

DIFFUSION IN GASES

This table gives binary diffusion coefficients D_{12} for a number of common gases as a function of temperature. Values refer to atmospheric pressure. The diffusion coefficient is inversely proportional to pressure as long as the gas is in a regime where binary collisions dominate. See Reference 1 for a discussion of the dependence of D_{12} on temperature and composition.

The first part of the table gives data for several gases in the presence of a large excess of air. The remainder applies to equimolar

mixtures of gases. Each gas pair is ordered alphabetically according to the most common way of writing the formula. The listing of pairs then follows alphabetical order by the first constituent.

References

1. Marrero, T. R., and Mason, E. A., *J. Phys. Chem. Ref. Data*, 1, 1, 1972.
2. Kestin, J., et al., *J. Phys. Chem. Ref. Data*, 13, 229, 1984.

$D_{12}/\text{cm}^2 \text{ s}^{-1}$ for $p = 101.325 \text{ kPa}$ and the Specified T/K

System	200	273.15	293.15	373.15	473.15	573.15	673.15
<i>Large Excess of Air</i>							
Ar-air		0.167	0.189	0.289	0.437	0.612	0.810
CH ₄ -air			0.210	0.321	0.485	0.678	0.899
CO-air			0.208	0.315	0.475	0.662	0.875
CO ₂ -air			0.160	0.252	0.390	0.549	0.728
H ₂ -air		0.668	0.756	1.153	1.747	2.444	3.238
H ₂ O-air			0.242	0.399	0.638	0.873	1.135
He-air		0.617	0.697	1.057	1.594	2.221	2.933
SF ₆ -air				0.150	0.233	0.329	0.438
<i>Equimolar Mixture</i>							
Ar-CH ₄				0.306	0.467	0.657	0.876
Ar-CO		0.168	0.190	0.290	0.439	0.615	0.815
Ar-CO ₂		0.129	0.148	0.235	0.365	0.517	0.689
Ar-H ₂		0.698	0.794	1.228	1.876	2.634	3.496
Ar-He	0.381	0.645	0.726	1.088	1.617	2.226	2.911
Ar-Kr	0.064	0.117	0.134	0.210	0.323	0.456	0.605
Ar-N ₂		0.168	0.190	0.290	0.439	0.615	0.815
Ar-Ne	0.160	0.277	0.313	0.475	0.710	0.979	1.283
Ar-O ₂		0.166	0.187	0.285	0.430	0.600	0.793
Ar-SF ₆				0.128	0.202	0.290	0.389
Ar-Xe	0.052	0.095	0.108	0.171	0.264	0.374	0.498
CH ₄ -H ₂			0.708	1.084	1.648	2.311	3.070
CH ₄ -He			0.650	0.992	1.502	2.101	2.784
CH ₄ -N ₂			0.208	0.317	0.480	0.671	0.890
CH ₄ -O ₂			0.220	0.341	0.523	0.736	0.978
CH ₄ -SF ₆				0.167	0.257	0.363	0.482
CO-CO ₂			0.162	0.250	0.384		
CO-H ₂	0.408	0.686	0.772	1.162	1.743	2.423	3.196
CO-He	0.365	0.619	0.698	1.052	1.577	2.188	2.882
CO-Kr		0.131	0.149	0.227	0.346	0.485	0.645
CO-N ₂	0.133	0.208	0.231	0.336	0.491	0.673	0.878
CO-O ₂			0.202	0.307	0.462	0.643	0.849
CO-SF ₆				0.144	0.226	0.323	0.432
CO ₂ -C ₃ H ₈			0.084	0.133	0.209		
CO ₂ -H ₂	0.315	0.552	0.627	0.964	1.470	2.066	2.745
CO ₂ -H ₂ O			0.162	0.292	0.496	0.741	1.021
CO ₂ -He	0.300	0.513	0.580	0.878	1.321		
CO ₂ -N ₂			0.160	0.253	0.392	0.553	0.733
CO ₂ -N ₂ O	0.055	0.099	0.113	0.177	0.276		
CO ₂ -Ne	0.131	0.227	0.258	0.395	0.603	0.847	
CO ₂ -O ₂			0.159	0.248	0.380	0.535	0.710
CO ₂ -SF ₆				0.099	0.155		
D ₂ -H ₂	0.631	1.079	1.219	1.846	2.778	3.866	5.103
H ₂ -He	0.775	1.320	1.490	2.255	3.394	4.726	6.242
H ₂ -Kr	0.340	0.601	0.682	1.053	1.607	2.258	2.999

System	200	273.15	293.15	373.15	473.15	573.15	673.15
H ₂ -N ₂	0.408	0.686	0.772	1.162	1.743	2.423	3.196
H ₂ -Ne	0.572	0.982	1.109	1.684	2.541	3.541	4.677
H ₂ -O ₂		0.692	0.782	1.188	1.792	2.497	3.299
H ₂ -SF ₆			0.412	0.649	0.998	1.400	1.851
H ₂ -Xe		0.513	0.581	0.890	1.349	1.885	2.493
H ₂ O-N ₂			0.242	0.399			
H ₂ O-O ₂			0.244	0.403	0.645	0.882	1.147
He-Kr	0.330	0.559	0.629	0.942	1.404	1.942	2.550
He-N ₂	0.365	0.619	0.698	1.052	1.577	2.188	2.882
He-Ne	0.563	0.948	1.066	1.592	2.362	3.254	4.262
He-O ₂		0.641	0.723	1.092	1.640	2.276	2.996
He-SF ₆			0.400	0.592	0.871	1.190	1.545
He-Xe	0.282	0.478	0.538	0.807	1.201	1.655	2.168
Kr-N ₂		0.131	0.149	0.227	0.346	0.485	0.645
Kr-Ne	0.131	0.228	0.258	0.392	0.587	0.812	1.063
Kr-Xe	0.035	0.064	0.073	0.116	0.181	0.257	0.344
N ₂ -Ne			0.317	0.483	0.731	1.021	1.351
N ₂ -O ₂			0.202	0.307	0.462	0.643	0.849
N ₂ -SF ₆				0.148	0.231	0.328	0.436
N ₂ -Xe		0.107	0.122	0.188	0.287	0.404	0.539
Ne-Xe	0.111	0.193	0.219	0.332	0.498	0.688	0.901
O ₂ -SF ₆			0.097	0.154	0.238	0.334	0.441