

## PROPERTIES OF CRYOGENIC FLUIDS

This table gives physical and thermodynamic properties of eight cryogenic fluids. The properties are:

$M$	Molar mass in grams per mole
$T_t$	Triple point temperature in kelvins
$P_t$	Triple point pressure in kilopascals
$\rho_t(l)$	Liquid density at the triple point in grams per milliliter
$\Delta_{\text{fus}}H @ T_t$	Enthalpy of fusion at the triple point in joules per gram
$T_b$	Normal boiling point in kelvins at a pressure of 101325 pascals (760 mmHg)
$\Delta_{\text{vap}}H @ T_b$	Enthalpy of vaporization at the normal boiling point in joules per gram
$\rho(l) @ T_b$	Liquid density at the normal boiling point in grams per milliliter
$\rho(g) @ T_b$	Vapor density at the normal boiling point in grams per liter
$C_p(l) @ T_b$	Liquid heat capacity at constant pressure at the normal boiling point in joules per gram kelvin
$C_p(g) @ T_b$	Vapor heat capacity at constant pressure at the normal boiling point in joules per gram kelvin
$T_c$	Critical temperature in kelvins
$P_c$	Critical pressure in megapascals
$\rho_c$	Critical density in grams per milliliter

In the case of air, the value given for the triple point temperature is the incipient solidification temperature, and the normal boiling point value is the incipient boiling (bubble) point. See Reference 3 for more details.

### References

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Property	Units	Air	$N_2$	$O_2$	$H_2$	He	Ne	Ar	Kr	Xe	$CH_4$
$M$	g/mol	28.96	28.014	31.999	2.0159	4.0026	20.180	39.948	83.800	131.290	16.043
$T_t$	K	59.75	63.15	54.3584	13.8		24.5561	83.8058	115.8	161.4	90.694
$P_t$	kPa		12.463	0.14633	7.042		50	68.95	72.92	81.59	11.696
$\rho_t(l)$	g/mL	0.959	0.870	1.306	0.0770		1.251	1.417	2.449	2.978	0.4515
$\Delta_{\text{fus}}H @ T_t$	J/g		25.3	13.7	59.5		16.8	28.0	16.3	13.8	58.41
$T_b$	K	78.67	77.35	90.188	20.28	4.2221	27.07	87.293	119.92	165.10	111.668
$\Delta_{\text{vap}}H @ T_b$	J/g	198.7	198.8	213.1	445	20.7	84.8	161.0	108.4	96.1	510.83
$\rho(l) @ T_b$	g/mL	0.8754	0.807	1.141	0.0708	0.124901	1.204	1.396	2.418	2.953	0.4224
$\rho(g) @ T_b$	g/L	3.199	4.622	4.467	1.3390	16.89	9.51	5.79	8.94		1.816
$C_p(l) @ T_b$	J/g K	1.865	2.042	1.699	9.668	4.545	1.877	1.078	0.533	0.340	3.481
$C_p(g) @ T_b$	J/g K		1.341	0.980	12.24	9.78		0.570	0.248	0.158	2.218
$T_c$	K	132.5	126.20	154.581	32.98	5.1953	44.40	150.663	209.40	289.73	190.56
$P_c$	MPa	3.766	3.390	5.043	1.293	0.227460	2.760	4.860	5.500	5.840	4.592
$\rho_c$	g/mL	0.316	0.313	0.436	0.031	0.06964	0.484	0.531	0.919	1.110	0.1627