

# IONIZATION ENERGIES OF GAS-PHASE MOLECULES

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This table presents values for the first ionization energies (IP) of approximately 1000 molecules and atoms. Substances are listed by molecular formula in the modified Hill order (see Preface). Values enclosed in parentheses are considered not to be well established. Data appearing in the 1988 reference, were updated in 1996 for inclusion in the database of ionization energies available at the Internet site of the Standard Reference Data program of the National Institute of Standards and Technology (<http://webbook.nist.gov>). The list appearing here includes these updates.

The list also includes values for enthalpies of formation of the ions at 298 K,  $\Delta_f H_{\text{ion}}$ , given according to the ion convention used by

mass spectrometrists; to convert these values to the electron convention used by thermodynamicists, add 6 kJ/mol. Details on the calculation of  $\Delta_f H_{\text{ion}}$ , as well as data for a much larger number of molecules, may be found in the reference and on the Internet site.

## Reference

Lias, S. G., Bartmess, J. E., Liebman, J. F., Holmes, J. L., Levin, R. D., and Mallard, W.G., *Gas-Phase Ion and Neutral Thermochemistry*, *J. Phys. Chem. Ref. Data*, Vol. 17, Suppl. No. 1, 1988.

Mol. form.	Name	IP/eV	$\Delta_f H_{\text{ion}}$ kJ/mol
<i>Substances not containing carbon</i>			
Ac	Actinium	5.17	905
Ag	Silver	7.57624	1016
AgCl	Silver(I) chloride	( $\leq 10.08$ )	$\leq 1065$
AgF	Silver(I) fluoride	( $11.0 \pm 0.3$ )	1071
Al	Aluminum	5.98577	905
AlBr	Aluminum monobromide	(9.3)	913
AlBr <sub>3</sub>	Aluminum tribromide	(10.4)	593
AlCl	Aluminum monochloride	9.4	855
AlCl <sub>3</sub>	Aluminum trichloride	(12.01)	573
AlF	Aluminum monofluoride	$9.73 \pm 0.01$	673
AlF <sub>3</sub>	Aluminum trifluoride	$\leq 15.45$	$\leq 282$
AlI	Aluminum monoiodide	$9.3 \pm 0.3$	965
AlI <sub>3</sub>	Aluminum triiodide	(9.1)	673
Am	Americium	$5.9738 \pm 0.0002$	860
Ar	Argon	15.75962	1521
As	Arsenic	9.8152	1250
AsCl <sub>3</sub>	Arsenic(III) chloride	( $10.55 \pm 0.025$ )	754
AsF <sub>3</sub>	Arsenic(III) fluoride	( $12.84 \pm 0.05$ )	452
AsH <sub>3</sub>	Arsine	(9.89)	1021
Au	Gold	9.22567	1254
B	Boron	8.29803	1363
BBr <sub>3</sub>	Boron tribromide	(10.51)	809
BCl <sub>3</sub>	Boron trichloride	$11.60 \pm 0.02$	718
BF	Fluoroborane	$11.12 \pm 0.01$	957
	Difluoroborane	(9.4)	317
BF <sub>3</sub>	Boron trifluoride	$15.7 \pm 0.3$	365
BH	Boron monohydride	(9.77)	1385
BH <sub>3</sub>	Borane	$12.026 \pm 0.024$	1261
BI <sub>3</sub>	Boron triiodide	( $9.25 \pm 0.03$ )	964
BO <sub>2</sub>	Boron dioxide	( $13.5 \pm 0.3$ )	1001
B <sub>2</sub> H <sub>6</sub>	Diborane	$11.38 \pm 0.05$	1134
B <sub>2</sub> O <sub>3</sub>	Boron oxide	$13.5 \pm 0.15$	460
B <sub>4</sub> H <sub>10</sub>	Tetraborane	$10.76 \pm 0.04$	1105
B <sub>5</sub> H <sub>9</sub>	Pentaborane(9)	$9.90 \pm 0.04$	1028
B <sub>6</sub> H <sub>10</sub>	Hexaborane	(9.0)	965
Ba	Barium	5.21170	683
BaO	Barium oxide	$6.91 \pm 0.06$	543
Be	Beryllium	9.32263	1224
BeO	Beryllium oxide	( $10.1 \pm 0.4$ )	1111
Bi	Bismuth	7.2855	908
BiCl <sub>3</sub>	Bismuth trichloride	(10.4)	736

Mol. form.	Name	IP/eV	$\Delta_f H_{\text{ion}}$ kJ/mol
Bk	Berkelium	6.23	911
Br	Bromine (atomic)	11.81381	1252
BrCl	Bromine chloride	11.01	1079
BrF	Bromine fluoride	11.86	1086
BrF <sub>5</sub>	Bromine pentafluoride	13.172 ± 0.002	840
BrH	Hydrogen bromide	11.66 ± 0.03	1087
BrH <sub>3</sub> Si	Bromosilane	10.6	943
BrI	Iodine bromide	9.790 ± 0.004	986
BrK	Potassium bromide	7.85 ± 0.1	578
BrLi	Lithium bromide	(8.7)	685
BrNO	Nitrosyl bromide	10.17 ± 0.03	1065
BrNa	Sodium bromide	8.31 ± 0.1	660
BrO	Bromine monoxide	10.46 ± 0.02	1135
BrRb	Rubidium bromide	7.94 ± 0.03	583
BrTl	Thallium(I) bromide	9.14 ± 0.02	844
Br <sub>2</sub>	Bromine	10.516 ± 0.005	1046
Br <sub>2</sub> Hg	Mercury(II) bromide	10.560 ± 0.003	935
Br <sub>2</sub> Sn	Tin(II) bromide	9.0	839
Br <sub>3</sub> Ga	Gallium(III) bromide	10.40	711
Br <sub>3</sub> P	Phosphorus(III) bromide	9.7	798
Br <sub>4</sub> Hf	Hafnium(IV) bromide	(10.9)	366
Br <sub>4</sub> Sn	Tin(IV) bromide	10.6	709
Br <sub>4</sub> Ti	Titanium(IV) bromide	10.3	375
Br <sub>4</sub> Zr	Zirconium(IV) bromide	(10.7)	388
Ca	Calcium	6.11316	768
CaCl	Calcium monochloride	5.86 ± 0.07	462
CaO	Calcium oxide	6.66 ± 0.18	668
Cd	Cadmium	8.99367	980
Ce	Cerium	5.5387	957
Cf	Californium	6.30	805
Cl	Chlorine (atomic)	12.96764	1373
ClCs	Cesium chloride	(7.84 ± 0.05)	510
ClF	Chlorine fluoride	12.66 ± 0.01	1171
ClFO <sub>3</sub>	Perchloryl fluoride	(12.945 ± 0.005)	1224
ClF <sub>2</sub>	Chlorine difluoride	(12.77 ± 0.05)	1128
ClF <sub>3</sub>	Chlorine trifluoride	(12.65 ± 0.05)	1057
ClF <sub>5</sub> S	Sulfur chloride pentafluoride	(12.335 ± 0.005)	144
ClH	Hydrogen chloride	12.749 ± 0.009	1137
ClHO	Hypochlorous acid	(11.12 ± 0.01)	993
ClH <sub>3</sub> Si	Chlorosilane	11.4	899
ClI	Iodine chloride	10.088 ± 0.01	991
ClIn	Indium(I) chloride	(9.51)	842
ClK	Potassium chloride	(8.0 ± 0.4)	557
CLi	Lithium chloride	9.57	727
ClNO	Nitrosyl chloride	10.87 ± 0.01	1099
ClNO <sub>2</sub>	Nitryl chloride	(11.84)	1155
ClNa	Sodium chloride	8.92 ± 0.06	681
ClO	Chlorine monoxide	10.95	1159
ClO <sub>2</sub>	Chlorine dioxide	10.33 ± 0.02	1093
ClRb	Rubidium chloride	(8.50 ± 0.03)	590
ClTl	Thallium(I) chloride	9.70 ± 0.03	869
Cl <sub>2</sub>	Chlorine	11.480 ± 0.005	1108
Cl <sub>2</sub> CrO <sub>2</sub>	Chromyl chloride	11.6	580
Cl <sub>2</sub> Ge	Germanium(II) chloride	(10.20 ± 0.05)	813
Cl <sub>2</sub> H <sub>2</sub> Si	Dichlorosilane	11.4	765
Cl <sub>2</sub> Hg	Mercury(II) chloride	11.380 ± 0.003	952
Cl <sub>2</sub> O	Chlorine oxide	10.94	1135
Cl <sub>2</sub> OS	Thionyl chloride	10.96	844
Cl <sub>2</sub> O <sub>2</sub> S	Sulfuryl chloride	12.05	807
Cl <sub>2</sub> Pb	Lead(II) chloride	(10.2)	791

Mol. form.	Name	IP/eV	$\Delta_f H_{\text{ion}}$ kJ/mol
Cl <sub>2</sub> S	Sulfur dichloride	9.45 ± 0.03	895
Cl <sub>2</sub> Si	Dichlorosilylene	(10.93 ± 0.10)	887
Cl <sub>2</sub> Sn	Tin(II) chloride	(10.0)	760
Cl <sub>3</sub> Ga	Gallium(III) chloride	11.52	664
Cl <sub>3</sub> HSi	Trichlorosilane	(11.7)	648
Cl <sub>3</sub> N	Nitrogen trichloride	(10.12 ± 0.1)	1244
Cl <sub>3</sub> OP	Phosphorus(V) oxychloride	11.36 ± 0.02	540
Cl <sub>3</sub> OV	Vanadyl trichloride	(11.6)	425
Cl <sub>3</sub> P	Phosphorus(III) chloride	9.91	668
Cl <sub>3</sub> PS	Phosphorus(V) sulfide trichloride	9.71 ± 0.03	573
Cl <sub>3</sub> Sb	Antimony(III) chloride	(≤ 10.7)	s719
Cl <sub>4</sub> Ge	Germanium(IV) chloride	11.68 ± 0.05	629
Cl <sub>4</sub> Hf	Hafnium(IV) chloride	(11.7)	246
Cl <sub>4</sub> Si	Tetrachlorosilane	11.79 ± 0.01	527
Cl <sub>4</sub> Sn	Tin(IV) chloride	11.7 ± 0.2	656
Cl <sub>4</sub> Ti	Titanium(IV) chloride	(11.5)	349
Cl <sub>4</sub> V	Vanadium(IV) chloride	(9.2)	361
Cl <sub>4</sub> Zr	Zirconium(IV) chloride	(11.2)	210
Cl <sub>5</sub> Mo	Molybdenum(V) chloride	(8.7)	392
Cl <sub>5</sub> Nb	Niobium(V) chloride	(10.97)	356
Cl <sub>5</sub> P	Phosphorus(V) chloride	(10.2)	608
Cl <sub>5</sub> Ta	Tantalum(V) chloride	(11.08)	303
Cl <sub>6</sub> W	Tungsten(VI) chloride	(9.5)	348
Cm	Curium	6.02	966
Co	Cobalt	7.8810	1187
Cr	Chromium	6.76664	1050
Cs	Cesium	3.89390	452
CsF	Cesium fluoride	(8.80 ± 0.10)	489
CsNa	Cesium sodium	(4.05 ± 0.04)	535
Cu	Copper	7.72638	1084
CuF	Copper(I) fluoride	10.15 ± 0.02	984
Dy	Dysprosium	5.9389	862
Er	Erbium	6.1078	907
Es	Einsteinium	6.42	753
Eu	Europium	5.6704	723
F	Fluorine (atomic)	17.42282	1761
FGa	Gallium monofluoride	(9.6 ± 0.5)	700
FH	Hydrogen fluoride	16.044 ± 0.003	1276
FHO	Hypofluorous acid	12.71 ± 0.01	1130
FH <sub>3</sub> Si	Fluorosilane	11.7	752
FI	Iodine fluoride	10.54 ± 0.01	922
FIn	Indium monofluoride	(9.6 ± 0.5)	740
FNO	Nitrosyl fluoride	12.63 ± 0.03	1152
FNO <sub>2</sub>	Nitryl fluoride	(13.09)	1154
FNS	Thionitrosyl fluoride (NSF)	11.51 ± 0.04	1090
FO	Fluorine monoxide	12.78 ± 0.03	1342
FO <sub>2</sub>	Fluorine superoxide (FOO)	(12.6 ± 0.2)	1228
FS	Sulfur fluoride	10.09	986
FTl	Thallium(I) fluoride	10.52	835
F <sub>2</sub>	Fluorine	15.697 ± 0.003	1515
F <sub>2</sub> Ge	Germanium(II) fluoride	(≤ 11.65)	551
F <sub>2</sub> HN	Difluoramine	(11.53 ± 0.08)	1046
F <sub>2</sub> H <sub>2</sub> Si	Difluorosilane	(12.2)	386
F <sub>2</sub> Mg	Magnesium fluoride	(13.6 ± 0.3)	588
F <sub>2</sub> N	Difluoroamidogen	11.628 ± 0.01	1155
F <sub>2</sub> N <sub>2</sub>	<i>trans</i> -Difluorodiazine	(12.8)	1315
F <sub>2</sub> O	Fluorine monoxide	13.11 ± 0.01	1290
F <sub>2</sub> OS	Thionyl fluoride	12.25	688
F <sub>2</sub> O <sub>2</sub> S	Sulfuryl fluoride	13.04 ± 0.01	501
F <sub>2</sub> Pb	Lead(II) fluoride	(11.5)	679

Mol. form.	Name	IP/eV	$\Delta_r H_{\text{ion}}$ kJ/mol
F <sub>2</sub> S	Sulfur difluoride	(10.08)	676
F <sub>2</sub> Si	Difluorosilylene	10.78 ± 0.05	450
F <sub>2</sub> Sn	Tin(II) fluoride	(11.1)	586
F <sub>2</sub> Xe	Xenon difluoride	12.35 ± 0.01	1083
F <sub>3</sub> HSi	Trifluorosilane	(14.0)	150
F <sub>3</sub> N	Nitrogen trifluoride	13.00 ± 0.02	1125
F <sub>3</sub> NO	Trifluoramine oxide	13.31 ± 0.06	1121
F <sub>3</sub> OP	Phosphorus(V) oxyfluoride	12.76 ± 0.01	-24
F <sub>3</sub> P	Phosphorus(III) fluoride	11.60 ± 0.05	161
F <sub>3</sub> PS	Phosphorus(V) sulfide trifluoride	≤ 11.05 ± 0.035	≤ 58
F <sub>3</sub> Si	Trifluorosilyl	(9.99)	-32
F <sub>4</sub> Ge	Germanium(IV) fluoride	(15.5)	307
F <sub>4</sub> N <sub>2</sub>	Tetrafluorohydrazine	11.94 ± 0.03	1119
F <sub>4</sub> S	Sulfur tetrafluoride	12.0 ± 0.3	399
F <sub>4</sub> Si	Tetrafluorosilane	15.24 ± 0.14	-144
F <sub>4</sub> Xe	Xenon tetrafluoride	12.65 ± 0.1	1016
F <sub>5</sub> I	Iodine pentafluoride	12.943 ± 0.005	408
F <sub>5</sub> P	Phosphorus(V) fluoride	(15.1)	-137
F <sub>5</sub> S	Sulfur pentafluoride	9.60 ± 0.05	10
F <sub>6</sub> Mo	Molybdenum(VI) fluoride	(14.5 ± 0.1)	-159
F <sub>6</sub> S	Sulfur hexafluoride	15.32 ± 0.02	258
F <sub>6</sub> U	Uranium(VI) fluoride	14.00 ± 0.10	-796
Fe	Iron	7.9024	1177
Fm	Fermium	6.50	627
Ga	Gallium	5.99930	851
GaI <sub>3</sub>	Gallium(III) iodide	9.40	765
Gd	Gadolinium	6.1500	991
Ge	Germanium	7.900	1139
GeH <sub>4</sub>	Germane	≤ 10.53	≤ 1108
GeI <sub>4</sub>	Germanium(IV) iodide	(9.42)	850
GeO	Germanium(II) oxide	11.25 ± 0.01	1044
GeS	Germanium(II) sulfide	(9.98)	1055
H	Hydrogen (atomic)	13.59844	1530
HI	Hydrogen iodide	10.386 ± 0.001	1028
HLi	Lithium hydride	7.7	882
HN	Imidogen	≤ 13.49 ± 0.01	1678
HNO	Nitrosyl hydride	(10.1)	1075
HNO <sub>2</sub>	Nitrous acid	≤ 11.3	≤ 1011
HNO <sub>3</sub>	Nitric acid	11.95 ± 0.01	1019
HN <sub>3</sub>	Hydrazoic acid	10.72 ± 0.025	1328
HO	Hydroxyl	13.0170 ± 0.0002	1294
HO <sub>2</sub>	Hydroperoxy	11.35 ± 0.01	1106
HS	Mercapto	10.4219 ± 0.0004	1145
H <sub>2</sub>	Hydrogen	15.42593 ± 0.00005	1488
H <sub>2</sub> N	Amidogen	11.14 ± 0.01	1264
H <sub>2</sub> O	Water	12.6206 ± 0.0020	976
H <sub>2</sub> O <sub>2</sub>	Hydrogen peroxide	10.58 ± 0.04	885
H <sub>2</sub> S	Hydrogen sulfide	10.457 ± 0.012	989
H <sub>2</sub> Se	Hydrogen selenide	9.892 ± 0.005	984
H <sub>2</sub> Si	Silylene	8.244 ± 0.025	1084
H <sub>3</sub> N	Ammonia	10.070 ± 0.020	925
H <sub>3</sub> NO	Hydroxylamine	(10.00)	923
H <sub>3</sub> P	Phosphine	9.869 ± 0.002	958
H <sub>3</sub> Sb	Stibine	9.54 ± 0.03	1067
H <sub>4</sub> N <sub>2</sub>	Hydrazine	8.1 ± 0.15	877
H <sub>4</sub> Si	Silane	11.00 ± 0.02	1095
H <sub>4</sub> Sn	Stannane	(10.75)	1200
H <sub>6</sub> Si <sub>2</sub>	Disilane	9.74 ± 0.02	1019
H <sub>8</sub> Si <sub>3</sub>	Trisilane	(9.2)	1009
He	Helium	24.58741	2372

Mol. form.	Name	IP/eV	$\Delta_f H_{\text{ion}}$ kJ/mol
Hf	Hafnium	6.82507 ± 0.00004	1278
Hg	Mercury	10.43750	1069
HgI <sub>2</sub>	Mercury(II) iodide	9.5088 ± 0.0022	900
Ho	Holmium	6.0216	882
I	Iodine (atomic)	10.45126	1115
IK	Potassium iodide	(7.21 ± 0.3)	570
ILi	Lithium iodide	(7.5)	633
INa	Sodium iodide	7.64 ± 0.02	659
ITl	Thallium(I) iodide	8.47 ± 0.02	826
I <sub>2</sub>	Iodine	9.3074 ± 0.0002	960
I <sub>4</sub> Ti	Titanium(IV) iodide	(9.1)	602
I <sub>4</sub> Zr	Zirconium(IV) iodide	(9.3)	500
In	Indium	5.78636	802
Ir	Iridium	9.1	1543
K	Potassium	4.34066	508
KLi	Lithium potassium	4.57 ± 0.04	512
KNa	Potassium sodium	4.41636 ± 0.00017	561
K <sub>2</sub>	Dipotassium	4.0637 ± 0.0002	519
Kr	Krypton	13.99961	1351
La	Lanthanum	5.5770	969
Li	Lithium	5.39172	680
LiNa	Lithium sodium	5.05 ± 0.04	571
LiO	Lithium monoxide	(8.44)	894
LiRb	Lithium rubidium	4.3 ± 0.1	486
Li <sub>2</sub>	Dilithium	5.1127 ± 0.0003	709
Lu	Lutetium	5.42585	950
Md	Mendelevium	6.58	635
Mg	Magnesium	7.64624	885
MgO	Magnesium oxide	(8.76 ± 0.22)	901
Mn	Manganese	7.43402	998
Mo	Molybdenum	7.09243	1343
N	Nitrogen (atomic)	14.53414	1875
NO	Nitric oxide	9.26438 ± 0.00005	985
NO <sub>2</sub>	Nitrogen dioxide	9.586 ± 0.002	958
NP	Phosphorus nitride	11.84 ± 0.04	1247
NS	Nitrogen sulfide	8.87 ± 0.01	1119
N <sub>2</sub>	Nitrogen	15.5808	1503
N <sub>2</sub> O	Nitrous oxide	12.886	1325
N <sub>2</sub> O <sub>4</sub>	Nitrogen tetroxide	(10.8)	1050
N <sub>2</sub> O <sub>5</sub>	Nitrogen pentoxide	(11.9)	1161
Na	Sodium	5.13908	603
NaRb	Rubidium sodium	4.32 ± 0.04	480
Na <sub>2</sub>	Disodium	4.894 ± 0.003	614
Nb	Niobium	6.75885	1384
Nd	Neodymium	5.5250	859
Ne	Neon	21.56454	2081
Ni	Nickel	7.6398	1167
No	Nobelium	6.65	642
Np	Neptunium	6.2657 ± 0.0003	1069
O	Oxygen (atomic)	13.61806	1563
OPb	Lead(II) oxide	9.08 ± 0.10	939
OS	Sulfur monoxide	10.294 ± 0.004	998
OS <sub>2</sub>	Sulfur oxide (SSO)	10.584 ± 0.005	971
OSi	Silicon monoxide	11.49 ± 0.20	1008
OSn	Tin(II) oxide	9.60 ± 0.02	944
OSr	Strontium oxide	6.6 ± 0.2	623
O <sub>2</sub>	Oxygen	12.0697 ± 0.0002	1165
O <sub>2</sub> S	Sulfur dioxide	12.349 ± 0.001	894
O <sub>2</sub> Th	Thorium(IV) oxide	(8.7 ± 0.15)	342
O <sub>2</sub> Ti	Titanium(IV) oxide	(9.54 ± 0.1)	623

Mol. form.	Name	IP/eV	$\Delta_r H_{\text{ion}}$ kJ/mol
O <sub>2</sub> U	Uranium(IV) oxide	(5.4 ± 0.1)	57
O <sub>3</sub>	Ozone	12.43	1342
O <sub>3</sub> S	Sulfur trioxide	12.82 ± 0.03	841
O <sub>3</sub> U	Uranium(VI) oxide	(10.5 ± 0.5)	214
O <sub>4</sub> Os	Osmium(VIII) oxide	(12.32)	850
O <sub>4</sub> Ru	Ruthenium(VIII) oxide	12.15 ± 0.03	988
O <sub>7</sub> Re <sub>2</sub>	Rhenium(VII) oxide	(12.7 ± 0.2)	125
Os	Osmium	8.7	1630
P	Phosphorus	10.48669	1328
P <sub>2</sub>	Diphosphorus	10.53	1160
Pa	Protactinium	5.89	1133
Pb	Lead	7.41666	911
PbS	Lead(II) sulfide	(8.5 ± 0.5)	954
Pd	Palladium	8.3367	1181
Pm	Promethium	5.55	536
Pr	Praseodymium	5.464	883
Pt	Platinum	9.0	1433
Pu	Plutonium	6.025	926
Ra	Radium	5.27892	668
Rb	Rubidium	4.17713	484
Re	Rhenium	7.88	1530
Rh	Rhodium	7.45890	1276
Rn	Radon	10.74850	1037
Ru	Ruthenium	7.36050	1355
S	Sulfur	10.36001	1277
SSn	Tin(II) sulfide	(8.8)	966
S <sub>2</sub>	Disulfur	9.356 ± 0.002	1031
Sb	Antimony	8.64	1096
Sc	Scandium	6.56144	1010
Se	Selenium	9.75238	1168
Si	Silicon	8.15169	1238
Sm	Samarium	5.6437	751
Sn	Tin	7.34381	1011
Sr	Strontium	5.69484	713
Ta	Tantalum	7.89	1544
Tb	Terbium	5.8639	955
Tc	Technetium	7.28	1380
Te	Tellurium	9.0096	1066
Th	Thorium	6.308 ± 0.003	1207
Ti	Titanium	6.8282	1127
Tl	Thallium	6.10829	771
Tm	Thulium	6.18431	827
U	Uranium	6.19405	1129
V	Vanadium	6.746 ± 0.002	1166
W	Tungsten	7.98	1621
Xe	Xenon	12.12987	1170
Y	Yttrium	6.217	1022
Yb	Ytterbium	6.25416	754
Zn	Zinc	9.39405	1037
Zr	Zirconium	6.63390	1251
<i>Substances containing carbon</i>			
C	Carbon	11.26030	1803
CBrClF <sub>2</sub>	Bromochlorodifluoromethane	(11.21)	642
CBrCl <sub>3</sub>	Bromotrichloromethane	(10.6)	980
CBrF <sub>3</sub>	Bromotrifluoromethane	(11.40)	451
CBr <sub>2</sub> F <sub>2</sub>	Dibromodifluoromethane	11.03 ± 0.04	683
CBr <sub>4</sub>	Tetrabromomethane	(10.31 ± 0.02)	1079
CCL	Chloromethylidyne	(8.9 ± 0.2)	1244
CClF <sub>3</sub>	Chlorotrifluoromethane	12.6 ± 0.2	505

Mol. form.	Name	IP/eV	$\Delta_r H_{\text{ion}}$ kJ/mol
CClN	Cyanogen chloride	12.34 ± 0.01	1329
CCl <sub>2</sub>	Dichloromethylene	(9.27)	1058
CCl <sub>2</sub> F <sub>2</sub>	Dichlorodifluoromethane	12.05 ± 0.24	685
CCl <sub>2</sub> O	Carbonyl chloride	(11.5)	888
CCl <sub>3</sub> F	Trichlorofluoromethane	11.77 ± 0.02	868
CCl <sub>4</sub>	Tetrachloromethane	11.47 ± 0.01	1010
CF	Fluoromethylidyne	9.11 ± 0.01	1134
CFN	Cyanogen fluoride	13.34 ± 0.02	1325
CF <sub>2</sub>	Difluoromethylene	11.44 ± 0.03	899
CF <sub>2</sub> O	Carbonyl fluoride	13.035 ± 0.030	617
CF <sub>3</sub>	Trifluoromethyl	8.7 ± 0.2	379
CF <sub>3</sub> I	Trifluoroiodomethane	10.23	397
CH	Methylidyne	10.64 ± 0.01	1622
CHBrCl <sub>2</sub>	Bromodichloromethane	10.6	973
CHBr <sub>2</sub> Cl	Chlorodibromomethane	10.59 ± 0.01	1030
CHBr <sub>3</sub>	Tribromomethane	10.48 ± 0.02	1035
CHCl	Chloromethylene	9.84	1247
CHClF <sub>2</sub>	Chlorodifluoromethane	(12.2)	693
CHCl <sub>2</sub> F	Dichlorofluoromethane	(11.5)	829
CHCl <sub>3</sub>	Trichloromethane	11.37 ± 0.02	992
CHF	Fluoromethylene	10.06 ± 0.05	1121
CHF <sub>3</sub>	Trifluoromethane	(13.86)	643
CHI <sub>3</sub>	Triiodomethane	9.25 ± 0.02	1010
CHN	Hydrogen cyanide	13.60 ± 0.01	1447
CHN	Hydrogen isocyanide	(12.5 ± 0.1)	1407
CHNO	Isocyanic acid	11.595 ± 0.005	1016
CHNO	Fulminic acid	(10.83)	1263
CHO	Oxomethyl (HCO)	(8.55)	826
CH <sub>2</sub>	Methylene	10.396 ± 0.003	1392
CH <sub>2</sub> BrCl	Bromochloromethane	10.77 ± 0.01	1085
CH <sub>2</sub> Br <sub>2</sub>	Dibromomethane	(10.50 ± 0.02)	1013
CH <sub>2</sub> ClF	Chlorofluoromethane	11.71 ± 0.01	870
CH <sub>2</sub> Cl <sub>2</sub>	Dichloromethane	11.32 ± .01	996
CH <sub>2</sub> F <sub>2</sub>	Difluoromethane	12.71	774
CH <sub>2</sub> I <sub>2</sub>	Diiodomethane	9.46 ± 0.02	1030
CH <sub>2</sub> N <sub>2</sub>	Diazomethane	8.999 ± 0.001	1098
CH <sub>2</sub> N <sub>2</sub>	Cyanamide	(10.4)	1137
CH <sub>2</sub> O	Formaldehyde	10.88 ± 0.01	941
CH <sub>2</sub> O <sub>2</sub>	Formic acid	11.33 ± 0.01	715
CH <sub>3</sub>	Methyl	9.843 ± 0.002	1095
CH <sub>3</sub> BO	Borane carbonyl	11.14 ± 0.02	962
CH <sub>3</sub> Br	Bromomethane	10.541 ± 0.003	979
CH <sub>3</sub> Cl	Chloromethane	11.22 ± 0.01	1001
CH <sub>3</sub> Cl <sub>3</sub> Si	Methyltrichlorosilane	(11.36 ± 0.03)	548
CH <sub>3</sub> F	Fluoromethane	12.47 ± 0.02	956
CH <sub>3</sub> I	Iodomethane	9.538	936
CH <sub>3</sub> NO	Formamide	10.16 ± 0.06	796
CH <sub>3</sub> NO <sub>2</sub>	Nitromethane	11.08 ± 0.07	994
CH <sub>3</sub> N <sub>3</sub>	Methyl azide	9.81 ± 0.02	1227
CH <sub>3</sub> O	Methoxy	(10.72)	1050
CH <sub>4</sub>	Methane	12.61 ± 0.01	1143
CH <sub>4</sub> N <sub>2</sub> O	Urea	9.7	690
CH <sub>4</sub> O	Methanol	10.85 ± 0.01	845
CH <sub>4</sub> S	Methanethiol	9.44 ± 0.005	888
CH <sub>5</sub> N	Methylamine	(8.80)	826
CH <sub>6</sub> N <sub>2</sub>	Methylhydrazine	7.7 ± 0.15	835
CH <sub>6</sub> Si	Methylsilane	(10.7)	1003
CN	Cyanide	13.5984	1748
CNO	Cyanate	11.76 ± 0.01	1290
CO	Carbon monoxide	14.014 ± 0.0003	1242

Mol. form.	Name	IP/eV	$\Delta_i H_{\text{ion}}$ kJ/mol
COS	Carbon oxysulfide	11.18 ± 0.01	936
COSe	Carbon oxyselenide	10.36 ± 0.01	929
CO <sub>2</sub>	Carbon dioxide	13.773 ± 0.002	935
CS	Carbon sulfide	11.33 ± 0.01	1361
CS <sub>2</sub>	Carbon disulfide	10.0685 ± 0.0020	1089
C <sub>2</sub>	Dicarbon	(11.4 ± 0.3)	2000
C <sub>2</sub> Br <sub>2</sub> F <sub>4</sub>	1,2-Dibromotetrafluoroethane	(11.1)	280
C <sub>2</sub> ClF <sub>3</sub>	Chlorotrifluoroethylene	9.81 ± 0.03	373
C <sub>2</sub> ClF <sub>5</sub>	Chloropentafluoroethane	(12.6)	99
C <sub>2</sub> Cl <sub>2</sub>	Dichloroacetylene	9.9	1165
C <sub>2</sub> Cl <sub>2</sub> F <sub>4</sub>	1,2-Dichlorotetrafluoroethane	12.2	252
C <sub>2</sub> Cl <sub>3</sub> F <sub>3</sub>	1,1,1-Trichlorotrifluoroethane	11.5	386
C <sub>2</sub> Cl <sub>3</sub> F <sub>3</sub>	1,1,2-Trichlorotrifluoroethane	11.99 ± 0.02	429
C <sub>2</sub> Cl <sub>4</sub>	Tetrachloroethylene	9.326 ± 0.001	887
C <sub>2</sub> Cl <sub>4</sub> F <sub>2</sub>	1,1,2,2-Tetrachloro-1,2-difluoroethane	(11.3)	563
C <sub>2</sub> Cl <sub>4</sub> O	Trichloroacetyl chloride	(11.0)	827
C <sub>2</sub> Cl <sub>6</sub>	Hexachloroethane	(11.1)	920
C <sub>2</sub> F <sub>3</sub> N	Trifluoroacetonitrile	13.93 ± 0.07	845
C <sub>2</sub> F <sub>4</sub>	Tetrafluoroethylene	10.12 ± 0.02	315
C <sub>2</sub> F <sub>6</sub>	Hexafluoroethane	(13.6)	-30
C <sub>2</sub> H	Ethynyl	(11.61 ± 0.07)	1685
C <sub>2</sub> HBr	Bromoacetylene	10.31 ± 0.02	1242
C <sub>2</sub> HBrClF <sub>3</sub>	2-Bromo-2-chloro-1,1,1-trifluoroethane	(11.0)	363
C <sub>2</sub> HCl	Chloroacetylene	10.58 ± 0.02	1276
C <sub>2</sub> HClF <sub>2</sub>	1-Chloro-2,2-difluoroethylene	9.80 ± 0.04	628
C <sub>2</sub> HCl <sub>3</sub>	Trichloroethylene	9.46 ± 0.02	894
C <sub>2</sub> HCl <sub>3</sub> O	Dichloroacetyl chloride	(10.9)	809
C <sub>2</sub> HCl <sub>5</sub>	Pentachloroethane	(11.0)	919
C <sub>2</sub> HF	Fluoroacetylene	11.26	1195
C <sub>2</sub> HF <sub>3</sub>	Trifluoroethylene	10.14	489
C <sub>2</sub> HF <sub>3</sub> O <sub>2</sub>	Trifluoroacetic acid	11.46	75
C <sub>2</sub> H <sub>2</sub>	Acetylene	11.400 ± 0.002	1328
C <sub>2</sub> H <sub>2</sub> Cl <sub>2</sub>	1,1-Dichloroethylene	9.81 ± 0.04	949
C <sub>2</sub> H <sub>2</sub> Cl <sub>2</sub>	<i>cis</i> -1,2-Dichloroethylene	9.66 ± 0.01	936
C <sub>2</sub> H <sub>2</sub> Cl <sub>2</sub>	<i>trans</i> -1,2-Dichloroethylene	9.64 ± 0.02	934
C <sub>2</sub> H <sub>2</sub> Cl <sub>2</sub> O	Chloroacetyl chloride	(≤ 10.3)	815
C <sub>2</sub> H <sub>2</sub> Cl <sub>4</sub>	1,1,1,2-Tetrachloroethane	(11.1)	920
C <sub>2</sub> H <sub>2</sub> Cl <sub>4</sub>	1,1,2,2-Tetrachloroethane	(≤ 11.62)	≤ 971
C <sub>2</sub> H <sub>2</sub> F <sub>2</sub>	1,1-Difluoroethylene	10.29 ± 0.01	650
C <sub>2</sub> H <sub>2</sub> F <sub>2</sub>	<i>cis</i> -1,2-Difluoroethylene	10.23 ± 0.02	690
C <sub>2</sub> H <sub>2</sub> O	Ketene	9.617 ± 0.003	880
C <sub>2</sub> H <sub>2</sub> O <sub>2</sub>	Glyoxal	10.2	773
C <sub>2</sub> H <sub>2</sub> S <sub>2</sub>	Thiirene	8.61	892
C <sub>2</sub> H <sub>3</sub> Br	Bromoethylene	9.83 ± 0.02	1028
C <sub>2</sub> H <sub>3</sub> Cl	Chloroethylene	9.99 ± 0.02	985
C <sub>2</sub> H <sub>3</sub> ClF <sub>2</sub>	1-Chloro-1,1-difluoroethane	(11.98)	626
C <sub>2</sub> H <sub>3</sub> ClO	Acetyl chloride	10.82 ± 0.04	801
C <sub>2</sub> H <sub>3</sub> ClO	Chloroacetaldehyde	(10.48)	815
C <sub>2</sub> H <sub>3</sub> ClO <sub>2</sub>	Chloroacetic acid	(10.7)	597
C <sub>2</sub> H <sub>3</sub> Cl <sub>3</sub>	1,1,1-Trichloroethane	(11.0)	917
C <sub>2</sub> H <sub>3</sub> Cl <sub>3</sub>	1,1,2-Trichloroethane	(11.0)	911
C <sub>2</sub> H <sub>3</sub> F	Fluoroethylene	10.36 ± 0.01	861
C <sub>2</sub> H <sub>3</sub> FO	Acetyl fluoride	(11.5)	667
C <sub>2</sub> H <sub>3</sub> F <sub>3</sub>	1,1,1-Trifluoroethane	13.3 ± 0.5	536
C <sub>2</sub> H <sub>3</sub> N	Acetonitrile	12.20 ± 0.01	1253
C <sub>2</sub> H <sub>3</sub> NO	Methylisocyanate	(10.67)	900
C <sub>2</sub> H <sub>4</sub>	Ethylene	10.5138 ± 0.0006	1067
C <sub>2</sub> H <sub>4</sub> Br <sub>2</sub>	1,2-Dibromoethane	10.35 ± 0.04	961
C <sub>2</sub> H <sub>4</sub> Cl <sub>2</sub>	1,1-Dichloroethane	11.04 ± 0.02	935
C <sub>2</sub> H <sub>4</sub> Cl <sub>2</sub>	1,2-Dichloroethane	11.04 ± 0.02	931



Mol. form.	Name	IP/eV	$\Delta_f H_{\text{ion}}$ kJ/mol
C <sub>2</sub> H <sub>4</sub> F <sub>2</sub>	1,1-Difluoroethane	(11.87)	643
C <sub>2</sub> H <sub>4</sub> O	Acetaldehyde	10.229 ± 0.0007	821
C <sub>2</sub> H <sub>4</sub> O	Ethylene oxide	10.56 ± 0.01	966
C <sub>2</sub> H <sub>4</sub> O <sub>2</sub>	Acetic acid	10.65 ± 0.02	595
C <sub>2</sub> H <sub>4</sub> O <sub>2</sub>	Methyl formate	10.835 ± 0.005	690
C <sub>2</sub> H <sub>5</sub> Br	Bromoethane	10.29 ± 0.01	931
C <sub>2</sub> H <sub>5</sub> Cl	Chloroethane	10.98 ± 0.02	947
C <sub>2</sub> H <sub>5</sub> ClO	2-Chloroethanol	(10.5)	756
C <sub>2</sub> H <sub>5</sub> F	Fluoroethane	(11.78)	873
C <sub>2</sub> H <sub>5</sub> I	Iodoethane	9.3492 ± 0.0006	893
C <sub>2</sub> H <sub>5</sub> N	Ethyleneimine	(9.5 ± 0.3)	1044
C <sub>2</sub> H <sub>5</sub> NO	Acetamide	9.65 ± 0.03	693
C <sub>2</sub> H <sub>5</sub> NO	<i>N</i> -Methylformamide	9.83 ± 0.04	760
C <sub>2</sub> H <sub>5</sub> NO <sub>2</sub>	Nitroethane	10.88 ± 0.05	948
C <sub>2</sub> H <sub>6</sub>	Ethane	11.56 ± 0.02	1031
C <sub>2</sub> H <sub>6</sub> Cl <sub>2</sub> Si	Dichlorodimethylsilane	(10.7)	576
C <sub>2</sub> H <sub>6</sub> O	Ethanol	10.43 ± 0.05	772
C <sub>2</sub> H <sub>6</sub> O	Dimethyl ether	10.025 ± 0.025	783
C <sub>2</sub> H <sub>6</sub> OS	Dimethyl sulfoxide	9.10 ± 0.03	727
C <sub>2</sub> H <sub>6</sub> O <sub>2</sub>	Ethylene glycol	10.16	593
C <sub>2</sub> H <sub>6</sub> S	Ethanethiol	9.31 ± 0.03	851
C <sub>2</sub> H <sub>6</sub> S	Dimethyl sulfide	8.69 ± 0.02	801
C <sub>2</sub> H <sub>6</sub> S <sub>2</sub>	Dimethyl disulfide	(7.4 ± 0.3)	690
C <sub>2</sub> H <sub>7</sub> N	Ethylamine	8.86 ± 0.02	808
C <sub>2</sub> H <sub>7</sub> N	Dimethylamine	8.24 ± 0.08	777
C <sub>2</sub> H <sub>7</sub> NO	Ethanolamine	8.96	664
C <sub>2</sub> H <sub>8</sub> N <sub>2</sub>	1,2-Ethanediamine	(8.6)	812
C <sub>2</sub> H <sub>8</sub> N <sub>2</sub>	1,1-Dimethylhydrazine	7.29 ± 0.05	787
C <sub>2</sub> N <sub>2</sub>	Cyanogen	13.37 ± 0.01	1597
C <sub>3</sub> F <sub>6</sub>	Perfluoropropene	10.60 ± 0.03	-103
C <sub>3</sub> F <sub>6</sub> O	Perfluoroacetone	(11.57 ± 0.13)	-282
C <sub>3</sub> F <sub>8</sub>	Perfluoropropane	(13.38)	-491
C <sub>3</sub> HN	Cyanoacetylene	11.64 ± 0.01	1475
C <sub>3</sub> H <sub>2</sub> O	2-Propynal	(10.7 ± 0.1)	1145
C <sub>3</sub> H <sub>3</sub> F <sub>3</sub>	3,3,3-Trifluoropropene	(10.9)	437
C <sub>3</sub> H <sub>3</sub> N	2-Propenenitrile	10.91 ± 0.01	1237
C <sub>3</sub> H <sub>3</sub> NO	Oxazole	(9.9)	940
C <sub>3</sub> H <sub>3</sub> NO	Isoxazole	(9.93)	1038
C <sub>3</sub> H <sub>4</sub>	Allene	9.692 ± 0.004	1126
C <sub>3</sub> H <sub>4</sub>	Propyne	10.37 ± 0.01	1187
C <sub>3</sub> H <sub>4</sub>	Cyclopropene	9.67 ± 0.01	1209
C <sub>3</sub> H <sub>4</sub> N <sub>2</sub>	Imidazole	(8.81)	997
C <sub>3</sub> H <sub>4</sub> O	Propargyl alcohol	10.49 ± 0.02	1060
C <sub>3</sub> H <sub>4</sub> O	Acrolein	10.103 ± 0.006	900
C <sub>3</sub> H <sub>4</sub> O	Cyclopropanone	(9.1 ± 0.1)	895
C <sub>3</sub> H <sub>4</sub> O <sub>2</sub>	Propenoic acid	10.60	701
C <sub>3</sub> H <sub>4</sub> O <sub>2</sub>	2-Oxetanone	(9.70 ± 0.01)	653
C <sub>3</sub> H <sub>5</sub> Br	3-Bromopropene	(9.96)	1008
C <sub>3</sub> H <sub>5</sub> Cl	3-Chloropropene	10.04 ± 0.01	965
C <sub>3</sub> H <sub>5</sub> ClO	Epichlorohydrin	(10.64)	919
C <sub>3</sub> H <sub>5</sub> ClO <sub>2</sub>	Methyl chloroacetate	(10.3)	575
C <sub>3</sub> H <sub>5</sub> F	3-Fluoropropene	(10.11)	821
C <sub>3</sub> H <sub>5</sub> N	Propanenitrile	11.84 ± 0.02	1194
C <sub>3</sub> H <sub>5</sub> NO	Acrylamide	(9.5)	720
C <sub>3</sub> H <sub>6</sub>	Propene	9.73 ± 0.02	959
C <sub>3</sub> H <sub>6</sub>	Cyclopropane	9.86	1005
C <sub>3</sub> H <sub>6</sub> Br <sub>2</sub>	1,2-Dibromopropane	(10.1)	903
C <sub>3</sub> H <sub>6</sub> Br <sub>2</sub>	1,3-Dibromopropane	(≤ 10.2)	≤ 919
C <sub>3</sub> H <sub>6</sub> Cl <sub>2</sub>	1,2-Dichloropropane	10.8 ± 0.1	886
C <sub>3</sub> H <sub>6</sub> Cl <sub>2</sub>	1,3-Dichloropropane	10.89 ± 0.04	892

Mol. form.	Name	IP/eV	$\Delta_f H_{\text{ion}}$ kJ/mol
C <sub>3</sub> H <sub>6</sub> O	Allyl alcohol	9.67 ± 0.05	808
C <sub>3</sub> H <sub>6</sub> O	Methyl vinyl ether	8.95 ± 0.01	763
C <sub>3</sub> H <sub>6</sub> O	Propanal	9.96 ± 0.01	772
C <sub>3</sub> H <sub>6</sub> O	Acetone	9.703 ± 0.006	719
C <sub>3</sub> H <sub>6</sub> O	Methyloxirane	(10.22)	892
C <sub>3</sub> H <sub>6</sub> O	Oxetane	9.65 ± 0.01	851
C <sub>3</sub> H <sub>6</sub> O <sub>2</sub>	Propanoic acid	10.525 ± 0.003	568
C <sub>3</sub> H <sub>6</sub> O <sub>2</sub>	Ethyl formate	10.61 ± 0.01	639
C <sub>3</sub> H <sub>6</sub> O <sub>2</sub>	Methyl acetate	10.25 ± 0.02	579
C <sub>3</sub> H <sub>6</sub> O <sub>2</sub>	1,3-Dioxolane	(9.9)	658
C <sub>3</sub> H <sub>6</sub> O <sub>3</sub>	1,3,5-Trioxane	(10.3)	528
C <sub>3</sub> H <sub>7</sub> Br	1-Bromopropane	10.18 ± 0.01	898
C <sub>3</sub> H <sub>7</sub> Br	2-Bromopropane	10.10 ± 0.03	877
C <sub>3</sub> H <sub>7</sub> Cl	1-Chloropropane	10.81 ± 0.01	911
C <sub>3</sub> H <sub>7</sub> Cl	2-Chloropropane	10.79 ± 0.02	896
C <sub>3</sub> H <sub>7</sub> F	1-Fluoropropane	(11.3)	806
C <sub>3</sub> H <sub>7</sub> F	2-Fluoropropane	(11.08)	776
C <sub>3</sub> H <sub>7</sub> I	1-Iodopropane	9.25 ± 0.01	860
C <sub>3</sub> H <sub>7</sub> I	2-Iodopropane	9.19 ± 0.02	845
C <sub>3</sub> H <sub>7</sub> N	Allylamine	(8.76)	891
C <sub>3</sub> H <sub>7</sub> N	Cyclopropylamine	(8.8)	926
C <sub>3</sub> H <sub>7</sub> N	Propyleneimine	(9.0)	960
C <sub>3</sub> H <sub>7</sub> NO	<i>N,N</i> -Dimethylformamide	(9.12)	688
C <sub>3</sub> H <sub>7</sub> NO <sub>2</sub>	1-Nitropropane	(10.81)	919
C <sub>3</sub> H <sub>7</sub> NO <sub>2</sub>	2-Nitropropane	(10.71)	894
C <sub>3</sub> H <sub>8</sub>	Propane	10.95 ± 0.05	952
C <sub>3</sub> H <sub>8</sub> O	1-Propanol	10.18 ± 0.06	727
C <sub>3</sub> H <sub>8</sub> O	2-Propanol	10.17 ± 0.02	709
C <sub>3</sub> H <sub>8</sub> O	Ethyl methyl ether	9.72 ± 0.07	722
C <sub>3</sub> H <sub>8</sub> O <sub>2</sub>	Dimethoxymethane	9.7	588
C <sub>3</sub> H <sub>8</sub> S	1-Propanethiol	9.20 ± 0.01	819
C <sub>3</sub> H <sub>8</sub> S	2-Propanethiol	9.145 ± 0.005	806
C <sub>3</sub> H <sub>8</sub> S	Ethyl methyl sulfide	(8.55)	765
C <sub>3</sub> H <sub>9</sub> BO <sub>3</sub>	Trimethyl borate	(10.0)	65
C <sub>3</sub> H <sub>9</sub> ClSi	Trimethylchlorosilane	(10.15)	624
C <sub>3</sub> H <sub>9</sub> N	Propylamine	(8.78)	777
C <sub>3</sub> H <sub>9</sub> N	Isopropylamine	(8.72)	758
C <sub>3</sub> H <sub>9</sub> N	Trimethylamine	7.82 ± 0.06	731
C <sub>3</sub> H <sub>9</sub> NO	3-Amino-1-propanol	(9.0)	651
C <sub>4</sub> H <sub>2</sub> O <sub>3</sub>	Maleic anhydride	(10.8)	645
C <sub>4</sub> H <sub>4</sub>	1-Buten-3-yne	9.58 ± 0.02	1230
C <sub>4</sub> H <sub>4</sub> N <sub>2</sub>	Succinonitrile	(12.1 ± 0.25)	1377
C <sub>4</sub> H <sub>4</sub> N <sub>2</sub>	Pyrimidine	9.23	1087
C <sub>4</sub> H <sub>4</sub> N <sub>2</sub>	Pyridazine	8.67 ± 0.03	1112
C <sub>4</sub> H <sub>4</sub> O	Furan	8.883 ± 0.003	822
C <sub>4</sub> H <sub>4</sub> O <sub>2</sub>	Diketene	(9.6 ± 0.02)	736
C <sub>4</sub> H <sub>4</sub> O <sub>3</sub>	Succinic anhydride	(10.6)	500
C <sub>4</sub> H <sub>4</sub> O <sub>4</sub>	Fumaric acid	(10.7)	355
C <sub>4</sub> H <sub>4</sub> S	Thiophene	8.86 ± 0.02	970
C <sub>4</sub> H <sub>5</sub> N	Methylacrylonitrile	10.34	1127
C <sub>4</sub> H <sub>5</sub> N	Pyrrole	8.207 ± 0.005	900
C <sub>4</sub> H <sub>5</sub> N	Cyclopropanecarbonitrile	(10.25)	1173
C <sub>4</sub> H <sub>6</sub>	1,2-Butadiene	(9.03)	1034
C <sub>4</sub> H <sub>6</sub>	1,3-Butadiene	9.082 ± 0.004	986
C <sub>4</sub> H <sub>6</sub>	1-Butyne	10.19 ± 0.02	1148
C <sub>4</sub> H <sub>6</sub>	2-Butyne	9.59 ± 0.03	1071
C <sub>4</sub> H <sub>6</sub>	Cyclobutene	9.43 ± 0.02	1067
C <sub>4</sub> H <sub>6</sub> O	Divinyl ether	(8.7)	827
C <sub>4</sub> H <sub>6</sub> O	<i>trans</i> -2-Butenal	9.73 ± 0.01	835
C <sub>4</sub> H <sub>6</sub> O	2-Methylpropenal	(9.92)	834

Mol. form.	Name	IP/eV	$\Delta_f H_{\text{ion}}$ kJ/mol
C <sub>4</sub> H <sub>6</sub> O	Cyclobutanone	(9.35)	815
C <sub>4</sub> H <sub>6</sub> O <sub>2</sub>	<i>cis</i> -Crotonic acid	(10.08)	625
C <sub>4</sub> H <sub>6</sub> O <sub>2</sub>	<i>trans</i> -Crotonic acid	(9.9)	604
C <sub>4</sub> H <sub>6</sub> O <sub>2</sub>	Methacrylic acid	(10.15)	611
C <sub>4</sub> H <sub>6</sub> O <sub>2</sub>	Vinyl acetate	9.19 ± 0.05	572
C <sub>4</sub> H <sub>6</sub> O <sub>2</sub>	Methyl acrylate	(9.9)	641
C <sub>4</sub> H <sub>6</sub> O <sub>3</sub>	Acetic anhydride	(10.0)	398
C <sub>4</sub> H <sub>6</sub> O <sub>4</sub>	Dimethyl oxalate	(10.0)	287
C <sub>4</sub> H <sub>6</sub> S	2,5-Dihydrothiophene	(8.4)	898
C <sub>4</sub> H <sub>7</sub> N	Butanenitrile	(11.2)	1110
C <sub>4</sub> H <sub>7</sub> N	2-Methylpropanenitrile	(11.3)	1115
C <sub>4</sub> H <sub>7</sub> NO	2-Pyrrolidone	(9.2)	674
C <sub>4</sub> H <sub>8</sub>	1-Butene	9.55 ± 0.06	921
C <sub>4</sub> H <sub>8</sub>	<i>cis</i> -2-Butene	9.11 ± 0.01	871
C <sub>4</sub> H <sub>8</sub>	<i>trans</i> -2-Butene	9.10 ± 0.01	866
C <sub>4</sub> H <sub>8</sub>	Isobutene	9.239 ± 0.003	875
C <sub>4</sub> H <sub>8</sub>	Cyclobutane	(9.82 ± 0.05)	976
C <sub>4</sub> H <sub>8</sub>	Methylcyclopropane	(9.46)	936
C <sub>4</sub> H <sub>8</sub> Br <sub>2</sub>	1,4-Dibromobutane	(10.15)	879
C <sub>4</sub> H <sub>8</sub> O	Ethyl vinyl ether	(8.98)	709
C <sub>4</sub> H <sub>8</sub> O	1,2-Epoxybutane	(≤ 10.15)	862
C <sub>4</sub> H <sub>8</sub> O	Butanal	9.84 ± 0.02	742
C <sub>4</sub> H <sub>8</sub> O	Isobutanal	9.71 ± 0.01	721
C <sub>4</sub> H <sub>8</sub> O	2-Butanone	9.52 ± 0.04	678
C <sub>4</sub> H <sub>8</sub> O	Tetrahydrofuran	9.38 ± 0.05	721
C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	Butanoic acid	10.17 ± 0.05	509
C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	2-Methylpropanoic acid	10.33 ± 0.03	516
C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	Propyl formate	10.52 ± 0.02	555
C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	Ethyl acetate	10.01 ± 0.05	522
C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	Methyl propanoate	10.15 ± 0.03	548
C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	1,3-Dioxane	9.8	607
C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	1,4-Dioxane	9.19 ± 0.01	571
C <sub>4</sub> H <sub>8</sub> O <sub>2</sub> S	Sulfolane	(9.8)	577
C <sub>4</sub> H <sub>8</sub> S	Tetrahydrothiophene	8.38	774
C <sub>4</sub> H <sub>9</sub> Br	1-Bromobutane	(10.12)	869
C <sub>4</sub> H <sub>9</sub> Br	2-Bromobutane	10.01 ± 0.02	845
C <sub>4</sub> H <sub>9</sub> Br	1-Bromo-2-methylpropane	10.09 ± 0.02	861
C <sub>4</sub> H <sub>9</sub> Br	2-Bromo-2-methylpropane	9.92 ± 0.03	823
C <sub>4</sub> H <sub>9</sub> Cl	1-Chlorobutane	10.67 ± 0.03	875
C <sub>4</sub> H <sub>9</sub> Cl	2-Chlorobutane	10.53	857
C <sub>4</sub> H <sub>9</sub> Cl	1-Chloro-2-methylpropane	10.73 ± 0.07	877
C <sub>4</sub> H <sub>9</sub> Cl	2-Chloro-2-methylpropane	(10.61)	842
C <sub>4</sub> H <sub>9</sub> I	1-Iodobutane	9.23 ± 0.01	840
C <sub>4</sub> H <sub>9</sub> I	2-Iodobutane	9.10 ± 0.02	815
C <sub>4</sub> H <sub>9</sub> I	1-Iodo-2-methylpropane	9.19 ± 0.01	824
C <sub>4</sub> H <sub>9</sub> I	2-Iodo-2-methylpropane	(9.02)	798
C <sub>4</sub> H <sub>9</sub> N	Pyrrolidine	(8.0)	769
C <sub>4</sub> H <sub>9</sub> NO	<i>N,N</i> -Dimethylacetamide	8.81 ± 0.03	616
C <sub>4</sub> H <sub>9</sub> NO	Morpholine	(8.2)	841
C <sub>4</sub> H <sub>10</sub>	Butane	10.53 ± 0.10	890
C <sub>4</sub> H <sub>10</sub>	Isobutane	(10.57)	886
C <sub>4</sub> H <sub>10</sub> O	1-Butanol	9.99 ± 0.05	689
C <sub>4</sub> H <sub>10</sub> O	2-Butanol	9.88 ± 0.03	658
C <sub>4</sub> H <sub>10</sub> O	2-Methyl-1-propanol	10.02 ± 0.04	683
C <sub>4</sub> H <sub>10</sub> O	2-Methyl-2-propanol	9.90 ± 0.02	642
C <sub>4</sub> H <sub>10</sub> O	Diethyl ether	9.51 ± 0.03	666
C <sub>4</sub> H <sub>10</sub> O	Methyl propyl ether	9.41 ± 0.07	670
C <sub>4</sub> H <sub>10</sub> O	Isopropyl methyl ether	9.45 ± 0.04	661
C <sub>4</sub> H <sub>10</sub> O <sub>2</sub>	Ethylene glycol monoethyl ether	(9.6)	529
C <sub>4</sub> H <sub>10</sub> O <sub>2</sub>	Ethylene glycol dimethyl ether	(9.3)	558

Mol. form.	Name	IP/eV	$\Delta_i H_{\text{ion}}$ kJ/mol
C <sub>4</sub> H <sub>10</sub> S	1-Butanethiol	9.14 ± 0.01	794
C <sub>4</sub> H <sub>10</sub> S	2-Butanethiol	(9.10)	781
C <sub>4</sub> H <sub>10</sub> S	2-Methyl-1-propanethiol	(9.12)	783
C <sub>4</sub> H <sub>10</sub> S	2-Methyl-2-propanethiol	(9.03)	762
C <sub>4</sub> H <sub>10</sub> S	Diethyl sulfide	(8.43)	730
C <sub>4</sub> H <sub>10</sub> S	Methyl propyl sulfide	(8.8)	767
C <sub>4</sub> H <sub>10</sub> S	Isopropyl methyl sulfide	(8.7)	749
C <sub>4</sub> H <sub>10</sub> S <sub>2</sub>	Diethyl disulfide	(8.27)	724
C <sub>4</sub> H <sub>11</sub> N	Butylamine	8.7 ± 0.1	748
C <sub>4</sub> H <sub>11</sub> N	<i>sec</i> -Butylamine	8.46 ± 0.1	711
C <sub>4</sub> H <sub>11</sub> N	<i>tert</i> -Butylamine	8.46 ± 0.1	695
C <sub>4</sub> H <sub>11</sub> N	Isobutylamine	8.50 ± 0.1	721
C <sub>4</sub> H <sub>11</sub> N	Diethylamine	7.85 ± 0.1	684
C <sub>4</sub> H <sub>12</sub> Si	Tetramethylsilane	9.80 ± 0.04	713
C <sub>4</sub> H <sub>12</sub> Sn	Tetramethylstannane	8.89 ± 0.05	837
C <sub>4</sub> NiO <sub>4</sub>	Nickel carbonyl	8.27 ± 0.04	200
C <sub>5</sub> H <sub>4</sub> O <sub>2</sub>	Furfural	9.22 ± 0.01	739
C <sub>5</sub> H <sub>5</sub> N	Pyridine	9.25	1031
C <sub>5</sub> H <sub>6</sub>	1-Penten-3-yne	9.00 ± 0.01	1119
C <sub>5</sub> H <sub>6</sub>	<i>cis</i> -3-Penten-1-yne	9.14 ± 0.04	1137
C <sub>5</sub> H <sub>6</sub>	<i>trans</i> -3-Penten-1-yne	9.05 ± 0.01	1128
C <sub>5</sub> H <sub>6</sub>	2-Methyl-1-buten-3-yne	9.25 ± 0.02	1152
C <sub>5</sub> H <sub>6</sub>	1,3-Cyclopentadiene	8.55 ± 0.02	955
C <sub>5</sub> H <sub>6</sub> O	2-Methylfuran	8.38 ± 0.02	729
C <sub>5</sub> H <sub>6</sub> O	3-Methylfuran	(8.64)	763
C <sub>5</sub> H <sub>6</sub> S	2-Methylthiophene	(8.14)	867
C <sub>5</sub> H <sub>6</sub> S	3-Methylthiophene	(8.40)	893
C <sub>5</sub> H <sub>8</sub>	<i>cis</i> -1,3-Pentadiene	8.63 ± 0.03	914
C <sub>5</sub> H <sub>8</sub>	<i>trans</i> -1,3-Pentadiene	8.59 ± 0.02	905
C <sub>5</sub> H <sub>8</sub>	1,4-Pentadiene	9.60 ± 0.02	1032
C <sub>5</sub> H <sub>8</sub>	2-Methyl-1,3-butadiene	8.84 ± 0.01	928
C <sub>5</sub> H <sub>8</sub>	1-Pentyne	10.10 ± 0.01	1119
C <sub>5</sub> H <sub>8</sub>	Cyclopentene	9.01 ± 0.01	905
C <sub>5</sub> H <sub>8</sub>	Spiropentane	(9.26)	1078
C <sub>5</sub> H <sub>8</sub> O	Cyclopropyl methyl ketone	(≤ 9.46)	796
C <sub>5</sub> H <sub>8</sub> O	Cyclopentanone	9.26 ± 0.01	701
C <sub>5</sub> H <sub>8</sub> O	3,4-Dihydro-2H-pyran	8.35 ± 0.01	681
C <sub>5</sub> H <sub>8</sub> O <sub>2</sub>	Ethyl acrylate	(≤ 10.3)	617
C <sub>5</sub> H <sub>8</sub> O <sub>2</sub>	Methyl methacrylate	(9.7)	589
C <sub>5</sub> H <sub>8</sub> O <sub>2</sub>	2,4-Pentanedione	8.85 ± 0.01	469
C <sub>5</sub> H <sub>9</sub> NO	<i>N</i> -Methyl-2-pyrrolidone	(≤ 9.17)	≤ 676
C <sub>5</sub> H <sub>10</sub>	1-Pentene	9.51 ± 0.01	896
C <sub>5</sub> H <sub>10</sub>	<i>cis</i> -2-Pentene	9.01 ± 0.03	843
C <sub>5</sub> H <sub>10</sub>	<i>trans</i> -2-Pentene	9.04 ± 0.01	841
C <sub>5</sub> H <sub>10</sub>	2-Methyl-1-butene	9.12 ± 0.01	844
C <sub>5</sub> H <sub>10</sub>	3-Methyl-1-butene	9.52 ± 0.01	891
C <sub>5</sub> H <sub>10</sub>	2-Methyl-2-butene	8.69 ± 0.01	796
C <sub>5</sub> H <sub>10</sub>	Cyclopentane	(10.33 ± 0.15)	918
C <sub>5</sub> H <sub>10</sub> O	2,2-Dimethylpropanal	9.51 ± 0.01	675
C <sub>5</sub> H <sub>10</sub> O	Cyclopentanol	(9.72)	695
C <sub>5</sub> H <sub>10</sub> O	Pentanal	9.74 ± 0.04	709
C <sub>5</sub> H <sub>10</sub> O	2-Pentanone	9.38 ± 0.01	646
C <sub>5</sub> H <sub>10</sub> O	3-Pentanone	9.31 ± 0.01	640
C <sub>5</sub> H <sub>10</sub> O	3-Methyl-2-butanone	9.30 ± 0.01	635
C <sub>5</sub> H <sub>10</sub> O	Tetrahydropyran	9.25 ± 0.01	670
C <sub>5</sub> H <sub>10</sub> O <sub>2</sub>	Pentanoic acid	(≤ 10.53)	≤ 527
C <sub>5</sub> H <sub>10</sub> O <sub>2</sub>	3-Methylbutanoic acid	(≤ 10.51)	≤ 499
C <sub>5</sub> H <sub>10</sub> O <sub>2</sub>	Butyl formate	10.52 ± 0.02	584
C <sub>5</sub> H <sub>10</sub> O <sub>2</sub>	Propyl acetate	(≤ 9.92)	501
C <sub>5</sub> H <sub>10</sub> O <sub>2</sub>	Isopropyl acetate	9.99 ± 0.03	482

Mol. form.	Name	IP/eV	$\Delta_r H_{\text{ion}}$ kJ/mol
C <sub>5</sub> H <sub>10</sub> O <sub>2</sub>	Ethyl propanoate	(10.00)	500
C <sub>5</sub> H <sub>10</sub> O <sub>2</sub>	Methyl butanoate	(10.07)	520
C <sub>5</sub> H <sub>10</sub> S	Thiacyclohexane	(8.2)	728
C <sub>5</sub> H <sub>11</sub> Br	1-Bromopentane	10.10 ± 0.01	846
C <sub>5</sub> H <sub>11</sub> I	1-Iodopentane	9.20 ± 0.01	817
C <sub>5</sub> H <sub>11</sub> N	Piperidine	8.03 ± 0.11	726
C <sub>5</sub> H <sub>11</sub> N	<i>N</i> -Methylpyrrolidine	≤ 8.41 ± 0.02	≤ 809
C <sub>5</sub> H <sub>12</sub>	Pentane	10.28 ± 0.10	845
C <sub>5</sub> H <sub>12</sub>	Isopentane	10.32 ± 0.05	843
C <sub>5</sub> H <sub>12</sub>	Neopentane	(≤ 10.2)	≤ 818
C <sub>5</sub> H <sub>12</sub> O	1-Pentanol	(10.00)	668
C <sub>5</sub> H <sub>12</sub> O	2-Pentanol	(9.78)	630
C <sub>5</sub> H <sub>12</sub> O	3-Pentanol	9.78	628
C <sub>5</sub> H <sub>12</sub> O	2-Methyl-1-butanol	(9.86)	649
C <sub>5</sub> H <sub>12</sub> O	2-Methyl-2-butanol	(9.8)	615
C <sub>5</sub> H <sub>12</sub> O	3-Methyl-2-butanol	(9.88 ± 0.13)	637
C <sub>5</sub> H <sub>12</sub> O	Butyl methyl ether	(9.4 ± 0.1)	648
C <sub>5</sub> H <sub>12</sub> O	Methyl <i>tert</i> -butyl ether	(9.24)	608
C <sub>5</sub> H <sub>12</sub> O	Ethyl propyl ether	(9.45)	640
C <sub>5</sub> H <sub>12</sub> S	<i>tert</i> -Butyl methyl sulfide	(8.38)	687
C <sub>5</sub> H <sub>12</sub> S	Ethyl propyl sulfide	(8.50)	716
C <sub>5</sub> H <sub>12</sub> S	Ethyl isopropyl sulfide	(8.35)	689
C <sub>6</sub> BrF <sub>5</sub>	Bromopentafluorobenzene	(9.67)	222
C <sub>6</sub> ClF <sub>5</sub>	Chloropentafluorobenzene	(9.72)	126
C <sub>6</sub> Cl <sub>6</sub>	Hexachlorobenzene	(8.98)	822
C <sub>6</sub> F <sub>6</sub>	Hexafluorobenzene	9.89 ± 0.04	8
C <sub>6</sub> F <sub>12</sub>	Perfluorocyclohexane	(13.2)	-1095
C <sub>6</sub> HF <sub>5</sub>	Pentafluorobenzene	(9.63)	122
C <sub>6</sub> HF <sub>5</sub> O	Pentafluorophenol	(9.20)	-71
C <sub>6</sub> H <sub>2</sub> F <sub>4</sub>	1,2,3,4-Tetrafluorobenzene	(9.53)	284
C <sub>6</sub> H <sub>2</sub> F <sub>4</sub>	1,2,3,5-Tetrafluorobenzene	(9.53)	263
C <sub>6</sub> H <sub>2</sub> F <sub>4</sub>	1,2,4,5-Tetrafluorobenzene	(9.35)	254
C <sub>6</sub> H <sub>3</sub> Cl <sub>3</sub>	1,2,4-Trichlorobenzene	(9.04)	880
C <sub>6</sub> H <sub>3</sub> Cl <sub>3</sub>	1,3,5-Trichlorobenzene	9.32 ± 0.02	899
C <sub>6</sub> H <sub>4</sub> ClNO <sub>2</sub>	1-Chloro-3-nitrobenzene	(9.92 ± 0.1)	995
C <sub>6</sub> H <sub>4</sub> ClNO <sub>2</sub>	1-Chloro-4-nitrobenzene	(9.96 ± 0.1)	999
C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub>	<i>o</i> -Dichlorobenzene	9.06 ± 0.02	907
C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub>	<i>m</i> -Dichlorobenzene	9.10 ± 0.02	906
C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub>	<i>p</i> -Dichlorobenzene	8.92 ± 0.02	885
C <sub>6</sub> H <sub>4</sub> FNO <sub>2</sub>	1-Fluoro-4-nitrobenzene	(9.90)	826
C <sub>6</sub> H <sub>4</sub> F <sub>2</sub>	<i>o</i> -Difluorobenzene	9.29 ± 0.01	602
C <sub>6</sub> H <sub>4</sub> F <sub>2</sub>	<i>m</i> -Difluorobenzene	9.33 ± 0.01	591
C <sub>6</sub> H <sub>4</sub> F <sub>2</sub>	<i>p</i> -Difluorobenzene	9.1589 ± 0.0003	577
C <sub>6</sub> H <sub>4</sub> O <sub>2</sub>	<i>p</i> -Benzoquinone	10.01 ± 0.06	844
C <sub>6</sub> H <sub>5</sub> Br	Bromobenzene	9.00 ± 0.02	971
C <sub>6</sub> H <sub>5</sub> Cl	Chlorobenzene	9.07 ± 0.02	930
C <sub>6</sub> H <sub>5</sub> ClO	<i>m</i> -Chlorophenol	8.655 ± 0.001	680
C <sub>6</sub> H <sub>5</sub> ClO	<i>p</i> -Chlorophenol	(≤ 8.69)	≤ 692
C <sub>6</sub> H <sub>5</sub> F	Fluorobenzene	9.20 ± 0.01	772
C <sub>6</sub> H <sub>5</sub> I	Iodobenzene	8.685	1003
C <sub>6</sub> H <sub>5</sub> NO <sub>2</sub>	Nitrobenzene	9.86 ± 0.02	1019
C <sub>6</sub> H <sub>5</sub> NO <sub>3</sub>	<i>o</i> -Nitrophenol	(9.1)	782
C <sub>6</sub> H <sub>5</sub> NO <sub>3</sub>	<i>m</i> -Nitrophenol	(9.0)	755
C <sub>6</sub> H <sub>5</sub> NO <sub>3</sub>	<i>p</i> -Nitrophenol	(9.1)	761
C <sub>6</sub> H <sub>6</sub>	Benzene	9.24378 ± 0.00007	975
C <sub>6</sub> H <sub>6</sub>	Fulvene	(8.36)	1031
C <sub>6</sub> H <sub>6</sub> ClN	<i>o</i> -Chloroaniline	(8.50)	883
C <sub>6</sub> H <sub>6</sub> ClN	<i>m</i> -Chloroaniline	(8.09)	835
C <sub>6</sub> H <sub>6</sub> ClN	<i>p</i> -Chloroaniline	(≤ 8.18)	≤ 844
C <sub>6</sub> H <sub>6</sub> N <sub>2</sub> O <sub>2</sub>	<i>o</i> -Nitroaniline	(8.27)	861

Mol. form.	Name	IP/eV	$\Delta_f H_{\text{ion}}$ kJ/mol
$C_6H_6N_2O_2$	<i>m</i> -Nitroaniline	(8.31)	865
$C_6H_6N_2O_2$	<i>p</i> -Nitroaniline	(8.34)	859
$C_6H_6O$	Phenol	8.49 ± 0.02	723
$C_6H_6O_2$	<i>p</i> -Hydroquinone	7.94 ± 0.01	503
$C_6H_6S$	Benzenethiol	(8.32)	915
$C_6H_7N$	Aniline	7.720 ± 0.002	832
$C_6H_7N$	2-Methylpyridine	(9.02)	970
$C_6H_7N$	3-Methylpyridine	(9.04)	979
$C_6H_7N$	4-Methylpyridine	(9.04)	976
$C_6H_8N_2$	<i>o</i> -Phenylenediamine	(7.2)	787
$C_6H_8N_2$	<i>m</i> -Phenylenediamine	(7.14)	777
$C_6H_8N_2$	<i>p</i> -Phenylenediamine	(6.87 ± 0.05)	759
$C_6H_{10}$	1,5-Hexadiene	9.27 ± 0.05	978
$C_6H_{10}$	1-Hexyne	10.03 ± 0.05	1089
$C_6H_{10}$	3,3-Dimethyl-1-butyne	9.90 ± 0.04	1060
$C_6H_{10}$	Cyclohexene	8.945 ± 0.01	859
$C_6H_{10}O$	Cyclohexanone	9.14 ± 0.01	656
$C_6H_{10}O$	Mesityl oxide	9.10 ± 0.01	694
$C_6H_{10}O_4$	Diethyl oxalate	(9.8)	205
$C_6H_{11}NO$	Caprolactam	(9.07 ± 0.02)	629
$C_6H_{12}$	1-Hexene	9.44 ± 0.04	869
$C_6H_{12}$	<i>cis</i> -2-Hexene	(8.97 ± 0.01)	818
$C_6H_{12}$	<i>trans</i> -2-Hexene	(8.97 ± 0.01)	814
$C_6H_{12}$	2-Methyl-1-pentene	(9.08 ± 0.01)	817
$C_6H_{12}$	4-Methyl-1-pentene	9.45 ± 0.01	862
$C_6H_{12}$	2-Methyl-2-pentene	(8.58)	761
$C_6H_{12}$	4-Methyl- <i>cis</i> -2-pentene	8.98 ± 0.01	809
$C_6H_{12}$	4-Methyl- <i>trans</i> -2-pentene	(8.97 ± 0.01)	804
$C_6H_{12}$	2-Ethyl-1-butene	(9.06 ± 0.02)	818
$C_6H_{12}$	2,3-Dimethyl-1-butene	(9.07 ± 0.01)	812
$C_6H_{12}$	2,3-Dimethyl-2-butene	8.27 ± 0.01	729
$C_6H_{12}$	Cyclohexane	9.86 ± 0.03	828
$C_6H_{12}$	Methylcyclopentane	(9.85)	845
$C_6H_{12}O$	Hexanal	9.72 ± 0.05	691
$C_6H_{12}O$	2-Hexanone	9.3 ± 0.1	626
$C_6H_{12}O$	3-Hexanone	9.12 ± 0.02	600
$C_6H_{12}O$	3-Methyl-2-pentanone	9.21 ± 0.01	600
$C_6H_{12}O$	4-Methyl-2-pentanone	9.30 ± 0.01	609
$C_6H_{12}O$	2-Methyl-3-pentanone	9.10 ± 0.01	592
$C_6H_{12}O$	3,3-Dimethyl-2-butanone	9.12 ± 0.02	589
$C_6H_{12}O$	Cyclohexanol	(9.75)	651
$C_6H_{12}O_2$	Hexanoic acid	≤ 10.12	≤ 463
$C_6H_{12}O_2$	Butyl acetate	(9.92 ± .05)	471
$C_6H_{12}O_2$	<i>sec</i> -Butyl acetate	9.90	453
$C_6H_{12}O_2$	Methyl 2,2-dimethylpropanoate	(9.90 ± 0.04)	466
$C_6H_{13}I$	1-Iodohexane	9.179	794
$C_6H_{13}N$	Cyclohexylamine	(8.86)	750
$C_6H_{14}$	Hexane	10.13	810
$C_6H_{14}$	2-Methylpentane	(10.12)	802
$C_6H_{14}$	3-Methylpentane	(10.08)	801
$C_6H_{14}$	2,2-Dimethylbutane	(10.06)	787
$C_6H_{14}$	2,3-Dimethylbutane	(10.02)	791
$C_6H_{14}O$	1-Hexanol	(9.89)	639
$C_6H_{14}O$	2-Hexanol	(9.80 ± 0.03)	611
$C_6H_{14}O$	3-Hexanol	(9.63 ± 0.03)	599
$C_6H_{14}O$	Dipropyl ether	(9.27)	602
$C_6H_{14}O$	Diisopropyl ether	9.20 ± 0.05	569
$C_6H_{14}O$	Butyl ethyl ether	(9.36)	610
$C_6H_{14}O$	Methyl pentyl ether	(≤ 9.67)	≤ 657
$C_6H_{14}O_2$	1,1-Diethoxyethane	(9.2)	434

Mol. form.	Name	IP/eV	$\Delta_r H_{\text{ion}}$ kJ/mol
$C_6H_{14}O_3$	Diethylene glycol dimethyl ether	$\leq 9.8$	$\leq 448$
$C_6H_{14}S$	Dipropyl sulfide	$8.30 \pm 0.02$	676
$C_6H_{14}S$	Diisopropyl sulfide	$(8.2 \pm 0.2)$	649
$C_6H_{15}N$	Hexylamine	$(8.63 \pm 0.05)$	699
$C_6H_{15}N$	Dipropylamine	$(7.84 \pm 0.02)$	641
$C_6H_{15}N$	Diisopropylamine	$(7.73 \pm 0.03)$	602
$C_6H_{15}N$	Triethylamine	$(7.50 \pm 0.02)$	631
$C_6H_{15}NO_3$	Triethanolamine	$(7.9)$	206
$C_7H_3F_5$	2,3,4,5,6-Pentafluorotoluene	$(9.4)$	64
$C_7H_5ClO$	Benzoyl chloride	$(9.53)$	815
$C_7H_5Cl_3$	(Trichloromethyl)benzene	$(\leq 9.60)$	$\leq 914$
$C_7H_5F_3$	(Trifluoromethyl)benzene	$9.685 \pm 0.005$	335
$C_7H_5N$	Benzonitrile	$9.70 \pm 0.01$	1154
$C_7H_6O$	Benzaldehyde	$9.49 \pm 0.02$	878
$C_7H_6O_2$	Benzoic acid	$(9.3)$	604
$C_7H_7Br$	<i>p</i> -Bromotoluene	$8.67 \pm 0.02$	908
$C_7H_7Cl$	<i>o</i> -Chlorotoluene	$(8.7 \pm 0.1)$	856
$C_7H_7Cl$	<i>m</i> -Chlorotoluene	$(8.83)$	869
$C_7H_7Cl$	<i>p</i> -Chlorotoluene	$(8.69)$	855
$C_7H_7Cl$	(Chloromethyl)benzene	$9.10 \pm 0.02$	897
$C_7H_7F$	<i>o</i> -Fluorotoluene	$8.91 \pm 0.01$	709
$C_7H_7F$	<i>m</i> -Fluorotoluene	$8.91 \pm 0.01$	709
$C_7H_7F$	<i>p</i> -Fluorotoluene	$8.79 \pm 0.01$	701
$C_7H_7NO$	Benzamide	$(9.25)$	792
$C_7H_7NO_2$	<i>o</i> -Nitrotoluene	9.24	946
$C_7H_7NO_2$	<i>m</i> -Nitrotoluene	$9.45 \pm 0.1$	941
$C_7H_7NO_2$	<i>p</i> -Nitrotoluene	$9.46 \pm 0.05$	942
$C_7H_8$	Toluene	$8.8276 \pm 0.0006$	901
$C_7H_8O$	<i>o</i> -Cresol	$(8.24)$	670
$C_7H_8O$	<i>m</i> -Cresol	$8.29 \pm 0.07$	668
$C_7H_8O$	<i>p</i> -Cresol	$(8.3)$	675
$C_7H_8O$	Benzyl alcohol	$(8.3)$	701
$C_7H_8O$	Anisole	$8.22 \pm 0.03$	725
$C_7H_9N$	Benzylamine	$(8.64)$	917
$C_7H_9N$	<i>o</i> -Methylaniline	$(7.44 \pm 0.02)$	772
$C_7H_9N$	<i>m</i> -Methylaniline	$(7.50 \pm 0.02)$	778
$C_7H_9N$	<i>p</i> -Methylaniline	$(7.24 \pm 0.02)$	753
$C_7H_9N$	<i>N</i> -Methylaniline	$7.34 \pm 0.04$	792
$C_7H_9N$	2,3-Dimethylpyridine	$(8.85 \pm 0.02)$	922
$C_7H_9N$	2,4-Dimethylpyridine	$(8.85 \pm 0.03)$	918
$C_7H_9N$	2,5-Dimethylpyridine	$(\leq 8.80 \pm 0.05)$	$\leq 916$
$C_7H_9N$	2,6-Dimethylpyridine	$8.86 \pm 0.03$	913
$C_7H_9N$	3,4-Dimethylpyridine	$(\leq 9.15)$	$\leq 953$
$C_7H_9N$	3,5-Dimethylpyridine	$(\leq 9.25)$	$\leq 965$
$C_7H_{10}O$	Dicyclopentyl ketone	$(9.1)$	1041
$C_7H_{14}$	1-Heptene	$9.34 \pm 0.10$	839
$C_7H_{14}$	<i>trans</i> -3-Heptene	$(8.92)$	790
$C_7H_{14}$	Cycloheptane	9.97	844
$C_7H_{14}$	Methylcyclohexane	9.64	775
$C_7H_{14}$	<i>cis</i> -1,2-Dimethylcyclopentane	$(9.92 \pm 0.05)$	828
$C_7H_{14}$	<i>trans</i> -1,2-Dimethylcyclopentane	$9.7 \pm 0.2$	799
$C_7H_{14}O$	1-Heptanal	$(9.65)$	668
$C_7H_{14}O$	2-Heptanone	$9.28 \pm 0.10$	594
$C_7H_{14}O$	3-Heptanone	$9.18 \pm 0.08$	589
$C_7H_{14}O$	4-Heptanone	$9.10 \pm 0.06$	577
$C_7H_{14}O$	5-Methyl-2-hexanone	$(9.28)$	586
$C_7H_{14}O$	2,4-Dimethyl-3-pentanone	$8.95 \pm 0.01$	552
$C_7H_{14}O$	1-Methylcyclohexanol	$(9.8 \pm 0.2)$	586
$C_7H_{16}$	Heptane	$9.93 \pm 0.10$	771
$C_7H_{16}O$	1-Heptanol	$(9.84)$	614

Mol. form.	Name	IP/eV	$\Delta_f H_{\text{ion}}$ kJ/mol
C <sub>7</sub> H <sub>16</sub> O	2-Heptanol	(9.70)	580
C <sub>7</sub> H <sub>16</sub> O	3-Heptanol	(9.68)	578
C <sub>7</sub> H <sub>16</sub> O	4-Heptanol	(9.61)	572
C <sub>7</sub> H <sub>16</sub> O	Ethyl pentyl ether	( $\leq 9.49$ )	$\leq 602$
C <sub>8</sub> H <sub>4</sub> O <sub>3</sub>	Phthalic anhydride	(10.1)	603
C <sub>8</sub> H <sub>6</sub> O <sub>4</sub>	Isophthalic acid	(9.98)	268
C <sub>8</sub> H <sub>6</sub> O <sub>4</sub>	Terephthalic acid	(9.86)	232
C <sub>8</sub> H <sub>7</sub> N	2-Methylbenzonitrile	( $\leq 9.38$ )	1085
C <sub>8</sub> H <sub>7</sub> N	3-Methylbenzonitrile	( $\leq 9.34$ )	1085
C <sub>8</sub> H <sub>7</sub> N	4-Methylbenzonitrile	9.32 $\pm$ 0.02	1083
C <sub>8</sub> H <sub>7</sub> N	Indole	7.7602 $\pm$ 0.0006	908
C <sub>8</sub> H <sub>8</sub>	Styrene	8.464 $\pm$ 0.001	964
C <sub>8</sub> H <sub>8</sub> O	<i>p</i> -Tolualdehyde	(9.33)	825
C <sub>8</sub> H <sub>8</sub> O	Acetophenone	9.29 $\pm$ 0.03	810
C <sub>8</sub> H <sub>8</sub> O <sub>2</sub>	<i>o</i> -Toluic acid	(9.1)	558
C <sub>8</sub> H <sub>8</sub> O <sub>2</sub>	<i>m</i> -Toluic acid	(9.43)	579
C <sub>8</sub> H <sub>8</sub> O <sub>2</sub>	<i>p</i> -Toluic acid	(9.23)	560
C <sub>8</sub> H <sub>8</sub> O <sub>2</sub>	Benzeneacetic acid	(8.26)	479
C <sub>8</sub> H <sub>8</sub> O <sub>2</sub>	Methyl benzoate	9.32 $\pm$ 0.03	611
C <sub>8</sub> H <sub>10</sub>	Ethylbenzene	8.77 $\pm$ 0.01	876
C <sub>8</sub> H <sub>10</sub>	<i>o</i> -Xylene	8.56 $\pm$ 0.01	844
C <sub>8</sub> H <sub>10</sub>	<i>m</i> -Xylene	8.56 $\pm$ 0.01	843
C <sub>8</sub> H <sub>10</sub>	<i>p</i> -Xylene	8.44 $\pm$ 0.01	832
C <sub>8</sub> H <sub>10</sub> O	<i>p</i> -Ethylphenol	(7.84)	613
C <sub>8</sub> H <sub>10</sub> O	2,3-Xylenol	(8.26)	640
C <sub>8</sub> H <sub>10</sub> O	2,4-Xylenol	(8.0)	609
C <sub>8</sub> H <sub>10</sub> O	2,6-Xylenol	(8.05)	615
C <sub>8</sub> H <sub>10</sub> O	3,4-Xylenol	(8.09)	624
C <sub>8</sub> H <sub>10</sub> O	Phenetole	(8.13)	683
C <sub>8</sub> H <sub>11</sub> N	2,4,6-Trimethylpyridine	( $\leq 8.9$ )	$\leq 880$
C <sub>8</sub> H <sub>11</sub> N	<i>N</i> -Ethylaniline	( $\leq 7.67$ )	$\leq 794$
C <sub>8</sub> H <sub>11</sub> N	<i>N,N</i> -Dimethylaniline	7.12 $\pm$ 0.02	787
C <sub>8</sub> H <sub>14</sub>	1-Octyne	(9.95 $\pm$ 0.02)	1040
C <sub>8</sub> H <sub>14</sub>	2-Octyne	9.31 $\pm$ 0.01	961
C <sub>8</sub> H <sub>14</sub>	3-Octyne	9.22 $\pm$ 0.01	952
C <sub>8</sub> H <sub>14</sub>	4-Octyne	9.20 $\pm$ 0.01	946
C <sub>8</sub> H <sub>16</sub>	1-Octene	9.43 $\pm$ 0.01	829
C <sub>8</sub> H <sub>16</sub>	Cyclooctane	9.75 $\pm$ 0.05	816
C <sub>8</sub> H <sub>16</sub>	Ethylcyclohexane	(9.54)	748
C <sub>8</sub> H <sub>16</sub>	1,1-Dimethylcyclohexane	(9.42)	728
C <sub>8</sub> H <sub>16</sub>	<i>cis</i> -1,2-Dimethylcyclohexane	(<9.78)	772
C <sub>8</sub> H <sub>16</sub>	<i>trans</i> -1,2-Dimethylcyclohexane	9.41	728
C <sub>8</sub> H <sub>16</sub>	<i>cis</i> -1,3-Dimethylcyclohexane	(<9.98)	778
C <sub>8</sub> H <sub>16</sub>	<i>trans</i> -1,3-Dimethylcyclohexane	9.53	743
C <sub>8</sub> H <sub>16</sub>	<i>cis</i> -1,4-Dimethylcyclohexane	(<9.93)	782
C <sub>8</sub> H <sub>16</sub>	<i>trans</i> -1,4-Dimethylcyclohexane	(9.56)	738
C <sub>8</sub> H <sub>16</sub>	Propylcyclopentane	(9.34)	753
C <sub>8</sub> H <sub>16</sub> O	2,2,4-Trimethyl-3-pentanone	(8.80)	511
C <sub>8</sub> H <sub>18</sub>	Octane	9.80 $\pm$ 0.10	737
C <sub>8</sub> H <sub>18</sub>	2-Methylheptane	(9.84)	734
C <sub>8</sub> H <sub>18</sub>	2,2,4-Trimethylpentane	(9.86)	713
C <sub>8</sub> H <sub>18</sub>	2,2,3,3-Tetramethylbutane	9.8	720
C <sub>8</sub> H <sub>18</sub> O	Dibutyl ether	(9.28)	$\leq 560$
C <sub>8</sub> H <sub>18</sub> O	Di- <i>sec</i> -butyl ether	(9.11)	511
C <sub>8</sub> H <sub>18</sub> O	Di- <i>tert</i> -butyl ether	8.88 $\pm$ 0.07	493
C <sub>8</sub> H <sub>18</sub> S	Dibutyl sulfide	(8.2)	624
C <sub>8</sub> H <sub>18</sub> S	Di- <i>tert</i> -butyl sulfide	(8.0)	583
C <sub>8</sub> H <sub>18</sub> S	Diisobutyl sulfide	(8.34)	625
C <sub>8</sub> H <sub>19</sub> N	Dibutylamine	(7.69)	586
C <sub>8</sub> H <sub>19</sub> N	Diisobutylamine	(7.8)	574



Mol. form.	Name	IP/eV	$\Delta_f H_{\text{ion}}$ kJ/mol
C <sub>8</sub> H <sub>20</sub> Si	Tetraethylsilane	(8.9)	595
C <sub>9</sub> H <sub>7</sub> N	Quinoline	8.62 ± 0.01	1041
C <sub>9</sub> H <sub>7</sub> N	Isoquinoline	8.53 ± 0.03	1032
C <sub>9</sub> H <sub>8</sub>	Indene	8.14 ± 0.01	949
C <sub>9</sub> H <sub>10</sub>	<i>o</i> -Methylstyrene	(8.20)	908
C <sub>9</sub> H <sub>10</sub>	<i>m</i> -Methylstyrene	(8.15)	899
C <sub>9</sub> H <sub>10</sub>	<i>p</i> -Methylstyrene	(8.1)	895
C <sub>9</sub> H <sub>10</sub>	Cyclopropylbenzene	(8.35)	956
C <sub>9</sub> H <sub>10</sub>	Indan	(8.3)	864
C <sub>9</sub> H <sub>10</sub> O <sub>2</sub>	Ethyl benzoate	(8.9)	537
C <sub>9</sub> H <sub>12</sub>	Propylbenzene	8.713 ± 0.010	848
C <sub>9</sub> H <sub>12</sub>	Isopropylbenzene	8.73 ± 0.01	847
C <sub>9</sub> H <sub>12</sub>	1,2,3-Trimethylbenzene	8.42 ± 0.02	803
C <sub>9</sub> H <sub>12</sub>	1,2,4-Trimethylbenzene	8.27 ± 0.01	784
C <sub>9</sub> H <sub>12</sub>	1,3,5-Trimethylbenzene	8.41 ± 0.01	796
C <sub>9</sub> H <sub>13</sub> N	<i>N,N</i> -Dimethyl- <i>o</i> -toluidine	7.40 ± 0.02	814
C <sub>9</sub> H <sub>14</sub> O	Isophorone	(≤ 9.07)	≤ 670
C <sub>9</sub> H <sub>18</sub>	Butylcyclopentane	(9.95)	793
C <sub>9</sub> H <sub>18</sub>	Propylcyclohexane	(9.46)	720
C <sub>9</sub> H <sub>18</sub>	Isopropylcyclohexane	(9.33)	704
C <sub>9</sub> H <sub>18</sub> O	2-Nonanone	(9.16)	545
C <sub>9</sub> H <sub>18</sub> O	5-Nonanone	(9.07)	530
C <sub>9</sub> H <sub>16</sub> O	2,6-Dimethyl-4-heptanone	9.01 ± 0.06	512
C <sub>9</sub> H <sub>20</sub>	Nonane	9.71 ± 0.10	709
C <sub>10</sub> F <sub>8</sub>	Perfluoronaphthalene	8.85	-368
C <sub>10</sub> H <sub>7</sub> Br	1-Bromonaphthalene	8.08 ± 0.03	955
C <sub>10</sub> H <sub>7</sub> Cl	1-Chloronaphthalene	(8.13)	906
C <sub>10</sub> H <sub>8</sub>	Naphthalene	8.1442 ± 0.0009	936
C <sub>10</sub> H <sub>8</sub>	Azulene	7.38 ± 0.05	1001
C <sub>10</sub> H <sub>8</sub> O	1-Naphthol	7.76 ± 0.03	719
C <sub>10</sub> H <sub>8</sub> O	2-Naphthol	7.87 ± 0.06	729
C <sub>10</sub> H <sub>10</sub> O <sub>4</sub>	Dimethyl phthalate	(9.64 ± 0.07)	277
C <sub>10</sub> H <sub>12</sub>	1,2,3,4-Tetrahydronaphthalene	8.46 ± 0.02	841
C <sub>10</sub> H <sub>14</sub>	Butylbenzene	8.69 ± 0.02	826
C <sub>10</sub> H <sub>14</sub>	<i>sec</i> -Butylbenzene	8.68 ± 0.02	820
C <sub>10</sub> H <sub>14</sub>	<i>tert</i> -Butylbenzene	8.68 ± 0.05	816
C <sub>10</sub> H <sub>14</sub>	Isobutylbenzene	8.69 ± 0.02	817
C <sub>10</sub> H <sub>14</sub>	<i>p</i> -Cymene	(8.29)	771
C <sub>10</sub> H <sub>14</sub>	<i>o</i> -Diethylbenzene	(≤ 8.51)	≤ 804
C <sub>10</sub> H <sub>14</sub>	<i>m</i> -Diethylbenzene	(8.49)	798
C <sub>10</sub> H <sub>14</sub>	<i>p</i> -Diethylbenzene	(8.40)	790
C <sub>10</sub> H <sub>14</sub>	1,2,4,5-Tetramethylbenzene	8.04 ± 0.02	730
C <sub>10</sub> H <sub>14</sub> O	<i>p</i> - <i>tert</i> -Butylphenol	(7.8)	552
C <sub>10</sub> H <sub>16</sub>	$\alpha$ -Pinene	(8.07)	808
C <sub>10</sub> H <sub>16</sub> O	Camphor	(8.76)	577
C <sub>10</sub> H <sub>18</sub>	<i>cis</i> -Decahydronaphthalene	9.36 ± 0.04	734
C <sub>10</sub> H <sub>18</sub>	<i>trans</i> -Decahydronaphthalene	9.34 ± 0.04	720
C <sub>10</sub> H <sub>20</sub>	1-Decene	9.42 ± 0.05	786
C <sub>10</sub> H <sub>20</sub>	Butylcyclohexane	(9.41)	695
C <sub>10</sub> H <sub>22</sub>	Decane	(9.65)	682
C <sub>11</sub> H <sub>10</sub>	1-Methylnaphthalene	7.97 ± 0.03	882
C <sub>11</sub> H <sub>10</sub>	2-Methylnaphthalene	7.91 ± 0.08	877
C <sub>11</sub> H <sub>16</sub>	<i>p</i> - <i>tert</i> -Butyltoluene	(8.12)	730
C <sub>11</sub> H <sub>24</sub>	Undecane	(9.56)	650
C <sub>11</sub> H <sub>24</sub>	2-Methyldecane	(9.7)	658
C <sub>12</sub> H <sub>8</sub>	Acenaphthylene	(8.22)	1053
C <sub>12</sub> H <sub>9</sub> N	Carbazole	(7.57)	961
C <sub>12</sub> H <sub>10</sub>	Acenaphthene	7.75 ± 0.07	903
C <sub>12</sub> H <sub>10</sub>	Biphenyl	8.23 ± 0.10	977
C <sub>12</sub> H <sub>10</sub> N <sub>2</sub> O	<i>trans</i> -Azoxybenzene	(8.1)	1123

Mol. form.	Name	IP/eV	$\Delta_f H_{\text{ion}}$ kJ/mol
C <sub>12</sub> H <sub>10</sub> O	Diphenyl ether	(8.09)	766
C <sub>12</sub> H <sub>11</sub> N	Diphenylamine	7.16 ± 0.04	908
C <sub>12</sub> H <sub>18</sub>	5,7-Dodecadiyne	(8.67)	1079
C <sub>12</sub> H <sub>18</sub>	Hexamethylbenzene	7.85 ± 0.01	670
C <sub>12</sub> H <sub>22</sub>	Cyclohexylcyclohexane	(9.41)	690
C <sub>12</sub> H <sub>27</sub> N	Tributylamine	(7.4)	492
C <sub>13</sub> H <sub>10</sub>	9 <i>H</i> -Fluorene	7.91 ± 0.02	952
C <sub>13</sub> H <sub>10</sub> O	Benzophenone	9.08 ± 0.05	926
C <sub>13</sub> H <sub>12</sub>	Diphenylmethane	(8.55)	963
C <sub>14</sub> H <sub>10</sub>	Anthracene	7.439 ± 0.006	948
C <sub>14</sub> H <sub>10</sub>	Phenanthrene	7.8914 ± 0.0006	966
C <sub>14</sub> H <sub>10</sub>	Diphenylacetylene	7.94 ± 0.03	1168
C <sub>14</sub> H <sub>12</sub>	<i>cis</i> -Stilbene	(7.80)	1005
C <sub>14</sub> H <sub>12</sub>	<i>trans</i> -Stilbene	7.656 ± 0.001	973
C <sub>14</sub> H <sub>14</sub>	1,2-Diphenylethane	8.9 ± 0.1	1002
C <sub>16</sub> H <sub>10</sub>	Fluoranthene	7.9 ± 0.1	1052
C <sub>16</sub> H <sub>10</sub>	Pyrene	7.4256 ± 0.0006	935
C <sub>18</sub> H <sub>12</sub>	Chrysene	7.60 ± 0.01	1017
C <sub>18</sub> H <sub>14</sub>	<i>o</i> -Terphenyl	(7.99)	1056
C <sub>18</sub> H <sub>14</sub>	<i>m</i> -Terphenyl	(8.01)	1057
C <sub>18</sub> H <sub>14</sub>	<i>p</i> -Terphenyl	7.80 ± 0.03	1037
C <sub>20</sub> H <sub>12</sub>	Perylene	6.960 ± 0.001	981
C <sub>24</sub> H <sub>12</sub>	Coronene	7.29 ± 0.01	1026