



BIOMATHEMATICS

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Lecture 38

Statistical thermodynamics of biological systems

Thermodynamics of protein organization along DNA

Total number of arrangements possible,
When there are N binding sites and m
proteins

$$\Omega = \frac{N!}{m!(N - m)!}$$

Entropy

$$S = k_B \ln \Omega = k_B \ln \left(\frac{N!}{m!(N-m)!} \right)$$

$$S = -k_B N [\rho \ln \rho + (1 - \rho) \ln (1 - \rho)]$$

When there are m proteins bound,
the energy is given by

$$E = -m \epsilon k_B T = -N \rho \epsilon k_B T$$

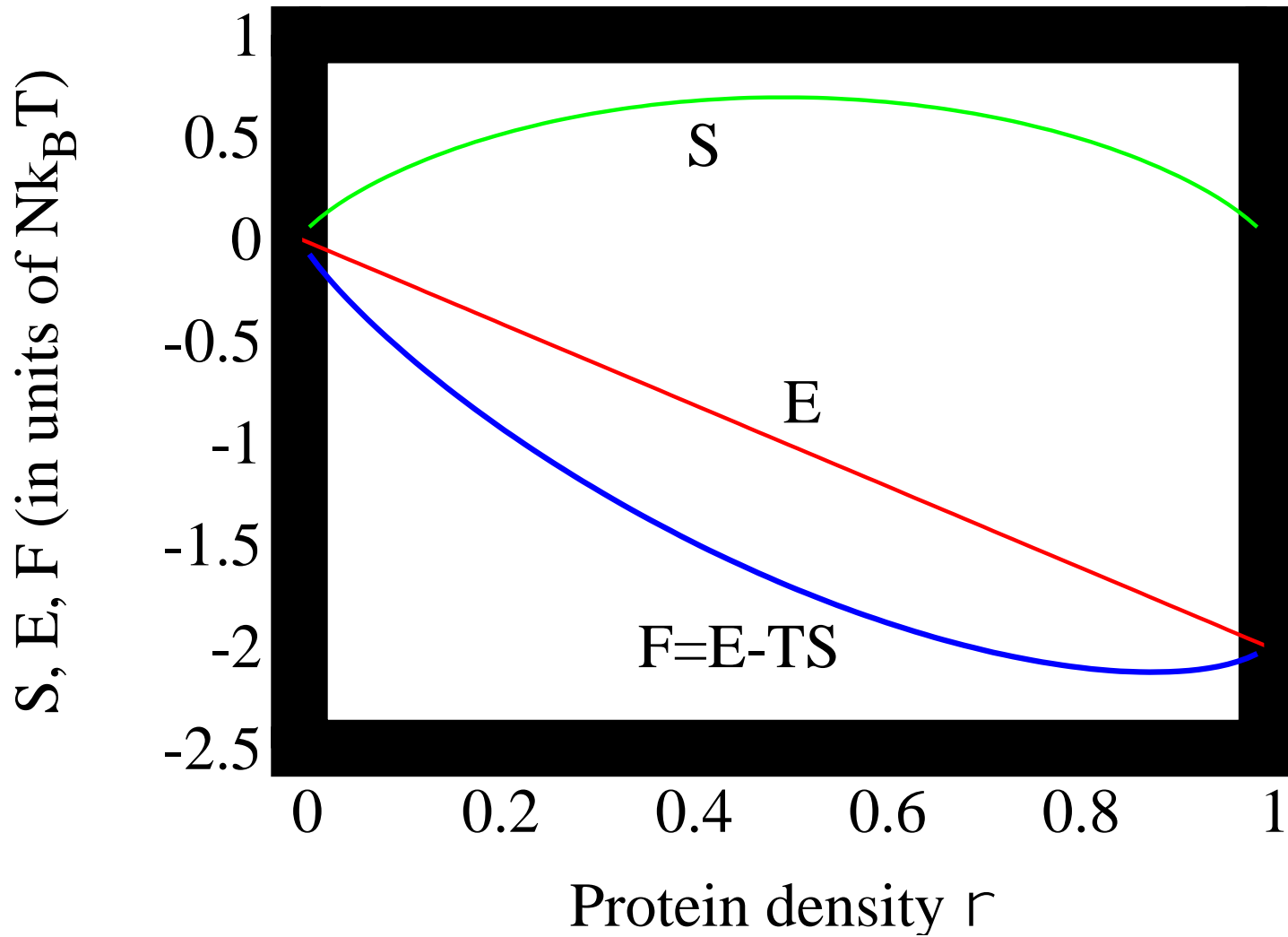
Free energy (F)

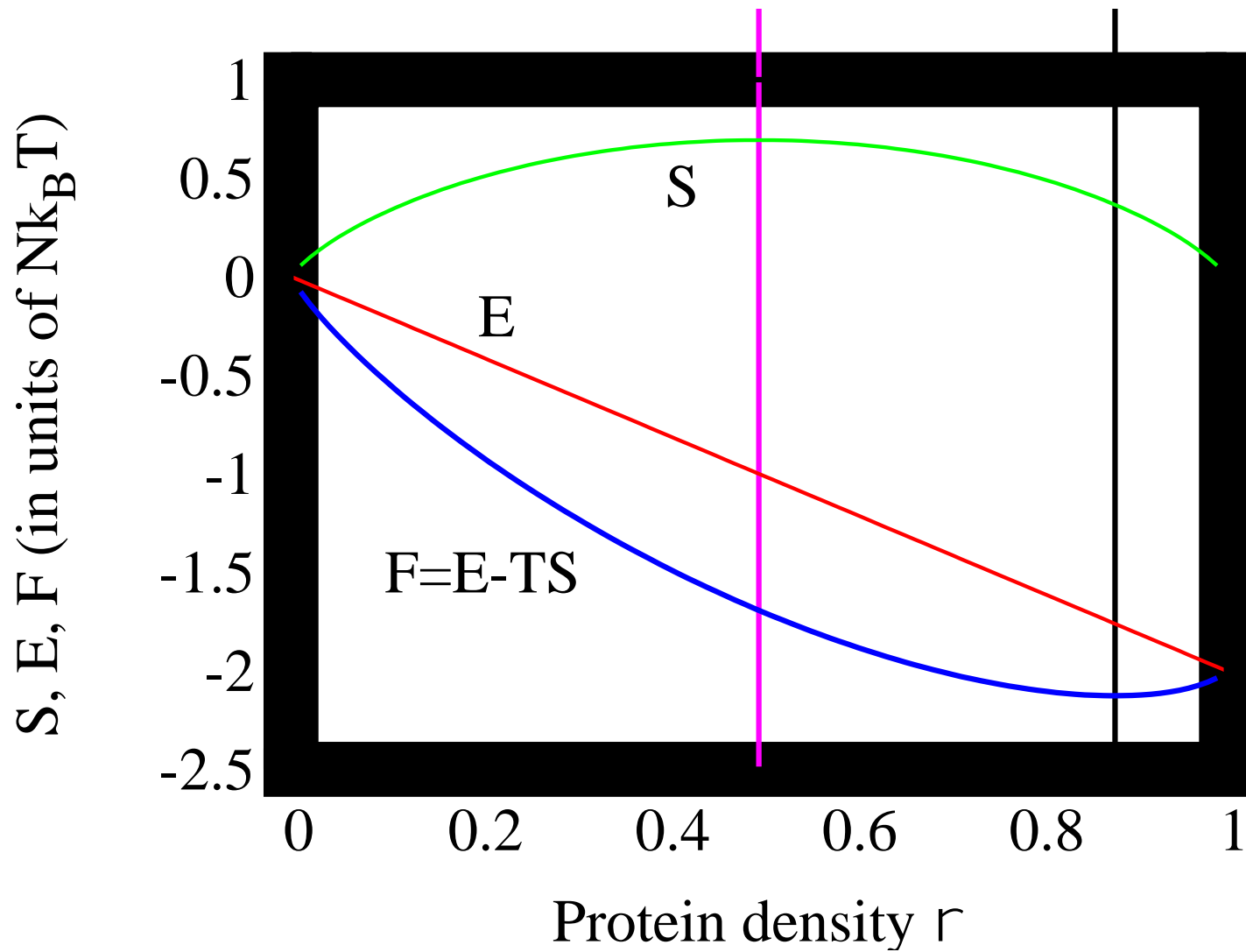
$$F = E - TS$$

$$F = -N\rho\epsilon k_B T + k_B T N [\rho \ln \rho + (1 - \rho) \ln (1 - \rho)]$$

Free energy per binding site, in units of $k_B T$, is given by

$$\frac{F}{Nk_B T} = -\rho\epsilon + \rho \ln \rho + (1 - \rho) \ln (1 - \rho)$$





The free energy is minimum when

$$\frac{\partial F}{\partial \rho} = 0$$

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$$\Rightarrow \rho = \frac{e^{\epsilon}}{1 + e^{\epsilon}}$$

For $\epsilon = 2$

