

# STANDARD TRANSFORMED GIBBS ENERGIES OF FORMATION FOR BIOCHEMICAL REACTANTS

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This table contains values of the standard transformed Gibbs energies of formation  $\Delta_f G'^{\circ}$  for 130 biochemical reactants. Values of  $\Delta_f G'^{\circ}$  are given at pH 7.0, the temperature 298.15 K, and the pressure 100 kPa for three ionic strengths:  $I = 0$ ,  $I = 0.1$  mol/L and  $I = 0.25$  mol/L. The table can be used for calculating apparent equilibrium constants  $K'$  and standard apparent reduction potentials  $E'^{\circ}$  for biochemical reactions. Such a listing is more compact than tabulating the actual apparent equilibrium constants or standard apparent reduction potentials, which would require a very large number of reactant-product combinations. In the table, all reactants are in aqueous solution unless indicated otherwise.

A biochemical reactant is a sum of species. For example, ATP consists of an equilibrium mixture of the aqueous species  $\text{ATP}^4$ ,  $\text{HATP}^3$ ,  $\text{H}_2\text{ATP}^2$ ,  $\text{MgATP}^2$ , etc. Similarly, phosphate refers to the equilibrium mixture of the aqueous species  $\text{PO}_4^{3-}$ ,  $\text{HPO}_4^{2-}$ ,  $\text{H}_2\text{PO}_4^-$ ,  $\text{H}_3\text{PO}_4$ ,  $\text{MgHPO}_4$ , etc. Biochemical reactions are written using biochemical reactants in terms of an apparent equilibrium constant  $K'$ , which is distinct from the standard equilibrium constant  $K$ . This subject is discussed in an IUPAC report (see Reference 1 below).

The apparent equilibrium constant  $K'$  and the standard transformed Gibbs energy change  $\Delta_f G'^{\circ}$  for a biochemical reaction can be calculated from the  $\Delta_f G'^{\circ}$  values by using the relationship

$$-RT \ln K' = \Delta_f G'^{\circ} = \sum \nu_i \Delta_f G'^{\circ}_i,$$

where the summation is over all of the biochemical reactants. The quantity  $\nu_i$  is the stoichiometric number of reactant  $i$  ( $\nu_i$  is positive for reactants on the right side of the equation and negative for reactants on the left side);  $R$  is the gas constant. As an example, the hydrolysis reaction of ATP is

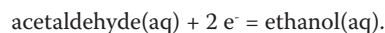


At pH 7.00 and  $I = 0.25$  M,  $\Delta_f G'^{\circ}$  and  $K'$  are calculated as follows:

$$\Delta_f G'^{\circ} = \{-1424.70 - 1059.49 - (-2292.50 - 155.66)\} \cdot (\text{kJ mol}^{-1}) = -36.03 \text{ kJ mol}^{-1}$$

$$K' = \exp[-(-36030 \text{ J mol}^{-1}) / \{(8.3145 \text{ J mol}^{-1} \text{ K}^{-1}) \cdot (298.15 \text{ K})\}] = 2.05 \cdot 10^6$$

An example involving a biochemical half-cell reaction is



At 298.15 K, pH 7.00, and  $I = 0$ , the standard apparent reduction potential  $E'^{\circ}$  can be calculated as follows

$$E'^{\circ} = -(1/nF) \cdot \{\Delta_f G'^{\circ}(\text{ethanol}) - \Delta_f G'^{\circ}(\text{acetaldehyde})\},$$

where  $n$  is the number of electrons in the half-cell reaction and  $F$  is the Faraday constant. Then,

$$E'^{\circ} = [-1 / (2 \cdot 9.6485 \cdot 10^4 \text{ C mol}^{-1})] \cdot (58.10 \cdot 10^3 \text{ J mol}^{-1} - 20.83 \cdot 10^3 \text{ J mol}^{-1}) = -0.193 \text{ V}$$

## References

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- Alberty, R.A., *BasicBiochemData2: Data and Programs for Biochemical Thermodynamics*, <<http://library.wolfram.com/infocenter/MathSource/797>>.

Reactant	$\Delta_f G'^{\circ}(I = 0)$ kJ mol <sup>-1</sup>	$\Delta_f G'^{\circ}(I = 0.1 \text{ M})$ kJ mol <sup>-1</sup>	$\Delta_f G'^{\circ}(I = 0.25 \text{ M})$ kJ mol <sup>-1</sup>
Acetaldehyde	20.83	23.27	24.06
Acetate	-249.46	-248.23	-247.83
Acetone	80.04	83.71	84.90
Acetyl Coenzyme A	-60.49	-58.65	-58.06
Acetylphosphate	-1109.34	-1107.57	-1107.02
cis-Aconitate	-797.26	-800.93	-802.12
Adenine	510.45	513.51	514.50
Adenosine	324.93	332.89	335.46
Adenosine 5'-diphosphate (ADP)	-1428.93	-1425.55	-1424.70
Adenosine 5'-monophosphate (AMP)	-562.04	-556.53	-554.83
Adenosine 5'-triphosphate (ATP)	-2292.61	-2292.16	-2292.50
D-Alanine	-91.31	-87.02	-85.64
Ammonia	80.50	82.34	82.93
D-Arabinose	-342.67	-336.55	-334.57
L-Asparagine	-206.28	-201.38	-199.80
L-Aspartate	-456.14	-453.08	-452.09
1,3-Biphosphoglycerate	-2202.06	-2205.69	-2207.30
Butanoate	-72.94	-69.26	-68.08
1-Butanol	227.72	233.84	235.82

Reactant	$\Delta_f G^{\circ}(I = 0)$ kJ mol <sup>-1</sup>	$\Delta_f G^{\circ\prime}(I = 0.1 \text{ M})$ kJ mol <sup>-1</sup>	$\Delta_f G^{\circ\prime\prime}(I = 0.25 \text{ M})$ kJ mol <sup>-1</sup>
Citrate	-963.46	-965.49	-966.23
Isocitrate	-956.82	-958.84	-959.58
Coenzyme A (CoA)	-7.98	-7.43	-7.26
CO(aq)	-119.90	-119.90	-119.90
CO(g)	-137.17	-137.17	-137.17
CO <sub>2</sub> (aq)[total]	-547.33	-547.15	-547.10
CO <sub>2</sub> (g)	-394.36	-394.36	-394.36
Creatine	100.41	105.92	107.69
Creatinine	256.55	260.84	262.22
<i>L</i> -Cysteine	-59.23	-55.01	-53.65
<i>L</i> -Cystine	-187.03	-179.69	-177.32
Cytochrome c [oxidized]	0.00	-5.51	-7.29
Cytochrome c [reduced]	-24.51	-26.96	-27.75
Dihydroxyacetone phosphate	-1096.60	-1095.91	-1095.70
Ethanol	58.10	61.77	62.96
Ethyl acetate	-18.00	-13.10	-11.52
Ferredoxin [oxidized]	0.00	-0.61	-0.81
Ferredoxin [reduced]	38.07	38.07	38.07
Flavine adenine dinucleotide (FAD) [oxidized]	1238.65	1255.17	1260.51
Flavine adenine dinucleotide (FAD) [reduced]	1279.68	1297.43	1303.16
Flavin adenine dinucleotide-enzyme (FADenz) [oxidized]	1238.65	1255.17	1260.51
Flavin adenine dinucleotide-enzyme (FADenz) [reduced]	1229.96	1247.71	1253.44
Flavin mononucleotide (FMN) [oxidized]	759.17	768.35	771.32
Flavin mononucleotide (FMN) [reduced]	800.20	810.61	813.97
Formate	-311.04	-311.04	-311.04
<i>D</i> -Fructose	-436.03	-428.69	-426.32
<i>D</i> -Fructose 1,6-diphosphate	-2202.84	-2205.66	-2206.78
<i>D</i> -Fructose 6-phosphate	-1321.71	-1317.16	-1315.74
Fumarate	-521.97	-523.19	-523.58
<i>D</i> -Galactose	-429.45	-422.11	-419.74
$\alpha$ - <i>D</i> -Galactose 1-phosphate	-1317.50	-1313.01	-1311.60
<i>D</i> -Glucose	-436.42	-429.08	-426.71
$\alpha$ - <i>D</i> -Glucose 1-phosphate	-1318.03	-1313.34	-1311.89
<i>D</i> -Glucose 6-phosphate	-1325.00	-1320.37	-1318.92
Glutamate	-377.82	-373.54	-372.16
<i>D</i> -Glutamine	-128.46	-122.34	-120.36
Glutathione [oxidized]	1198.69	1214.60	1219.74
Glutathione [reduced]	625.75	634.76	637.62
Glutathione-coenzyme A	563.49	572.06	574.83
<i>D</i> -Glyceraldehyde 3-phosphate	-1088.94	-1088.25	-1088.04
Glycerol	-177.83	-172.93	-171.35
<i>sn</i> -Glycerol 3-phosphate	-1080.22	-1077.83	-1077.13
Glycine	-180.13	-177.07	-176.08
Glycolate	-411.08	-409.86	-409.46
Glycylglycine	-200.55	-195.65	-194.07
Glyoxylate	-428.64	-428.64	-428.64
H <sub>2</sub> (aq)	97.51	98.74	99.13
H <sub>2</sub> (g)	79.91	81.14	81.53
H <sub>2</sub> O(l)	-157.28	-156.05	-155.66
H <sub>2</sub> O <sub>2</sub> (aq)	-54.12	-52.89	-52.50
3-Hydroxypropanoate	-318.62	-316.17	-315.38
Hypoxanthine	249.33	251.77	252.56
Indole	503.49	507.78	509.16
Lactate	-316.94	-314.49	-313.70
Lactose	-688.29	-674.83	-670.48
<i>L</i> -Leucine	167.18	175.14	177.71
<i>L</i> -Isoleucine	175.53	183.49	186.06
<i>D</i> -Lyxose	-349.58	-343.46	-341.48
Malate	-682.88	-682.85	-682.85
Maltose	-695.65	-682.19	-677.84

Reactant	$\Delta_f G'^{\circ}(I = 0)$ kJ mol <sup>-1</sup>	$\Delta_f G'^{\circ}(I = 0.1 \text{ M})$ kJ mol <sup>-1</sup>	$\Delta_f G'^{\circ}(I = 0.25 \text{ M})$ kJ mol <sup>-1</sup>
<i>D</i> -Mannitol	-383.22	-374.65	-371.89
Mannose	-430.52	-423.18	-420.81
Methane(aq)	125.50	127.94	128.73
Methane(g)	109.11	111.55	112.34
Methanol	-15.48	-13.04	-12.25
<i>L</i> -Methionine	-63.40	-56.67	-54.49
N <sub>2</sub> (aq)	18.70	18.70	18.70
N <sub>2</sub> (g)	0.00	0.00	0.00
Nicotinamide Adenine Dinucleotide (NAD) [oxidized]	1038.86	1054.17	1059.11
Nicotinamide Adenine Dinucleotide (NAD) [reduced]	1101.47	1115.55	1120.09
Nicotinamide Adenine Dinucleotide Phosphate (NADP) [oxidized]	163.73	173.52	176.68
Nicotinamide Adenine Dinucleotide Phosphate (NADP) [reduced]	229.67	235.79	237.77
O <sub>2</sub> (aq)	16.40	16.40	16.40
O <sub>2</sub> (g)	0.00	0.00	0.00
Oxalate	-673.90	-676.35	-677.14
Oxaloacetate	-713.38	-714.60	-715.00
Oxalosuccinate	-979.05	-979.05	-979.05
2-Oxoglutarate	-633.58	-633.58	-633.58
Palmitate	979.25	997.61	1003.54
<i>L</i> -Phenylalanine	232.42	239.15	241.33
Phosphate	-1058.56	-1059.17	-1059.49
2-Phospho- <i>D</i> -glycerate	-1340.72	-1341.32	-1341.79
3-Phospho- <i>D</i> -glycerate	-1346.38	-1347.19	-1347.73
Phosphoenolpyruvate	-1185.46	-1188.53	-1189.73
1-Propanol	143.84	148.74	150.32
2-Propanol	134.42	139.32	140.90
Pyrophosphate	-1934.95	-1939.13	-1940.66
Pyruvate	-352.40	-351.18	-350.78
Retinal	1118.78	1135.91	1141.45
Retinol	1170.78	1189.14	1195.07
Ribose	-339.23	-333.11	-331.13
Ribose 1-phosphate	-1215.87	-1212.24	-1211.14
Ribose 5-phosphate	-1223.95	-1220.32	-1219.22
Ribulose	-336.38	-330.26	-328.28
<i>L</i> -Serine	-231.18	-226.89	-225.51
Sorbose	-432.47	-425.13	-422.76
Succinate	-530.72	-530.65	-530.64
Succinyl Coenzyme A	-349.90	-348.06	-347.47
Sucrose	-685.66	-672.20	-667.85
Thioredoxin [oxidized]	0.00	0.00	0.00
Thioredoxin [reduced]	54.32	55.41	55.74
<i>L</i> -Tryptophan	364.78	372.12	374.49
<i>L</i> -Tyrosine	68.82	75.55	77.73
Ubiquinone [oxidized]	3596.07	3651.15	3668.94
Ubiquinone [reduced]	3586.06	3642.37	3660.55
Urate	-206.03	-204.81	-204.41
Urea	-42.97	-40.53	-39.74
Uric acid	-197.07	-194.63	-193.84
<i>L</i> -Valine	80.87	87.60	89.78
<i>D</i> -Xylose	-350.93	-344.81	-342.83
<i>D</i> -Xylulose	-346.59	-340.47	-338.49