

Chapter 9

1. At isoelectric pH, protein is

- (a) Positively charged (b) Negatively charged (c) Neutral

Ans: (c)

2. What major problems in electrophoresis arises due to joule heating effects?

Ans: (a) Density difference occurs leading to free convection.

(b) Temperature increase degrades protein.

3. How joule heating is prevented?

Ans: (a) Efficient cooling.

(b) Electrophoresis is used in a gel/membrane to prevent cooling.

4. How electroosmosis is prevented in electrophoresis?

Ans: Use of material for gel having low residual charge like ployacrylamide.

5. In a batch page electrophoresis, $\mu_{\text{osm}}=0$. Mobilities of two proteins are $\mu_A = 5 \times 10^{-5}$

$\text{cm}^2/\text{V}\cdot\text{s}$ and $\mu_B = 3 \times 10^{-5} \text{cm}^2/\text{V}\cdot\text{s}$. $D_A = 1.5 \times 10^{-7} \text{cm}^2/\text{s}$ and $D_B = 1.2 \times 10^{-7} \text{cm}^2/\text{s}$.

$E = 200 \text{ V/cm}$; $t = 4 \text{ hours}$. Find resolution (R).

Ans

$$\mu_A = 5 \times 10^{-5} \frac{\text{cm}^2}{\text{v}\cdot\text{s}}$$

$$\mu_B = 3 \times 10^{-5} \frac{\text{cm}^2}{\text{v}\cdot\text{s}}$$

$$Z_A = \mu_A E t_{\text{exp}}$$

$$= 5 \times 10^{-5} \frac{\text{cm}^2}{\text{v}\cdot\text{s}} \times 200 \frac{\text{v}}{\text{cm}} \times (4 \times 3600) \text{s}$$

$$= 144 \text{cm}$$

$$Z_B = \mu_B E t_{\text{exp}}$$

$$= 3 \times 10^{-5} \frac{cm^2}{v.s} \times 200 \frac{v}{cm} \times (4 \times 3600) s$$

$$= 86.4 cm$$

$$D_{eff} = \left(\frac{1.5 + 1.2}{2} \right) \times 10^{-7} cm^2 / sec$$

$$= 1.35 \times 10^{-7} cm^2 / sec$$

$$\sigma = \sqrt{2D_{eff} t_{exp}}$$

$$\sigma = \sqrt{2 \times 1.35 \times 10^{-7} \times (4 \times 3600)}$$

$$= 0.0039 cm$$

$$R = \frac{1}{4} E \sqrt{\frac{t_{exp}}{2D_{eff}}} (\mu_1 - \mu_2)$$

$$= \frac{1}{4} \times 200 \sqrt{\frac{4 \times 3600}{2 \times 1.35 \times 10^{-7}}} (2 \times 10^{-5}) = 230.94$$