Rubidium bromate	$RbBrO_3$	13446-70-3	213.370	cub cry	430		3.68	2.95^{25}	
2308	Rubidium bromide	RbBr	7789-39-1	165.372	wh cub cry; hyg	692	1340	3.35	116^{25}	
2309	Rubidium carbonate	Rb_2CO_3	584-09-8	230.945	col monocl cry; hyg	837			223^{20}	
2310	Rubidium chlorate	$RbClO_3$	13446-71-4	168.919	col cry	324	dec 480	3.19	6.63^{25}	sl H_2O
2311	Rubidium chloride	RbCl	7791-11-9	120.921	wh cub cry; hyg	724	1390	2.76	93.9^{25}	sl EtOH
2312	Rubidium chromate	Rb_2CrO_4	13446-72-5	286.930	yel rhom cry			3.518	76.2^{25}	
2313	Rubidium dichromate	$Rb_2Cr_2O_7$	13446-73-6	386.924	red tricl or yel monocl cry			3.1		s H_2O
2314	Rubidium cyanide	RbCN	19073-56-4	111.486	wh cub cry			2.3		s H_2O; i EtOH, eth
2315	Rubidium fluoride	RbF	13446-74-7	104.466	wh cub cry; hyg	795	1410	3.2	300^{20}	i EtOH
2316	Rubidium fluoroborate	$RbBF_4$	18909-68-7	172.273	orth cry	612 dec		2.82		sl H_2O
2317	Rubidium formate	$RbCHO_2$	3495-35-0	130.486	wh hyg cry	dec				
2318	Rubidium hexafluorogermanate	Rb_2GeF_6	16962-48-4	357.57	wh cry	696				s H_2O
2319	Rubidium hydride	RbH	13446-75-8	86.476	wh cub cry; flam	≈170 dec		2.60		reac H_2O
2320	Rubidium hydrogen carbonate	$RbHCO_3$	19088-74-5	146.485	wh rhomb cry	175 dec			116^{20}	
2321	Rubidium hydrogen fluoride	$RbHF_2$	12280-64-7	124.473	tetr cry	188		3.3		
2322	Rubidium hydrogen sulfate	$RbHSO_4$	15587-72-1	182.539	col monocl cry	208		2.9		s H_2O
2323	Rubidium hydroxide	RbOH	1310-82-3	102.475	gray-wh orth cry; hyg	385		3.2	173^{30}	s EtOH
2324	Rubidium iodate	$RbIO_3$	13446-76-9	260.370	monocl or cub cry	dec		4.33	2.44^{25}	vs HCl
2325	Rubidium iodide	RbI	7790-29-6	212.372	wh cub cry	656	1300	3.55	165^{25}	s EtOH
2326	Rubidium molybdate	Rb_2MoO_4	13718-22-4	330.87	wh cry	958				s H_2O
2327	Rubidium nitrate	$RbNO_3$	13126-12-0	147.473	wh hex cry; hyg	310		3.11	65.0^{25}	vs H_2O
2328	Rubidium nitrite	$RbNO_2$	13825-25-7	131.474	wh cry	422				vs H_2O
2329	Rubidium oxide	Rb_2O	18088-11-4	186.935	yel-brn cub cry; hyg	400 dec		4.0		reac H_2O
2330	Rubidium perchlorate	$RbClO_4$	13510-42-4	184.919	wh hyg cry	597	dec 900	2.9	1.5^{25}	
2331	Rubidium permanganate	$RbMnO_4$	13465-49-1	204.404	dark purp cry	300 dec		3.24		sl H_2O
2332	Rubidium peroxide	Rb_2O_2	23611-30-5	202.935	wh orth cry	570		3.8		reac H_2O
2333	Rubidium selenide	Rb_2Se	31052-43-4	249.90	wh cub cry	733		3.22		reac H_2O

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2334	Rubidium sulfate	Rb_2SO_4	7488-54-2	266.999	wh orth cry	1066		3.61	50.8^{25}	
2335	Rubidium sulfide	Rb_2S	31083-74-6	203.001	wh cub cry	425		2.91		s H_2O
2336	Rubidium superoxide	RbO_2	12137-25-6	117.467	tetr cry	412		≈3.0		
2337	Ruthenium	Ru	7440-18-8	101.07	silv-wh metal; hex	2334	4150	12.1		i acid, aqua regia
2338	Ruthenium dodecacarbonyl	$Ru_3(CO)_{12}$	15243-33-1	639.33	oran cry	150 dec				
2339	Ruthenium pentacarbonyl	$Ru(CO)_5$	16406-48-7	241.12	col liq	-22	dec 50			i H_2O; s EtOH, bz, chl, hc
2340	Ruthenium nonacarbonyl	$Ru_2(CO)_9$	63128-11-0	454.23	stab below -40					s hex
2341	Ruthenium nitrosyl chloride monohydrate	$Ru(NO)Cl_3 \cdot H_2O$	18902-42-6	255.45	hyg red cry					
2342	Hexaammineruthenium(III) chloride	$Ru(NH_3)_6Cl_3$	14282-91-8	309.61	col monocl cry					s H_2O
2343	Ruthenium(III) bromide	$RuBr_3$	14014-88-1	340.78	brn hex cry	dec 500		5.3		i H_2O, acid, EtOH
2344	Ruthenium(III) chloride	$RuCl_3$	10049-08-8	207.43	blk-brn hex cry	≈500 dec		3.1		i H_2O; sl EtOH
2345	Ruthenium(III) fluoride	RuF_3	51621-05-7	158.07	brn rhomb cry	≈600 dec		5.36		i H_2O, dil acid
2346	Ruthenium(III) iodide	RuI_3	13896-65-6	481.78	blk hex cry	dec 300		6.0		sl H_2O
2347	Ruthenium(III) 2,4-pentanedioate	$Ru(CH_3COCHCOCH_3)_3$	14284-93-6	398.39	red-brn cry	230				
2348	Ruthenium(IV) fluoride	RuF_4	71500-16-8	177.06	yel-red cry					reac H_2O
2349	Ruthenium(IV) oxide	RuO_2	12036-10-1	133.07	gray-blk tetr cry	dec 1300		7.05		i H_2O, acid
2350	Ruthenium(V) fluoride	RuF_5	14521-18-7	196.06	grn monocl cry	86.5	227	3.90		
2351	Ruthenium(VI) fluoride	RuF_6	13693-08-8	215.06	dark brn orth cry	54	200 dec	3.54		reac H_2O
2352	Ruthenium(VIII) oxide	RuO_4	20427-56-9	165.07	yel monocl prisms	25.4	40	3.29	2.03^{20}	sl H_2O; vs ctc; reac EtOH
2353	Samarium	Sm	7440-19-9	150.36	silv metal; rhomb	1072	1794	7.52		
2354	Samarium boride	SmB_6	12008-30-9	215.23	refrac solid	2580		5.07		
2355	Samarium silicide	$SmSi_2$	12300-22-0	206.53	orth cry			5.14		
2356	Samarium(II) bromide	$SmBr_2$	50801-97-3	310.17	brn cry	669				reac H_2O
2357	Samarium(II) chloride	$SmCl_2$	13874-75-4	221.27	brn cry	855		3.69		reac H_2O
2358	Samarium(II) fluoride	SmF_2	15192-17-3	188.36	purp cry					reac H_2O
2359	Samarium(II) iodide	SmI_2	32248-43-4	404.17	grn cry	520				reac H_2O
2360	Samarium(III) acetate trihydrate	$Sm(C_2H_3O_2)_3 \cdot 3H_2O$	17829-86-6	381.54	hyg yel-wh solid			1.94		s H_2O
2361	Samarium(III) bromate nonahydrate	$Sm(BrO_3)_3 \cdot 9H_2O$	63427-22-5	696.20	pink hex cry	75 dec				vs H_2O; sl EtOH
2362	Samarium(III) bromide	$SmBr_3$	13759-87-0	390.07	yel cry	640				reac H_2O
2363	Samarium(III) carbonate	$Sm_2(CO_3)_3$	5895-47-6	480.75	wh-yel pow	dec >500				
2364	Samarium(III) chloride	$SmCl_3$	10361-82-7	256.72	yel cry	682		4.46	93.8^{25}	
2365	Samarium(III) chloride hexahydrate	$SmCl_3 \cdot 6H_2O$	13465-55-9	364.81	yel cry	dec		2.38	93.8^{25}	
2366	Samarium(III) fluoride	SmF_3	13765-24-7	207.36	wh cry	1306				reac H_2O
2367	Samarium(III) iodide	SmI_3	13813-25-7	531.07	oran cry	850				reac H_2O
2368	Samarium(III) nitrate	$Sm(NO_3)_3$	10361-83-8	336.38	yel-wh hyg solid				144^{25}	s EtOH
2369	Samarium(III) nitrate hexahydrate	$Sm(NO_3)_3 \cdot 6H_2O$	13759-83-6	444.47	pale yel cry	78				s H_2O, MeOH, ace
2370	Samarium(III) oxide	Sm_2O_3	12060-58-1	348.72	yel-wh cub cry	2269	3780	7.6		
2371	Samarium(III) sulfate octahydrate	$Sm_2(SO_4)_3 \cdot 8H_2O$	13465-58-2	733.03	yel cry			2.93	2.67^{20}	
2372	Samarium(III) sulfide	Sm_2S_3	12067-22-0	396.92	gray-brn cub cry	1720		5.87		
2373	Samarium(III) telluride	Sm_2Te_3	12040-00-5	683.52	orth cry			7.31		
2374	Scandium	Sc	7440-20-2	44.956	silv metal; hex	1541	2836	2.99		
2375	Scandium boride	ScB_2	12007-34-0	66.578	refrac solid	2250		3.17		
2376	Scandium bromide	$ScBr_3$	13465-59-3	284.668	wh hyg cry	969		9.33		s H_2O
2377	Scandium chloride	$ScCl_3$	10361-84-9	151.315	wh hyg cry	967		2.4		s H_2O; i EtOH
2378	Scandium fluoride	ScF_3	13709-47-2	101.951	wh powder	1552				sl H_2O
2379	Scandium hydroxide	$Sc(OH)_3$	17674-34-9	95.978	col amorp solid					i H_2O; s dil acid
2380	Scandium iodide	ScI_3	14474-33-0	425.669	hyg yel cry	953	subl			s H_2O, EtOH, CCD
2381	Scandium nitrate	$Sc(NO_3)_3$	13465-60-6	230.971	wh cry				169^{25}	s EtOH
2382	Scandium oxide	Sc_2O_3	12060-08-1	137.910	wh cub cry	2489		3.864		s conc acid
2383	Scandium sulfate pentahydrate	$Sc_2(SO_4)_3 \cdot 5H_2O$	15292-44-1	468.176	col cry	dec 110				vs H_2O
2384	Scandium sulfide	Sc_2S_3	12166-29-9	186.107	yel orth cry	1775		2.91		
2385	Scandium telluride	Sc_2Te_3	12166-44-8	472.71	blk hex cry			5.29		
2386	Selenium (gray)	Se	7782-49-2	78.96	gray metallic cry; hex	220.8	685	4.809		i H_2O, CS_2
2387	Selenium (α form)	Se	7782-49-2	78.96	red monocl cry	trans gray Se >120	685	4.39		i H_2O, EtOH; sl eth
2388	Selenium (vitreous)	Se	7782-49-2	78.96	blk amorp solid	trans gray Se 180	685	4.28		i H_2O; sl CS_2
2389	Selenic acid	H_2SeO_4	7783-08-6	144.97	wh hyg solid	58	260 dec	2.95		vs H_2O; reac EtOH
2390	Pentafluoroorthoselenic acid	$HOSeF_5$	38989-47-8	190.96	col solid	38	47			
2391	Selenous acid	H_2SeO_3	7783-00-8	128.97	wh hyg cry	70 dec		3.0		vs H_2O; s EtOH
2392	Selenium dioxide	SeO_2	7446-08-4	110.96	wh tetr needles or powder	340 tp	315 sp	3.95	264^{22}	s EtOH, MeOH; sl ace
2393	Selenium trioxide	SeO_3	13768-86-0	126.96	wh tetr cry; hyg	118	subl	3.44		s H_2O, os

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2394	Selenium bromide	Se_2Br_2	7789-52-8	317.73	red liq	5	225 dec	3.60		reac H_2O; s CS_2, chl
2395	Selenium chloride	Se_2Cl_2	10025-68-0	228.83	yel-brn oily liq	-85	127 dec	2.774		reac H_2O; s CS_2, bz, ctc, chl
2396	Selenium tetrabromide	$SeBr_4$	7789-65-3	398.58	oran-red cry	123				reac H_2O; s CS_2, chl
2397	Selenium tetrachloride	$SeCl_4$	10026-03-6	220.77	wh-yel cry	305 tp	191.4 sp	2.6		reac H_2O
2398	Selenium tetrafluoride	SeF_4	13465-66-2	154.95	col liq	-9.5	101.6	2.75		reac H_2O; vs EtOH, eth
2399	Selenium hexafluoride	SeF_6	7783-79-1	192.95	col gas	-34.6 tp	-46.6 sp	7.887 g/L		i H_2O
2400	Selenium chloride pentafluoride	SeF_5Cl	34979-62-9	209.41	col gas	-19	4.5			
2401	Selenium oxybromide	$SeOBr_2$	7789-51-7	254.77	red-yel solid	41.6	220 dec	3.38		reac H_2O; s CS_2, bz, ctc
2402	Selenium oxychloride	$SeOCl_2$	7791-23-3	165.87	col or yel liq	8.5	177	2.44		reac H_2O; s ctc, chl, bz, tol
2403	Selenium oxyfluoride	$SeOF_2$	7783-43-9	132.96	col liq	15	125	2.8		reac H_2O
2404	Selenium oxytetrafluoride	$SeOF_4$	53319-44-1	170.95	unstab col liq	12	65			
2405	Selenium dioxydifluoride	SeO_2F_2	14984-81-7	148.96	col gas	-99.5	-8.4	6.089 g/L		reac H_2O
2406	Selenium monosulfide	SeS	7446-34-6	111.03						
2407	Selenium disulfide	SeS_2	7488-56-4	143.09	red-yel cry	100				i H_2O; s acid
2408	Selenium sulfide (Se_2S_6)	Se_2S_6	75926-26-0	350.31	oran needles	121.5		2.44		s CS_2; sl bz
2409	Selenium sulfide (Se_4S_4)	Se_4S_4	75926-28-2	444.10	red cry	113 dec		3.29		s bz; sl CS_2
2410	Selenium sulfide (Se_6S_2)	Se_6S_2	75926-30-6	537.89	oran cry	121.5				s CS_2
2411	Silicon	Si	7440-21-3	28.086	gray cry or brn amorp solid	1414	3265	2.3296		i H_2O, acid; s alk
2412	Silane	SiH_4	7803-62-5	32.118	col gas; flam	-185	-111.9	1.313 g/L		reac H_2O; i EtOH, bz
2413	Disilane	Si_2H_6	1590-87-0	62.219	col gas; flam	-129.4	-14.8	2.543 g/L		reac H_2O, ctc, chl; s EtOH, bz
2414	Trisilane	Si_3H_8	7783-26-8	92.321	flam col liq	-117.4	52.9	0.739		reac H_2O
2415	Tetrasilane	Si_4H_{10}	7783-29-1	122.421	col liq; flam	-89.9	108.1	0.792		reac H_2O
2416	2-Silyltrisilane	Si_4H_{10}	13597-87-0	122.421	col liq	-99.4	101.7	0.792		reac H_2O
2417	Cyclopentasilane	Si_5H_{10}	289-22-5	150.507	col liq	-10.5	194.3	0.963		reac H_2O
2418	Pentasilane	Si_5H_{12}	14868-53-2	152.523	col liq	-72.8	153.2	0.827		reac H_2O
2419	2-Silyltetrasilane	Si_5H_{12}	14868-54-3	152.523	col liq	-109.9	146.2	0.820		reac H_2O
2420	2,2-Disilyltrisilane	Si_5H_{12}	15947-57-6	152.523	col liq	-57.8	134.3	0.815		reac H_2O
2421	Cyclohexasilane	Si_6H_{12}	291-59-8	180.608	col liq	16.5	226			reac H_2O
2422	Hexasilane	Si_6H_{14}	14693-61-9	182.624	col liq	-44.7	193.6	0.847		reac H_2O
2423	2-Silylpentasilane	Si_6H_{14}	14868-55-4	182.624	col liq	-78.4	185.2	0.840		
2424	3-Silylpentasilane	Si_6H_{14}	14868-55-4	182.624	col liq	-69	179.5	0.843		reac H_2O
2425	Heptasilane	Si_7H_{16}	14693-65-3	212.726	col liq	-30.1	226.8	0.859		reac H_2O
2426	Bromosilane	SiH_3Br	13465-73-1	111.014	col gas	-94	1.9	4.538 g/L		
2427	Dibromosilane	SiH_2Br_2	13768-94-0	189.910	liq	-70.1	66			
2428	Tribromosilane	$SiHBr_3$	7789-57-3	268.806	flam liq	-73	109	2.7		reac H_2O
2429	Tetrabromosilane	$SiBr_4$	7789-66-4	347.702	col fuming liq	5.39	154	2.8		reac H_2O
2430	Bromotrichlorosilane	$SiBrCl_3$	13465-74-2	214.349	col liq	-62	80.3	1.826		reac H_2O
2431	Dibromodichlorosilane	$SiBr_2Cl_2$	13465-75-3	258.800	col liq	-45.5	104	2.172		reac H_2O
2432	Tribromochlorosilane	$SiBr_3Cl$	13465-76-4	303.251	col liq	-20.8	127	2.497		reac H_2O
2433	Hexabromosilane	Si_2Br_6	13517-13-0	535.595	col cry	95	265			
2434	Octabromotrisilane	Si_3Br_8	54804-32-9	723.489	col liq	46				
2435	Chlorosilane	SiH_3Cl	13465-78-6	66.563	col gas	-118	-30.4	2.721 g/L		
2436	Dichlorosilane	SiH_2Cl_2	4109-96-0	101.008	col gas; flam	-122	8.3	4.129 g/L		reac H_2O
2437	Trichlorosilane	$SiHCl_3$	10025-78-2	135.453	fuming liq	-128.2	33	1.331		reac H_2O
2438	Tetrachlorosilane	$SiCl_4$	10026-04-7	169.898	col fuming liq	-68.74	57.65	1.5		reac H_2O
2439	Chlorotrifluorosilane	$SiClF_3$	14049-36-6	120.534	col gas	-138	-70.0	4.927 g/L		reac H_2O
2440	Dichlorodifluorosilane	$SiCl_2F_2$	18356-71-3	136.989	col gas	-44	-32	5.599 g/L		reac H_2O
2441	Trichlorofluorosilane	$SiCl_3F$	14965-52-7	153.443	col gas		12.25	6.272 g/L		
2442	Trichloroiodosilane	$SiCl_3I$	13465-85-5	261.349	col liq	-60	113.5			reac H_2O
2443	Hexachlorodisilane	Si_2Cl_6	13465-77-5	268.889	col liq	2.5	146			reac H_2O
2444	Octachlorotrisilane	Si_3Cl_8	13596-23-1	367.881	col liq	-67	216			
2445	Fluorosilane	SiH_3F	13537-33-2	50.108	col gas		-98.6	2.048 g/L		
2446	Difluorosilane	SiH_2F_2	13824-36-7	68.099	col gas	-122	-77.8	2.783 g/L		
2447	Trifluorosilane	$SiHF_3$	13465-71-9	86.089	col gas	-131	-95	3.519 g/L		
2448	Tetrafluorosilane	SiF_4	7783-61-1	104.080	col gas	-90.2	-86	4.254 g/L		reac H_2O
2449	Hexafluorodisilane	Si_2F_6	13830-68-7	170.161	col gas	-18.7 tp @ 780 mmHg	-19.1 sp			reac H_2O
2450	Octafluorotrisilane	Si_3F_8	14521-14-3	236.244	col liq	-1.2	42			
2451	Decafluorotetrasilane	Si_4F_{10}	14521-15-4	302.326	col cry	68	85.1			
2452	Iodosilane	SiH_3I	13598-42-0	158.014	col liq	-57	45.6			

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2453	Diiodosilane	SiH_2I_2	13760-02-6	283.911	col liq	-1	150			
2454	Triiodosilane	$SiHI_3$	13465-72-0	409.807	liq	8	220 dec			
2455	Tetraiodosilane	SiI_4	13465-84-4	535.704	wh powder	120.5	287.35	4.1		
2456	Hexaiododisilane	Si_2I_6	13510-43-5	817.598	pale yel cry	250				
2457	Disiloxane	$(SiH_3)_2O$	13597-73-4	78.218	gas	-144	-15.2	3.197 g/L		
2458	Hexachlorodisiloxane	$(SiCl_3)_2O$	14986-21-1	284.888	liq	-28	137			
2459	Methylsilane	SiH_3CH_3	992-94-9	46.145	col gas	-156.5	-57.5			
2460	Metasilicic acid	H_2SiO_3	7699-41-4	78.100	wh amorp powder					i H_2O; s HF
2461	Orthosilicic acid	H_4SiO_4	10193-36-9	96.116	exists only in soln					
2462	Fluorosilicic acid	H_2SiF_6	16961-83-4	144.092	stab only in aq soln					s H_2O
2463	Silicon carbide (hexagonal)	SiC	409-21-2	40.097	hard grn-black hex cry	2830		3.16		i H_2O, EtOH
2464	Silicon nitride (Si_3N_4)	Si_3N_4	12033-89-5	140.284	gray refrac solid; hex	1900		3.17		
2465	Silicon monoxide	SiO	10097-28-6	44.085	blk cub cry, stable >1200			2.18		
2466	Silicon dioxide (α-quartz)	SiO_2	14808-60-7	60.085	col hex cry	trans to beta quartz 573	2950	2.648		i H_2O, acid; s HF
2467	Silicon dioxide (β-quartz)	SiO_2	14808-60-7	60.085	col hex cry	trans to tridymite 867	2950	2.533^{800}		i H_2O, acid; s HF
2468	Silicon dioxide (tridymite)	SiO_2	15468-32-3	60.085	col hex cry	trans cristobalite 1470	2950	2.265		i H_2O, acid; s HF
2469	Silicon dioxide (cristobalite)	SiO_2	14464-46-1	60.085	col hex cry	1722	2950	2.334		i H_2O, acid; s HF
2470	Silicon dioxide (vitreous)	SiO_2	60676-86-0	60.085	col amorp solid	1713	2950	2.196		i H_2O, acid; s HF
2471	Silicon monosulfide	SiS	12504-41-5	60.151	yel-red hyg powder	1090	940	1.85		reac H_2O
2472	Silicon disulfide	SiS_2	13759-10-9	92.216	wh rhomb cry	1090	subl	2.04		reac H_2O, EtOH; i bz
2473	Silicon tetraacetate	$Si(C_2H_3O_2)_4$	562-90-3	264.262	wh hyg cry	110				reac H_2O; s ace, bz
2474	Silicon tetraboride	SiB_4	12007-81-7	71.330	gray refrac solid	1870 dec		2.4		
2475	Silicotungstic acid	$H_4SiO_4 \cdot (W_3O_9)_4$	12520-88-6	2878.17	hyg yel cry					vs H_2O, EtOH
2476	Silver	Ag	7440-22-4	107.868	silv metal; cub	961.78	2162	10.5		
2477	Silver azide	AgN_3	13863-88-2	149.888	orth cry; exp	exp ≈250		4.9	0.00081^{20}	
2478	Silver subfluoride	Ag_2F	1302-01-8	234.734	yel hex cry	100 dec		8.6		reac H_2O
2479	Silver(I) acetate	$AgC_2H_3O_2$	563-63-3	166.912	wh needles or powder	dec		3.26	1.04^{20}	
2480	Silver(I) acetylide	Ag_2C_2	7659-31-6	239.757	wh powder; exp					
2481	Silver(I) acetylide (AgC_2H)	AgC_2H	13092-75-6	132.897	wh powder; exp					
2482	Silver(I) arsenate	Ag_3AsO_4	13510-44-6	462.524	red cub cry	dec		6.657	0.00085	s NH_4OH
2483	Silver(I) benzoate	$Ag(C_6H_5CO_2)$	532-31-0	228.982	powder				30^{20}	
2484	Silver(I) bromate	$AgBrO_3$	7783-89-3	235.770	wh tetr cry	360 dec		5.21	0.193^{25}	
2485	Silver(I) bromide	AgBr	7785-23-1	187.772	yel cub cry	430	1502	6.47	0.000014^{25}	i H_2O, acid, EtOH
2486	Silver(I) carbonate	Ag_2CO_3	534-16-7	275.745	yel monocl cry	218		6.077	0.0036^{20}	s acid
2487	Silver(I) chlorate	$AgClO_3$	7783-92-8	191.319	wh tetr cry	230	270 dec	4.430	17.6^{25}	sl EtOH
2488	Silver(I) chloride	AgCl	7783-90-6	143.321	wh cub cry	455	1547	5.56	0.00019^{25}	
2489	Silver(I) chlorite	$AgClO_2$	7783-91-7	175.320	yel cry	105 exp			0.55^{25}	
2490	Silver(I) chromate	Ag_2CrO_4	7784-01-2	331.730	brn-red monocl cry			5.625	0.000014^{0}	
2491	Silver(I) citrate	$Ag_3C_6H_5O_7$	126-45-4	512.705	wh cry powder					i H_2O; s HNO_3
2492	Silver(I) cyanide	AgCN	506-64-9	133.886	wh-gray hex cry	320 dec		3.95	0.0000011	i EtOH, dil acid
2493	Silver(I) dichromate	$Ag_2Cr_2O_7$	7784-02-3	431.724	red cry			4.770		sl H_2O
2494	Silver(I) diethyldithiocarbamate	$Ag(C_2H_5)_2NCS_2$	1470-61-7	256.138	powder	173				s py
2495	Silver(I) fluoride	AgF	7775-41-9	126.866	yel-brn cub cry; hyg	435	1159	5.852	172^{20}	
2496	Silver(I) hexafluoroantimonate	$AgSbF_6$	26042-64-8	343.618	powder					
2497	Silver(I) hexafluoroarsenate	$AgAsF_6$	12005-82-2	296.780	powder					
2498	Silver(I) hexafluorophosphate	$AgPF_6$	26042-63-7	252.832	powder	102 dec				
2499	Silver(I) hydrogen fluoride	$AgHF_2$	12249-52-4	146.873	hyg cry	dec				
2500	Silver(I) iodate	$AgIO_3$	7783-97-3	282.770	wh orth cry	>200		5.53	0.053^{25}	
2501	Silver(I) iodide	AgI	7783-96-2	234.772	yel powder; hex	558	1506	5.68	0.000003	i acid
2502	Silver(I) lactate monohydrate	$AgC_3H_5O_3 \cdot H_2O$	128-00-7	214.954	gray cry powder					sl H_2O, EtOH
2503	Silver(I) metaphosphate	$AgPO_3$	13465-96-8	186.840	grn glass	490		6.37		i H_2O; s HNO_3, NH_4OH
2504	Silver(I) molybdate	Ag_2MoO_4	13765-74-7	375.67	yel cub cry	483		6.18		sl H_2O
2505	Silver(I) nitrate	$AgNO_3$	7761-88-8	169.873	col rhomb cry	210	440 dec	4.35	234^{25}	sl EtOH, ace
2506	Silver(I) nitrite	$AgNO_2$	7783-99-5	153.874	yel needles	140 dec		4.453	0.415^{25}	i EtOH; reac acid
2507	Silver(I) oxalate	$Ag_2C_2O_4$	533-51-7	303.755	wh cry powder	exp 140		5.03	0.0043^{20}	

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2508	Silver(I) oxide	Ag_2O	20667-12-3	231.735	brn-blk cub cry	≈200 dec		7.2	0.0025	i EtOH; s acid, alk
2509	Silver(I) perchlorate	$AgClO_4$	7783-93-9	207.319	col cub cry; hyg	486 dec		2.806	558^{25}	s bz, py, os
2510	Silver(I) perchlorate monohydrate	$AgClO_4 \cdot H_2O$	14242-05-8	225.334	hyg wh cry	43 dec			558^{25}	
2511	Silver(I) permanganate	$AgMnO_4$	7783-98-4	226.804	viol monocl cry	dec		4.49	0.91^{18}	reac EtOH
2512	Silver(I) phosphate	Ag_3PO_4	7784-09-0	418.576	yel powder	849		6.37	0.0064	sl dil acid
2513	Silver(I) picrate monohydrate	$AgC_6H_2N_3O_7 \cdot H_2O$	146-84-9	353.979	yel cry					sl H_2O, EtOH; i chl, eth
2514	Silver(I) selenate	Ag_2SeO_4	7784-07-8	358.69	orth cry			5.72	0.118^{20}	
2515	Silver(I) selenide	Ag_2Se	1302-09-6	294.70	gray hex needles	880		8.216		i H_2O
2516	Silver(I) selenite	Ag_2SeO_3	7784-05-6	342.69	needles	530	>550 dec	5.930		sl H_2O; s acid
2517	Silver(I) sulfate	Ag_2SO_4	10294-26-5	311.799	col cry or powder	660		5.45	0.84^{25}	
2518	Silver(I) sulfide	Ag_2S	21548-73-2	247.801	gray-blk orth powder	825 (high press.)		7.23		i H_2O; s acid
2519	Silver(I) sulfite	Ag_2SO_3	13465-98-0	295.799	wh cry	100 dec			0.00046^{20}	s acid, NH_4OH
2520	Silver(I) telluride	Ag_2Te	12002-99-2	343.34	blk orth cry	955		8.4		
2521	Silver(I) tetraiodomercurate(II)	Ag_2HgI_4	7784-03-4	923.94	yel tetr cry	trans to red cub ≈40		6.1		i H_2O, dil acid
2522	Silver(I) thiocyanate	AgSCN	1701-93-5	165.951	wh powder	dec				i H_2O
2523	Silver(I) thiosulfate	$Ag_2S_2O_3$	23149-52-2	327.864	wh cry	dec				sl H_2O; s NH_4OH
2524	Silver(II) oxide	AgO	1301-96-8	123.867	gray powder; monocl or cub	>100 dec		7.5	0.0027^{25}	s alk; reac acid
2525	Silver(I) tungstate	Ag_2WO_4	13465-93-5	463.57	yel cry	620			0.015	s HNO_3, NH_4OH
2526	Silver(II) fluoride	AgF_2	7783-95-1	145.865	wh or gray hyg cry	690		4.58		reac H_2O
2527	Silver(II) oxide (Ag_2O_2)	Ag_2O_2	25455-73-6	247.735	gray-blk cub cry	>100		7.44		i H_2O; s acid, NH_4OH
2528	Sodium	Na	7440-23-5	22.990	soft silv met; cub	97.794	882.940	0.97		reac H_2O
2529	Sodium acetate	$NaC_2H_3O_2$	127-09-3	82.034	col cry	328.2		1.528	50.4^{25}	
2530	Sodium acetate trihydrate	$NaC_2H_3O_2 \cdot 3H_2O$	6131-90-4	136.079	col cry	58 dec		1.45	50.4^{25}	sl EtOH
2531	Sodium aluminate	$NaAlO_2$	1302-42-7	81.971	wh orth cry; hyg	1650		4.63		vs H_2O; i EtOH
2532	Sodium aluminum hydride	$NaAlH_4$	13770-96-2	54.004	wh hyg solid	174 dec		1.24		i eth; s thf
2533	Sodium aluminum sulfate dodecahydrate	$NaAl(SO_4)_2 \cdot 12H_2O$	10102-71-3	458.281	col cry	≈60		1.61	39.7^{20}	i EtOH
2534	Sodium amide	$NaNH_2$	7782-92-5	39.013	wh-grn orth cry	210	500 dec	1.39		reac H_2O
2535	Sodium ammonium phosphate tetrahydrate	$NaNH_4HPO_4 \cdot 4H_2O$	13011-54-6	209.069	monocl cry	≈80 dec		1.54		s H_2O; i EtOH
2536	Sodium arsenate dodecahydrate	$Na_3AsO_4 \cdot 12H_2O$	7778-43-0	424.072	col monocl prism	86 dec				s H_2O; sl EtOH; i eth
2537	Sodium arsenite	$NaAsO_2$	7784-46-5	129.911	wh-gray hyg powder			1.87		vs H_2O; i EtOH
2538	Sodium azide	NaN_3	26628-22-8	65.010	col hex cry	300 dec		1.846	40.8^{20}	sl EtOH; i eth
2539	Sodium borohydride	$NaBH_4$	16940-66-2	37.833	wh cub cry; hyg	≈400 dec		1.07	55^{20}	reac EtOH
2540	Sodium bromate	$NaBrO_3$	7789-38-0	150.892	col cub cry	381		3.34	39.4^{25}	i EtOH
2541	Sodium bromide	NaBr	7647-15-6	102.894	wh cub cry	747	1390	3.200	94.6^{25}	s EtOH
2542	Sodium bromide dihydrate	$NaBr \cdot 2H_2O$	13466-08-5	138.925	wh cry	36 dec		2.18	94.6^{25}	sl EtOH
2543	Sodium carbonate	Na_2CO_3	497-19-8	105.989	wh hyg powder	856		2.54	30.7^{25}	i EtOH
2544	Sodium carbonate monohydrate	$Na_2CO_3 \cdot H_2O$	5968-11-6	124.005	col orth cry	100 dec		2.25	30.7^{25}	i EtOH
2545	Sodium carbonate decahydrate	$Na_2CO_3 \cdot 10H_2O$	6132-02-1	286.142	col cry	34 dec		1.46	30.7^{25}	i EtOH
2546	Sodium chlorate	$NaClO_3$	7775-09-9	106.441	col cub cry	248	dec 630	2.5	100^{25}	sl EtOH
2547	Sodium chloride	NaCl	7647-14-5	58.443	col cub cry	800.7	1465	2.17	36.0^{25}	sl EtOH
2548	Sodium chlorite	$NaClO_2$	7758-19-2	90.442	wh hyg cry	≈180 dec			64^{17}	
2549	Sodium chromate	Na_2CrO_4	7775-11-3	161.974	yel orth cry	794		2.72	87.6^{25}	sl EtOH
2550	Sodium chromate tetrahydrate	$Na_2CrO_4 \cdot 4H_2O$	10034-82-9	234.035	yel hyg cry	dec			87.6^{25}	sl EtOH
2551	Sodium citrate dihydrate	$Na_3C_6H_5O_7 \cdot 2H_2O$	6132-04-3	294.099	wh cry	150 dec				vs H_2O; i EtOH, eth
2552	Sodium citrate pentahydrate	$Na_3C_6H_5O_7 \cdot 5H_2O$	6858-44-2	348.145	hyg col cry	dec 150		1.86	92^{25}	vs H_2O; sl EtOH; i eth
2553	Sodium cyanate	NaCNO	917-61-3	65.007	col needles	550		1.89		s H_2O; sl EtOH; i eth
2554	Sodium cyanide	NaCN	143-33-9	49.008	wh cub cry; hyg	562		1.6	58.2^{20}	sl EtOH
2555	Sodium cyanoborohydride	$NaBH_3(CN)$	25895-60-7	62.843	wh hyg powder	240 dec		1.12		vs H_2O; s thf; sl EtOH; i bz, eth
2556	Sodium dichromate	$Na_2Cr_2O_7$	10588-01-9	261.968	red hyg cry	357	400 dec		187^{25}	
2557	Sodium dichromate dihydrate	$Na_2Cr_2O_7 \cdot 2H_2O$	7789-12-0	297.999	oran-red monocl cry	85 dec		2.35		vs H_2O; S HOAc
2558	Sodium dihydrogen phosphate	NaH_2PO_4	7558-80-7	119.977	col monocl cry	200 dec			94.9^{25}	
2559	Sodium dihydrogen phosphate monohydrate	$NaH_2PO_4 \cdot H_2O$	10049-21-5	137.993	wh hyg cry	100 dec			94.9^{25}	i EtOH
2560	Sodium dihydrogen phosphate dihydrate	$NaH_2PO_4 \cdot 2H_2O$	13472-35-0	156.008	col orth cry	60 dec		1.91	94.9^{25}	i EtOH
2561	Sodium dihydrogen hypophosphate hexahydrate	$Na_2H_2P_2O_6 \cdot 6H_2O$	7782-95-8	314.031	monocl plates	110 dec		1.849	2.0^{25}	i EtOH

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2562	Sodium dihydrogen pyrophosphate	$Na_2H_2P_2O_7$	7758-16-9	221.939	wh powder	220 dec		≈1.9		s H_2O
2563	Sodium dithionate	$Na_2S_2O_4$	7775-14-6	174.108	gray-wh powder	52 dec			24.1^{20}	sl EtOH
2564	Sodium dithionate dihydrate	$Na_2S_2O_6 \cdot 2H_2O$	7631-94-9*	242.137	col orth cry	110 dec		2.19	15.1^{20}	i EtOH
2565	Sodium ethanolate	NaC_2H_5O	141-52-6	68.050	wh-yel hyg powder	260 dec				reac H_2O; s EtOH
2566	Sodium ferricyanide monohydrate	$Na_3Fe(CN)_6 \cdot H_2O$	14217-21-1*	298.933	red hyg cry					s H_2O; i EtOH
2567	Sodium ferrocyanide decahydrate	$Na_4Fe(CN)_6 \cdot 10H_2O$	13601-19-9	484.061	yel monocl cry	≈50 dec		1.46	20^{20}	i os
2568	Sodium fluoride	NaF	7681-49-4	41.988	col cub or tetr cry	996	1704	2.78	4.13^{25}	i EtOH
2569	Sodium fluorophosphate	Na_2PO_3F	10163-15-2	143.950	powder					
2570	Sodium formate	$NaCHO_2$	141-53-7	68.008	wh hyg cry	257.3	dec	1.92	94.9^{25}	sl EtOH
2571	Sodium germanate	Na_2GeO_3	12025-19-3	166.62	wh monocl hyg cry	1083		3.31		
2572	Sodium gold cyanide	$NaAu(CN)_2$	15280-09-8	271.991	wh-yel cry pow					s H_2O, NH_4OH
2573	Sodium gold thiosulfate dihydrate	$Na_3Au(S_2O_3)_2 \cdot 5H_2O$	10233-88-2	526.223	wh needles or prisms	dec 150		3.09		vs H_2O; i EtOH
2574	Sodium hexabromoplatinate(IV) hexahydrate	$Na_2PtBr_6 \cdot 6H_2O$	39277-13-9	828.579	cry					
2575	Sodium hexachloroiridate(IV) hexahydrate	$Na_2IrCl_6 \cdot 6H_2O$	19567-78-3	559.006	cry	600 dec				
2576	Sodium hexachloroplatinate(IV)	Na_2PtCl_6	16923-58-3	453.782	yel hyg cry				53^{16}	s EtOH
2577	Sodium hexachloroplatinate(IV) hexahydrate	$Na_2PtCl_6 \cdot 6H_2O$	16923-58-3	561.873	yel cry	110 dec		2.50	53^{16}	s EtOH; i eth
2578	Sodium hexafluoroaluminate	Na_3AlF_6	13775-53-6	209.941	col monocl cry; trans cub 560	1013		2.97		i H_2O
2579	Sodium hexafluoroantimonate	$NaSbF_6$	16925-25-0	258.740	wh cub cry			3.375	129^{20}	s EtOH, ace
2580	Sodium hexafluorophosphate monohydrate	$NaPF_6 \cdot H_2O$	20644-15-9	185.969	col orth cry			2.369	103^{0}	s EtOH, MeOH, ace
2581	Sodium hexafluorosilicate	Na_2SiF_6	16893-85-9	188.056	wh hex cry	847		2.7	0.67^{20}	i EtOH
2582	Sodium hexanitrocobaltate(III)	$Na_3Co(NO_2)_6$	14649-73-1	403.935	yel-brn cry powder					vs H_2O; sl EtOH
2583	Sodium hydride	NaH	7646-69-7	23.998	silv cub cry; flam	425 dec		1.39		reac H_2O, EtOH
2584	Sodium hydrogen arsenate	Na_2HAsO_4	7778-43-0	185.908	wh powder	≈195 dec			51^{20}	sl EtOH
2585	Sodium hydrogen arsenate heptahydrate	$Na_2HAsO_4 \cdot 7H_2O$	10048-95-0	312.014	wh monocl cry	≈50 dec		1.87	51^{20}	sl EtOH
2586	Sodium hydrogen carbonate	$NaHCO_3$	144-55-8	84.007	wh monocl cry	≈50 dec		2.20	10.3^{25}	i EtOH
2587	Sodium hydrogen fluoride	$NaHF_2$	1333-83-1	61.995	wh hex cry	>160 dec		2.08	3.25^{20}	
2588	Sodium hydrogen phosphate	Na_2HPO_4	7558-79-4	141.959	wh hyg powder			1.7	11.8^{25}	
2589	Sodium hydrogen phosphate heptahydrate	$Na_2HPO_4 \cdot 7H_2O$	7782-85-6	268.066	col cry			≈1.7	11.8^{25}	i EtOH
2590	Sodium hydrogen phosphate dodecahydrate	$Na_2HPO_4 \cdot 12H_2O$	10039-32-4	358.143	col cry	≈35 dec		≈1.5	11.8^{25}	i EtOH
2591	Sodium hydrogen sulfate	$NaHSO_4$	7681-38-1	120.061	wh hyg cry	≈315		2.43	28.5^{25}	
2592	Sodium hydrogen sulfate monohydrate	$NaHSO_4 \cdot H_2O$	10034-88-5	138.076	wh monocl cry			2.10	28.5^{25}	reac EtOH
2593	Sodium hydrogen sulfide	NaHS	16721-80-5	56.063	col rhomb cry	350		1.79		s H_2O, EtOh, eth
2594	Sodium hydrogen sulfide dihydrate	$NaHS \cdot 2H_2O$	16721-80-5	92.094	yel hyg needles	55 dec				vs H_2O, EtOH, eth
2595	Sodium hydrogen sulfite	$NaHSO_3$	7631-90-5	104.061	wh cry			1.48		s H_2O; sl EtOH
2596	Sodium hydroxide	NaOH	1310-73-2	39.997	wh orth cry; hyg	323	1388	2.13	100^{25}	s EtOH, MeOH
2597	Sodium hypochlorite	NaClO	7681-52-9	74.442	stab in aq soln	anh form exp			79.9^{25}	
2598	Sodium hypochlorite pentahydrate	$NaOCl \cdot 5H_2O$	10022-70-5	164.518	pale grn orth cry	18		1.6		s H_2O
2599	Sodium iodate	$NaIO_3$	7681-55-2	197.892	wh orth cry	422		4.28	9.47^{25}	i EtOH
2600	Sodium iodide	NaI	7681-82-5	149.894	wh cub cry; hyg	661	1304	3.67	184^{25}	s EtOH, ace
2601	Sodium iodide dihydrate	$NaI \cdot 2H_2O$	13517-06-1	185.925	hyg col monocl cry	69 dec		2.45	318^{0}	vs H_2O
2602	Sodium bismuthate	$NaBiO_3$	12232-99-4	279.968	yel-brn hyg cry					i cold H_2O, reac acid
2603	Sodium metabisulfite	$Na_2S_2O_5$	7681-57-4	190.107	wh cry				66.7^{25}	sl EtOH
2604	Sodium metaborate	$NaBO_2$	7775-19-1	65.800	wh hex cry	966	1434	2.46		s H_2O
2605	Sodium metasilicate	Na_2SiO_3	6834-92-0	122.064	wh amorp solid; hyg	1089		2.61		s cold H_2O; reac hot H_2O; i EtOH
2606	Sodium metasilicate pentahydrate	$Na_2SiO_3 \cdot 5H_2O$	13517-24-3	212.140	wh pow	72 dec				s H_2O
2607	Sodium molybdate	Na_2MoO_4	7631-95-0	205.92	col cub cry	687		≈3.5	65.0^{25}	
2608	Sodium molybdate dihydrate	$Na_2MoO_4 \cdot 2H_2O$	10102-40-6	241.95	cry powder	100 dec		≈3.5	65.0^{25}	
2609	Sodium molybdophosphate	$Na_3PO_4 \cdot 12MoO_3$	1313-30-0	1891.20	hyg solid			2.83		vs H_2O, EtOH
2610	Sodium niobate	$NaNbO_3$	12034-09-2	163.894	rhom cry	1422		4.55		i H_2O
2611	Sodium nitrate	$NaNO_3$	7631-99-4	84.995	col hex cry; hyg	306.5		2.261	91.2^{25}	sl EtOH, MeOH
2612	Sodium nitrite	$NaNO_2$	7632-00-0	68.996	wh orth cry; hyg	284	>320 dec	2.17	84.8^{25}	sl EtOH; reac acid
2613	Sodium nitroferricyanide dihydrate	$Na_2[Fe(CN)_5NO] \cdot 2H_2O$	13755-38-9	297.949	red cry			1.72	40^{16}	sl EtOH
2614	Sodium orthovanadate	Na_3VO_4	13721-39-6	183.909	col hex prisms	860				s H_2O; i EtOH
2615	Sodium oxalate	$Na_2C_2O_4$	62-76-0	133.999	wh powder	≈250 dec		2.34	3.61^{25}	i EtOH

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2616	Sodium oxide	Na_2O	1313-59-3	61.979	wh amorp powder	1134		2.27		reac H_2O
2617	Sodium perborate tetrahydrate	$NaBO_3 \cdot 4H_2O$	7632-04-4	153.861	wh cry	60 dec				reac H_2O
2618	Sodium perchlorate	$NaClO_4$	7601-89-0	122.441	wh orth cry; hyg	482 dec		2.52	205^{25}	
2619	Sodium perchlorate monohydrate	$NaClO_4 \cdot H_2O$	7791-07-3	140.456	wh hyg cry	≈130 dec		2.02	205^{25}	
2620	Sodium periodate	$NaIO_4$	7790-28-5	213.892	wh tetr cry	≈300 dec		3.86	14.4^{25}	s acid
2621	Sodium periodate trihydrate	$NaIO_4 \cdot 3H_2O$	13472-31-6	267.938	wh hex cry	175 dec		3.22	14.4^{25}	
2622	Sodium permanganate trihydrate	$NaMnO_4 \cdot 3H_2O$	10101-50-5*	195.972	red-blk hyg cry	170 dec		2.47	144^{20}	reac EtOH
2623	Sodium peroxide	Na_2O_2	1313-60-6	77.979	yel hyg powder	675		2.805		reac H_2O
2624	Sodium perrhenate	$NaReO_4$	13472-33-8	273.195	cry	300		5.39		
2625	Sodium persulfate	$Na_2S_2O_8$	7775-27-1	238.105	wh hyg cry					vs H_2O; reac EtOH
2626	Sodium phosphate	Na_3PO_4	7601-54-9	163.940	col cry	1583		2.54	14.5^{25}	s H_2O
2627	Sodium phosphate dodecahydrate	$Na_3PO_4 \cdot 12H_2O$	10101-89-0	380.124	col hex cry	≈75		1.62	14.4^{25}	i EtOH
2628	Sodium phosphate, chlorinated	$Na_3PO_4 \cdot NaOCl$	56802-99-4	238.383	wh cry				25^{25}	
2629	Sodium phosphide	Na_3P	12058-85-4	99.943	red solid	>650				reac H_2O
2630	Sodium phosphinate	NaH_2PO_2	7681-53-0	87.979	wh cry				100^{25}	
2631	Sodium phosphinate monohydrate	$NaH_2PO_2 \cdot H_2O$	10039-56-2	105.994	col hyg cry	310 dec			100^{25}	s EtOH
2632	Sodium phosphonate pentahydrate	$Na_2HPO_3 \cdot 5H_2O$	13517-23-2	216.036	wh hex plates	dec 200			429^{20}	vs H_2O; i EtOH
2633	Sodium pyrophosphate	$Na_4P_2O_7$	7722-88-5	265.902	col cry	988		2.53	7.09^{25}	
2634	Sodium selenate	Na_2SeO_4	13410-01-0	188.94	col orth cry				58.5^{25}	
2635	Sodium selenate decahydrate	$Na_2SeO_4 \cdot 10H_2O$	10102-23-5	369.09	wh cry			1.61	58.5^{25}	
2636	Sodium selenide	Na_2Se	1313-85-5	124.94	amorp solid	>875		2.62		reac H_2O
2637	Sodium selenite	Na_2SeO_3	10102-18-8	172.94	wh tetr cry				89.8^{25}	i EtOH
2638	Sodium selenite pentahydrate	$Na_2SeO_3 \cdot 5H_2O$	26970-82-1	184.054	wh tetr cry	dec				s H_2O; i EtOH
2639	Sodium stannate trihydrate	$Na_2SnO_3 \cdot 3H_2O$	12209-98-2	266.734	col hex cry	dec 140			61^{15}	vs H_2O; i EtOH, ace
2640	Sodium stearate	$NaC_{18}H_{35}O_2$	822-16-2	306.460	wh powder					sl H_2O, EtOH; vs hot H_2O
2641	Sodium succinate hexahydrate	$Na_2C_4H_4O_4 \cdot 6H_2O$	150-90-3	270.144	cry powder	120 dec			20	i EtOH
2642	Sodium sulfate	Na_2SO_4	7757-82-6	142.043	wh orth cry or powder	884		2.7	28.1^{25}	i EtOH
2643	Sodium sulfate heptahydrate	$Na_2SO_4 \cdot 7H_2O$	13472-39-4	204.152	wh cry	dec				vs H_2O
2644	Sodium sulfate decahydrate	$Na_2SO_4 \cdot 10H_2O$	7727-73-3	322.196	col monocl cry	32 dec		1.46	28.1^{25}	i EtOH
2645	Sodium sulfide	Na_2S	1313-82-2	78.045	wh cub cry; hyg	1172		1.856	20.6^{25}	sl EtOH; i eth
2646	Sodium sulfide pentahydrate	$Na_2S \cdot 5H_2O$	1313-83-3	168.121	col orth cry	120 dec		1.58	20.6^{25}	s EtOH; i eth
2647	Sodium sulfide nonahydrate	$Na_2S \cdot 9H_2O$	1313-84-4	240.183	wh-yel hyg cry	≈50 dec		1.43	20.6^{25}	sl EtOH; i eth
2648	Sodium sulfite	Na_2SO_3	7757-83-7	126.043	wh hex cry	911		2.63	30.7^{25}	i EtOH
2649	Sodium sulfite heptahydrate	$Na_2SO_3 \cdot 7H_2O$	10102-15-5	252.150	wh monocl cry; unstab			1.56	30.7^{25}	sl EtOH
2650	Sodium superoxide	NaO_2	12034-12-7	54.989	yel cub cry	552		2.2		reac H_2O
2651	Sodium tartrate dihydrate	$Na_2C_4H_4O_6 \cdot 2H_2O$	6106-24-7	230.082				1.545		s H_2O; i EtOH
2652	Sodium tellurate	Na_2TeO_4	10101-83-4	237.58	wh powder				0.8	
2653	Sodium tellurite	Na_2TeO_3	10102-20-2	221.58	wh rhomb prisms					sl H_2O
2654	Sodium tetraborate	$Na_2B_4O_7$	1330-43-4	201.220	col gl solid; hyg	743	1575	2.4	3.17^{25}	sl MeOH
2655	Sodium tetraborate tetrahydrate	$Na_2B_4O_7 \cdot 4H_2O$	12045-87-3	273.281	wh monocl cry			1.95	3.17^{25}	
2656	Sodium tetraborate pentahydrate	$Na_2B_4O_7 \cdot 5H_2O$	12045-88-4	291.296	hex cry	dec		1.88	3.17^{25}	
2657	Sodium tetraborate decahydrate	$Na_2B_4O_7 \cdot 10H_2O$	1303-96-4	381.373	wh monocl cry	75 dec		1.73	3.17^{25}	i EtOH
2658	Sodium tetrachloroaluminate	$NaAlCl_4$	7784-16-9	191.784	orth cry			2.01		s H_2O
2659	Sodium tetrachloroaurate(III) dihydrate	$NaAuCl_4 \cdot 2H_2O$	13874-02-7	397.800	oran-yel rhom cry	100 dec			150^{10}	s EtOH, eth
2660	Sodium tetrachloropalladate(II) trihydrate	$Na_2PdCl_4 \cdot 3H_2O$	13820-53-6	348.26	brn-red hyg cry					vs H_2O; s EtOH
2661	Sodium tetrachloroplatinate(II) tetrahydrate	$Na_2PtCl_4 \cdot 4H_2O$	10026-00-3	454.938	red prisms	100				s H_2O, EtOH
2662	Sodium tetrafluoroberyllate	Na_2BeF_4	13871-27-7	130.986	orth cry	575		2.47		sl H_2O
2663	Sodium tetrafluoroborate	$NaBF_4$	13755-29-8	109.795	wh orth prisms	384		2.47	108^{20}	sl EtOH
2664	Sodium thioantimonate nonahydrate	$Na_3SbS_4 \cdot 9H_2O$	10101-91-4	481.127	yel cry	dec 108		1.8	28^{20}	i EtOH
2665	Sodium thiocyanate	$NaSCN$	540-72-7	81.073	col hyg cry	287			151^{25}	
2666	Sodium thiophosphate dodecahydrate	$Na_3PO_3S \cdot 12H_2O$	10101-88-9	396.190	hex hyg leaflets	60				vs hot H_2O
2667	Sodium thiosulfate	$Na_2S_2O_3$	7772-98-7	158.108	col monocl cry	100 dec		1.69	76.4^{25}	i EtOH
2668	Sodium thiosulfate pentahydrate	$Na_2S_2O_3 \cdot 5H_2O$	10102-17-7	248.184	col cry	≈50 dec		1.69	76.4^{25}	i EtOH
2669	Sodium trimetaphosphate	$Na_3(PO_3)_3$	7785-84-4	305.885	wh cry			2.49	22	
2670	Sodium trimetaphosphate hexahydrate	$Na_3(PO_3)_3 \cdot 6H_2O$	7785-84-4	413.976	tricl-rhom hyg prisms	53		1.786	22	i EtOH
2671	Sodium tripolyphosphate	$Na_5P_3O_{10}$	7758-29-4	367.864	wh hyg powder	622			20^{25}	
2672	Sodium tungstate	Na_2WO_4	13472-45-2	293.82	wh rhom cry	695		4.18	74.2^{25}	
2673	Sodium tungstate dihydrate	$Na_2WO_4 \cdot 2H_2O$	10213-10-2	329.85	wh orth cry	100 dec		3.25	74.2^{25}	i EtOH
2674	Sodium uranate(VI) monohydrate	$Na_2U_2O_7 \cdot H_2O$	13721-34-1	652.049	yel powder					i H_2O; s acid

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2675	Sodium vanadate(V)	$NaVO_3$	13718-26-8	121.930	col monocl prisms	630			21^{25}	
2676	Sodium vanadate(V) tetrahydrate	$NaVO_3 \cdot 4H_2O$	13718-26-8	193.992	yel-wh cry powder				21^{25}	
2677	Strontium	Sr	7440-24-6	87.62	silv-wh metal; cub	777	1382	2.64		reac H_2O; s EtOH
2678	Strontium acetate	$Sr(C_2H_3O_2)_2$	543-94-2	205.71	col hyg cry	dec		2.1	40^{25}	vs H_2O
2679	Strontium arsenite tetrahydrate	$Sr(AsO_2)_2 \cdot 4H_2O$	10378-48-0	373.52	wh powder					sl H_2O, EtOH; sol dil acid
2680	Strontium bromate monohydrate	$Sr(BrO_3)_2 \cdot H_2O$	14519-18-7	361.44	yel hyg monocl cry	120 dec		3.773	39.0^{25}	
2681	Strontium bromide	$SrBr_2$	10476-81-0	247.43	wh tetr cry	657		4.216	107^{25}	
2682	Strontium bromide hexahydrate	$SrBr_2 \cdot 6H_2O$	7789-53-9	355.52	col hyg cry	88 dec			107^{25}	s EtOH; i eth
2683	Strontium carbide	SrC_2	12071-29-3	111.64	blk tetr cry	>1700		3.19		i H_2O
2684	Strontium carbonate	$SrCO_3$	1633-05-2	147.63	wh orth cry; hyg	1494		3.785	0.00034^{20}	s dil acid
2685	Strontium chlorate	$Sr(ClO_3)_2$	7791-10-8	254.52	col cry	120 dec		3.15	176^{25}	sl EtOH
2686	Strontium chloride	$SrCl_2$	10476-85-4	158.53	wh cub cry; hyg	874	1250	3.052	54.7^{25}	
2687	Strontium chloride hexahydrate	$SrCl_2 \cdot 6H_2O$	10025-70-4	266.62	col hyg cry	100 dec		1.96	54.7^{25}	s EtOH
2688	Strontium chromate	$SrCrO_4$	7789-06-2	203.61	yel monocl cry	dec		3.9	0.106^{20}	s dil acid
2689	Strontium cyanide dihydrate	$Sr(CN)_2 \cdot 4H_2O$	52870-08-3	211.72	wh hyg cry	dec				vs H_2O
2690	Strontium ferrocyanide pentadecahydrate	$SrFe(CN)_6 \cdot 15H_2O$	14654-44-5	569.80	yel monocl cry				50	
2691	Strontium fluoride	SrF_2	7783-48-4	125.62	wh cub cry or powder	1477	2460	4.24	0.021^{25}	s dil acid
2692	Strontium formate	$Sr(CHO_2)_2$	592-89-2	177.66	wh cry	71.9		2.693	9.1^{0}	
2693	Strontium formate dihydrate	$Sr(CHO_2)_2 \cdot 2H_2O$	6160-34-5	213.69	col rhom cry	100 dec		2.25	9.1^{37}	i EtOH, eth
2694	Strontium hexaboride	SrB_6	12046-54-7	152.49	blk cub cry	2235		3.39		i H_2O; s HNO_3
2695	Strontium hydride	SrH_2	13598-33-9	89.64	orth cry	1050		3.26		reac H_2O
2696	Strontium hydroxide	$Sr(OH)_2$	18480-07-4	121.64	col orth cry; hyg	535	710 dec	3.625	2.25^{25}	
2697	Strontium iodate	$Sr(IO_3)_2$	13470-01-4	437.43	tricl cry			5.045	0.165^{25}	
2698	Strontium iodide	SrI_2	10476-86-5	341.43	wh hyg cry	538	1773 dec	4.55	177^{25}	
2699	Strontium iodide hexahydrate	$SrI_2 \cdot 6H_2O$	73796-25-5	449.52	wh-yel hex cry; hyg	120 dec		4.4	177^{25}	s EtOH
2700	Strontium molybdate	$SrMoO_4$	13470-04-7	247.56	wh cry pow	1040		4.54		i H_2O
2701	Strontium niobate	$SrNb_2O_6$	12034-89-8	369.43	monocl cry	1225		5.11		i H_2O
2702	Strontium nitrate	$Sr(NO_3)_2$	10042-76-9	211.63	wh cub cry	570		2.99	80.2^{25}	sl EtOH, ace
2703	Strontium nitride	Sr_3N_2	12033-82-8	290.87	refrac solid	1200				reac H_2O; s HCl
2704	Strontium nitrite	$Sr(NO_2)_2$	13470-06-9	179.63	wh-yel hyg needles	240 dec		2.8	72.1^{30}	s H_2O
2705	Strontium orthosilicate	Sr_2SiO_4	13597-55-2	267.32	orth cry			4.5		
2706	Strontium oxalate monohydrate	$SrC_2O_4 \cdot H_2O$	814-95-9	193.65	cry pow	dec 150			0.00005^{20}	sl dil acid
2707	Strontium oxide	SrO	1314-11-0	103.62	col cub cry	2531		5.1		reac H_2O
2708	Strontium perchlorate	$Sr(ClO_4)_2$	13450-97-0	286.52	col hyg cry				306^{25}	s EtOH, MeOH
2709	Strontium permanganate trihydrate	$Sr(MnO_4)_2 \cdot 3H_2O$	14446-13-0	379.54	purp cub cry	175 dec		2.75	250^{18}	
2710	Strontium peroxide	SrO_2	1314-18-7	119.62	wh tetr cry; unstab	215 dec		4.78		reac H_2O
2711	Strontium phosphate	$Sr_3(PO_4)_2$	7446-28-8	452.80	wh powder				0.000011^{20}	s acid
2712	Strontium selenate	$SrSeO_4$	7446-21-1	230.58	orth cry			4.25	0.115^{20}	s hot HCl
2713	Strontium selenide	SrSe	1315-07-7	166.58	wh cub cry	1600		4.54		
2714	Strontium silicide	$SrSi_2$	12138-28-2	143.79	silv-gray cub cry	1100		3.35		
2715	Strontium sulfate	$SrSO_4$	7759-02-6	183.68	wh orth cry	1606		3.96	0.0135^{25}	i EtOH; sl acid
2716	Strontium sulfide	SrS	1314-96-1	119.69	gray cub cry	2226		3.70		sl H_2O; s acid
2717	Strontium sulfite	$SrSO_3$	13451-02-0	167.68	col cry	dec			0.0015^{25}	s H_2SO_4, HCl
2718	Strontium telluride	SrTe	12040-08-3	215.22	wh cub cry			4.83		
2719	Strontium thiosulfate pentahydrate	$SrS_2O_3 \cdot 5H_2O$	15123-90-7	289.82	monocl needles	100 dec		2.17	36.3^{25}	i EtOH
2720	Strontium titanate	$SrTiO_3$	12060-59-2	183.49	wh cub cry	2080		5.1		i H_2O
2721	Strontium tungstate	$SrWO_4$	13451-05-3	335.46	col tetr cry	dec		6.187	0.14^{15}	i EtOH
2722	Strontium zirconate	$SrZrO_3$	12036-39-4	226.84	col cry	2600				
2723	Sulfur (rhombic)	S	7704-34-9	32.065	yel orth cry	95.3 (trans to monocl)	444.61	2.07		i H_2O; sl EtOH, bz, eth; s CS_2
2724	Sulfur (monoclinic)	S	7704-34-9	32.065	yel monocl needles, stable 95.3-120	115.21	444.61	2.07		i H_2O; sl EtOH, bz, eth; s CS_2
2725	Sulfuric acid	H_2SO_4	7664-93-9	98.079	col oily liq	10.31	337	1.8302		vs H_2O
2726	Peroxysulfuric acid	H_2SO_5	7722-86-3	114.078	wh cry; unstab	45 dec				vs H_2O
2727	Nitrosylsulfuric acid	$HNOSO_4$	7782-78-7	127.077	prisms	73 dec				reac H_2O; s H_2SO_4
2728	Chlorosulfonic acid	$SO_2(OH)Cl$	7790-94-5	116.524	col-yel liq	-80	152	1.75		reac H_2O; s py
2729	Fluorosulfonic acid	$SO_2(OH)F$	7789-21-1	100.069	col liq	-89	163	1.726		reac H_2O
2730	Sulfurous acid	H_2SO_3	7782-99-2	82.079	exists only in aq soln					
2731	Sulfamic acid	H_2NSO_3H	5329-14-6	97.094	orth cry	≈205 dec		2.15	14.7^{0}	sl ace; i eth
2732	Sulfur dioxide	SO_2	7446-09-5	64.064	col gas	-75.5	-10.05	2.619 g/L		s H_2O, EtOh, eth, chl

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2733	Sulfur trioxide	SO_3	7446-11-9	80.063	wh needles	62.2	subl			reac H_2O
2734	Sulfur trioxide (γ-form)	SO_3	7446-11-9	80.063	col solid or liq	16.8	44.5	1.90		reac H_2O
2735	Sulfur trioxide (β-form)	SO_3	7446-11-9	80.063	wh needles	30.5	44.5			reac H_2O
2736	Sulfur bromide ($SSBr_2$)	$SSBr_2$	13172-31-1	223.938	red oily liq	-46	>25 dec	2.63		reac H_2O
2737	Sulfur chloride ($SSCl_2$)	$SSCl_2$	10025-67-9	135.036	yel-red oily liq	-77	137	1.69		reac H_2O; s EtOH, bz, eth, ctc
2738	Sulfur fluoride (SSF_2)	SSF_2	16860-99-4	102.127	col gas	-164.6	-10.6	4.174 g/L		reac H_2O
2739	Sulfur fluoride (FSSF)	FSSF	13709-35-8	102.127	col gas	-133	15	4.174 g/L		reac H_2O
2740	Sulfur dichloride	SCl_2	10545-99-0	102.971	red visc liq	-122	59.6	1.62		reac H_2O
2741	Sulfur tetrafluoride	SF_4	7783-60-0	108.059	col gas	-125	-40.45	4.417 g/L		reac H_2O
2742	Sulfur hexafluoride	SF_6	2551-62-4	146.055	col gas	-49.596 tp	-63.8 sp	5.970 g/L		sl H_2O; s EtOH
2743	Sulfur bromide pentafluoride	SF_5Br	15607-89-3	206.961	col gas	-79	3.1	8.459 g/L		
2744	Sulfur chloride pentafluoride	SF_5Cl	13780-57-9	162.510	col gas	-64	-19.05	6.642 g/L		
2745	Sulfur decafluoride	S_2F_{10}	5714-22-7	254.114	liq	-52.7	30; dec 150	2.08		i H_2O
2746	Sulfuryl amide	$(NH_2)_2SO_2$	7803-58-9	96.109	orth plates	93	250 dec			vs H_2O; sl EtOH
2747	Sulfuryl chloride	SO_2Cl_2	7791-25-5	134.970	col liq	-51	69.4	1.680		reac H_2O; s bz, tol, eth
2748	Sulfuryl fluoride	SO_2F_2	2699-79-8	102.061	col gas	-135.8	-55.4	4.172 g/L		sl H_2O, EtOH; s tol, ctc
2749	Sulfuryl bromide fluoride	SO_2BrF	13536-61-3	162.966	col liq	-86	41			reac H_2O
2750	Sulfuryl chloride fluoride	SO_2ClF	13637-84-8	118.515	col gas	-124.7	7.1	1.62^{0}		reac H_2O
2751	Pyrosulfuryl chloride	$S_2O_5Cl_2$	7791-27-7	215.033	col fuming liq	-37	151	1.837		reac H_2O
2752	Thionyl bromide	$SOBr_2$	507-16-4	207.872	yel liq	-50	140			reac H_2O
2753	Thionyl chloride	$SOCl_2$	7719-09-7	118.970	yel fuming liq	-101	75.6	1.631		reac H_2O; s bz, ctc, chl
2754	Thionyl fluoride	SOF_2	7783-42-8	86.061	col gas	-129.5	-43.8	3.518 g/L		reac H_2O; s bz, eth
2755	Sulfur fluoride oxide (SOF_4)	SOF_4	13709-54-1	124.058	col gas	-99.6	-48.5	1.95^{-82}		reac H_2O
2756	Sulfur fluoride hypofluorite	F_5SOF	15179-32-5	162.054	col gas	-86	-35.1	6.624 g/L		
2757	Tetrasulfur tetranitride	S_4N_4	28950-34-7	184.287	yel-oran cry	178.2	subl			i H_2O; reac alk, acid
2758	Tantalum	Ta	7440-25-7	180.948	gray metal; cub	3017	5458	16.4		reac HF
2759	Tantalum aluminide	$TaAl_3$	12004-76-1	261.893	gray refrac powder	≈1400		7.02		i H_2O, acid, alk
2760	Tantalum boride (TaB)	TaB	12007-07-7	191.759	refrac orth cry	2040		14.2		
2761	Tantalum boride (TaB_2)	TaB_2	12007-35-1	202.570	blk hex cry	3100		11.2		i H_2O, acid, alk
2762	Tantalum carbide (TaC)	TaC	12070-06-3	192.959	gold-brown powder; cub	3880	4780	14.3		s HF-HNO_3 mixture
2763	Tantalum carbide (Ta_2C)	Ta_2C	12070-07-4	373.907	refrac hex cry	3327		15.1		
2764	Tantalum hydride	TaH	13981-95-8	181.956	gray metallic solid			15.1		i acid
2765	Tantalum nitride	TaN	12033-62-4	194.955	blk hex cry	3090		13.7		i H_2O; sl aqua regia; reac alk
2766	Tantalum silicide	$TaSi_2$	12039-79-1	237.119	gray powder	2200		9.14		
2767	Tantalum(III) bromide	$TaBr_3$	13842-73-4	420.660	gray-grn solid	dec 220				
2768	Tantalum(III) chloride	$TaCl_3$	13569-67-0	287.307	blk-grn solid	dec 440				s H_2O
2769	Tantalum(IV) bromide	$TaBr_4$	13842-76-7	500.564	dark blue solid	392		5.77		reac H_2O
2770	Tantalum(IV) chloride	$TaCl_4$	13569-72-7	322.760	dark grn solid	dec 300	subl	4.35		reac H_2O
2771	Tantalum(IV) iodide	TaI_4	14693-80-2	688.566	gray-blk solid	400 dec				reac H_2O
2772	Tantalum(IV) oxide	TaO_2	12036-14-5	212.947	tetr cry			10.0		
2773	Tantalum(IV) selenide	$TaSe_2$	12039-55-3	338.87	hex cry			6.7		
2774	Tantalum(IV) sulfide	TaS_2	12143-72-5	245.078	blk hex cry	>3000		6.86		i H_2O
2775	Tantalum(IV) telluride	$TaTe_2$	12067-66-2	436.15	monocl cry			9.4		
2776	Tantalum(V) bromide	$TaBr_5$	13451-11-1	580.468	yel cry powder	265.8	348.8	4.67		
2777	Tantalum(V) chloride	$TaCl_5$	7721-01-9	358.213	yel-wh monocl cry; hyg	216.6	239	3.68		reac H_2O; s EtOH
2778	Tantalum(V) fluoride	TaF_5	7783-71-3	275.940	wh monocl cry; hyg	96.9	229.5	4.74		s H_2O, eth; sl CS_2, ctc
2779	Tantalum(V) iodide	TaI_5	14693-81-3	815.470	blk hex cry; hyg	496	543	5.80		reac H_2O
2780	Tantalum(V) oxide	Ta_2O_5	1314-61-0	441.893	wh rhomb cry or powder	1875		8.24		i H_2O, EtOH, acid; s HF
2781	Technetium	Tc	7440-26-8	98	hex cry	2157	4265	11		
2782	Technetium(V) fluoride	TcF_5	31052-14-9	193	yel solid	50	dec			
2783	Technetium(VI) fluoride	TcF_6	13842-93-8	212	yel cub cry	37.4	55.3	3.0		
2784	Tellurium	Te	13494-80-9	127.60	gray-wh rhomb cry	449.51	988	6.232		i H_2O, bz, CS_2
2785	Telluric(VI) acid	H_6TeO_6	7803-68-1	229.64	wh monocl cry	136		3.07	50.1^{30}	
2786	Tellurous acid	H_2TeO_3	10049-23-7	177.61	wh cry	40 dec		3.0		sl H_2O; s dil acid, alk
2787	Tellurium dioxide	TeO_2	7446-07-3	159.60	wh orth cry	733	1245	5.9		i H_2O; s alk, acid
2788	Tellurium trioxide	TeO_3	13451-18-8	175.60	yel-oran cry	430		5.07		i H_2O

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2789	Tellurium dibromide	$TeBr_2$	7789-54-0	287.41	grn-brn hyg cry	210	339			reac H_2O; s eth; sl chl
2790	Tellurium dichloride	$TeCl_2$	10025-71-5	198.51	blk amorp solid; hyg	208	328	6.9		reac H_2O; i ctc
2791	Tellurium tetrabromide	$TeBr_4$	10031-27-3	447.22	yel-oran monocl cry	380	≈420 dec	4.3		reac H_2O; s eth
2792	Tellurium tetrachloride	$TeCl_4$	10026-07-0	269.41	wh monocl cry; hyg	224	387	3.0		reac H_2O; s EtOH, tol
2793	Tellurium tetrafluoride	TeF_4	15192-26-4	203.59	col cry	129	195 dec			reac H_2O
2794	Tellurium decafluoride	Te_2F_{10}	53214-07-6	445.18	col liq	-33.7	59			
2795	Tellurium tetraiodide	TeI_4	7790-48-9	635.22	blk orth cry	280		5.05		reac H_2O; sl ace
2796	Tellurium hexafluoride	TeF_6	7783-80-4	241.59	col gas	-37.6 tp	-38.9 sp	9.875 g/L		reac H_2O
2797	Terbium	Tb	7440-27-9	158.925	silv metal; hex	1359	3230	8.23		
2798	Terbium nitride	TbN	12033-64-6	172.932	cub cry			9.55		
2799	Terbium silicide	$TbSi_2$	12039-80-4	215.096	orth cry			6.66		
2800	Terbium(III) bromide	$TbBr_3$	14456-47-4	398.637	wh hex cry	830	1490			s H_2O
2801	Terbium(III) chloride	$TbCl_3$	10042-88-3	265.284	wh orth cry; hyg	582		4.35		s H_2O
2802	Terbium(III) chloride hexahydrate	$TbCl_3 \cdot 6H_2O$	13798-24-8	373.375	hyg cry			4.35		vs H_2O
2803	Terbium(III) iodide	TbI_3	13813-40-6	539.638	hex cry; hyg	955		≈5.2		s H_2O
2804	Terbium(III) nitrate	$Tb(NO_3)_3$	10043-27-3	344.940	pink hyg solid				157^{25}	s EtOH
2805	Terbium(III) nitrate hexahydrate	$Tb(NO_3)_3 \cdot 6H_2O$	13451-19-9	453.031	col needles	89				s H_2O, EtOH, ace
2806	Terbium(III) oxide	Tb_2O_3	12036-41-8	365.849	wh cub cry	2303		7.91		
2807	Terbium(III) sulfate octahydrate	$Tb_2(SO_4)_3 \cdot 8H_2O$	13842-67-6	750.161	wh cry	dec 360				sl H_2O
2808	Terbium(III) sulfide	Tb_2S_3	12138-11-3	414.046	cub cry			6.35		
2809	Terbium(III) fluoride	TbF_3	13708-63-9	215.920	wh solid	1175	2280			i H_2O
2810	Terbium(IV) fluoride	TbF_4	36781-15-4	234.919	wh monocl cry	dec 300				i H_2O
2811	Thallium	Tl	7440-28-0	204.383	soft blue-wh metal	304	1473	11.8		i H_2O; reac acid
2812	Thallium(I) acetate	$TlC_2H_3O_2$	563-68-8	263.427	hyg wh cry	131		3.68		s H_2O, EtOH
2813	Thallium(I) azide	TlN_3	13847-66-0	246.403	yel cry	334	exp			s H_2O
2814	Thallium(I) bromate	$TlBrO_3$	14550-84-6	332.285	col needles	120 dec			0.49^{30}	s EtOH
2815	Thallium(I) bromide	TlBr	7789-40-4	284.287	yel cub cry	460	819	7.5	0.059^{20}	
2816	Thallium(I) carbonate	Tl_2CO_3	6533-73-9	468.776	wh monocl cry	273		7.11	4.69^{20}	i EtOH
2817	Thallium(I) chlorate	$TlClO_3$	13453-30-0	287.834	col hex cry	dec 500		5.5	3.92^{20}	
2818	Thallium(I) chloride	TlCl	7791-12-0	239.836	wh cub cry	431	720	7.0	0.33^{20}	i EtOH
2819	Thallium(I) chromate	Tl_2CrO_4	13473-75-1	524.761	yel cry				0.003^{20}	sl acid, alk
2820	Thallium(I) cyanide	TlCN	13453-34-4	230.401	wh hex plates			6.523		s H_2O, acid, EtOH
2821	Thallium(I) ethanolate	TlC_2H_5O	20398-06-5	249.443	cloudy liq	-3	130 dec	3.49		reac H_2O
2822	Thallium(I) fluoride	TlF	7789-27-7	223.381	wh orth cry	326	826	8.36	245^{25}	
2823	Thallium(I) formate	$TlCHO_2$	992-98-3	249.401	hyg col needles	101		4.97		vs H_2O; s MeOH
2824	Thallium(I) hexafluorophosphate	$TlPF_6$	60969-19-9	349.347	wh cub cry			4.6		
2825	Thallium(I) hydroxide	TlOH	12026-06-1	221.390	yel needles	139 dec		7.44	34.3^{18}	
2826	Thallium(I) iodate	$TlIO_3$	14767-09-0	379.285	wh needles				0.058	sl HNO_3
2827	Thallium(I) iodide	TlI	7790-30-9	331.287	yel cry powder	441.7	824	7.1	0.0085^{20}	i EtOH
2828	Thallium(I) molybdate	Tl_2MoO_4	34128-09-1	568.71	yel-wh cub cry					i H_2O
2829	Thallium(I) nitrate	$TlNO_3$	10102-45-1	266.388	wh cry	206	450 dec	5.55	9.55^{20}	s H_2O; i EtOH
2830	Thallium(I) nitrite	$TlNO_2$	13826-63-6	250.389	yel cub cry	186		5.7	32.1^{25}	s H_2O
2831	Thallium(I) oxalate	$Tl_2C_2O_4$	30737-24-7	496.786	wh powder			6.31	1.83^{20}	
2832	Thallium(I) oxide	Tl_2O	1314-12-1	424.766	blk rhomb cry; hyg	579	≈1080	9.52		s H_2O, EtOH
2833	Thallium(I) perchlorate	$TlClO_4$	13453-40-2	303.834	col orth cry	501		4.89	19.7^{30}	
2834	Thallium(I) selenate	Tl_2SeO_4	7446-22-2	551.73	orth cry	>400		6.875	2.8^{20}	i EtOH, eth
2835	Thallium(I) selenide	Tl_2Se	15572-25-5	487.73	gray plates	340				i H_2O, acid
2836	Thallium(I) sulfate	Tl_2SO_4	7446-18-6	504.830	wh rhomb prisms	632		6.77	5.47^{25}	
2837	Thallium(I) sulfide	Tl_2S	1314-97-2	440.832	blue-blk cry	457	1367	8.39	0.02^{20}	sl alk; s acid
2838	Thallium(III) acetate	$Tl(C_2H_3O_2)_3$	2570-63-0	381.514	hyg wh platelets	182 dec				
2839	Thallium(III) bromide tetrahydrate	$TlBr_3 \cdot 4H_2O$	13701-90-1	516.157	yel orth cry			3.65		s H_2O, EtOH
2840	Thallium(III) chloride	$TlCl_3$	13453-32-2	310.742	monocl cry	155		4.7		vs H_2O, EtOH, eth
2841	Thallium(III) chloride tetrahydrate	$TlCl_3 \cdot 4H_2O$	13453-32-2*	382.804	orth cry			3.00		s H_2O
2842	Thallium(III) fluoride	TlF_3	7783-57-5	261.378	wh orth cry; hyg	550 dec		8.65		reac H_2O
2843	Thallium(III) nitrate	$Tl(NO_3)_3$	13746-98-0	390.398	col cry					reac H_2O
2844	Thallium(III) oxide	Tl_2O_3	1314-32-5	456.765	brn cub cry	834		10.2		i H_2O; reac acid
2845	Thallium(III) sulfate	$Tl_2(SO_4)_3$	16222-66-5	696.955	col leaflets					reac H_2O
2846	Thallium selenide	TlSe	12039-52-0	283.34	blk solid	330				i H_2O, acid
2847	Thorium	Th	7440-29-1	232.038	soft gray-wh metal; cub	1750	4788	11.7		s acid
2848	Thorium hydride	ThH_2	16689-88-6	234.054	tetr cry			9.5		
2849	Thorium boride	ThB_6	12229-63-9	296.904	refrac solid	2450		6.99		
2850	Thorium(IV) bromide	$ThBr_4$	13453-49-1	551.654	wh hyg cry	679			65^{20}	

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2851	Thorium carbide	ThC	12012-16-7	244.049	cub cry	2500		10.6		reac H_2O
2852	Thorium dicarbide	ThC_2	12071-31-7	256.059	yel monocl cry	≈2650		9.0		reac H_2O
2853	Thorium(IV) chloride	$ThCl_4$	10026-08-1	373.850	gray-wh tetr needles; hyg	770	921	4.59		s H_2O, EtOH
2854	Thorium(IV) fluoride	ThF_4	13709-59-6	308.032	wh monocl cry; hyg	1110	1680	6.1		
2855	Thorium(IV) iodide	ThI_4	7790-49-0	739.656	wh-yel monocl cry	566	837			
2856	Thorium(IV) nitrate	$Th(NO_3)_4$	13823-29-5	480.058	hyg wh plates	55 dec				vs H_2O, EtOH
2857	Thorium(IV) nitrate tetrahydrate	$Th(NO_3)_4 \cdot 4H_2O$	13470-07-0	552.119	wh hyg cry	500 dec			191^{20}	s EtOH
2858	Thorium nitride	ThN	12033-65-7	246.045	refrac cub cry	2820		11.6		reac H_2O
2859	Thorium(IV) oxide	ThO_2	1314-20-1	264.037	wh cub cry	3350	4400	10.0		i H_2O, alk; sl acid
2860	Thorium(IV) selenide	$ThSe_2$	60763-24-8	389.96	orth cry			8.5		
2861	Thorium orthosilicate	$ThSiO_4$	14553-44-7	324.122	brn tetr cry			6.7		
2862	Thorium silicide	$ThSi_2$	12067-54-8	288.209	tetr cry	1850		7.9		
2863	Thorium(IV) sulfate nonahydrate	$Th(SO_4)_2 \cdot 9H_2O$	10381-37-0	586.301	wh monocl cry	dec		2.8	4.2^{20}	
2864	Thorium(IV) sulfide	ThS_2	12138-07-7	296.168	dark brn cry	1905		7.30		i H_2O; s acid
2865	Thulium	Tm	7440-30-4	168.934	silv metal; hex	1545	1950	9.32		s dil acid
2866	Thulium(II) bromide	$TmBr_2$	64171-97-7	328.742	dark grn solid	619				
2867	Thulium(II) chloride	$TmCl_2$	22852-11-5	239.840	red or grn cry	718				reac H_2O
2868	Thulium(II) iodide	TmI_2	60864-26-8	422.743	blk hyg solid	756				reac H_2O
2869	Thulium(III) bromide	$TmBr_3$	14456-51-0	408.646	wh hyg cry	954				s H_2O
2870	Thulium(III) chloride	$TmCl_3$	13537-18-3	275.293	yel hyg cry	845				s H_2O
2871	Thulium(III) chloride heptahydrate	$TmCl_3 \cdot 7H_2O$	13778-39-7	401.400	hyg cry					s H_2O, EtOH
2872	Thulium(III) fluoride	TmF_3	13760-79-7	225.929	wh cry	1158				s H_2O
2873	Thulium(III) hydroxide	$Tm(OH)_3$	1311-33-7	219.956	wh or grn prec					i H_2O
2874	Thulium(III) iodide	TmI_3	13813-43-9	549.647	yel hyg cry	1021				
2875	Thulium(III) nitrate	$Tm(NO_3)_3$	14985-19-4	354.949	grn hyg solid				212^{25}	s EtOH
2876	Thulium(III) nitrate pentahydrate	$Tm(NO_3)_3 \cdot 5H_2O$	36548-87-5	445.025	grn hyg cry					s H_2O, EtOH, ace
2877	Thulium(III) oxalate hexahydrate	$Tm_2(C_2O_4)_3 \cdot 6H_2O$	26677-68-9	710.016	grn solid	dec 50				s alk oxalates
2878	Thulium(III) oxide	Tm_2O_3	12036-44-1	385.866	grn-wh cub cry	2341	3945	8.6		sl acid
2879	Tin (gray)	Sn	7440-31-5	118.710	cub cry	trans to wh Sn 13.2	2602	5.769		
2880	Tin (white)	Sn	7440-31-5	118.710	silv tetr cry	231.93	2602	7.287		
2881	Stannane	SnH_4	2406-52-2	122.742	unstab col gas	-146	-51.8	5.017 g/L		
2882	Methylstannane	SnH_3CH_3	1631-78-3	136.769	col gas		1.4	5.590 g/L		reac H_2O
2883	(Dimethylamino)trimethystannane	$Sn(CH_3)_3N(CH_3)_2$	993-50-0	207.890	liq	1	126	1.22		reac H_2O
2884	Tin monophosphide	SnP	25324-56-5	149.684	dull metallic solid	540				
2885	Tin triphosphide	Sn_4P_3	12286-33-8	567.761	wh cry	≈550		5.2		
2886	Tin(II) acetate	$Sn(C_2H_3O_2)_2$	638-39-1	236.799	wh orth cry	183	subl	2.31		s dil HCl
2887	Tin(II) bromide	$SnBr_2$	10031-24-0	278.518	yel powder	215	639	5.12	85^0	s EtOH, eth, ace
2888	Tin(II) chloride	$SnCl_2$	7772-99-8	189.616	wh orth cry	247.0	623	3.90	178^{10}	s EtOH, ace, eth; i xyl
2889	Tin(II) chloride dihydrate	$SnCl_2 \cdot 2H_2O$	10025-69-1	225.647	wh monocl cry	37 dec		2.71	178^{10}	s EtOH, NaOH; vs HCl
2890	Tin(II) fluoride	SnF_2	7783-47-3	156.707	wh monocl cry; hyg	215	850	4.57		s H_2O; i EtOH, eth, chl
2891	Tin(II) hexafluorozirconate	$SnZrF_6$	12419-43-1	323.924	cry			4.21		s H_2O
2892	Tin(II) hydroxide	$Sn(OH)_2$	12026-24-3	152.725	wh amorp solid					
2893	Tin(II) iodide	SnI_2	10294-70-9	372.519	red-oran powder	320	714	5.28	0.98^{20}	s bz, chl, CS_2
2894	Tin(II) oxalate	SnC_2O_4	814-94-8	206.729	wh powder	280 dec		3.56		i H_2O; s dil HCl
2895	Tin(II) oxide	SnO	21651-19-4	134.709	blue-blk tetr cry	1080 dec		6.45		i H_2O, EtOH; s acid
2896	Tin(II) pyrophosphate	$Sn_2P_2O_7$	15578-26-4	411.363	wh amorp powder	400 dec		4.009		i H_2O; s conc acid
2897	Tin(II) selenide	SnSe	1315-06-6	197.67	gray orth cry	861		6.18		i H_2O; s aqua regia
2898	Tin(II) sulfate	$SnSO_4$	7488-55-3	214.773	wh orth cry	378 dec		4.15	18.8^{19}	
2899	Tin(II) sulfide	SnS	1314-95-0	150.775	gray orth cry	881	1210	5.08		i H_2O; s conc acid
2900	Tin(II) tartrate	$SnC_4H_4O_6$	815-85-0	266.781	wh cry powder					s H_2O, dil HCl
2901	Tin(II) telluride	SnTe	12040-02-7	246.31	gray cub cry	806		6.5		
2902	Tin(IV) bromide	$SnBr_4$	7789-67-5	438.326	wh cry	29.1	205	3.34		vs H_2O; s EtOH
2903	Tin(IV) chloride	$SnCl_4$	7646-78-8	260.522	col fuming liq	-34.07	114.15	2.234		reac H_2O; s EtOH, ctc, bz, ace
2904	Tin(IV) chloride pentahydrate	$SnCl_4 \cdot 5H_2O$	10026-06-9	350.598	wh-yel cry	56 dec		2.04		vs H_2O; s EtOH
2905	Tin(IV) chromate	$Sn(CrO_4)_2$	38455-77-5	350.697	brn-yel cry powder	dec				s H_2O
2906	Tin(IV) fluoride	SnF_4	7783-62-2	194.704	wh tetr cry	442	705 subl	4.78		reac H_2O
2907	Tin(IV) iodide	SnI_4	7790-47-8	626.328	yel-brn cub cry	143	364.35	4.46		reac H_2O; s EtOH, bz, chl, eth
2908	Tin(IV) oxide	SnO_2	18282-10-5	150.709	gray tetr cry	1630		6.85		i H_2O, EtOH; s hot conc alk
2909	Tin(IV) selenide	$SnSe_2$	20770-09-6	276.63	red-brn cry	650		≈5.0		i H_2O; s alk, conc acid

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2910	Tin(IV) selenite	$Sn(SeO_3)_2$	7446-25-5	372.63	cry powder					i H_2O; s hot HCl
2911	Tin(IV) sulfide	SnS_2	1315-01-1	182.840	gold-yel hex cry	600 dec		4.5		i H_2O; s alk, aqua regia
2912	Titanium	Ti	7440-32-6	47.867	gray metal; hex	1668	3287	4.506		
2913	Titanocene dichloride	$Ti(C_5H_5)_2Cl_2$	1271-19-8	248.959	red cry	289		1.60		sl H_2O, bz; s chl, EtOH, tol
2914	Titanium hydride	TiH_2	7704-98-5	49.883	gray-blk powder	≈450 dec		3.75		i H_2O
2915	Titanium boride	TiB_2	12045-63-5	69.489	gray refrac solid; hex	3225		4.38		
2916	Titanium carbide	TiC	12070-08-5	59.878	cub cry	3067		4.93		i H_2O; s HNO_3
2917	Titanium nitride	TiN	25583-20-4	61.874	yel-brn cub cry	2947		5.21		i H_2O; s aqua regia
2918	Titanium phosphide	TiP	12037-65-9	78.841	gray hex cry	1990		4.08		
2919	Titanium silicide	$TiSi_2$	12039-83-7	104.038	blk orth cry	1500		4.0		i H_2O, acid, alk; s HF
2920	Titanium(II) bromide	$TiBr_2$	13783-04-5	207.675	blk powder	dec 400		4.0		reac H_2O
2921	Titanium(II) chloride	$TiCl_2$	10049-06-6	118.773	blk hex cryc	1035	1500	3.13		reac H_2O; s EtOH; i chl, eth
2922	Titanium(II) iodide	TiI_2	13783-07-8	301.676	blk hex cry	dec 400		5.02		reac H_2O
2923	Titanium(II) oxide	TiO	12137-20-1	63.866	yel cub cry	1770	3227	4.95		
2924	Titanium(II) sulfide	TiS	12039-07-5	79.932	brn hex cry	1927		3.85		i H_2O; s conc acid
2925	Titanium(III) bromide	$TiBr_3$	13135-31-4	287.579	viol hex cry	dec 400				s H_2O
2926	Titanium(III) chloride	$TiCl_3$	7705-07-9	154.226	red-viol hex cry; hyg	425 dec	960	2.64		reac H_2O
2927	Titanium(III) fluoride	TiF_3	13470-08-1	104.862	viol hex cry	950 dec		2.98		i H_2O, dil acid, alk
2928	Titanium(III) iodide	TiI_3	13783-08-9	428.580	viol cry	dec 350				
2929	Titanium(III) oxide	Ti_2O_3	1344-54-3	143.732	blk hex cry	1842		4.486		s hot HF
2930	Titanium(III) sulfate	$Ti_2(SO_4)_3$	10343-61-0	383.922	grn cry					i H_2O, EtOH; s dil HCl
2931	Titanium(III) sulfide	Ti_2S_3	12039-16-6	191.929	blk hex cry			3.56		
2932	Titanium(III,IV) oxide	Ti_3O_5	12065-65-5	223.598	blk monocl cry	1777		4.24		
2933	Titanium(IV) bromide	$TiBr_4$	7789-68-6	367.483	yel-oran cub cry; hyg	38.3	233.5	3.37		reac H_2O
2934	Titanium(IV) chloride	$TiCl_4$	7550-45-0	189.679	col or yel liq	-24.12	136.45	1.73		reac H_2O; s EtOH
2935	Titanium(IV) fluoride	TiF_4	7783-63-3	123.861	wh hyg powder	377	subl 284	2.798		reac H_2O; s EtOH, py
2936	Titanium(IV) iodide	TiI_4	7720-83-4	555.485	red hyg powder	155	377	4.3		reac H_2O
2937	Titanium(IV) oxide (anatase)	TiO_2	1317-70-0	79.866	brn tetr cry	1560		3.9		
2938	Titanium(IV) oxide (brookite)	TiO_2	12188-41-9	79.866	wh orth cry			4.17		
2939	Titanium(IV) oxide (rutile)	TiO_2	1317-80-2	79.866	wh tetr cry	1843	≈3000	4.17		i H_2O, dil acid; s conc acid
2940	Titanium(IV) oxysulfate monohydrate	$TiOSO_4 \cdot H_2O$	13825-74-6*	177.944	col orth cry			2.71		reac H_2O
2941	Titanium(IV) sulfate	$Ti(SO_4)_2$	13693-11-3	239.992	wh-yel hyg cry	150 dec				s H_2O
2942	Titanium(IV) sulfide	TiS_2	12039-13-3	111.997	yel-brn hex cry; hyg			3.37		s H_2SO_4
2943	Tungsten	W	7440-33-7	183.84	gray-wh metal; cub	3422	5555	19.3		
2944	Tungstic acid	H_2WO_4	7783-03-1	249.85	yel amorp powder	100 dec		5.5		i H_2O, acid; s alk
2945	Tungsten boride (W_2B)	W_2B	12007-10-2	378.49	refrac blk powder	2670		16.0		i H_2O
2946	Tungsten boride (WB)	WB	12007-09-9	194.65	blk refrac powder	2665		15.2		i H_2O
2947	Tungsten boride (W_2B_5)	W_2B_5	12007-98-6	421.74	refrac solid	2370		11.0		i H_2O
2948	Tungsten carbide (W_2C)	W_2C	12070-13-2	379.69	refrac hex cry	≈2800		14.8		i H_2O
2949	Tungsten carbide (WC)	WC	12070-12-1	195.85	gray hex cry	2785		15.6		i H_2O; s HNO_3/HF
2950	Tungsten carbonyl	$W(CO)_6$	14040-11-0	351.90	wh cry	170 dec	subl	2.65		i H_2O; s os
2951	Tungsten nitride (WN_2)	WN_2	60922-26-1	211.85	hex cry	600 dec		7.7		
2952	Tungsten nitride (W_2N)	W_2N	12033-72-6	381.69	gray cub cry	dec		17.8		
2953	Tungsten silicide (WSi_2)	WSi_2	12039-88-2	240.01	blue-gray tetr cry	2160		9.3		i H_2O
2954	Tungsten silicide (W_5Si_3)	W_5Si_3	12039-95-1	1003.46	blue-gray refrac solid	2320		14.4		
2955	Tungsten(II) bromide	WBr_2	13470-10-5	343.65	yel powder	dec 400				i H_2O
2956	Tungsten(II) chloride	WCl_2	13470-12-7	254.75	gray solid	dec 500		5.44		sl H_2O
2957	Tungsten(II) iodide	WI_2	13470-17-2	437.65	oran-brn cry	dec 800		6.79		i H_2O
2958	Tungsten(III) bromide	WBr_3	15163-24-3	423.55	blk hex cry	dec 180				i H_2O
2959	Tungsten(III) chloride	WCl_3	20193-56-0	290.20	red solid	550 dec	subl			reac H_2O
2960	Tungsten(III) iodide	WI_3	15513-69-6	564.55	blk solid	dec r.t.				i H_2O; s ace; sl EtOH, chl
2961	Tungsten(IV) bromide	WBr_4	14055-81-3	503.46	blk orth cry		240 subl			reac H_2O
2962	Tungsten(IV) chloride	WCl_4	13470-13-8	325.65	blk hyg powder	450 dec		4.62		reac H_2O
2963	Tungsten(IV) fluoride	WF_4	13766-47-7	259.83	red-brn cry	dec 800				reac H_2O; s MeCN; i bz, tol, ctc

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
2964	Tungsten(IV) iodide	WI_4	14055-84-6	691.46	blk cry	dec				reac H_2O; s EtOH; i eth chl
2965	Tungsten(IV) oxide	WO_2	12036-22-5	215.84	brn monocl cry	≈1500 dec	1730	10.8		i H_2O, os
2966	Tungsten(IV) selenide	WSe_2	12067-46-8	341.76	gray hex cry			9.2		
2967	Tungsten(IV) sulfide	WS_2	12138-09-9	247.97	gray hex cry	1250 dec		7.6		i H_2O, HCl, alk
2968	Tungsten(IV) telluride	WTe_2	12067-76-4	439.04	gray orth cry	1020		9.43		
2969	Tungsten(V) bromide	WBr_5	13470-11-6	583.36	brn-blk hyg solid	286	333			reac H_2O
2970	Tungsten(V) chloride	WCl_5	13470-14-9	361.11	blk-grn hyg cry	253	286	3.88		reac H_2O
2971	Tungsten(V) ethanolate	$W(C_2H_5O)_5$	62571-53-3	409.14	powder		105(0.05 mmHg)			s EtAc
2972	Tungsten(V) fluoride	WF_5	19357-83-6	278.83	yel solid	dec 20				reac H_2O
2973	Tungsten(V) oxytribromide	$WOBr_3$	20213-56-3	439.55	dark brn tetr cry			≈5.9		
2974	Tungsten(V) oxytrichloride	$WOCl_3$	14249-98-0	306.20	grn tetr cry			≈4.6		
2975	Tungsten(VI) bromide	WBr_6	13701-86-5	663.26	blue-blk cry	309		6.9		reac H_2O
2976	Tungsten(VI) chloride	WCl_6	13283-01-7	396.56	purp hex cry; hyg	282	337	3.52		reac H_2O; s EtOH, os
2977	Tungsten(VI) dioxydibromide	WO_2Br_2	13520-75-7	375.65	red cry		440 subl			
2978	Tungsten(VI) dioxydichloride	WO_2Cl_2	13520-76-8	286.75	yel orth cry	265		4.67		i H_2O
2979	Tungsten(VI) dioxydiiodide	WO_2I_2	14447-89-3	469.65	grn monocl cry	400 dec		6.39		
2980	Tungsten(VI) fluoride	WF_6	7783-82-6	297.83	yel liq or col gas	1.9	17.1	3.44		reac H_2O; vs ctc, cyhex
2981	Tungsten(VI) oxide	WO_3	1314-35-8	231.84	yel powder	1473	≈1700	7.2		i H_2O, os; sl acid; s alk
2982	Tungsten(VI) oxytetrabromide	$WOBr_4$	13520-77-9	519.46	red tetr cry	277	327	≈5.5		reac H_2O
2983	Tungsten(VI) oxytetrachloride	$WOCl_4$	13520-78-0	341.65	red hyg cry	210	230	11.92		reac H_2O; s bz, CS_2
2984	Tungsten(VI) oxytetrafluoride	WOF_4	13520-79-1	275.83	wh monocl cry	105	185.9	5.07		reac H_2O
2985	Tungsten(VI) sulfide	WS_3	12125-19-8	280.04	brn powder					sl H_2O; s alk
2986	Uranium	U	7440-61-1	238.029	silv-wh orth cry	1135	4131	19.1		
2987	Uranium boride (UB_2)	UB_2	12007-36-2	259.651	refrac solid	2430		12.7		
2988	Uranium boride (UB_4)	UB_4	12007-84-0	281.273	refrac solid	2530		9.32		i H_2O
2989	Uranium carbide (UC)	UC	12070-09-6	250.040	gray cub cry	2790				
2990	Uranium carbide (UC_2)	UC_2	12071-33-9	262.050	gray tetr cry	2350	4370	11.3		reac H_2O; sl EtOH
2991	Uranium carbide (U_2C_3)	U_2C_3	12076-62-9	512.090	gray cub cry	≈1700 dec		12.7		
2992	Uranium nitride (UN)	UN	25658-43-9	252.036	gray cub cry	2805		14.3		i H_2O
2993	Uranium nitride (U_2N_3)	U_2N_3	12033-83-9	518.078	cub cry	dec		11.3		
2994	Uranium(III) bromide	UBr_3	13470-19-4	477.741	red hyg cry	727				s H_2O
2995	Uranium(III) chloride	UCl_3	10025-93-1	344.388	grn hyg cry	837		5.51		vs H_2O; i bz, ctc
2996	Uranium(III) fluoride	UF_3	13775-06-9	295.024	blk hex cry	1495		8.9		i H_2O; s acid
2997	Uranium(III) hydride	UH_3	13598-56-6	241.053	gray-blk cub cry			11.1		
2998	Uranium(III) iodide	UI_3	13775-18-3	618.742	blk hyg cry	766				s H_2O
2999	Uranium(IV) bromide	UBr_4	13470-20-7	557.645	brn hyg cry	519				s H_2O, EtOH
3000	Uranium(IV) chloride	UCl_4	10026-10-5	379.841	grn octahed cry	590	791	4.72		reac H_2O; s EtOH
3001	Uranium(IV) fluoride	UF_4	10049-14-6	314.023	grn monocl cry	1036	1417	6.7	0.01^{25}	s conc acid, alk
3002	Uranium(IV) iodide	UI_4	13470-22-9	745.647	blk hyg cry	506				s H_2O, EtOH
3003	Uranium(IV) oxide	UO_2	1344-57-6	270.028	brn cub cry	2847		10.97		i H_2O, dil acid; s conc acid
3004	Uranium(IV,V) oxide	U_4O_9	12037-15-9	1096.111	cub cry			11.2		
3005	Uranium(V) bromide	UBr_5	13775-16-1	637.549	brn hyg cry					reac H_2O
3006	Uranium(V) chloride	UCl_5	13470-21-8	415.294	brn hyg cry	287				reac H_2O
3007	Uranium(V) fluoride	UF_5	13775-07-0	333.021	pale blue tetr cry; hyg	348		5.81		s H_2O
3008	Uranium(V,VI) oxide	U_3O_8	1344-59-8	842.082	grn-blk orth cry	1300 dec		8.38		
3009	Uranium(VI) chloride	UCl_6	13763-23-0	450.747	grn hex cry	177		3.6		
3010	Uranium(VI) fluoride	UF_6	7783-81-5	352.019	wh monocl solid	64.06 tp	56.5 sp	5.09		reac H_2O; s ctc, chl
3011	Uranium(VI) oxide	UO_3	1344-58-7	286.027	oran-yel cry			≈7.3		i H_2O; s acid
3012	Uranium(VI) oxide monohydrate	$UO_3 \cdot H_2O$	12326-21-5	304.043	yel orth cry	570 dec		7.05		
3013	Uranium peroxide dihydrate	$UO_4 \cdot 2H_2O$	19525-15-6	338.057	yel hyg cry	115 dec				i H_2O
3014	Uranyl acetate dihydrate	$UO_2(C_2H_3O_2)_2 \cdot 2H_2O$	6159-44-0	424.146	ye cry (HOAc)	80 dec		2.89		sl EtOH
3015	Uranyl chloride	UO_2Cl_2	7791-26-6	340.934	yel orth cry; hyg	577				vs H_2O; s EtOH, ace; i bz
3016	Uranyl fluoride	UO_2F_2	13536-84-0	308.025	yel hyg solid				64.4^{20}	i bz
3017	Uranyl hydrogen phosphate tetrahydrate	$UO_2HPO_4 \cdot 4H_2O$	18433-48-2	438.068	yel cry pow					i H_2O; s acid
3018	Uranyl nitrate	$UO_2(NO_3)_2$	10102-06-4	394.037	yel cry				127^{25}	s eth
3019	Uranyl nitrate hexahydrate	$UO_2(NO_3)_2 \cdot 6H_2O$	13520-83-7	502.129	yel orth cry; hyg	60	118 dec	2.81	127^{25}	s EtOH, eth
3020	Uranyl sulfate	UO_2SO_4	1314-64-3	366.090	yel cry					
3021	Uranyl sulfate trihydrate	$UO_2SO_4 \cdot 3H_2O$	20910-28-5	420.137	yel cry			3.28	152^{16}	sl EtOH
3022	Vanadium	V	7440-62-2	50.942	gray-wh metal; cub	1910	3407	6.0		i H_2O; s acid

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
3023	Vanadocene	$V(C_5H_5)_2$	1277-47-0	181.128	viol cry; hyg	167				s bz, thf
3024	Vanadocene dichloride	$V(C_5H_5)_2Cl_2$	12083-48-6	252.034	dark grn cry	205 dec				s H_2O, chl, EtOH
3025	Vanadium boride (VB)	VB	12045-27-1	61.753	refrac solid	2250				i H_2O
3026	Vanadium boride (VB_2)	VB_2	12007-37-3	72.564	refrac solid	2450				
3027	Vanadium carbide (VC)	VC	12070-10-9	62.953	refrac blk cry; cub	2810		5.77		i H_2O
3028	Vanadium carbide (V_2C)	V_2C	12012-17-8	113.894	hex cry	2167				
3029	Vanadium carbonyl	$V(CO)_6$	14024-00-1	219.002	blue-grn cry; flam	60 dec	subl			
3030	Vanadium nitride	VN	24646-85-3	64.949	blk powder; cub	2050		6.13		i H_2O; s aqua regia
3031	Vanadium silicide (VSi_2)	VSi_2	12039-87-1	107.113	metallic prisms			4.42		s HF
3032	Vanadium silicide (V_3Si)	V_3Si	12039-76-8	180.911	cub cry	1935		5.70		
3033	Vanadium(II) bromide	VBr_2	14890-41-6	210.750	oran-brn hex cry		800 subl	4.58		reac H_2O
3034	Vanadium(II) chloride	VCl_2	10580-52-6	121.848	grn hex plates	1350	910 subl	3.23		reac H_2O; s EtOH, eth
3035	Vanadium(II) fluoride	VF_2	13842-80-3	88.939	blue hyg cry	1490				reac H_2O
3036	Vanadium(II) iodide	VI_2	15513-84-5	304.751	red-viol hex cry		subl 800	5.44		reac H_2O
3037	Vanadium(II) oxide	VO	12035-98-2	66.941	gray-blk cry	1790		5.758		s acid
3038	Vanadium(II) sulfate heptahydrate	$VSO_4 \cdot 7H_2O$	36907-42-3	273.111	viol cry					
3039	Vanadium(III) bromide	VBr_3	13470-26-3	290.654	blk-grn hyg cry	dec 500	subl	4.00		reac H_2O
3040	Vanadium(III) chloride	VCl_3	7718-98-1	157.301	red-viol hex cry; hyg	500 dec		3.00		reac H_2O; s EtOH, eth
3041	Vanadium(III) fluoride	VF_3	10049-12-4	107.937	yel-grn hex cry	1395	subl	3.363		i H_2O, EtOH
3042	Vanadium(III) fluoride trihydrate	$VF_3 \cdot 3H_2O$	10049-12-4*	161.983	grn rhomb cry	≈100 dec				sl H_2O
3043	Vanadium(III) iodide	VI_3	15513-94-7	431.655	brn-blk rhomb cry; hyg	dec 300		5.21		reac H_2O
3044	Vanadium(III) oxide	V_2O_3	1314-34-7	149.881	blk powder	1957	≈3000	4.87		i H_2O
3045	Vanadium(III) 2,4-pentanedioate	$V(CH_3COCHCOCH_3)_3$	13476-99-8	348.266	brn cry	≈185	subl	≈1.0		s MeOH, ace, bz chl
3046	Vanadium(III) sulfate	$V_2(SO_4)_3$	13701-70-7	390.071	yel powder	≈400 dec				sl H_2O
3047	Vanadium(III) sulfide	V_2S_3	1315-03-3	198.078	grn-blk powder	dec		4.7		i H_2O; s hot HCl
3048	Vanadium(IV) bromide	VBr_4	13595-30-7	370.558	unstab purp cry	-23 dec				
3049	Vanadium(IV) chloride	VCl_4	7632-51-1	192.754	red-brn liq	-28	151	1.816		reac H_2O; s EtOH, eth
3050	Vanadium(IV) fluoride	VF_4	10049-16-8	126.936	grn hyg powder	325 dec	subl	3.15		vs H_2O
3051	Vanadium(IV) oxide	VO_2	12036-21-4	82.941	blue-blk powder	1967		4.339		i H_2O; s acid, alk
3052	Vanadium(V) fluoride	VF_5	7783-72-4	145.934	col liq	19.5	48.3	2.50		reac H_2O
3053	Vanadium(V) dioxide fluoride	VO_2F	14259-82-6	101.939	brn hyg cry	350 dec				reac H_2O
3054	Vanadium(V) dioxide chloride	VO_2Cl	13759-30-3	118.394	oran hyg cry	dec 180				s thf
3055	Vanadium(V) oxide	V_2O_5	1314-62-1	181.880	yel-brn orth cry	681	1750	3.35	0.07^{25}	s conc acid, alk; i EtOH
3056	Vanadium(V) sulfide	V_2S_5	12138-17-9	262.208	grn-blk pow	dec		3.0		i H_2O; s acid, alk
3057	Vanadyl bromide	VOBr	13520-88-2	146.845	viol cry	480 dec				
3058	Vanadyl chloride	VOCl	13520-87-1	102.394	brn orth cry		127	1.72		
3059	Vanadyl dibromide	$VOBr_2$	13520-89-3	226.749	yel-brn cry	180 dec				
3060	Vanadyl dichloride	$VOCl_2$	10213-09-9	137.847	grn hyg cry	380 dec		2.88		reac H_2O; s EtOH
3061	Vanadyl difluoride	VOF_2	13814-83-0	104.938	yel cry					
3062	Vanadyl selenite hydrate	$VOSeO_3 \cdot H_2O$	133578-89-9	211.92	grn tricl plates			3.506		
3063	Vanadyl sulfate dihydrate	$VOSO_4 \cdot 2H_2O$	27774-13-6	199.035	blue cry powder					s H_2O
3064	Vanadyl tribromide	$VOBr_3$	13520-90-6	306.653	deep red liq	-59	170			reac H_2O
3065	Vanadyl trichloride	$VOCl_3$	7727-18-6	173.300	fuming red-yel liq	-79	127	1.829		reac H_2O; s MeOH, eth, ace
3066	Vanadyl trifluoride	VOF_3	13709-31-4	123.936	yel hyg powder	300	480	2.459		reac H_2O
3067	Water	H_2O	7732-18-5	18.015	col liq	0.00	99.974	0.9970^{25}		vs EtOH, MeOH, ace
3068	Water-d_2	D_2O	7789-20-0	20.027	col liq	3.82	101.42	1.1044^{25}		
3069	Water-t_2	T_2O	14940-65-9	22.032	col liq	4.48	101.51	1.2138^{25}		
3070	Xenon	Xe	7440-63-3	131.293	col gas	-111.745 tp (81.6 kPa)	-108.09	5.366 g/L		sl H_2O
3071	Xenon trioxide	XeO_3	13776-58-4	179.291	col orth cry	exp ≈25		4.55		s H_2O
3072	Xenon tetroxide	XeO_4	12340-14-6	195.291	yel solid or col gas; exp	-35.9	≈0 dec			
3073	Xenon difluoride	XeF_2	13709-36-9	169.290	col tetr cry	129.03 tp	114.35 sp	4.32		sl H_2O
3074	Xenon tetrafluoride	XeF_4	13709-61-0	207.287	col monocl cry	117.10 tp	115.75 sp	4.04		reac H_2O
3075	Xenon hexafluoride	XeF_6	13693-09-9	245.283	col monocl cry	49.48	75.6	3.56		reac H_2O
3076	Xenon fluoride oxide	$XeOF_2$	13780-64-8	185.289	yel solid, stab <-25	exp ≈0				reac H_2O
3077	Xenon oxytetrafluoride	$XeOF_4$	13774-85-1	223.286	col liq	-46.2		3.17^0		reac H_2O
3078	Xenon dioxydifluoride	XeO_2F_2	13875-06-4	201.289	col orth cry	30.8 exp		4.10		
3079	Xenon difluoride trioxide	XeO_3F_2	15192-14-0	217.288	unstab at r.t.	-54.1	exp			
3080	Xenon pentafluoride hexafluoroarsenate	XeF_5AsF_6	20328-94-3	415.197	wh monocl cry	130.5		3.51		

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
3081	Xenon pentafluoride hexafluororuthenate	XeF_5RuF_6	39796-98-0	441.35	grn orth cry	152		3.79		
3082	Xenon fluoride hexafluoroantimonate	XeF_3SbF_6	39797-63-2	424.039	yel-grn monocl cry	≈110		3.92		
3083	Xenon fluoride hexafluoroarsenate	$Xe_2F_3AsF_6$	50432-32-1	508.494	yel-grn monocl cry	99		3.62		reac H_2O
3084	Xenon fluoride hexafluororuthenate	$XeFRuF_6$	22527-13-5	365.35	yel-grn monocl cry	110		3.78		
3085	Xenon fluoride undecafluoroantimonate	$XeFSb_2F_{11}$	15364-10-0	602.794	yel monocl cry	63		3.69		
3086	Xenon trifluoride undecafluoroantimonate	$XeF_3Sb_2F_{11}$	35718-37-7	640.791	yel-grn tricl cry	82		3.98		
3087	Ytterbium	Yb	7440-64-4	173.04	silv metal; cub	824	1196	6.90		s dil acid
3088	Ytterbium silicide	$YbSi_2$	12039-89-3	229.21	hex cry			7.54		
3089	Ytterbium(II) bromide	$YbBr_2$	25502-05-0	332.85	yel cry	673				reac H_2O
3090	Ytterbium(II) chloride	$YbCl_2$	13874-77-6	243.95	grn cry	721		5.27		reac H_2O
3091	Ytterbium(II) fluoride	YbF_2	15192-18-4	211.04	gray solid	1407				i H_2O
3092	Ytterbium(II) iodide	YbI_2	19357-86-9	426.85	blk cry	772				reac H_2O
3093	Ytterbium(III) acetate tetrahydrate	$Yb(C_2H_3O_2)_3 \cdot 4H_2O$	15280-58-7	422.23	hyg col cry	dec 70		2.09		vs H_2O
3094	Ytterbium(III) bromide	$YbBr_3$	13759-89-2	412.75	col cry	956 dec				s H_2O
3095	Ytterbium(III) chloride	$YbCl_3$	10361-91-8	279.40	wh hyg powder	854				s H_2O
3096	Ytterbium(III) chloride hexahydrate	$YbCl_3 \cdot 6H_2O$	19423-87-1	387.49	grn hyg cry	150 dec		2.57		vs H_2O
3097	Ytterbium(III) fluoride	YbF_3	13760-80-0	230.04	wh cry	1157		8.2		i H_2O
3098	Ytterbium(III) iodide	YbI_3	13813-44-0	553.75	yel cry	dec 700				s H_2O
3099	Ytterbium(III) nitrate	$Yb(NO_3)_3$	13768-67-7	359.06	col hyg solid				239^{25}	s EtOH
3100	Ytterbium(III) oxide	Yb_2O_3	1314-37-0	394.08	col cub cry	2355	4070	9.2		s dil acid
3101	Ytterbium(III) sulfate octahydrate	$Yb_2(SO_4)_3 \cdot 8H_2O$	10034-98-7	778.39	col cry			3.3	38.4^{20}	
3102	Yttrium	Y	7440-65-5	88.906	silv metal; hex	1522	3345	4.47		reac H_2O; s dil acid
3103	Yttrium aluminum oxide	$Y_3Al_5O_{12}$	12005-21-9	593.619	grn cub cry			≈4.5		
3104	Yttrium antimonide	YSb	12186-97-9	210.666	cub cry	2310		5.97		
3105	Yttrium arsenide	YAs	12255-48-0	163.828	cub cry			5.59		
3106	Yttrium boride	YB_6	12008-32-1	153.772	refrac solid	2600		3.72		
3107	Yttrium bromide	YBr_3	13469-98-2	328.618	col hyg cry	904			83.3^{30}	
3108	Yttrium carbide	YC_2	12071-35-1	112.927	refrac solid	≈2400		4.13		
3109	Yttrium carbonate trihydrate	$Y_2(CO_3)_3 \cdot 3H_2O$	5970-44-5	411.885	red-brn powder					i H_2O; s dil acid
3110	Yttrium chloride	YCl_3	10361-92-9	195.265	wh monocl cry; hyg	721	1482	2.61	75.1^{20}	vs H_2O
3111	Yttrium chloride hexahydrate	$YCl_3 \cdot 6H_2O$	10025-94-2	303.356	hyg col cry	dec 100				vs H_2O; s EtOH
3112	Yttrium fluoride	YF_3	13709-49-4	145.901	wh hyg powder	1155		4.0		i H_2O
3113	Yttrium hydroxide	$Y(OH)_3$	16469-22-0	139.928	wh prec or pow	dec 190				i H_2O
3114	Yttrium iodide	YI_3	13470-38-7	469.619	hyg wh-yel cry	997				s H_2O, ace, EtOH
3115	Yttrium iron oxide	$Y_3Fe_5O_{12}$	12063-56-8	737.936	cub cry	1555				
3116	Yttrium nitrate	$Y(NO_3)_3$	10361-93-0	274.921	wh hyg solid				149^{25}	s EtOH
3117	Yttrium nitrate tetrahydrate	$Y(NO_3)_3 \cdot 4H_2O$	13773-69-8	346.982	red-wh prisms			2.68	149^{25}	
3118	Yttrium nitrate hexahydrate	$Y(NO_3)_3 \cdot 6H_2O$	13494-98-9	383.012	hyg cry				149^{25}	
3119	Yttrium oxide	Y_2O_3	1314-36-9	225.810	wh cry; cub	2439		5.03		s dil acid
3120	Yttrium phosphide	YP	12294-01-8	119.880	cub cry			≈4.4		
3121	Yttrium sulfate octahydrate	$Y_2(SO_4)_3 \cdot 8H_2O$	7446-33-5	610.122	red monocl cry			2.6	7.47^{16}	
3122	Yttrium sulfide	Y_2S_3	12039-19-9	274.007	yel cub cry	1925		3.87		
3123	Zinc	Zn	7440-66-6	65.409	blue-wh metal; hex	419.53	907	7.134		s acid, alk
3124	Zinc acetate dihydrate	$Zn(C_2H_3O_2)_2 \cdot 2H_2O$	5970-45-6	219.527	wh powder	237 dec		1.735	30.0^{20}	s EtOH
3125	Zinc ammonium sulfate	$Zn(NH_4)_2(SO_4)_2$	7783-24-6	293.611	wh cry				9.2^{20}	
3126	Zinc antimonide	ZnSb	12039-35-9	187.169	silv-wh orth cry	565		6.33		reac H_2O
3127	Zinc arsenate	$Zn_3(AsO_4)_2$	13464-44-3	474.065	wh powder				0.000078^{20}	s acid, alk
3128	Zinc arsenate octahydrate	$Zn_3(AsO_4)_2 \cdot 8H_2O$	13464-45-4	618.187	wh monocl cry			3.33	0.000078^{20}	s acid, alk
3129	Zinc arsenide	Zn_3As_2	12006-40-5	346.070	powder	1015		5.528		
3130	Zinc arsenite	$Zn(AsO_2)_2$	10326-24-6	279.250	col powder					i H_2O; s acid
3131	Zinc borate	$3ZnO \cdot 2B_2O_3$	27043-84-1	383.466	wh amorp powder			3.64		sl H_2O; s dil acid
3132	Zinc borate hemiheptahydrate	$2ZnO \cdot 3B_2O_3 \cdot 3.5H_2O$	12513-27-8	434.69	wh cry	980		4.22		i H_2O
3133	Zinc borate pentahydrate	$2ZnO \cdot 3B_2O_3 \cdot 5H_2O$	12536-65-1	461.753	wh powder			3.64	0.007^{25}	sl HCl
3134	Zinc bromate hexahydrate	$Zn(BrO_3)_2 \cdot 6H_2O$	13517-27-6	429.305	wh hyg solid	100		2.57		vs H_2O
3135	Zinc bromide	$ZnBr_2$	7699-45-8	225.217	wh hex cry; hyg	402	≈670	4.5	488^{25}	vs EtOH; s eth
3136	Zinc caprylate	$Zn(C_8H_{15}O_2)_2$	557-09-5	351.816	wh hyg cry	136				sl H_2O
3137	Zinc carbonate	$ZnCO_3$	3486-35-9	125.418	wh hex cry	140 dec		4.434	0.000091^{20}	s dil acid, alk
3138	Zinc carbonate hydroxide	$3Zn(OH)_2 \cdot 2ZnCO_3$	12070-69-8	549.107	wh powder					
3139	Zinc chlorate	$Zn(ClO_3)_2$	10361-95-2	232.311	yel hyg cry	60 dec		2.15	200^{20}	
3140	Zinc chloride	$ZnCl_2$	7646-85-7	136.315	wh hyg cry	290	732	2.907	408^{25}	vs H_2O; s EtOH, ace

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
3141	Zinc chromate	$ZnCrO_4$	13530-65-9	181.403	yel prisms	316		3.40	3.08	s acid; i ace
3142	Zinc chromite	$ZnCr_2O_4$	12018-19-8	233.399	grn cub cry			5.29		
3143	Zinc citrate dihydrate	$Zn_3(C_6H_5O_7)_2 \cdot 2H_2O$	546-46-3	610.456	col powder					sl H_2O; s dil acid, alk
3144	Zinc cyanide	$Zn(CN)_2$	557-21-1	117.443	wh powder			1.852	0.00047^{20}	reac acid
3145	Zinc diethyl	$Zn(C_2H_5)_2$	557-20-0	123.531	col liq	-28	118	1.2065		reac H_2O; msc eth, peth, bz
3146	Zinc dithionate	ZnS_2O_4	7779-86-4	193.537	wh amorp solid	200 dec			40^{20}	
3147	Zinc fluoride	ZnF_2	7783-49-5	103.406	wh tetr needles; hyg	872	1500	4.9	1.55^{25}	sl H_2O
3148	Zinc fluoride tetrahydrate	$ZnF_2 \cdot 4H_2O$	13986-18-0	175.468	wh orth cry			2.30	1.55^{25}	
3149	Zinc fluoroborate hexahydrate	$Zn(BF_4)_2 \cdot 6H_2O$	27860-83-9	347.109	hex cry			2.12		vs H_2O; s EtOH
3150	Zinc formate dihydrate	$Zn(CHO_2)_2 \cdot 2H_2O$	5970-62-7	191.474	wh cry			2.207	5.2^{20}	i EtOH
3151	Zinc hexafluorosilicate hexahydrate	$ZnSiF_6 \cdot 6H_2O$	16871-71-9	315.576	wh cry					s H_2O
3152	Zinc hydroxide	$Zn(OH)_2$	20427-58-1	99.424	col orth cry	125 dec		3.05	0.000042^{20}	
3153	Zinc iodate	$Zn(IO_3)_2$	7790-37-6	415.214	wh cry powder				0.64^{25}	
3154	Zinc iodide	ZnI_2	10139-47-6	319.218	wh-yel hyg cry	450	625	4.74	438^{25}	vs H_2O; s EtOH, eth
3155	Zinc laurate	$Zn(C_{12}H_{23}O_2)_2$	2452-01-9	464.029	wh powder	128				sl H_2O
3156	Zinc molybdate	$ZnMoO_4$	13767-32-3	225.35	wh tetr cry	>700		4.3		i H_2O
3157	Zinc nitrate	$Zn(NO_3)_2$	7779-88-6	189.418	wh powder				120^{25}	
3158	Zinc nitrate hexahydrate	$Zn(NO_3)_2 \cdot 6H_2O$	10196-18-6	297.510	col orth cry	36 dec		2.067	120^{25}	vs EtOH
3159	Zinc nitride	Zn_3N_2	1313-49-1	224.240	blue-gray cub cry	700 dec		6.22		i H_2O
3160	Zinc nitrite	$Zn(NO_2)_2$	10102-02-0	157.420	hyg solid					reac H_2O
3161	Zinc oleate	$Zn(C_{18}H_{33}O_2)_2$	557-07-3	628.316	wh powder	70 dec				i H_2O; s EtOH, eth, bz
3162	Zinc oxalate	ZnC_2O_4	547-68-2	153.428	wh pwd				0.0026^{25}	
3163	Zinc oxalate dihydrate	$ZnC_2O_4 \cdot 2H_2O$	4255-07-6	189.458	wh powder	100 dec		2.56	0.0026^{25}	s dil acid
3164	Zinc tartrate dihydrate	$ZnC_4H_4O_6 \cdot 2H_2O$	22570-08-7	249.511	wh cry pow	dec 150			0.022^{20}	
3165	Zinc oxide	ZnO	1314-13-2	81.408	wh powder; hex	1974		5.6		i H_2O; s dil acid
3166	Zinc 2,4-pentanedioate	$Zn(CH_3COCHCOCH_3)_2$	14024-63-6	263.625	cry	137 dec				sl H_2O; s EtOH, DMSO
3167	Zinc pentanoate dihydrate	$Zn(C_5H_9O_2)_2 \cdot 2H_2O$	556-38-7	303.687	scales or powder					sl H_2O; reac acid; s EtOH
3168	Zinc perchlorate hexahydrate	$Zn(ClO_4)_2 \cdot 6H_2O$	10025-64-6	372.402	wh cub cry; hyg	106 dec		2.2	121.3^{25}	s EtOH
3169	Zinc permanganate hexahydrate	$Zn(MnO_4)_2 \cdot 6H_2O$	23414-72-4	411.372	blk orth cry; hyg			2.45		s H_2O; reac EtOH
3170	Zinc peroxide	ZnO_2	1314-22-3	97.408	yel-wh powder	>150 dec	212 exp	1.57		i H_2O; reac acid, EtOH, ace
3171	Zinc phosphate	$Zn_3(PO_4)_2$	7779-90-0	386.170	wh monocl cry	900		4.0		i H_2O
3172	Zinc phosphate tetrahydrate	$Zn_3(PO_4)_2 \cdot 4H_2O$	7543-51-3	458.231	col orth cry			3.04		i H_2O, EtOH; s dil acid, alk
3173	Zinc phosphide	Zn_3P_2	1314-84-7	258.175	gray tetr cry	1160		4.55		i H_2O, EtOH; reac acid; s bz
3174	Zinc pyrophosphate	$Zn_2P_2O_7$	7446-26-6	304.761	wh cry powder			3.75		i H_2O; s dil acid
3175	Zinc selenate pentahydrate	$ZnSeO_4 \cdot 5H_2O$	13597-54-1	298.44	tricl cry	50 dec		2.59	63.4^{25}	
3176	Zinc selenide	ZnSe	1315-09-9	144.37	yel-red cub cry	>1100	subl	5.65		i H_2O; s dil acid
3177	Zinc orthosilicate	Zn_2SiO_4	13597-65-4	222.902	wh hex cry	1509		4.1		i H_2O, dil acid
3178	Zinc selenite	$ZnSeO_3$	13597-46-1	192.37	wh powder	621				
3179	Zinc stearate	$Zn(C_{18}H_{35}O_2)_2$	557-05-1	632.348	wh powder	130		1.095		i H_2O, EtOH, eth; s bz
3180	Zinc sulfate	$ZnSO_4$	7733-02-0	161.472	col orth cry	680 dec		3.8	57.7^{25}	
3181	Zinc sulfate monohydrate	$ZnSO_4 \cdot H_2O$	7446-19-7	179.487	wh monocl cry	238 dec		3.20	57.7^{25}	i EtOH
3182	Zinc sulfate heptahydrate	$ZnSO_4 \cdot 7H_2O$	7446-20-0	287.578	col orth cry	100 dec		1.97	57.7^{25}	i EtOH
3183	Zinc sulfide (sphalerite)	ZnS	1314-98-3	97.474	gray-wh cub cry	trans wurtzite 1020		4.04		i H_2O, EtOH; s dil acid
3184	Zinc sulfide (wurtzite)	ZnS	1314-98-3	97.474	wh hex cry	1700	subl	4.09		i H_2O; s dil acid
3185	Zinc sulfite dihydrate	$ZnSO_3 \cdot 2H_2O$	7488-52-0	181.503	wh powder	200 dec			0.224^{25}	i EtOH
3186	Zinc telluride	ZnTe	1315-11-3	193.01	red cub cry	1239		5.9		i H_2O
3187	Zinc thiocyanate	$Zn(SCN)_2$	557-42-6	181.573	wh hyg cry					sl H_2O; s EtOH
3188	Zirconium	Zr	7440-67-7	91.224	gray-wh metal; hex	1854.7	4409	6.52		s hot conc acid
3189	Zirconocene dichloride	$Zr(C_5H_5)_2Cl_2$	1291-32-3	292.316	col cry	248	subl 150			
3190	Zirconium boride	ZrB_2	12045-64-6	112.846	gray refrac solid; hex	3050		6.17		
3191	Zirconium carbide	ZrC	12020-14-3	103.235	gray refrac solid; cub	3532		6.73		s HF
3192	Zirconium nitride	ZrN	25658-42-8	105.231	yel cub cry	2952		7.09		s conc HF; sl dil acid
3193	Zirconium phosphide	ZrP_2	12037-80-8	153.172	orth cry			≈5.1		
3194	Zirconium silicide	$ZrSi_2$	12039-90-6	147.395	gray powder	1620		4.88		i H_2O, aqua regia; s HF

No.	Name	Formula	CAS Reg No.	Mol. weight	Physical form	mp/°C	bp/°C	Density g cm^{-3}	Solubility g/100 g H_2O	Qualitative solubility
3195	Zirconium(II) bromide	$ZrBr_2$	24621-17-8	251.032	blue-blk cry	dec 400				
3196	Zirconium(II) chloride	$ZrCl_2$	13762-26-0	162.130	blk cry	772		3.16		reac H_2O
3197	Zirconium(II) fluoride	ZrF_2	13842-94-9	129.221	blk cry	902				
3198	Zirconium(II) hydride	ZrH_2	7704-99-6	93.240	gray tetr cry	800 dec		5.6		i H_2O
3199	Zirconium(II) iodide	ZrI_2	15513-85-6	345.033	blk cry	827				
3200	Zirconium(III) bromide	$ZrBr_3$	24621-18-9	330.936	dark blue cry	dec 300				
3201	Zirconium(III) chloride	$ZrCl_3$	10241-03-9	197.583	dark blue cry	627		3.05		reac H_2O
3202	Zirconium(III) fluoride	ZrF_3	13814-22-7	148.219	blue-grn cry	927		4.26		i H_2O; s acid
3203	Zirconium(III) iodide	ZrI_3	13779-87-8	471.937	dark blue cry	727				
3204	Zirconium(IV) acetate hydroxide	$Zr(C_2H_3O_2)_2(OH)_2$	14311-93-4	243.327	wh amorp solid					s H_2O
3205	Zirconium(IV) ammonium carbonate dihydrate	$Zr(NH_4)_3OH(CO_3)_3 \cdot 2H_2O$	12616-24-9*	362.404	prisms; unstab					s H_2O
3206	Zirconium(IV) bromide	$ZrBr_4$	13777-25-8	410.840	wh cub cry	450 tp	360 sp	3.98		reac H_2O
3207	Zirconium(IV) chloride	$ZrCl_4$	10026-11-6	233.036	wh monocl cry; hyg	437 tp	331 sp	2.80		reac H_2O; s EtOH, eth
3208	Zirconium(IV) fluoride	ZrF_4	7783-64-4	167.218	wh monocl cry	910	912 sp	4.43	1.5^{25}	
3209	Zirconium(IV) hydroxide	$Zr(OH)_4$	14475-63-9	159.254	wh amorp powder	dec		3.25		i H_2O; s acid
3210	Zirconium(IV) iodide	ZrI_4	13986-26-0	598.842	yel-oran cub cry	500	431 sp	4.85		vs H_2O
3211	Zirconium(IV) nitrate pentahydrate	$Zr(NO_3)_4 \cdot 5H_2O$	13746-89-9	429.320	wh hyg cry	100 dec				vs H_2O; s EtOH
3212	Zirconium(IV) orthosilicate	$ZrSiO_4$	10101-52-7	183.308	wh tetr cry	1540 dec		4.6		i H_2O, acid
3213	Zirconium(IV) oxide	ZrO_2	1314-23-4	123.223	wh amorp powder	2710	4300	5.68		i H_2O; sl acid
3214	Zirconium(IV) pyrophosphate	ZrP_2O_7	13565-97-4	265.167	wh refrac solid	dec 1550				i H_2O, dil acid; s HF
3215	Zirconium(IV) sulfate	$Zr(SO_4)_2$	14644-61-2	283.349	wh hyg cry	410 dec		3.22		s H_2O; sl EtOH
3216	Zirconium(IV) sulfate tetrahydrate	$Zr(SO_4)_2 \cdot 4H_2O$	7446-31-3	355.411	wh tetr cry	100 dec		2.80		vs H_2O
3217	Zirconium(IV) sulfide	ZrS_2	12039-15-5	155.354	red-brn hex cry	1550		3.87		i H_2O
3218	Zirconium(IV) tungstate	$Zr(WO_4)_2$	16853-74-0	586.90	grn pow					
3219	Zirconyl chloride	$ZrOCl_2$	7699-43-6	178.129	wh solid	250 dec				s H_2O, EtOH
3220	Zirconyl chloride octahydrate	$ZrOCl_2 \cdot 8H_2O$	13520-92-8	322.252	tetr cry	400 dec		1.91		vs H_2O, EtOH