

Tutorial

- A. Determine $\mathbf{W/m_f}$ for a Claude Cycle with N_2 as working fluid. The system operates between 1.013 bar (1 atm) and 40.52 bar (40 atm). The expander inlet \mathbf{T}_3 is at 225 K. The expander flow ratio is varied between 0.1 and 0.9.
- B. Repeat the above problem for $\mathbf{T}_3 = 300$ K, 275 K, 250 K and 225 K. Plot the data \mathbf{y} , $\mathbf{W/m_f}$ versus \mathbf{x} graphically and comment on the results.

Tutorial

Given

Cycle : Claude System

Working Pressure : 1 atm \rightarrow 40 atm

Working Fluid : Nitrogen

T_3 : 300 K, 275 K, 250 K, 225 K

Mass flow ratio : $x = 0.1 \rightarrow 0.9$

For above System, Calculate

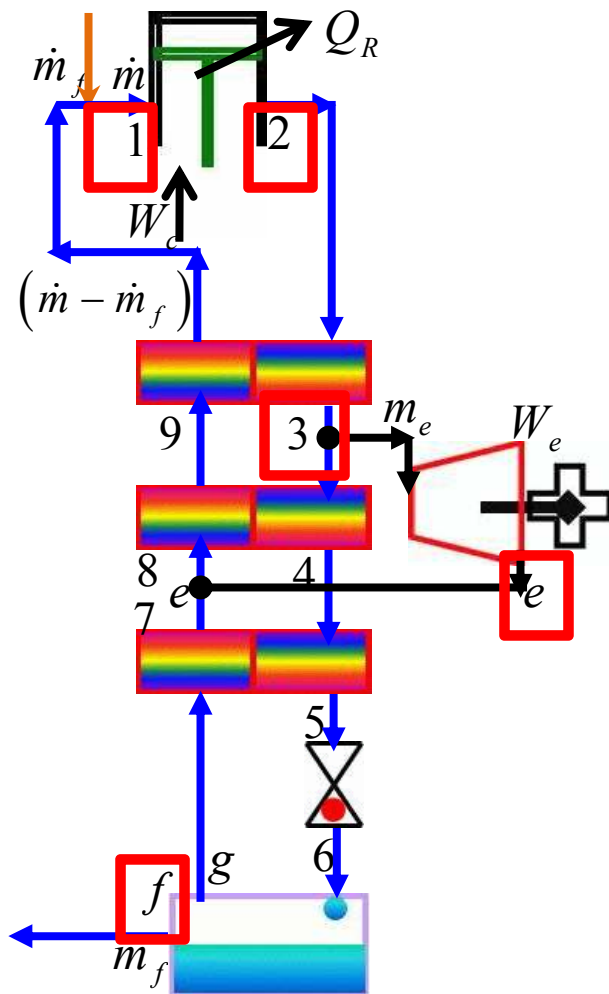
1 Work/unit mass of gas liquefied

N₂	Point 3
I	300 K
II	275 K
III	250 K
IV	225 K

Methodology

- In the part **A**, the expander inlet condition under study is 225 K at 40.52 bar.
- The expander mass flow ratio varies between 0.1 and 0.9.
- In this tutorial, the **y** , **W/m_f** is calculated only for **$x = 0.2$** and **225 K** as inlet condition.

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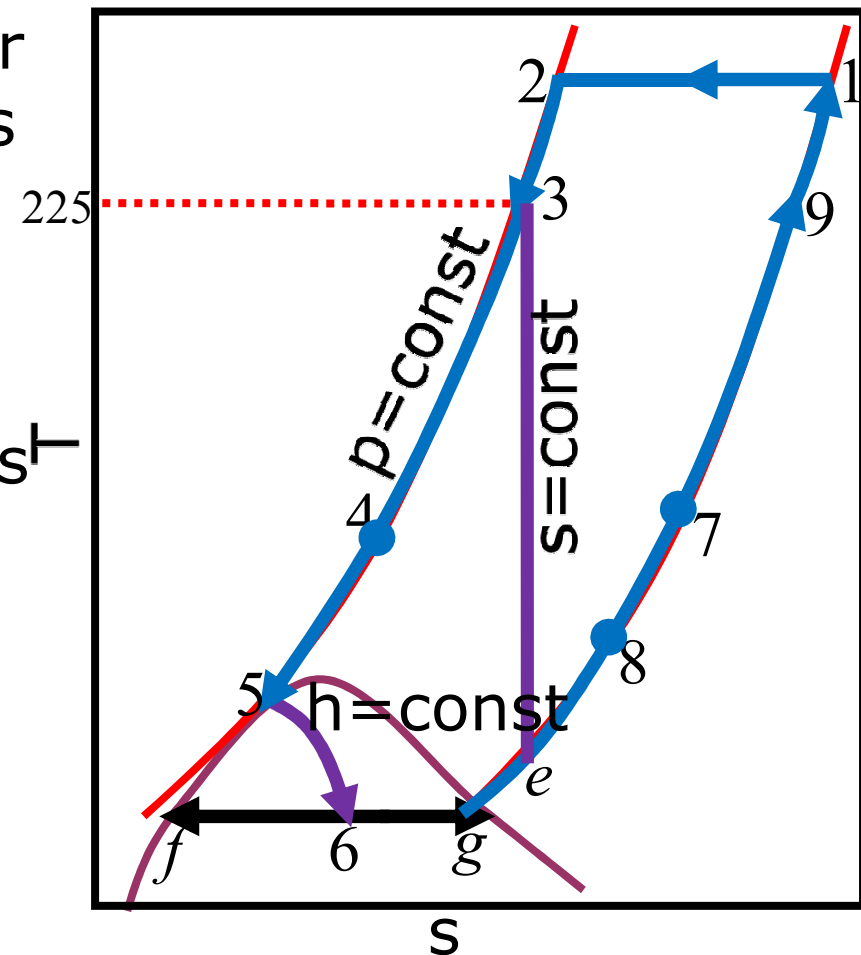
	1	2	3
p (bar)	1.013	40.52	40.52
T (K)	300	300	225
h (J/g)	462	453	369
s (J/gK)	4.42	3.3	3.0

	e	f
p (bar)	1.013	1.013
T (K)	80*	77
h (J/g)	228	29
s (J/gK)	3.0	.25

* The point **e** is located on p=1bar isobar by drawing a vertical line from point **3**.

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- The $T - s$ diagram for a Claude System is as shown.
- The expander inlet condition and its mass flow ratio are **225 K** and **0.2** respectively.



Tutorial

- Liquid yield**

$$y = \frac{h_1 - h_2}{h_1 - h_f} + x \left(\frac{h_3 - h_e}{h_1 - h_f} \right)$$

x	Point 3
0.2	225 K, 40 atm

	1	2	3	e	f
p (bar)	1.013	40.52	40.52	1.013	1.013
T (K)	300	300	225	80	77
h (J/g)	462	453	369	228	29
s (J/gK)	4.42	3.3	3.1	3.1	3.0

$$y = \frac{(462 - 453)}{(462 - 29)} + 0.2 \frac{(369 - 228)}{(462 - 29)} = 0.021 + 0.065 = 0.086$$

Tutorial

- **Work/unit mass of N₂ compressed**

$$-\frac{W_c}{\dot{m}} = T_1(s_1 - s_2) - (h_1 - h_2) - x(h_3 - h_e)$$

	1	2	3	e	f
p (bar)	1.013	40.52	40.52	1.013	1.013
T (K)	300	300	225	80	77
h (J/g)	462	453	369	228	29
s (J/gK)	4.42	3.3	3.1	3.1	3.0

$$-\frac{W_c}{\dot{m}} = 300(4.42 - 3.3) - (462 - 453) - 0.2(369 - 228)$$

$$= 299 \text{ J / g}$$

Tutorial

- **Work/unit mass of N₂ liquefied**

$$-\frac{W_c}{\dot{m}} = 299$$

$$y = 0.086$$

$$-\frac{W_c}{\dot{m}_f} = -\frac{W_c}{y\dot{m}} = \frac{299}{0.086} = 3476.7 \text{ J / g}$$

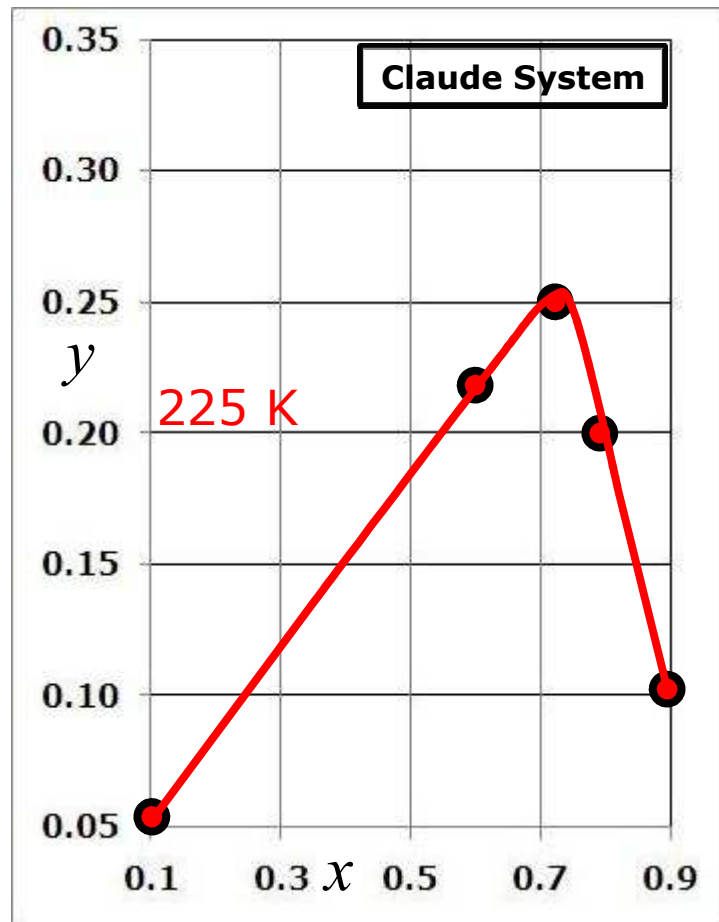
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- Extending the calculations for all other values of x and tabulating the results, we have
- In the adjacent table, the equation for y is used from $x=0.1$ to **0.73**. Thereafter, **$y=0.99-x$** is used.
- Actual y may be less than this value.

225		
x	y	W/m_f
0.10	0.05	5865.2
0.20	0.09	3478.0
0.30	0.12	2403.0
0.40	0.15	1791.6
0.50	0.18	1397.0
0.60	0.22	1121.4
0.70	0.25	917.9
0.73	0.26	866.8
0.80	0.19	1127.4
0.90	0.09	2223.3

Tutorial

- Liquid yield $v/s. x$

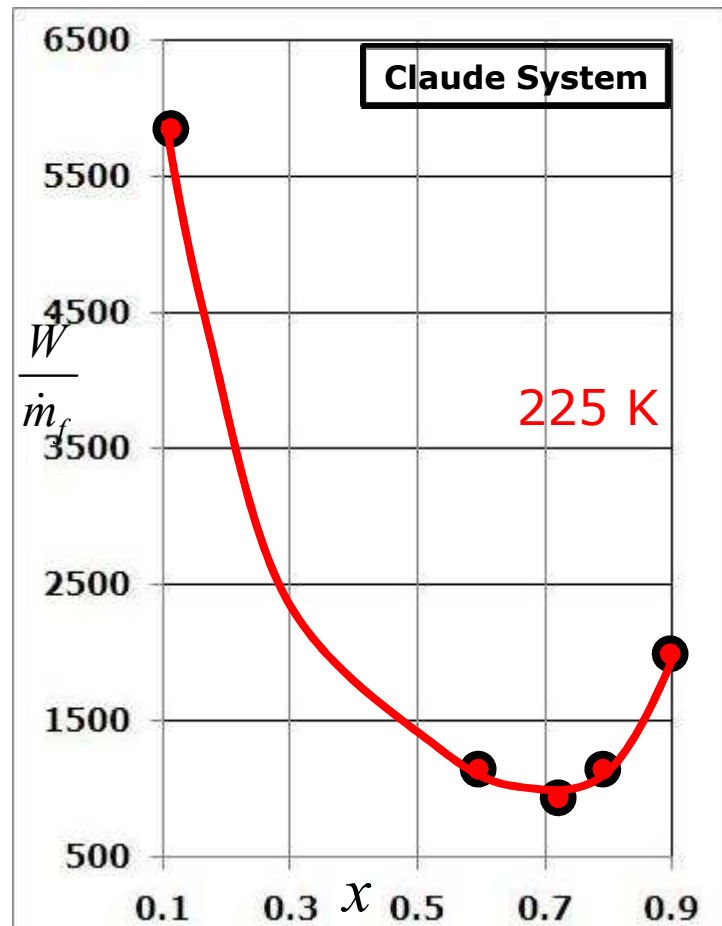


225 K	x	y
I	0.1	0.05
II	0.6	0.22
III	0.73	0.26
IV	0.8	0.19
V	0.9	0.09

- From the plot, it is clear that y crosses a maxima with the increasing x .
- Beyond this maxima, the y is estimated as limiting value of $y=0.99-x$.

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- W/m_f v/s. x



225 K	x	$-W / m_f$
I	0.1	5865.2
II	0.6	1121.4
III	0.73	866.8
IV	0.8	1127.4
V	0.9	2223.3

- The trend shows that the W/m_f crosses a minima with the increasing x .
- Beyond this minima, the W/m_f is estimated based on limiting value of y .

Tutorial

- All the calculations pertaining to part **B** are left as an exercise.

300		275		250	
x	y	x	y	x	y
0.10	0.07	0.10	0.06	0.10	0.06
0.20	0.11	0.20	0.10	0.20	0.09
0.30	0.16	0.30	0.15	0.30	0.13
0.40	0.20	0.40	0.19	0.40	0.16
0.50	0.25	0.50	0.23	0.50	0.20
0.60	0.29	0.60	0.27	0.60	0.24
0.67	0.32	0.69	0.30	0.70	0.27
0.70	0.29	0.70	0.29	0.72	0.27
0.80	0.19	0.80	0.19	0.80	0.19
0.90	0.09	0.90	0.09	0.90	0.09

- The results for these calculations are as shown.

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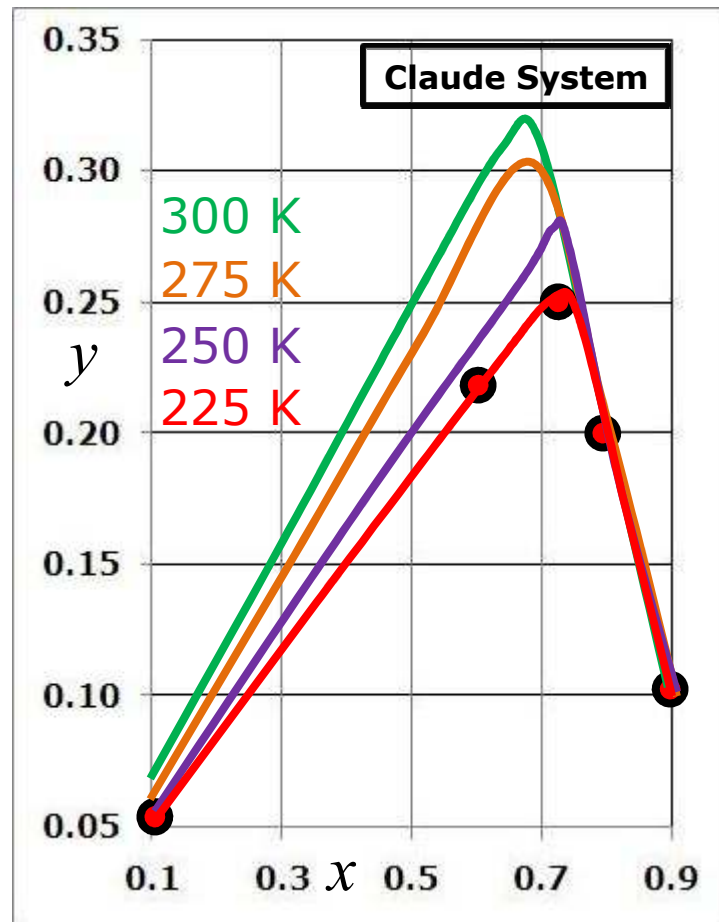
- All the calculations pertaining to part **B** are left as an exercise.

- The results for these calculations are as shown.

300		275		250	
x	W/m ^f	x	W/m _f	x	W/m _f
0.10	4671.8	0.10	4955.4	0.10	5505.3
0.20	2598.0	0.20	2800.1	0.20	3204.2
0.30	1722.4	0.30	1876.3	0.30	2188.4
0.40	1239.3	0.40	1363.1	0.40	1616.1
0.50	933.1	0.50	1036.6	0.50	1248.9
0.60	721.7	0.60	810.5	0.60	993.4
0.67	608.8	0.69	659.3	0.70	805.2
0.70	656.9	0.70	693.1	0.72	773.4
0.80	900.0	0.80	963.2	0.80	1068.4
0.90	1683.3	0.90	1833.3	0.90	2083.3

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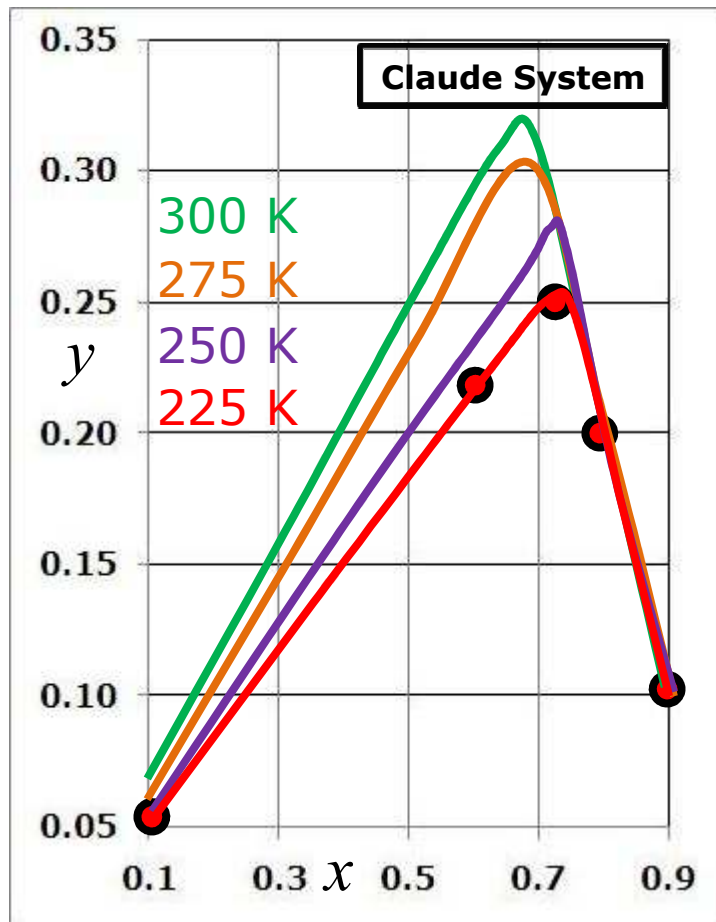
- **Liquid yield v/s. x** • The plot for y versus x for all other values of T_3 is as shown.



- As mentioned earlier, y crosses a maxima with the increase in the value of x for each of the T_3 .
- Also, the position of this maxima shifts to the right with the decrease in the value of T_3 .

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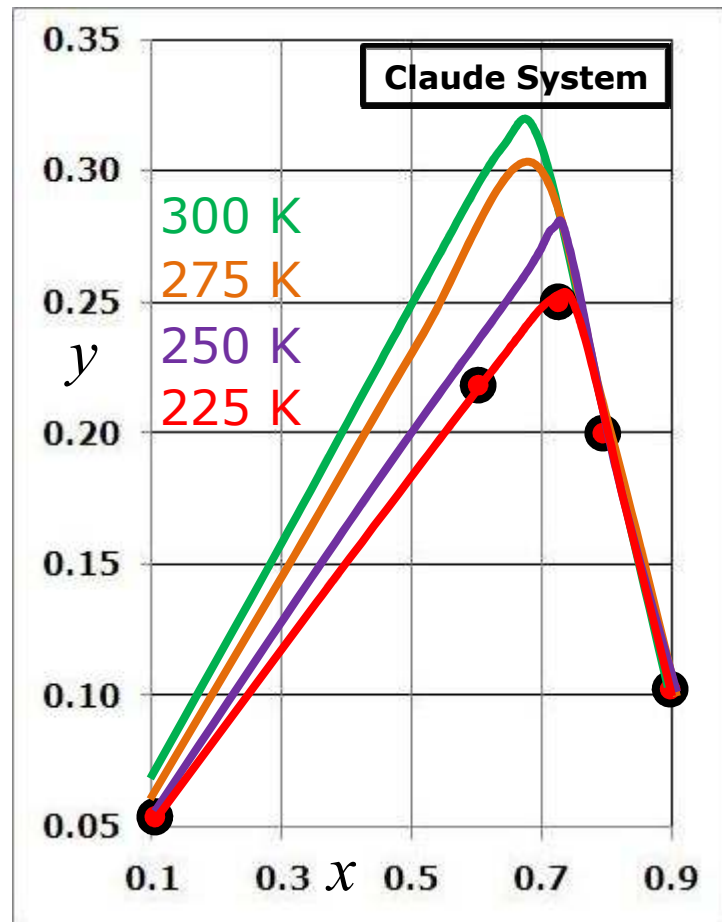
- **Liquid yield v/s. x**



- This occurs because the expander work ($h_3 - h_e$) decreases with the decrease in T_3 .
- Also at the lower values of T_3 , more amount of the gas can be diverted to the expander engine.
- This is because the product $x(h_3 - h_e)$ is maximized.

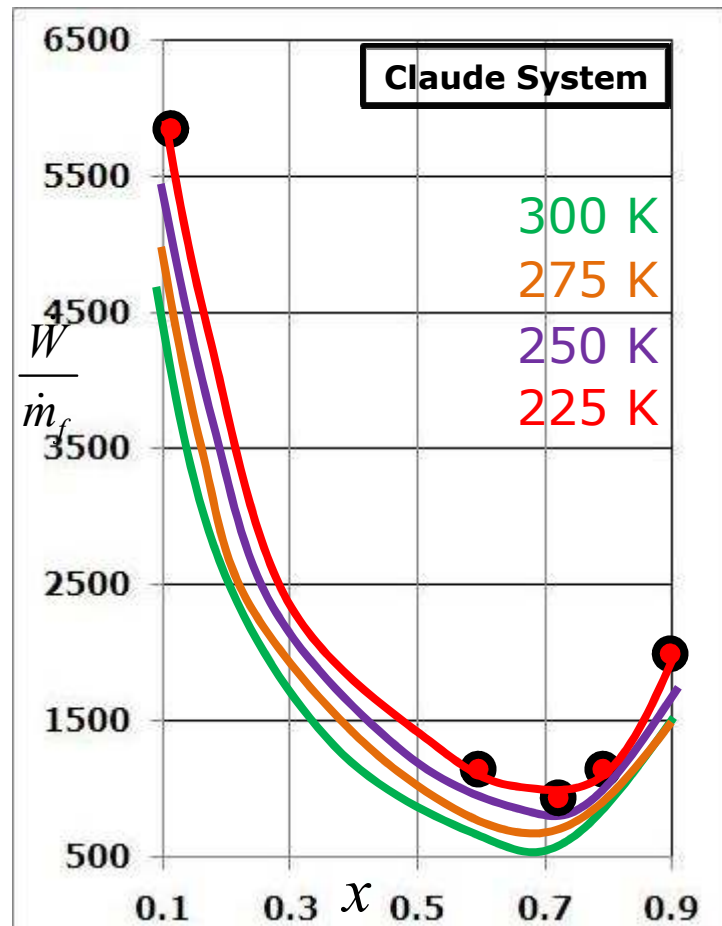
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- **Liquid yield v/s. x** • However, T_3 is limited by the position of the point **e** on the T – s diagram.



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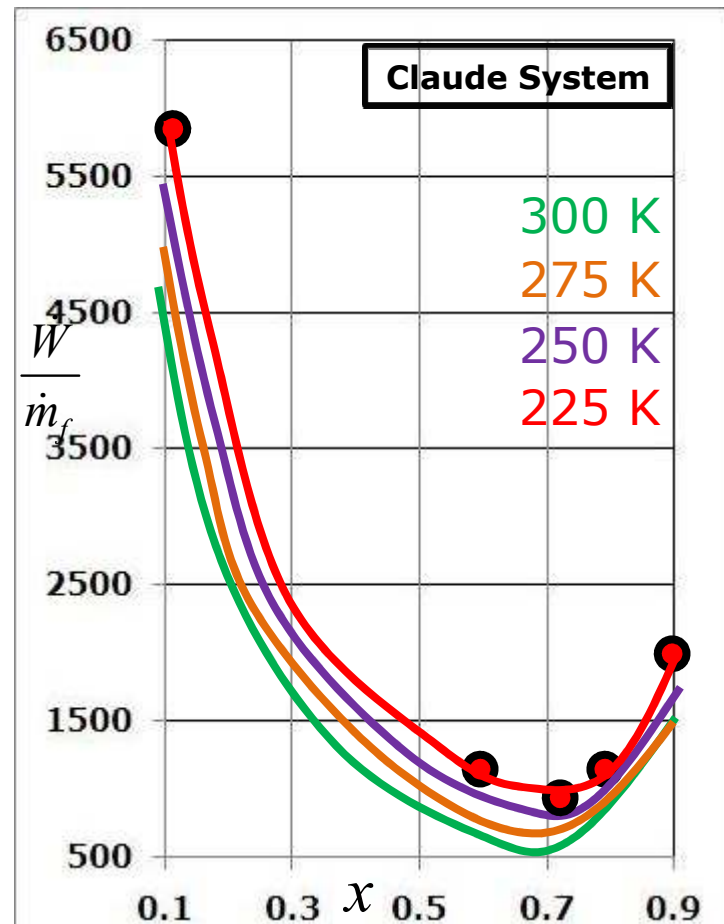
- W/m_f v/s. x



- The plot for W/m_f versus x for all other values of T_3 is as shown.
- As stated earlier, the W/m_f crosses a minima with the increase in the x , for each of the T_3 .
- Also, the position of this minima shifts to the right with the decrease in the value of T_3 .

Tutorial

- W/m_f v/s. x



- The minima shifts to the right because the expander work ($h_3 - h_e$) decreases with the decrease in T_3 .
- Also, at the lower values of T_3 , more amount of the gas can be diverted so that the product $x(h_3 - h_e)$ is maximized.