

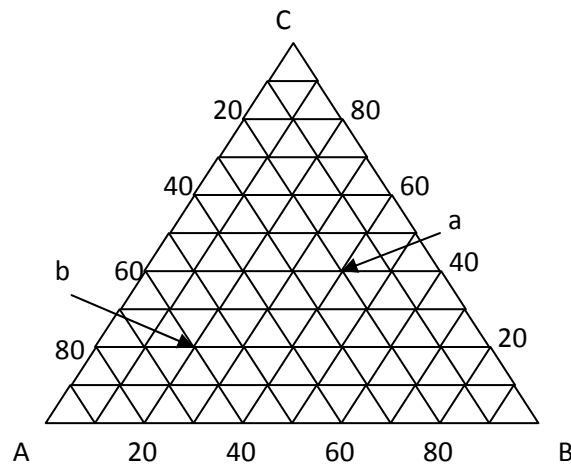
Lecture 25 Ternary diagram

Question:

1. Sketch a composition triangle to represent composition of a ternary alloy and locate the alloys having following compositions (a) 20A 40B 40C (b) 60A20B20C
2. In a ternary system estimate amount of α & β in an alloy having 22.5%A, 23.5%B, 50%C at a given temperature T where three phases α , β and liquid are in equilibrium. Composition of these three phases are liquid: 15A 15B70C, α : 60A20B20C, β :20A40B20C
3. Find out the composition of the alloy which has 50% α & 50% β at temperature T for system described in problem 2.
4. A ternary system consisting of three metals A, B & C has two binary eutectics; one between A & B and the other between B & C and one binary isomorphous system between A & C. The system has two terminal solid solutions (α and β). Assume that the temperature of binary eutectic between A & B is higher than that of B & C. (a) Sketch a space model of this system indicating single phase, two phase and three phase regions. (b) Sketch an isothermal section at a temperature where there are single phase, two phase and three phase regions. (c) Sketch a vertical section passing through B and the midpoint between A & C of the composition triangle.

Answer:

1.



2. Isothermal section at temperature T is given in the following figure. Alloy composition is represented by point x. This lies within a triangle whose vertices denote the composition of α , β and liquid. Join the points α & β with x and extend these lines to meet the opposite side at points m & n. Measure the lengths αn , xn , βm & xm .

