

SPECTROSCOPIC CONSTANTS OF DIATOMIC MOLECULES

This table lists the leading spectroscopic constants and equilibrium internuclear distance r_e in the ground electronic state for selected diatomic molecules. The constants are those describing the vibrational and rotational energy through the expressions:

$$E_{\text{vib}}/hc = \omega_e(v + 1/2) - \omega_e x_e(v + 1/2)^2 + \dots$$

$$E_{\text{rot}}/hc = B_v J(J + 1) - D_v [J(J + 1)]^2 + \dots$$

where

$$B_v = B_e - \alpha_e(v + 1/2) + \dots$$

$$D_v = D_e + \dots$$

Here v and J are the vibrational and rotational quantum numbers, respectively, h is Planck's constant, and c is the speed of light. In this customary formulation the constants ω_e , B_e , etc. have dimensions of inverse length; in this table they are given in units of cm^{-1} .

Users should note that higher order terms in the above energy expressions are required for very precise calculations; constants for many of these terms can be found in the references. Also, if the ground electronic state is not $^1\Sigma$, additional terms are needed to account for the interaction between electronic and pure rotational angular momentum. For some molecules in the table the data have been analyzed in terms of the Dunham series expansion:

$$E/hc = \sum_{lm} Y_{lm}(v + 1/2)^l J^m (J + 1)^m$$

In such cases it has been assumed that $Y_{10} = \omega_e$, $Y_{01} = B_e$, etc., although in the highest approximations these identities are not precisely correct. Some of the values of r_e in the table have been corrected for breakdown of the Born-Oppenheimer approximation, which can affect the last decimal place. Because of differences in the method of data analysis and limitations in the model, care should be taken in comparing r_e values for different molecules to a precision beyond 0.001 Å.

Molecules are listed in alphabetical order by formula as written in the most common form. In most cases this form places the more electropositive element first, but there are exceptions such as OH, NH, CH, etc.

* Indicates a value for the interval between $v = 0$ and $v = 1$ states instead of a value of ω_e .

References

1. Huber, K. P., and Herzberg, G., *Molecular Spectra and Molecular Structure IV. Constants of Diatomic Molecules*, Van Nostrand Reinhold, New York, 1979.
2. Lovas, F. J., and Tiemann, E., *J. Phys. Chem. Ref. Data*, 3, 609, 1974.
3. *Landolt-Börnstein, Numerical Data and Functional Relationships in Science and Technology, New Series*, II/6 (1974), II/14a (1982), II/14b (1983), II/19a (1992), II/19d-1 (1995), *Molecular Constants*, Springer-Verlag, Heidelberg.

Molecule	State	ω_e cm^{-1}	$\omega_e x_e$ cm^{-1}	B_e cm^{-1}	α_e cm^{-1}	D_e 10^{-6}cm^{-1}	r_e Å
¹⁰⁷ Ag ⁷⁹ Br	¹ Σ ⁺	249.57	0.63	0.064833	0.0002361	0.0175	2.39311
¹⁰⁷ Ag ³⁵ Cl	¹ Σ ⁺	343.49	1.17	0.12298388	0.00059541	0.06305	2.28079
¹⁰⁷ Ag ¹⁹ F	¹ Σ ⁺	513.45	2.59	0.2657020	0.0019206	0.284	1.98318
¹⁰⁷ Ag ¹ H	¹ Σ ⁺	1759.9	34.06	6.449	0.201	344	1.618
¹⁰⁷ Ag ² H	¹ Σ ⁺	1250.70	17.17	3.2572	0.0722	85.9	1.6180
¹⁰⁷ Ag ¹²⁷ I	¹ Σ ⁺	206.50	0.46	0.04486821	0.0001414	0.00847	2.54463
¹⁰⁷ Ag ¹⁶ O	² Π _{1/2}	490.2	3.1	0.3020	0.0025	0.45	2.003
²⁷ Al ₂	³ Σ _g ⁻	350.01	2.02	0.2054	0.0012	0.31	2.466
²⁷ Al ⁷⁹ Br	¹ Σ ⁺	378.0	1.28	0.15919713	0.00086045	0.11285	2.29481
²⁷ Al ³⁵ Cl	¹ Σ ⁺	481.30	1.95	0.24393012	0.00161113	0.2503	2.13011
²⁷ Al ¹⁹ F	¹ Σ ⁺	802.3	4.77	0.5524798	0.0049841	1.0464	1.65437
²⁷ Al ¹ H	¹ Σ ⁺	1682.56	29.09	6.3907	0.1858	356.5	1.6478
²⁷ Al ² H	¹ Σ ⁺	1211.95	15.14	3.3186	0.0697	97	1.6463
²⁷ Al ¹²⁷ I	¹ Σ ⁺	316.1	1.0	0.11769985	0.00055859		2.53710
²⁷ Al ¹⁶ O	² Σ ⁺	979.23	6.97	0.6414	0.0058	1.08	1.6179
²⁷ Al ³² S	² Σ ⁺	617.1	3.33	0.2799	0.0018	0.22	2.029
⁷⁵ As ₂	¹ Σ _g ⁺	429.55	1.12	0.10179	0.000333		2.1026
⁷⁵ As ¹ H	³ Σ ⁻	2130*		7.3067	0.2117	327	1.52315
⁷⁵ As ² H	³ Σ ⁻	1484*		3.6688		90	1.5306
⁷⁵ As ¹⁴ N	¹ Σ ⁺	1068.54	5.41	0.54551	0.003366	0.53	1.6184
⁷⁵ As ¹⁶ O	² Π _{1/2}	967.08	4.85	0.48482	0.003299	0.49	1.6236
¹⁹⁷ Au ₂	¹ Σ _g ⁺	190.9	0.42	0.028013	0.0000723	0.00250	2.4719
¹⁹⁷ Au ¹ H	¹ Σ ⁺	2305.01	43.12	7.2401	0.2136	279	1.5239
¹⁹⁷ Au ² H	¹ Σ ⁺	1634.98	21.65	3.6415	0.07614	70.9	1.5238
¹¹ B ₂	³ Σ _g ⁻	1051.3	9.35	1.212	0.014		1.590
¹¹ B ⁷⁹ Br	¹ Σ ⁺	684.31	3.52	0.4894	0.0035	1.00	1.888
¹¹ B ³⁵ Cl	¹ Σ ⁺	840.29	5.49	0.684282	0.006812	1.84	1.71528
¹¹ B ¹⁹ F	¹ Σ ⁺	1402.1	11.8	1.516950	0.019056	7.105	1.26267
¹¹ B ¹ H	¹ Σ ⁺	2366.9	49.40	12.021	0.412	1242	1.2324
¹¹ B ² H	¹ Σ ⁺	1703.3	28	6.54	0.17	400	1.2324
¹¹ B ¹⁴ N	³ Π	1514.6	12.3	1.666	0.025	8.1	1.281
¹¹ B ¹⁶ O	² Σ ⁺	1885.69	11.81	1.7820	0.0166	6.32	1.2045

Molecule	State	ω_e cm^{-1}	$\omega_e x_e$ cm^{-1}	B_e cm^{-1}	α_e cm^{-1}	D_e 10^{-6}cm^{-1}	r_e \AA
$^{11}\text{B}^{32}\text{S}$	$2\Sigma^+$	1180.17	6.31	0.7949	0.0061	1.40	1.6092
$^{138}\text{Ba}^{79}\text{Br}$	$2\Sigma^+$	193.77	0.41	0.0415082	0.0001219	0.00762	2.84449
$^{138}\text{Ba}^{35}\text{Cl}$	$2\Sigma^+$	279.92	0.82	0.08396717	0.00033429	0.03022	2.68276
$^{138}\text{Ba}^{19}\text{F}$	$2\Sigma^+$	468.9	1.79	0.2159	0.0012	0.175	2.163
$^{138}\text{Ba}^1\text{H}$	$2\Sigma^+$	1168.31	14.50	3.38285	0.06599	112.67	2.23175
$^{138}\text{Ba}^2\text{H}$	$2\Sigma^+$	829.77	7.32	1.7071	0.02363	28.77	2.2304
$^{138}\text{Ba}^{127}\text{I}$	$2\Sigma^+$	152.14	0.27	0.02680587	0.00006634	0.00333	3.08476
$^{138}\text{Ba}^{16}\text{O}$	$1\Sigma^+$	669.76	2.03	0.3126140	0.0013921	0.2724	1.93969
$^{138}\text{Ba}^{32}\text{S}$	$1\Sigma^+$	379.42	0.88	0.10331	0.0003188	0.0306	2.5074
$^9\text{Be}^{19}\text{F}$	$2\Sigma^+$	1247.36	9.12	1.4889	0.0176	8.28	1.3610
$^9\text{Be}^1\text{H}$	$2\Sigma^+$	2060.78	36.31	10.3164	0.3030	1022.1	1.3426
$^9\text{Be}^2\text{H}$	$2\Sigma^+$	1530.32	20.71	5.6872	0.1225	313.8	1.3419
$^9\text{Be}^{16}\text{O}$	$1\Sigma^+$	1487.32	11.83	1.6510	0.0190	8.20	1.3309
$^9\text{Be}^{32}\text{S}$	$1\Sigma^+$	997.94	6.14	0.79059	0.00664	2.00	1.7415
$^{209}\text{Bi}_2$	$1\Sigma_g^+$	172.71	0.34	0.022781	0.000055	0.00150	2.6596
$^{209}\text{Bi}^1\text{H}$	$3\Sigma^-$	1635.73	31.6	5.137	0.148	183	1.805
$^{209}\text{Bi}^2\text{H}$	$3\Sigma^-$	1173.32	16.1	2.592	0.054	50.6	1.804
$^{79}\text{Br}_2$	$1\Sigma_g^+$	325.32	1.08	0.082107	0.0003187	0.02092	2.2811
$^{79}\text{Br}^{35}\text{Cl}$	$1\Sigma^+$	444.28	1.84	0.152470	0.000770	0.07183	2.13607
$^{79}\text{Br}^{19}\text{F}$	$1\Sigma^+$	670.75	4.05	0.35584	0.00261	0.401	1.75894
$^{79}\text{Br}^{16}\text{O}$	$2\Pi_{3/2}$	779	6.8	0.429598	0.003639	0.523	1.717
$^{12}\text{C}_2$	$1\Sigma_g^+$	1854.71	13.34	1.8198	0.0177	6.92	1.2425
$^{12}\text{C}^{35}\text{Cl}$	$2\Pi_{1/2}$	866.72*	6.2	0.6936	0.00672	1.9	1.6450
$^{12}\text{C}^{19}\text{F}$	$2\Pi_{1/2}$	1308.1	11.10	1.4172	0.0184	6.5	1.2718
$^{12}\text{C}^1\text{H}$	$2\Pi_{1/2}$	2858.5	63.0	14.457	0.534	1450	1.1199
$^{12}\text{C}^2\text{H}$	$2\Pi_{1/2}$	2099.8	34.02	7.806	0.208	420	1.1190
$^{12}\text{C}^{14}\text{N}$	$2\Sigma^+$	2068.59	13.09	1.8997830	0.0173717	6.4034	1.17181
$^{12}\text{C}^{16}\text{O}$	$1\Sigma^+$	2169.81	13.29	1.93128075	0.01750390	6.1216	1.12823
$^{12}\text{C}^{31}\text{P}$	$2\Sigma^+$	1239.67	6.86	0.7986	0.00597	1.33	1.562
$^{12}\text{C}^{32}\text{S}$	$1\Sigma^+$	1285.15	6.50	0.8200434	0.0059182	1.336	1.53482
$^{12}\text{C}^{80}\text{Se}$	$1\Sigma^+$	1035.36	4.86	0.5750	0.00379	0.71	1.67609
$^{40}\text{Ca}^{35}\text{Cl}$	$2\Sigma^+$	367.53	1.31	0.1522302	0.0007990	0.1029	2.43676
$^{40}\text{Ca}^{19}\text{F}$	$2\Sigma^+$	581.1	2.74	0.339	0.0026	0.45	1.967
$^{40}\text{Ca}^1\text{H}$	$2\Sigma^+$	1298.34	19.10	4.2766	0.0970	183.7	2.0025
$^{40}\text{Ca}^2\text{H}$	$2\Sigma^+$	910*		2.1769	0.035	47.9	2.002
$^{40}\text{Ca}^{127}\text{I}$	$2\Sigma^+$	238.70	0.63	0.0693263	0.0002634	0.0234	2.82859
$^{40}\text{Ca}^{16}\text{O}$	$1\Sigma^+$	732.03	4.83	0.444441	0.003282	0.6541	1.8221
$^{40}\text{Ca}^{32}\text{S}$	$1\Sigma^+$	462.23	1.78	0.1766757	0.0008270	0.1032	2.31775
$^{114}\text{Cd}^1\text{H}$	$2\Sigma^+$	1337.1*		5.323		314	1.781
$^{114}\text{Cd}^2\text{H}$	$2\Sigma^+$			2.704		76	1.775
$^{35}\text{Cl}_2$	$1\Sigma_g^+$	559.7	2.68	0.2440	0.0015	0.186	1.988
$^{35}\text{Cl}^{19}\text{F}$	$1\Sigma^+$	786.15	6.16	0.516479	0.004358	0.88	1.62831
$^{35}\text{Cl}^{16}\text{O}$	$2\Pi_{3/2}$	853.8	5.5	0.62345	0.0058	1.33	1.56963
$^{52}\text{Cr}^1\text{H}$	$6\Sigma^+$	1581*	32	6.220	0.179	347	1.656
$^{52}\text{Cr}^2\text{H}$	$6\Sigma^+$	1182*		3.14		88.8	1.664
$^{52}\text{Cr}^{16}\text{O}$	5Π	898.4	6.8	0.5231	0.0070		1.615
$^{133}\text{Cs}_2$	$1\Sigma_g^+$	42.02	0.08	0.0127	0.0000264	0.00464	4.47
$^{133}\text{Cs}^{79}\text{Br}$	$1\Sigma^+$	149.66	0.37	0.03606925	0.00012401	0.00838	3.07225
$^{133}\text{Cs}^{35}\text{Cl}$	$1\Sigma^+$	214.17	0.73	0.07209149	0.00033756	0.03268	2.90627
$^{133}\text{Cs}^{19}\text{F}$	$1\Sigma^+$	352.56	1.62	0.18436969	0.0011756	0.20168	1.34535
$^{133}\text{Cs}^1\text{H}$	$1\Sigma^+$	891.0	12.9	2.7099	0.0579	113	2.4938
$^{133}\text{Cs}^2\text{H}$	$1\Sigma^+$	619.1*		1.354		20	2.505
$^{133}\text{Cs}^{127}\text{I}$	$1\Sigma^+$	119.18	0.25	0.02362736	0.00006826	0.00371	3.31519
$^{133}\text{Cs}^{16}\text{O}$	$2\Sigma^+$	357.5*		0.223073	0.001303	0.348	2.3007
$^{63}\text{Cu}_2$	$1\Sigma_g^+$	264.55	1.02	0.10874	0.000614	0.0716	2.2197
$^{63}\text{Cu}^{79}\text{Br}$	$1\Sigma^+$	314.8	0.96	0.10192625	0.00045214	0.04274	2.17344
$^{63}\text{Cu}^{35}\text{Cl}$	$1\Sigma^+$	415.29	1.58	0.17628802	0.00099647	0.12706	2.05118
$^{63}\text{Cu}^{19}\text{F}$	$1\Sigma^+$	622.7	3.95	0.3794029	0.0032298	0.563	1.74493
$^{63}\text{Cu}^1\text{H}$	$1\Sigma^+$	1941.26	37.51	7.9441	0.2563	520	1.46263
$^{63}\text{Cu}^2\text{H}$	$1\Sigma^+$	1384.14	18.97	4.0381	0.0917	136.2	1.4626

Molecule	State	ω_e cm ⁻¹	$\omega_e x_e$ cm ⁻¹	B_e cm ⁻¹	α_e cm ⁻¹	D_e 10 ⁻⁶ cm ⁻¹	r_e Å
⁶³ Cu ¹²⁷ I	¹ Σ ⁺	264.5	0.60	0.07328742	0.00028390	0.02244	2.33832
⁶³ Cu ¹⁶ O	² Π _{3/2}	640.17	4.43	0.44454	0.00456	0.85	1.7244
⁶³ Cu ³² S	² Π _{3/2}	415.0	1.75	0.1891		0.18	2.051
¹⁹ F ₂	¹ Σ _g ⁺	916.64	11.24	0.89019	0.013847	3.3	1.41193
⁵⁶ Fe ¹⁶ O	⁵ Δ	965*		0.650		0.72	1.444
⁶⁹ Ga ⁸¹ Br	¹ Σ ⁺	263.0	0.81	0.081839	0.0003207	0.032	2.35248
⁶⁹ Ga ³⁵ Cl	¹ Σ ⁺	365.67	1.25	0.1499046	0.0007936	0.1008	2.20169
⁶⁹ Ga ¹⁹ F	¹ Σ ⁺	622.2	3.2	0.3595161	0.0028642	0.50	1.77437
⁶⁹ Ga ¹ H	¹ Σ ⁺	1604.52	28.77	6.137	0.181	342	1.663
⁶⁹ Ga ² H	¹ Σ ⁺			3.083	0.06	84	1.663
⁶⁹ Ga ¹²⁷ I	¹ Σ ⁺	216.38	0.47	0.0569359	0.0001897	0.015770	2.57464
⁶⁹ Ga ¹⁶ O	² Σ	767.5	6.24	0.4271		0.37	1.744
⁷⁴ Ge ⁷⁹ Br	² Π _{1/2}	295	0.7				
⁷⁴ Ge ³⁵ Cl	² Π _{1/2}	407.6	1.36				
⁷² Ge ¹ H	² Π _{1/2}	1833.77	37	6.726	0.192	326	1.5880
⁷² Ge ² H	² Π _{1/2}	1320.09	19	3.415	0.070	83.2	1.5874
⁷⁴ Ge ¹⁶ O	¹ Σ ⁺	986.49	4.47	0.4856981	0.0030787	0.4709	1.62464
⁷⁴ Ge ³² S	¹ Σ ⁺	575.8	1.80	0.18656576	0.00074910	0.07883	2.01209
⁷⁴ Ge ⁸⁰ Se	¹ Σ ⁺	408.7	1.36	0.09634051	0.00028904	0.02207	2.13463
⁷⁴ Ge ¹³⁰ Te	¹ Σ ⁺	323.9	0.75	0.06533821	0.00017246	0.012	2.34017
¹ H ₂	¹ Σ _g ⁺	4401.21	121.34	60.853	3.062	47100	0.74144
² H ₂	¹ Σ _g ⁺	3115.50	61.82	30.444	1.0786	11410	0.74152
³ H ₂	¹ Σ _g ⁺	2546.5	41.23	20.335	0.5887		0.74142
¹ H ⁸¹ Br	¹ Σ ⁺	2648.97	45.22	8.46488	0.23328	345.8	1.41444
² H ⁸¹ Br	¹ Σ ⁺	1884.75	22.72	4.245596	0.084	88.32	1.4145
¹ H ³⁵ Cl	¹ Σ ⁺	2990.95	52.82	10.59342	0.30718	531.94	1.27455
² H ³⁵ Cl	¹ Σ ⁺	2145.16	27.18	5.448796	0.113292	140	1.27458
¹ H ¹⁹ F	¹ Σ ⁺	4138.32	89.88	20.9557	0.798	2151	0.91681
² H ¹⁹ F	¹ Σ ⁺	2998.19	45.76	11.0102	0.3017	594	0.91694
¹ H ¹²⁷ I	¹ Σ ⁺	2309.01	39.64	6.4263650	0.1689	206.9	1.60916
²⁰² Hg ¹ H	² Σ ⁺	1203.24*		5.3888		395.3	1.7662
²⁰² Hg ² H	² Σ ⁺	896.12*		2.739		91	1.757
¹²⁷ I ₂	¹ Σ _g ⁺	214.50	0.61	0.03737	0.000114	0.0043	2.666
¹²⁷ I ⁷⁹ Br	¹ Σ ⁺	268.64	0.81	0.0568325	0.0001969	0.0102	2.46899
¹²⁷ I ³⁵ Cl	¹ Σ ⁺	384.29	1.50	0.1141587	0.0005354	0.0403	2.32088
¹²⁷ I ¹⁹ F	¹ Σ ⁺	610.24	3.12	0.2797111	0.0018738	0.2356	1.90976
¹²⁷ I ¹⁶ O	² Π _{3/2}	681.5	4.3	0.34026	0.00270	0.36	1.8676
¹¹⁵ In ⁸¹ Br	¹ Σ ⁺	221.0	0.65	0.05489468	0.00018672	0.01350	2.54315
¹¹⁵ In ³⁵ Cl	¹ Σ ⁺	317.39	1.03	0.1090583	0.0005177	0.0515	2.40117
¹¹⁵ In ¹⁹ F	¹ Σ ⁺	535.4	2.6	0.2623241	0.0018798	0.252	1.98540
¹¹⁵ In ¹ H	¹ Σ ⁺	1476.0	25.61	4.995	0.143	223	1.8380
¹¹⁵ In ² H	¹ Σ ⁺	1048.2	12.4	2.523	0.051	58	1.837
¹¹⁵ In ¹²⁷ I	¹ Σ ⁺	177.08	0.34	0.03686702	0.00010411	0.00639	2.75364
³⁹ K ₂	¹ Σ _g ⁺	92.02	0.28	0.056743	0.000165	0.0863	3.9051
³⁹ K ⁷⁹ Br	¹ Σ ⁺	213	0.80	0.08122109	0.00040481	0.04462	2.82078
³⁹ K ³⁵ Cl	¹ Σ ⁺	281	1.30	0.1286348	0.0007899	0.1087	2.66665
³⁹ K ¹⁹ F	¹ Σ ⁺	426.26	2.45	0.27993741	0.00233492	0.4829	2.17146
³⁹ K ¹ H	¹ Σ ⁺	983.6	14.3	3.416400	0.085313	163.55	2.243
³⁹ K ² H	¹ Σ ⁺	707	7.7	1.754	0.0318	50	2.240
³⁹ K ¹²⁷ I	¹ Σ ⁺	186.53	0.57	0.06087473	0.00026776	0.02593	3.04784
¹³⁹ La ¹⁶ O	² Σ ⁺	812.8	2.22	0.35252001	0.00142365	0.2626	1.82591
⁷ Li ₂	¹ Σ _g ⁺	351.43	2.61	0.67264	0.00704	9.87	2.6729
⁷ Li ⁷⁹ Br	¹ Σ ⁺	563.2	3.5	0.555399	0.005644	2.159	2.17043
⁷ Li ³⁵ Cl	¹ Σ ⁺	642.95	4.47	0.7065225	0.0080102	3.409	2.02067
⁷ Li ¹⁹ F	¹ Σ ⁺	910.57	8.21	1.3452583	0.0202887	11.745	1.56386
⁷ Li ¹ H	¹ Σ ⁺	1405.65	23.20	7.51373	0.21665	862	1.59490
⁷ Li ² H	¹ Σ ⁺	1054.80	12.94	4.23310	0.09155	276	1.5941
⁷ Li ¹²⁷ I	¹ Σ ⁺	496.85	2.85	0.4431766	0.0040862	1.4104	2.39192
⁷ Li ¹⁶ O	² Π	814.62	7.78	1.212830	0.017899	0.1079	1.68822
²⁴ Mg ₂	¹ Σ _g ⁺	51.12	1.64	0.09287	0.00378	1.22	3.891

Molecule	State	ω_e cm ⁻¹	$\omega_e x_e$ cm ⁻¹	B_e cm ⁻¹	α_e cm ⁻¹	D_e 10 ⁻⁶ cm ⁻¹	r_e Å
²⁴ Mg ³⁵ Cl	2Σ ⁺	462.12*	2.1	0.2456154	0.0016204	0.2723	2.19639
²⁴ Mg ¹⁹ F	2Σ ⁺	711.69*	4.9	0.51922	0.00470	1.080	1.7500
²⁴ Mg ¹ H	2Σ ⁺	1495.20	31.89	5.8257	0.1859	344	1.7297
²⁴ Mg ² H	2Σ ⁺	1077.9	16.1	3.0306	0.06289	92	1.7302
²⁴ Mg ¹⁶ O	1Σ ⁺	784.78	5.26	0.57470436	0.00532377	1.2328	1.74838
⁵⁵ Mn ¹ H	7Σ	1548.0	28.8	5.6841	0.1570	303.9	1.7311
⁵⁵ Mn ² H	7Σ	1103	13.9	2.8957	0.051	79.5	1.7310
¹⁴ N ₂	1Σ _g ⁺	2358.57	14.32	1.99824	0.017318	5.76	1.09769
¹⁴ N ⁷⁹ Br	3Σ ⁻	691.75	4.72	0.444	0.0040		1.79
¹⁴ N ³⁵ Cl	3Σ ⁻	827.96	5.30	0.649770	0.006414	1.598	1.61071
¹⁴ N ¹⁹ F	3Σ ⁻	1141.37	8.99	1.2057	0.01492	5.39	1.3170
¹⁴ N ¹ H	3Σ ⁻	3282.3	78.4	16.6993	0.6490	1709.7	1.0362
¹⁴ N ² H	3Σ ⁻	2398	42	8.7913	0.2531	490.4	1.0361
¹⁴ N ¹⁶ O	2Π _{1/2}	1904.20	14.07	1.67195	0.0171	0.5	1.15077
¹⁴ N ³² S	2Π _{1/2}	1218.7	7.28	0.769602	0.0064	1.2	1.4940
²³ Na ₂	1Σ _g ⁺	159.13	0.72	0.154707	0.008736	0.581	3.0789
²³ Na ⁷⁹ Br	1Σ ⁺	302	1.5	0.1512533	0.0009410	0.1554	2.50204
²³ Na ³⁵ Cl	1Σ ⁺	366	2.05	0.2180631	0.0016248	0.3120	2.36080
²³ Na ¹⁹ F	1Σ ⁺	535.66	3.57	0.4369011	0.0045580	1.163	1.92595
²³ Na ¹ H	1Σ ⁺	1172.2	19.72	4.9033634	0.1370919	343.40	1.88654
²³ Na ² H	1Σ ⁺	826.1*		2.557089	0.051600	93.46	1.88654
²³ Na ¹²⁷ I	1Σ ⁺	258	1.1	0.1178056	0.0006478	0.0973	2.71145
²³ Na ¹⁶ O	2Π	492.3		0.424630	0.004506	1.2638	2.05155
⁹³ Nb ¹⁶ O	4Σ ⁻	989.0	3.8	0.4321	0.0021	0.22	1.691
⁵⁸ Ni ¹ H	2Δ _{5/2}	1926.6	38	7.700	0.23	481	1.476
⁵⁸ Ni ² H	2Δ _{5/2}	1390.1	19	3.992	0.092	130	1.465
¹⁶ O ₂	3Σ _g ⁻	1580.19	11.98	1.44563	0.0159	4.839	1.20752
¹⁶ O ¹ H	2Π _{3/2}	3737.76	84.88	18.911	0.7242	1938	0.96966
¹⁶ O ² H	2Π _{3/2}	2720.24	44.05	10.021	0.276	537.4	0.9698
³¹ P ₂	1Σ _g ⁺	780.77	2.84	0.30362	0.00149	0.188	1.8934
³¹ P ³⁵ Cl	3Σ ⁻	551.38	2.23	0.2528748	0.0015119	0.2124	2.01461
³¹ P ¹⁹ F	3Σ ⁻	846.75	4.49	0.5665	0.00456		1.58938
³¹ P ¹ H	3Σ ⁻	2365.2	44.5	8.5371	0.2514	436	1.42140
³¹ P ² H	3Σ ⁻	1699.2	23.0	4.4081	0.0928	116	1.4220
³¹ P ¹⁴ N	1Σ ⁺	1337.24	6.98	0.7864854	0.0055364	1.091	1.49087
³¹ P ¹⁶ O	2Π _{1/2}	1233.34	6.56	0.7337	0.0055	1.3	1.4759
²⁰⁸ Pb ₂		110.5	0.35				
²⁰⁸ Pb ⁷⁹ Br	2Π _{1/2}	207.5	0.50				
²⁰⁸ Pb ³⁵ Cl	2Π _{1/2}	303.9	0.88				
²⁰⁸ Pb ¹⁹ F	2Π _{1/2}	502.73	2.28	0.22875	0.001473	0.183	2.0575
²⁰⁸ Pb ¹ H	2Π _{1/2}	1564.1	29.75	4.971	0.144	201	1.839
²⁰⁸ Pb ¹⁶ O	1Σ ⁺	720.96	3.52	0.30730373	0.00190977	0.2138	1.92181
²⁰⁸ Pb ³² S	1Σ ⁺	429.17	1.26	0.11632307	0.00043510	0.03418	2.28678
²⁰⁸ Pb ⁸⁰ Se	1Σ ⁺	277.6	0.51	0.05059953	0.00012993	0.0070	2.40218
²⁰⁸ Pb ¹³⁰ Te	1Σ ⁺	212.0	0.43	0.03130774	0.00006743	0.0027	2.59492
¹⁹⁵ Pt ¹² C	1Σ ⁺	1051.13	4.86	0.53044	0.003273	0.546	1.6767
¹⁹⁵ Pt ¹ H	2Δ _{5/2}	2294.68*	46	7.1963	0.1996	261	1.52852
¹⁹⁵ Pt ² H	2Δ _{5/2}	1644.3*	23	3.640	0.071	66	1.524
⁸⁵ Rb ⁷⁹ Br	1Σ ⁺	169.46	0.46	0.04752798	0.00018596	0.01496	2.94474
⁸⁵ Rb ³⁵ Cl	1Σ ⁺	228	0.92	0.0876404	0.0004537	0.04947	2.78673
⁸⁵ Rb ¹⁹ F	1Σ ⁺	376	1.9	0.2106640	0.0015228	0.2684	2.27033
⁸⁵ Rb ¹ H	1Σ ⁺	936.9	14.21	3.020	0.072	123	2.367
⁸⁵ Rb ¹²⁷ I	1Σ ⁺	138.51	0.33	0.03283293	0.00010946	0.00738	3.17688
⁸⁵ Rb ¹⁶ O	2Σ ⁺	388.4*		0.246481	0.002174	0.397	2.25420
³² S ₂	3Σ _g ⁻	725.65	2.84	0.2955	0.001570	0.19	1.8892
³² S ¹⁹ F	2Π _{3/2}			0.552174			1.60058
³² S ¹ H	2Π _{3/2}	2711.6	59.9	9.5995	0.2785	480.6	1.34066
³² S ² H	2Π _{3/2}	1885	31	4.95130	0.10308	130	1.34049
³² S ¹⁶ O	3Σ ⁻	1149.2	5.6	0.7208171	0.005737	1.134	1.48109
¹²¹ Sb ³⁵ Cl	3Σ ⁻	374.7	0.6				

Molecule	State	ω_e cm^{-1}	$\omega_e x_e$ cm^{-1}	B_e cm^{-1}	α_e cm^{-1}	D_e 10^{-6}cm^{-1}	r_e \AA
¹²¹ Sb ¹⁹ F	³ Σ^-	605.0	2.6	0.2792	0.0020	0.23	1.918
¹²¹ Sb ¹ H	³ Σ^-			5.684		240	1.723
¹²¹ Sb ² H	³ Σ^-			2.8782		45	1.7194
¹²¹ Sb ¹⁴ N	¹ Σ^+	942.0	5.6				
¹²¹ Sb ¹⁶ O	² $\Pi_{1/2}$	816	4.2	0.3580	0.0022	0.270	1.826
⁴⁵ Sc ¹⁹ F	¹ Σ^+	735.6	3.8	0.3950	0.00266		1.788
⁸⁰ Se ₂	³ Σ_g^-	385.30	0.96	0.08992	0.000288	0.024	2.166
⁸⁰ Se ¹ H	² $\Pi_{3/2}$	2400*		8.02	0.23	330	1.48
⁸⁰ Se ² H	² $\Pi_{3/2}$	1708*		3.94			1.48
⁸⁰ Se ¹⁶ O	³ Σ^-	914.69	4.52	0.4655	0.00323	0.5	1.648
²⁸ Si ₂	³ Σ_g^-	510.98	2.02	0.2390	0.0014	0.21	2.246
²⁸ Si ³⁵ Cl	² $\Pi_{1/2}$	535.60	2.17	0.2561	0.0016	0.25	2.058
²⁸ Si ¹⁹ F	² $\Pi_{1/2}$	857.19	4.73	0.5812	0.00494	1.07	1.6011
²⁸ Si ¹ H	² $\Pi_{1/2}$	2041.80	35.51	7.4996	0.2190	397	1.5201
²⁸ Si ² H	² $\Pi_{1/2}$	1469.32	18.23	3.8840	0.0781	105.4	1.5199
²⁸ Si ¹⁴ N	² Σ^+	1151.4	6.47	0.7311	0.00565	1.2	1.572
²⁸ Si ¹⁶ O	¹ Σ^+	1241.54	5.97	0.7267521	0.0050379	0.9923	1.50975
²⁸ Si ³² S	¹ Σ^+	749.64	2.58	0.30352788	0.00147308	0.201	1.92926
²⁸ Si ⁸⁰ Se	¹ Σ^+	580.0	1.78	0.1920117	0.0007767	0.0842	2.05832
¹²⁰ Sn ⁷⁹ Br	² $\Pi_{1/2}$	247.2	0.6				
¹²⁰ Sn ³⁵ Cl	² $\Pi_{1/2}$	351.1	1.06	0.1117	0.0004		2.361
¹¹⁸ Sn ¹⁹ F	² $\Pi_{1/2}$	577.6	2.69	0.2727	0.0014	0.26	1.944
¹²⁰ Sn ¹ H	² $\Pi_{1/2}$			5.31488		207.5	1.78146
¹²⁰ Sn ² H	² $\Pi_{1/2}$	1188.0*		2.6950	0.049	53.4	1.7770
¹²⁰ Sn ¹²⁷ I	² $\Pi_{1/2}$	199.0	0.6				
¹²⁰ Sn ¹⁶ O	¹ Σ^+	822.13	3.72	0.35571998	0.00214432	0.26638	1.83251
¹²⁰ Sn ³² S	¹ Σ^+	487.26	1.36	0.13686139	0.00050563	0.0424	2.20898
¹²⁰ Sn ⁸⁰ Se	¹ Σ^+	331.2	0.74	0.0649978	0.0001705	0.011	2.32557
¹²⁰ Sn ¹³⁰ Te	¹ Σ^+	259.5	0.50	0.04247917	0.00009543	0.0055	2.52280
⁸⁸ Sr ⁷⁹ Br	² Σ^+	216.60	0.52	0.0541847	0.0001827	0.01356	2.73522
⁸⁸ Sr ³⁵ Cl	² Σ^+	302.3	0.95				
⁸⁸ Sr ¹⁹ F	² Σ^+	502.4	2.3	0.2505346	0.0015513	0.2498	2.07537
⁸⁸ Sr ¹ H	² Σ^+	1206.2	17.0	3.6751	0.0814	135	2.1456
⁸⁸ Sr ² H	² Σ^+	841	8.6	1.8609	0.0292	34.7	2.1449
⁸⁸ Sr ¹²⁷ I	² Σ^+	173.77	0.35	0.0367097	0.0001060	0.00655	2.94364
⁸⁸ Sr ¹⁶ O	¹ Σ^+	653.5	3.96	0.33798	0.00219	0.36	1.91983
¹⁸¹ Ta ¹⁶ O	² $\Delta_{3/2}$	1028.69	3.51	0.40284	0.00182	0.2450	1.68746
¹³⁰ Te ₂	³ Σ_g^-	247.07	0.51	0.039681	0.000106	0.0044	2.5574
¹³⁰ Te ¹ H	² $\Pi_{3/2}$			5.56			1.74
¹³⁰ Te ¹⁶ O	⁰ Σ^+	797.11	4.00	0.3554	0.00237	0.27	1.825
²³² Th ¹⁶ O	¹ Σ^+	895.77	2.39	0.332644	0.001302	0.1833	1.84032
⁴⁸ Ti ¹⁶ O	³ Δ_1	1009.02	4.50	0.53541	0.00301	0.603	1.6202
²⁰⁵ Tl ⁸¹ Br	¹ Σ^+	192.10	0.39	0.0423899	0.0001276	0.0083	1.61817
²⁰⁵ Tl ³⁵ Cl	¹ Σ^+	284.71	0.86	0.09139702	0.00039784	0.0377	2.48483
²⁰⁵ Tl ¹⁹ F	¹ Σ^+	476.86	2.24	0.22315014	0.00150380	0.1955	2.08439
²⁰⁵ Tl ¹ H	¹ Σ^+	1390.7	22.7	4.806	0.154	254	1.870
²⁰⁵ Tl ² H	¹ Σ^+	987.7	12.04	2.419	0.057	60	1.869
²⁰⁵ Tl ¹²⁷ I	¹ Σ^+	150*		0.0271676	0.0000664	0.0036	2.81361
⁵¹ V ¹⁶ O	⁴ Σ^-	1011.3	4.86	0.54825	0.00352	0.6	1.5893
⁸⁹ Y ³⁵ Cl	¹ Σ	380.7	1.3	0.1160	0.0003	0.09	2.41
⁸⁹ Y ¹⁹ F	¹ Σ^+	631.29	2.50	0.29042	0.00163	0.237	1.9257
⁸⁹ Y ¹⁶ O	² Σ^+	861.0	2.9	0.3881	0.0018	0.32	1.790
¹⁷⁴ Yb ¹ H	² Σ^+	1249.54	21.06	3.9931	0.0957	161.8	2.0526
¹⁷⁴ Yb ² H	² Σ^+	886.6	10.57	2.01162	0.03425	41.60	2.0516
⁶⁴ Zn ³⁵ Cl	² Σ	390.5	1.6				
⁶⁴ Zn ¹⁹ F	² Σ	628	3.5				
⁶⁴ Zn ¹ H	² Σ^+	1607.6	55.14	6.6794	0.2500	466	1.5949
⁶⁴ Zn ² H	² Σ^+	1072	28	3.350		124	1.6054
⁶⁴ Zn ¹²⁷ I	² Σ	223.4	0.6				
⁹⁰ Zr ¹⁶ O	¹ Σ^+	969.8	4.9	0.42263	0.0023	0.319	1.7116